

Enhanced photoluminescence stability of CdS nanocrystals through a zinc acetate reagent†

M. Banski,^a M. Chrzanowski,^a G. Zatoryb,^a J. Misiewicz,^a A. Podhorodecki,^a

^a Department of Experimental Physics, Wrocław University of Science and Technology, Wybrzeże Wyspiańskiego 27, 50-370 Wrocław, Poland.

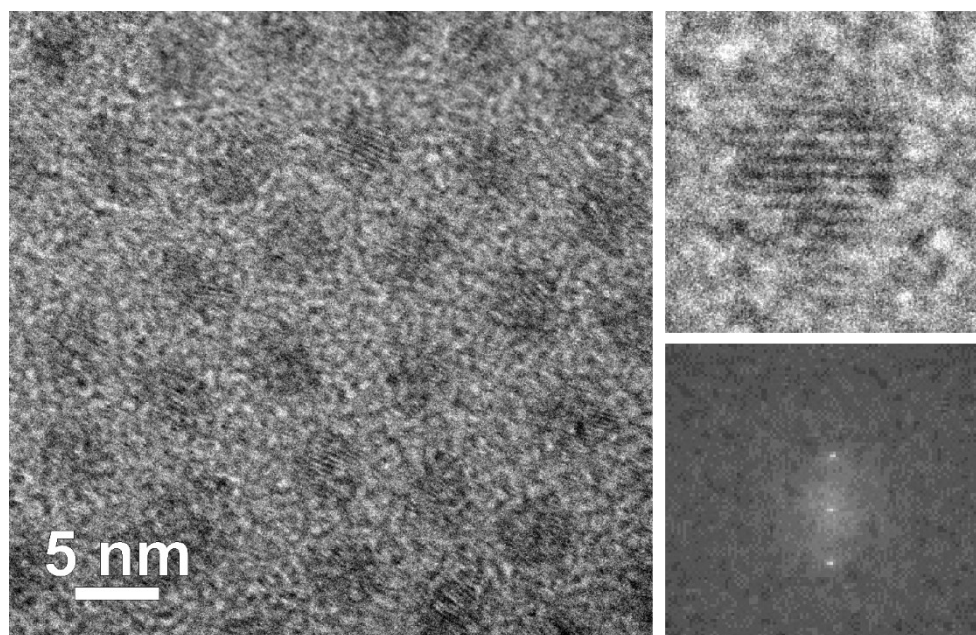


Fig. S1. HR-TEM images and electron diffraction of CdS:Zn NCs (0.2 mmol of Zn precursor).



Fig. S2. EDX of CdS:Zn synthesized with 0.2 mmol of Zn(OAc)₂.

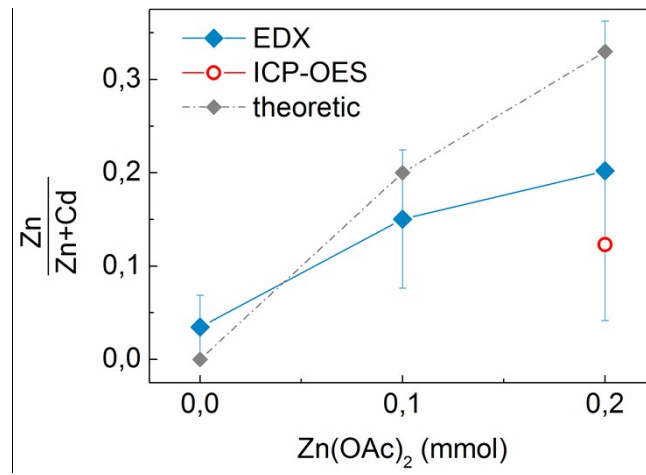


Fig. S3. Molar ratio of zinc to total cations of CdS:Zn nanocrystals determined from EDX spectra and ICP-OES in a function of $Zn(OAc)_2$ dose.

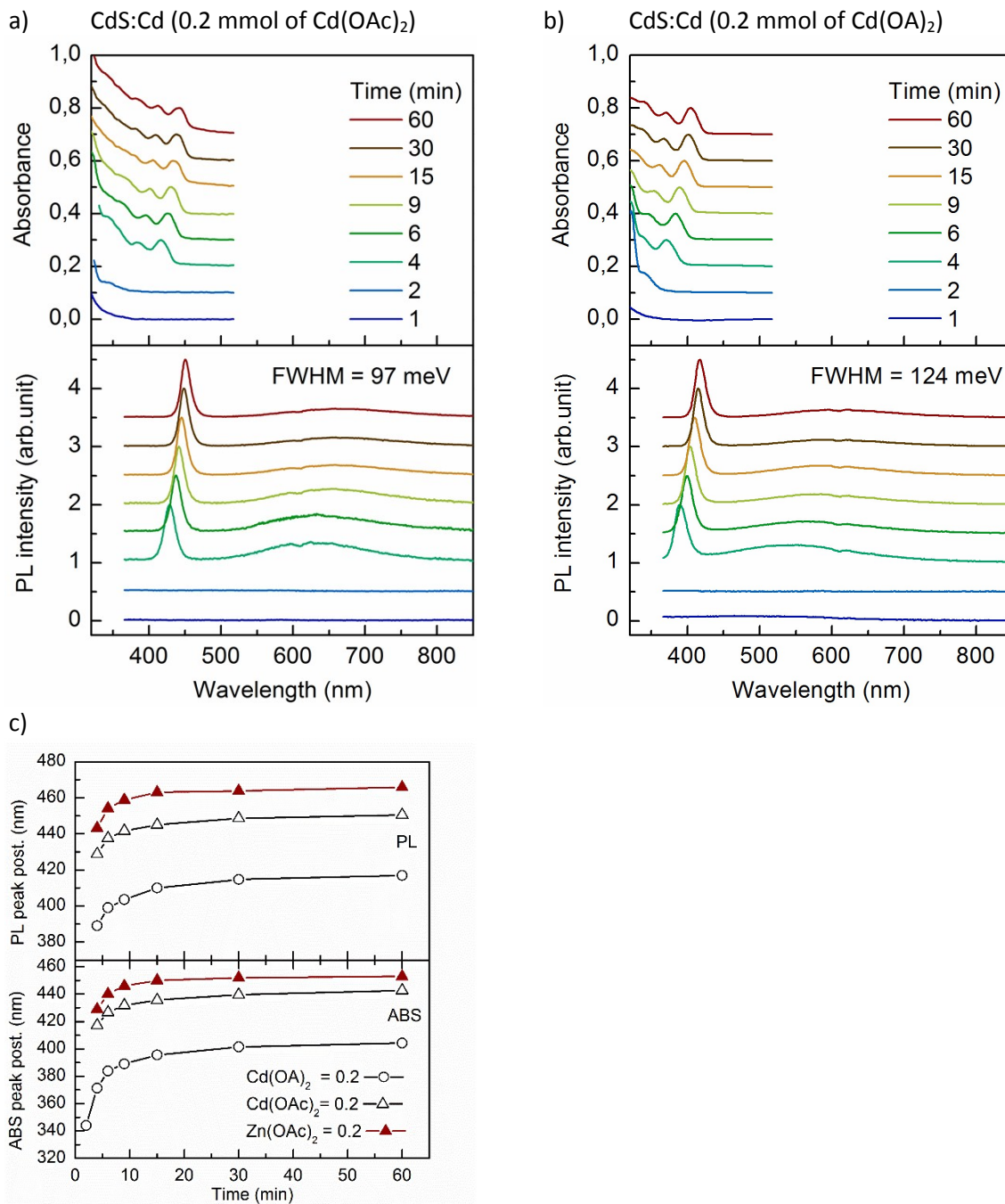


Fig. S4. ABS and PL spectra of CdS NCs synthesized with (a) 0.2 mmol Cd(OAc)₂ and (b) 0.2 mmol Cd(OA)₂. (c) Positions of the first exciton transition of above samples determined from ABS and PL spectra in a function of synthesis time compared to results of CdS NCs with 0.2 mmol Zn(OAc)₂ addition.

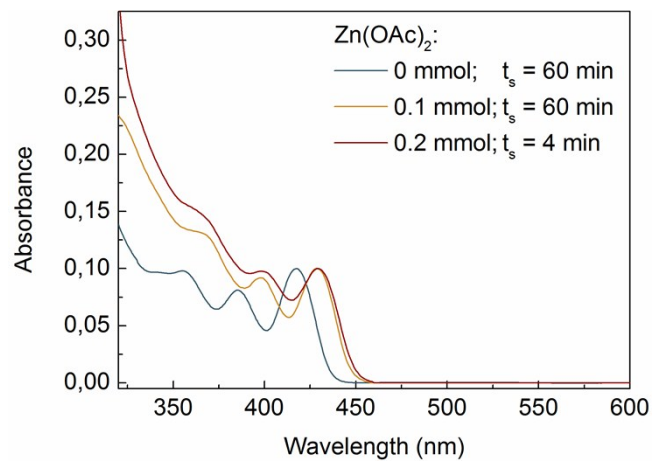


Fig. S5. ABS spectra of selected NCs samples having similar size. A decrease of ABS bands separation due to increase of Zn(OAc)₂ concentration is visible.