The crystallographic phase transition for a ferric

thiosemicarbazone spin crossover complex studied by X-Ray

powder diffraction.

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Supplementary Data

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Figure S1 : IR spectra of compound Li[Fe(5BrThsa)₂].H₂O in Heating mode from 170 to 380 K (a) and in cooling mode from 380 to 120 K (b) (From Ref. S1).

Dotted lines are only guides for eyes.

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Figure S2 : Distortion of $[FeN_2O_2S_2]$ core in HS (a) and LS (b) phases.



Figure S3 : Structureless whole pattern profile refinements for the LS phase (a) and the HS phase (b).

Supplementary Material (ESI) for New Journal of Chemistry This journal is (c) The Royal Society of Chemistry and The Centre National de la Recherche Scientifique, 2006 **Table S1.** Selected angles of LS and HS ferric thiosemicarbazone complexes. Labels are given in agreement with those used for

Li[Fe(5BrThsa)₂].H₂O.

K[Fe(3,5Cl ₂ Thsa) ₂] ,1.5H ₂ O Ref. S5 Site Fe2	103 K	Almost LS	87.1(5)	92.4(4)	90.3(6)	90.5(6)	93.1(5)	90.8(6)	90.7(6)	88.0(2)	83.7(5)	95.5(5)	94.9(5)	83.6(5)	174.0(5)	174.2(4)	178.3(6)
K[Fe(3,5Cl ₂ Thsa) ₂] ,1.5H ₂ O. Ref. S5 Site Fe1	103 K	ΓS	86.6(6)	91.8(7)	91.8(7)	89.6(6)	91.3(5)	89.2(7)	94.6(7)	90.3(2)	83.9(6)	94.9(5)	91.4(4)	84.9(5)	175.2(5)	178.4(4)	176.1(8)
NH4[Fe(5CIThsa) ₂] Ref. S4	135 K	LS	87.3	89.6	88.9	93.4	86.6	93.4	88.9	93.5	86.0	91.7	91.7	86.0	176.8	176.8	176.8
NH4[Fe(5BrThsa) ₂] Ref. S3	$300\mathrm{K}$	LS	86.4(3)	·	89.7(3)	93.3(3)	89.9(2)	ı	ı	93.8(1)	85.8(2)	91.4(2)	ı	ı	176.2(2)	ı	175.9(3)
Cs[Fe(Thsa) ₂] Ref. S2	298 K	SH	84.3(5)	90.9(4)	86.8(5)	109.8(5)	89.4(4)	108.0(5)	85.6(5)	100.3(2)	78.4(4)	87.2(4)	89.9(4)	78.9(4)	161.3(4)	161.2(4)	159.9(5)
Compounds	T/K	Spin State	O(1)FeO(2)	O(1)FeS(2)	O(1)FeN(11)	O(1)FeN(21)	O(2)FeS(1)	O(2)FeN(11)	O(2)FeN(21)	S(1)FeS(2)	S(1)FeN(11)	S(1)FeN(21)	S(2)FeN(11)	S(2)FeN(21)	O(1)FeS(1)	O(2)FeS(2)	N(11)FeN(21)

Table S2. Cell parameters reported in literature for ferric substituted salicylaldehyde-thiosemicarbazone complexes Supplementary Material (ESI) for New Journal of Chemistry This journal is (c) The Royal Society of Chemistry and The Centre National de la Recherche Scientifique, 2006

NH4[Fe(3,5Cl ₂ Thsa) ₂] ,1.5H ₂ O Ref S6	298 K	Monoclinic	P21/a	20.273(7)	27.437(9)	8.852(5)	98.80(8)	4866
NH4[Fe(3,5Cl ₂ Thsa) ₂] ,1.5H ₂ O Ref S6	103 K	Monoclinic	P2 ₁ /a	20.203(7)	27.117(9)	8.705(4)	99.05(9)	4710
K[Fe(3,5Cl ₂ Thsa) ₂] ,1.5H ₂ O Ref S5	298 K	Monoclinic	$P2_1/a$	20.221(8)	27.210(9)	8.916(5)	98.08	4857
K[Fe(3,5Cl ₂ Thsa) ₂] ,1.5H ₂ O Ref S5	103 K	Monoclinic	$P2_1/a$	20.090(7)	26.996(10)	8.865(4)	98.37	4666
NH4[Fe(5CIThsa)2] Ref. S4	298 K	Orthorhombic	Pbcn	20.348(9)	11.791(5)	8.548(4)	ı	2050.9
NH4[Fe(5ClThsa)2] Ref. S4	135 K	Orthorhombic	Pbcn	20.186(9)	11.729(8)	8.490(10)	I	2010.1
NH4[Fe(5BrThsa)2] Mica-Like crystals Ref. S3	300 K	Orthorhombic	Pnca	20.837(8)	11.761(7)	8.619(6)	ı	2112.2
NH4[Fe(5BrThsa) ₂] Tabular crystals Ref. S3	300 K	Orthorhombic	Pnca	21.179(2)	11.755(7)	8.560(5)	,	2131.1
Cs[Fe(Thsa) ₂] Ref S2	298 K	Orthorhombic	Pna21	15.285(3)	13.402(4)	9.449(8)	ı	1936
Cs[Fe(Thsa) ₂] Ref S2	103 K	Orthorhombic	Pna21	15.161(3)	13.340(3)	9.394(7)	I	1900
Compounds	T/K	System	Space	a (Å)	b (Å)	c (Å)	β(°)	V (Å ³)

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