Supporting Information

PTSA-Catalyzed Green Synthesis of 1,3,5-Triarylbenzene under Solvent-Free Conditions

Yanan Zhao, ^a Jian Li, ^a Chunju Li, ^a Kun Yin, ^a Dongyan Ye ^a and Xueshun Jia*, ^{a, b}

^a Department of Chemistry, Shanghai University, 99 Shangda Road, Shanghai, 200444, P. R. China

^b Key Laboratory of Synthetic Chemistry of Natural Substances, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, P. R. China Fax: +86-21-66132408;E-mail: xsjia@mail.shu.edu.cn.

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3. Copies of product ¹H NMR and ¹³C NMRS7-S28

1. General experimental details

All chemicals were purchased from commercial vendors and used without further purification, unless indicated otherwise. EI-MS were determined with an Agilent 5975N mass spectrometer. Flash column chromatography was performed using silica gel (300–400 mesh). Analytical thinlayer chromatography was performed using glass plates precoated with 200–400 mesh silica gel impregnated with a fluorescent indicator (254 nm). Melting points were obtained on a Fisher-Johns apparatus without correction. All NMR spectra (¹H-NMR, ¹³C-NMR) were recorded on a Bruker 500 MHz NMR spectrometer, using CDCl₃ or DMSO-*d6* as the solvent with tetramethylsilane (TMS) as the internal standard at room temperature. Chemical shifts are given in δ relative to TMS, the coupling constants *J* are given in Hz. All reactions were conducted under air atmosphere.

1.1 General procedure:

To a 15 mL two necked round bottom bottle, 360.45 mg (3 mmol) of acetophenone **1a** and 57.07 mg (0.30 mmol) of TsOH.H₂O was added. The reaction mixture was stirred at 130 °C for 10 h,determinated by TLC, After completion of the reaction, the reaction mixture was neutralized by saturated NaHCO₃, extracted with DCM (3 x 20 mL) or (3 x 50 mL). The combined organic layers were dried by MgSO₄, filtered, and concentrated in vacuo. The residue was purified by flash column chromatography (PE/EA = 20:1) and gave 278.82 mg (0.91 mmol, 91 %) of **2a**, as white solid.

2. Experimental characterization data for compounds



Mp:171-173 °C (Lit ¹ 170-172 °C). ¹H NMR (DMSO-*d6*, 500 MHz): δ 7.88 (s, 3H), 7.87-7.86 (m, 6H), 7.52-7.49 (t, *J* =7.5 Hz, 6H), 7.43-7.40 (t, *J* =7.5 Hz, 3H).¹³C NMR (DMSO-*d6*, 125 MHz): δ 141.6, 140.1, 128.9, 127.7, 127.2, 124.4.



Mp:176-178 °C (Lit ² 177-178 °C). ¹H NMR (CDCl₃, 500 MHz): δ 7.74 (s, 3H), 7.60 (d, *J* = 8.0 Hz, 6H), 7.29 (d, *J* = 7.5 Hz, 6H), 2.43 (s, 9H). ¹³C NMR (CDCl₃, 125 MHz):142.3, 138.5, 137.4, 129.6, 127.3, 124.7, 21.2. MS (EI) *m/z* (M⁺) 348.



Mp:164-166 °C (Lit ³ 161-165 °C). ¹H NMR (CDCl₃, 500 MHz): δ 7.59 (s, 3H), 7.74 (d, *J* = 8.5 Hz, 6H), 6.78 (d, *J* = 8.0 Hz, 6H), 3.76 (d, *J* = 10.5 Hz, 6H). ¹³C NMR (CDCl₃, 125 MHz):146.0, 142.1, 132.0, 128.3, 123.0, 115.5. MS (EI) *m/z* (M⁺) 351.



Mp:141-143 °C (Lit ⁴ 140-142 °C). ¹H NMR (CDCl₃, 500 MHz): δ 7.57 (d, *J* = 9.0 Hz, 6H), 7.32 (s, 3H), 6.99 (d, *J* = 8.5 Hz, 6H), 3.86 (s, 9H). ¹³C NMR (CDCl₃, 125 MHz): 159.2, 141.4, 134.0, 128.3, 122.7, 114.2, 55.4.



Mp:238-239 °C (Lit ¹ 238-240 °C). ¹H NMR (CDCl₃, 500 MHz): δ 7.69 (s, 3H), 7.60 (d, J = 8.5 Hz, 6H), 7.53 (d, J = 8.5 Hz, 6H). ¹³C NMR (CDCl₃, 125 MHz): 161.8, 137.1, 132.2, 129.0 (d, J = 7.5 Hz), 125.0, 115.9 (d, J = 22.5 Hz).



Mp:245-247 °C (Lit ⁵ 246 °C). ¹H NMR (CDCl₃, 500 MHz): δ 7.69 (s, 3H), 7.60 (d, *J* = 8.5 Hz, 6H), 7.45 (d, *J* = 8.5 Hz, 6H). ¹³C NMR (CDCl₃, 125 MHz):141.5, 139.3, 134.0, 129.2, 128.7, 125.1. MS (EI) *m/z* (M⁺) 408.



Mp:261-262 oC (Lit 1 260-261 °C). ¹H NMR (CDCl₃, 500 MHz): δ 7.68 (s, 3H), 7.60 (d, J = 8.5 Hz, 6H), 7.53 (d, J = 8.5 Hz, 6H). ¹³C NMR (CDCl₃, 125 MHz): 141.6, 139.7, 132.2, 129.0, 125.1, 122.2.



Mp:264-265 °C (Lit ⁶ 264.6-265.9 °C). ¹H NMR (CDCl₃, 500 MHz): δ 7.80 (d, J = 8.5 Hz, 6H), 7.67 (s, 3H), 7.39 (d, J = 8.5 Hz, 6H).



Mp:135-137 °C (Lit ⁶ 135.6-135.8 °C). ¹H NMR (CDCl₃, 500 MHz): δ 7.43-7.41 (m, 3H), 7.38 (s, 3H), 7.37-7.32 (m, 9H), 2.47 (s, 9H). ¹³C NMR (CDCl₃, 125 MHz):141.8, 141.5, 135.5, 130.5, 130.0, 128.6, 127.4, 125.9, 20.8.



Mp:117-118 °C (Lit ⁶ 116.8-118.1 °C). ¹H NMR (CDCl3, 500 MHz): ^δ 7.77 (s, 3H), 7.52-7.51 (m, 6H), 7.386 (m, 3H), 7.23 (m, 3H), 2.46 (s, 9H). ¹³C NMR (CDCl3, 125 MHz):142.4, 141.3, 138.5, 128.8, 128.3, 128.3, 125.2, 124.6, 21.7.



Mp:169-170 °C (Lit ¹ 168-170 °C). ¹H NMR (CDCl₃, 500 MHz): δ 7.72 (s, 3H), 7.67-7.66 (m, 3H), 7.57-7.55(m, 3H), 7.44-7.37 (m, 6H). ¹³C NMR (CDCl₃, 125 MHz):142.5, 142.5, 141.4, 134.9, 130.3, 127.9, 127.5, 125.6.MS (EI) *m/z* (M⁺) 408.



Mp:164-166 °C (Lit ⁶ 165.1-165.5 °C). ¹H NMR (CDCl₃, 500 MHz): δ 7.58 (s, 3H), 7.50-7.46 (m, 6H), 7.35-7.28(m, 6H). ¹³C NMR (CDCl₃, 125 MHz):140.0, 140.0, 132.7, 131.7, 130.2, 129.9, 128.8, 127.0. MS (EI) *m/z* (M⁺) 408.

3. Copies of product ¹H NMR and ¹³C NMR

















Supplementary Material (ESI) for Green Chemistry This journal is (c) The Royal Society of Chemistry 2010









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Reference:

1. Kumar, A.; Dixit, M.; Singh, S. P.; Goel, A.; Raghunandan, R.; Maulik, P. R. *Tetrahedron Lett.* **2009**, *50*, 4335.

2. Kothe, G.; Zimmermann, H. Tetrahedron 1973, 29, 2305.

3. Ishi-i, T.; Kuwahara, R.; Takata, A.; Jeong, Y.; Sakurai, K.; Mataka, S. *Chem. Eur. J.* **2006**, *12*, 763.

4. Karger, M. H.; Mazur. Y, J. Org. Chem. 1971, 36, 540.

5. Jing, X; Xu, F.; Zhu, Q.; Ren, X.; Yan, Ch.; Wang, L.; Wang, J. Syn. Commun. 2005, 35, 3167.

6. Ono, F.; Ishikura, Y.; Tada, Y.; Endo, M.; Sato, T. Synlett 2008, 15, 2365.