

Hemoglobin immobilized within mesoporous TiO₂-SiO₂ material with high loading and enhanced catalytic activity

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Supplementary Information

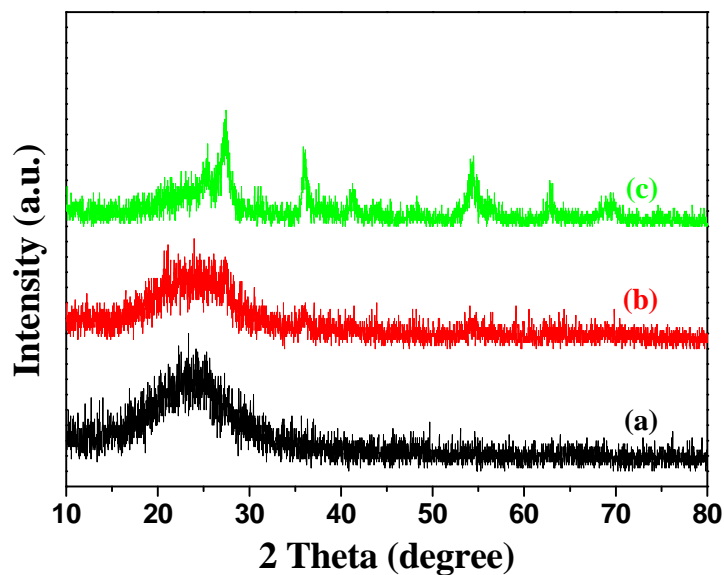


Fig.S1 Wide angle X-ray diffraction patterns of (a) 20TiO₂-SiO₂, (b) 40TiO₂-SiO₂ and (c) 60TiO₂-SiO₂

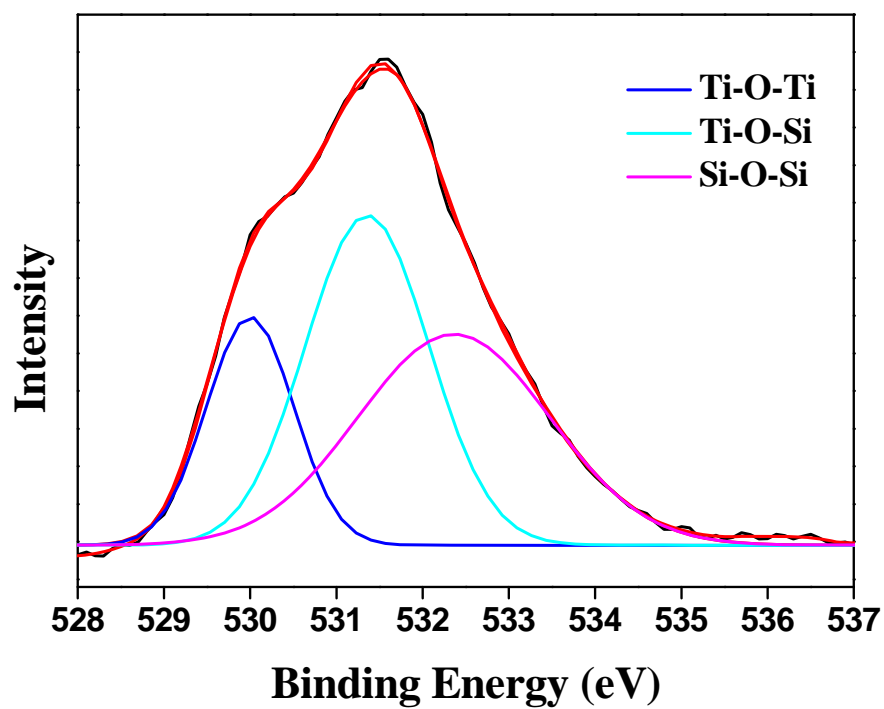


Fig.S2 X-ray photoelectron spectrum (XPS) of 20TiO₂-SiO₂

Table. S1 The Zeta potential of SiO_2 and $x\text{TiO}_2\text{-SiO}_2$ suspensions in pH
5.4 PBS

Sample	Zeta potential (mV)
10SiO_2	-17.04
$20\text{TiO}_2\text{-SiO}_2$	-18.76
$40\text{TiO}_2\text{-SiO}_2$	-20.57
$60\text{TiO}_2\text{-SiO}_2$	-28.02

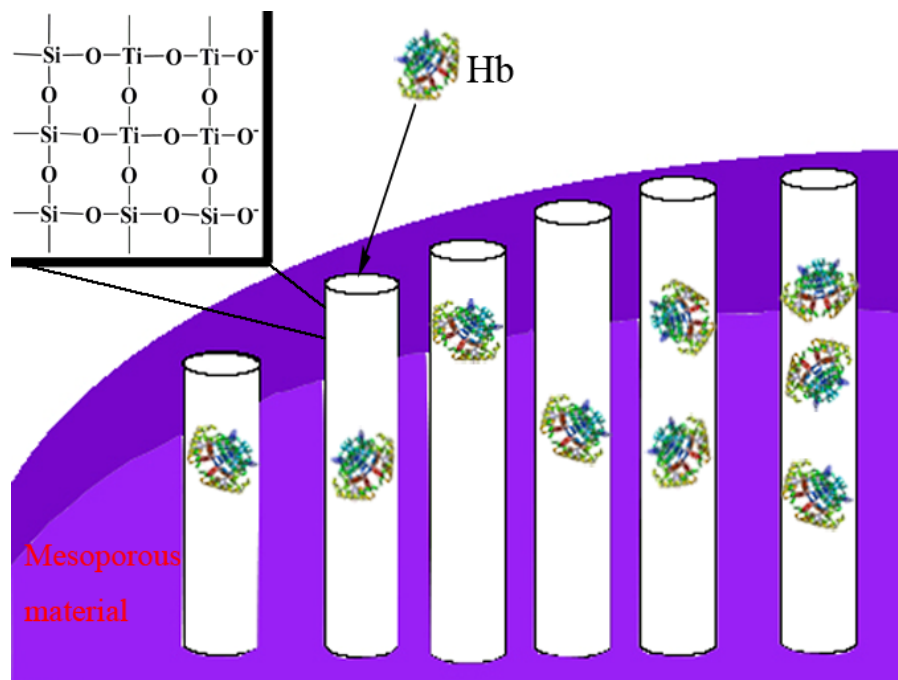


Fig.S3 Schematic illustration of encapsulation of Hb from pH 5.4 PBS into the pore of $\text{TiO}_2\text{-SiO}_2$. The wall of $\text{TiO}_2\text{-SiO}_2$ mesoporous material consists of Ti and Si which connect with each other by Ti-O-Si bond. As the pore size is larger than the Hb dimension, Hb can be adsorbed within the pore smoothly.

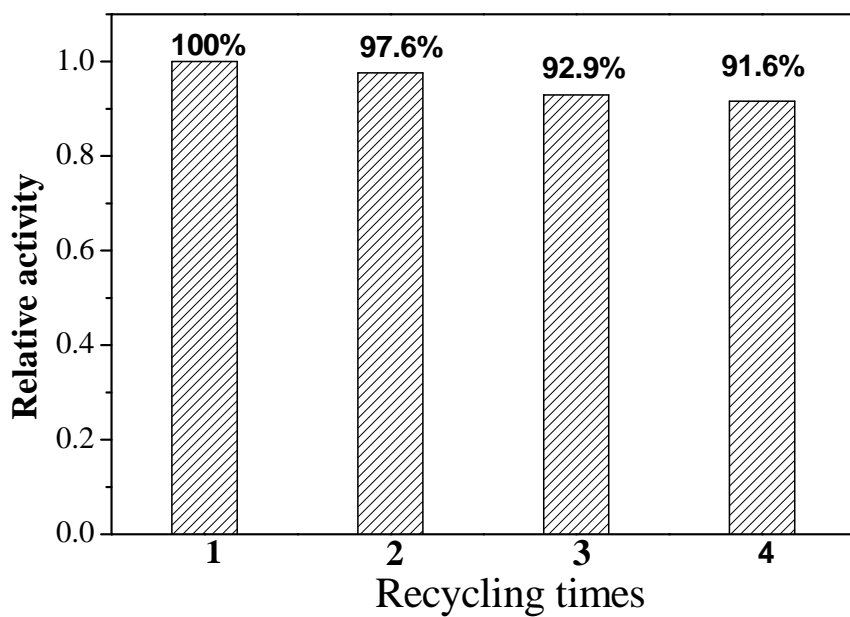


Fig.S4 Relative activity of 40% TiO₂-SiO₂-Hb in four times usage; the reaction possesses 0.0025M BPO and 0.001M OPD catalyzed by 0.01g

40% TiO₂-SiO₂-Hb in CHCl₃.