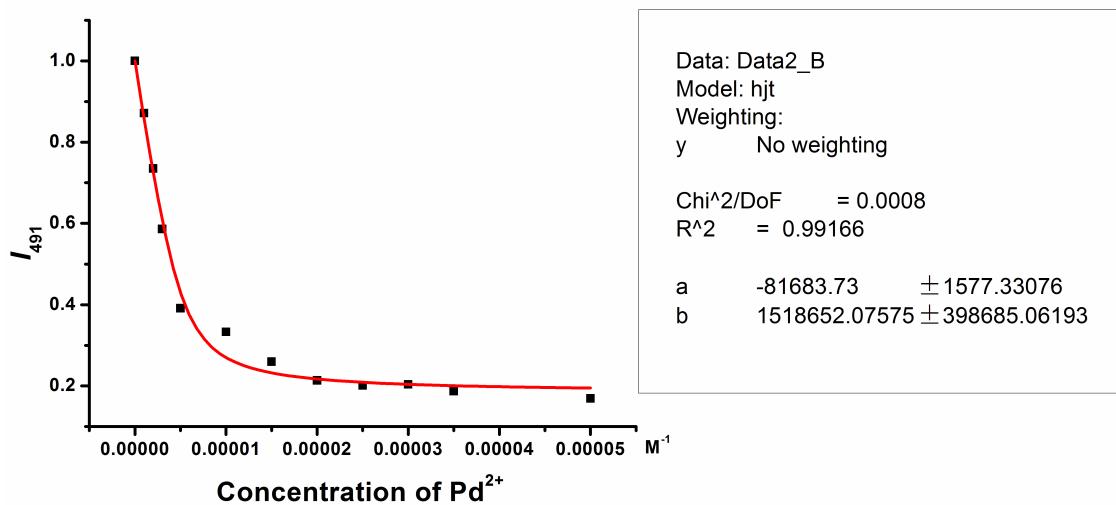


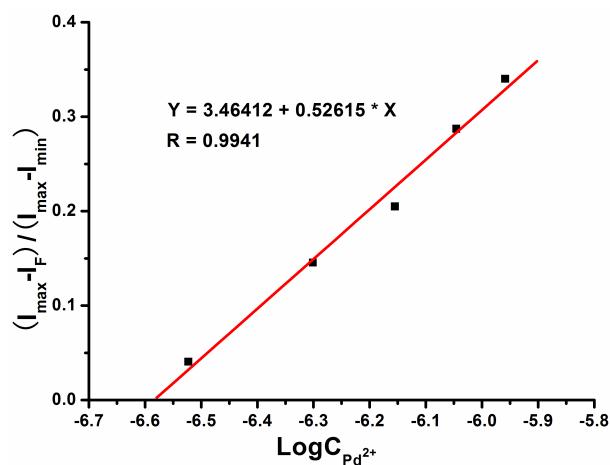
# Novel Triazole-Based Fluorescent Probe for Pd<sup>2+</sup> in Aqueous Solutions: Design, Theoretical Calculations and Imaging

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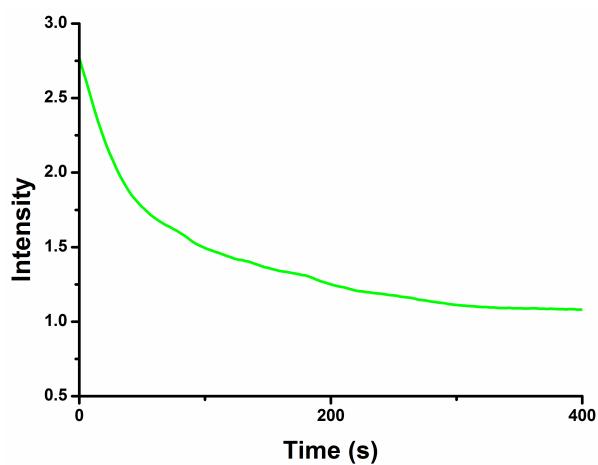
Figure S1	2
Figure S2	2
Figure S3	3
Figure S4	3
Figure S5	4
Figure S6	4
Figure S7	5
Figure S8	5
Figure S9	6
Figure S10	6
Figure S11	7
Figure S12	7
Figure S13	8
Table S1	8
Table S2	8
NMR and HRMS copies of various compounds	9-23



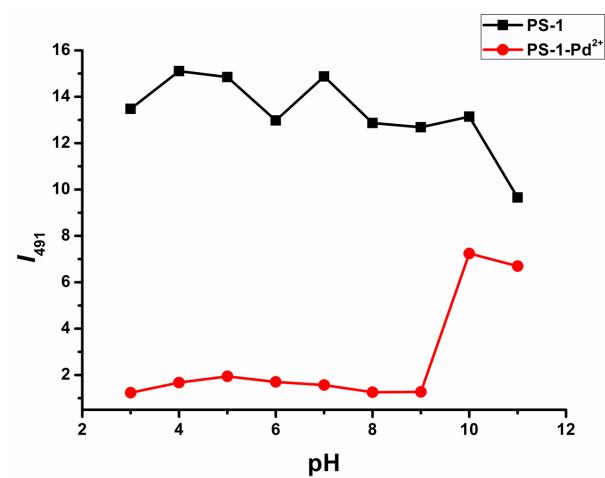
**Figure S1** Nonlinear least square analysis of **PS-1** and  $Pd^{2+}$ .



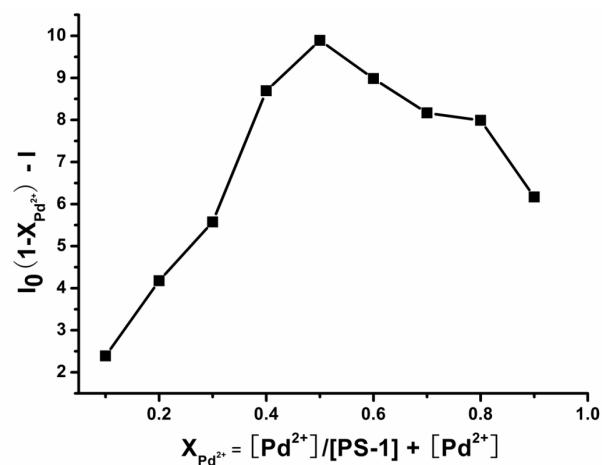
**Figure S2** Normalized response of fluorescence signal to changing  $Pd^{2+}$  concentrations in the PBS buffer (10 mM, pH = 7.2, containing 0.5% DMF). ( $\lambda_{ex} = 410$  nm,  $\lambda_{em} = 491$  nm ).



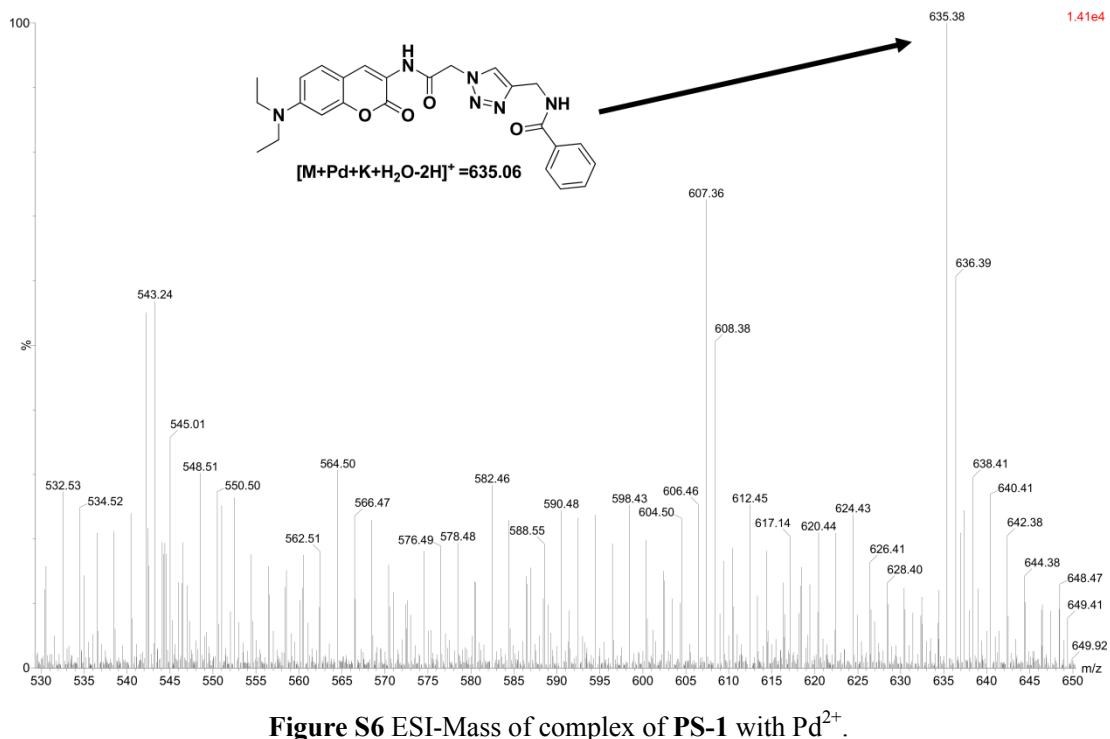
**Figure S3** Time-dependent fluorescence intensity changes at 491 nm of **PS-1** (5  $\mu\text{M}$ ) with  $\text{Pd}^{2+}$  (5  $\mu\text{M}$ )



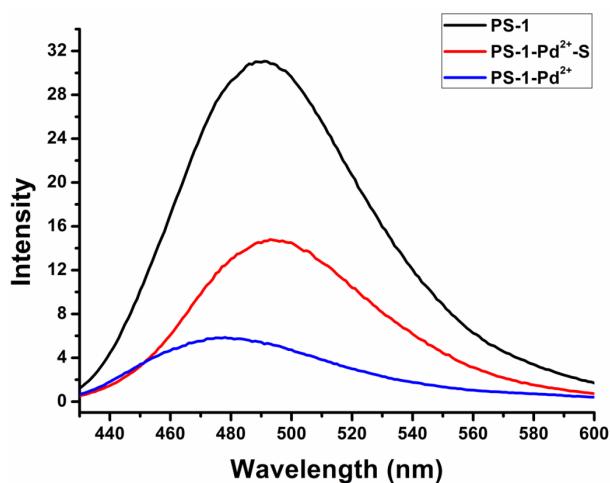
**Figure S4** pH titration of **PS-1** (5  $\mu\text{M}$ ) with or without  $\text{Pd}^{2+}$  (50  $\mu\text{M}$ ) during pH 3.0-11.0.



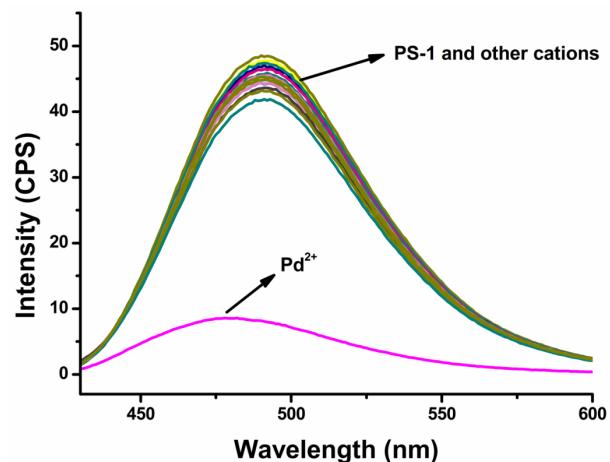
**Figure S5** The job plot of a 1:1 complex of **PS-1** with  $\text{Pd}^{2+}$ .



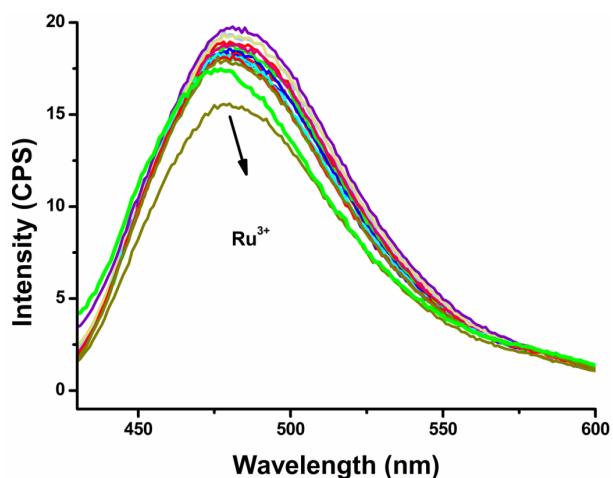
**Figure S6** ESI-Mass of complex of **PS-1** with  $\text{Pd}^{2+}$ .



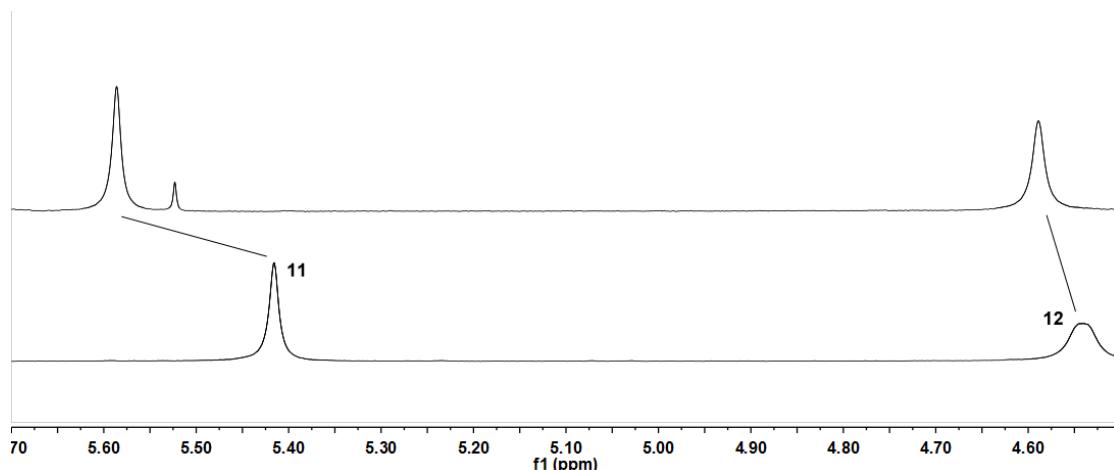
**Figure S7** Fluorescence spectral changes of **PS-1-Pd<sup>2+</sup>** solution upon addition of an excess amount of S<sup>2-</sup> ion.



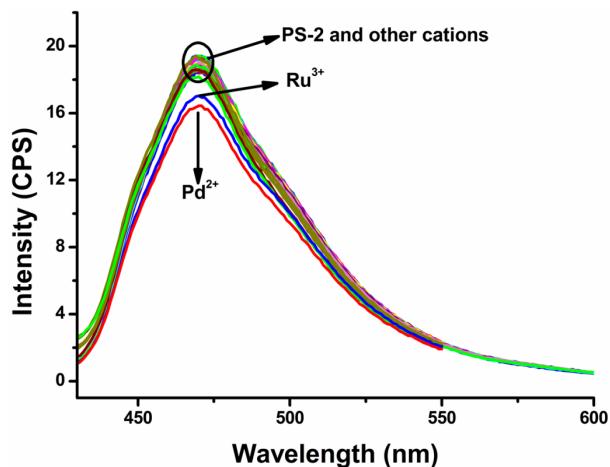
**Figure S8** Fluorescent spectra of compound **PS-1** (5  $\mu$ M) upon addition of 10 equiv metal ions in PBS buffer (10 mM, pH = 7.2, containing 0.5% DMF). ( $\lambda_{\text{ex}} = 410$  nm, slit = 3.0 nm/3.0 nm)



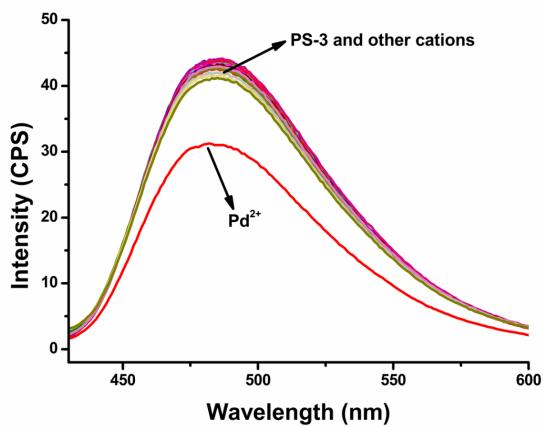
**Figure S9** Fluorescence response of **PS-1** (50  $\mu$ M) to various metal ions (10 equiv) in the presence of Pd<sup>2+</sup> (10 equiv) PBS buffer (10 mM, pH = 7.2, containing 0.5% DMF). ( $\lambda_{\text{ex}} = 410$  nm, slit = 3.0 nm/3.0 nm).



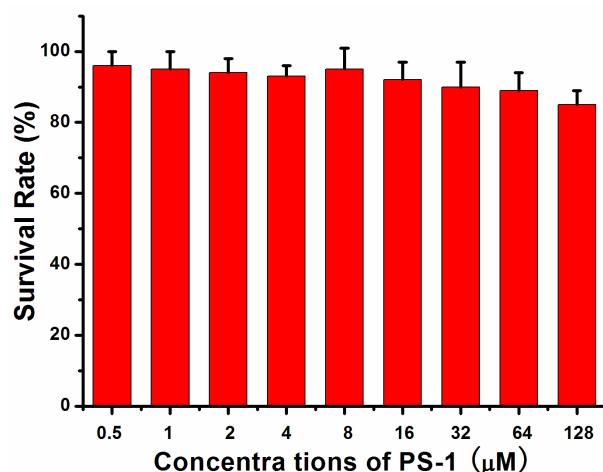
**Figure S10** Partial <sup>1</sup>H NMR spectra of **PS-1** in the absence or presence of 3 equiv Pd<sup>2+</sup> in DMSO-*d*<sub>6</sub> (Alkyl area). (top : PS-1 with Pd<sup>2+</sup>; bottom: PS-1)



**Figure S11** Fluorescent spectra of compound **PS-2** (5  $\mu$ M) upon addition of 10 equiv metal ions in PBS buffer (10 mM, pH = 7.2, containing 0.5% DMF). ( $\lambda_{\text{ex}} = 410$  nm, slit = 3.0 nm/3.0 nm)



**Figure S12** Fluorescent spectra of compound **PS-3** (5  $\mu$ M) upon addition of 10 equiv metal ions in PBS buffer (10 mM, pH = 7.2, containing 0.5% DMF). ( $\lambda_{\text{ex}} = 410$  nm, slit = 3.0 nm/3.0 nm)



**Figure S13** Effects of **PS-1** at varied concentrations on the viability of Hela cells. The results are the mean standard deviation of three separate measurements.

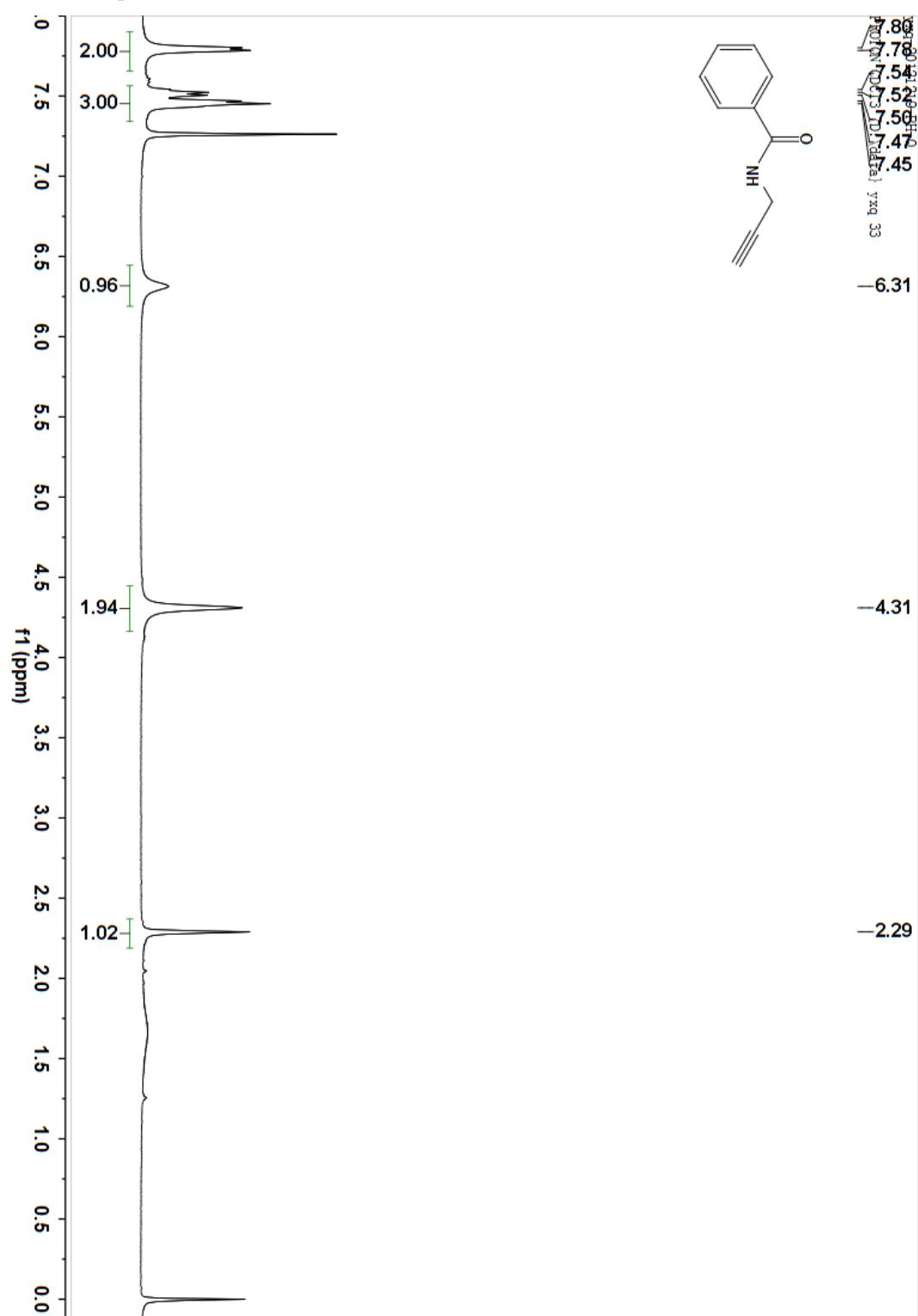
**Table S1** Quantum yields of **PS-1/2/3** with or without  $\text{Pd}^{2+}$ .

Species	$\Phi_F$
<b>PS-1</b>	0.077
<b>PS-1-Pd<sup>2+</sup></b>	0.015
<b>PS-2</b>	0.045
<b>PS-2-Pd<sup>2+</sup></b>	0.039
<b>PS-3</b>	0.021
<b>PS-3-Pd<sup>2+</sup></b>	0.017

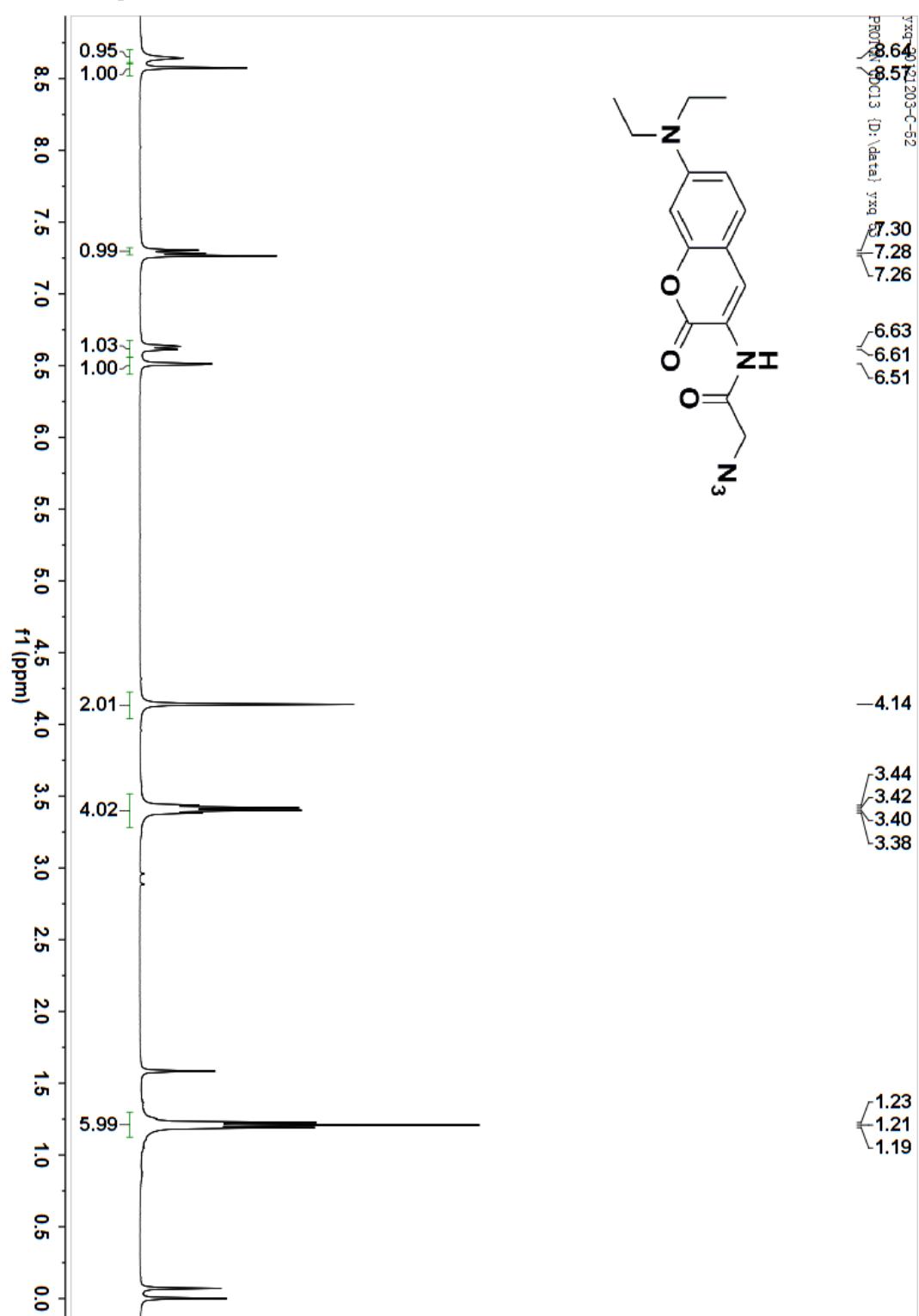
**Table S2** B3LYP optimized bond distances ( $\text{\AA}$ ) of **PS-1/2/3** with  $\text{Pd}^{2+}$ .

Bond	Bond lengths ( $\text{\AA}$ )		
	PS-1-Pd	PS-2-Pd	PS-3-Pd
Pd-N1	<b>2.016</b>	<b>2.011</b>	4.267
Pd-N2	2.871	2.946	2.918
Pd-N3	3.442	<b>2.064</b>	3.525
Pd-N4	3.287	2.965	3.433
Pd-C1	2.534	4.059	2.702
Pd-C2	<b>2.181</b>	4.053	<b>2.23</b>
Pd-O1	4.207	4.066	<b>2.095</b>
Pd-N5	<b>2.095</b>		3.515
Pd-O2	3.166		<b>2.069</b>

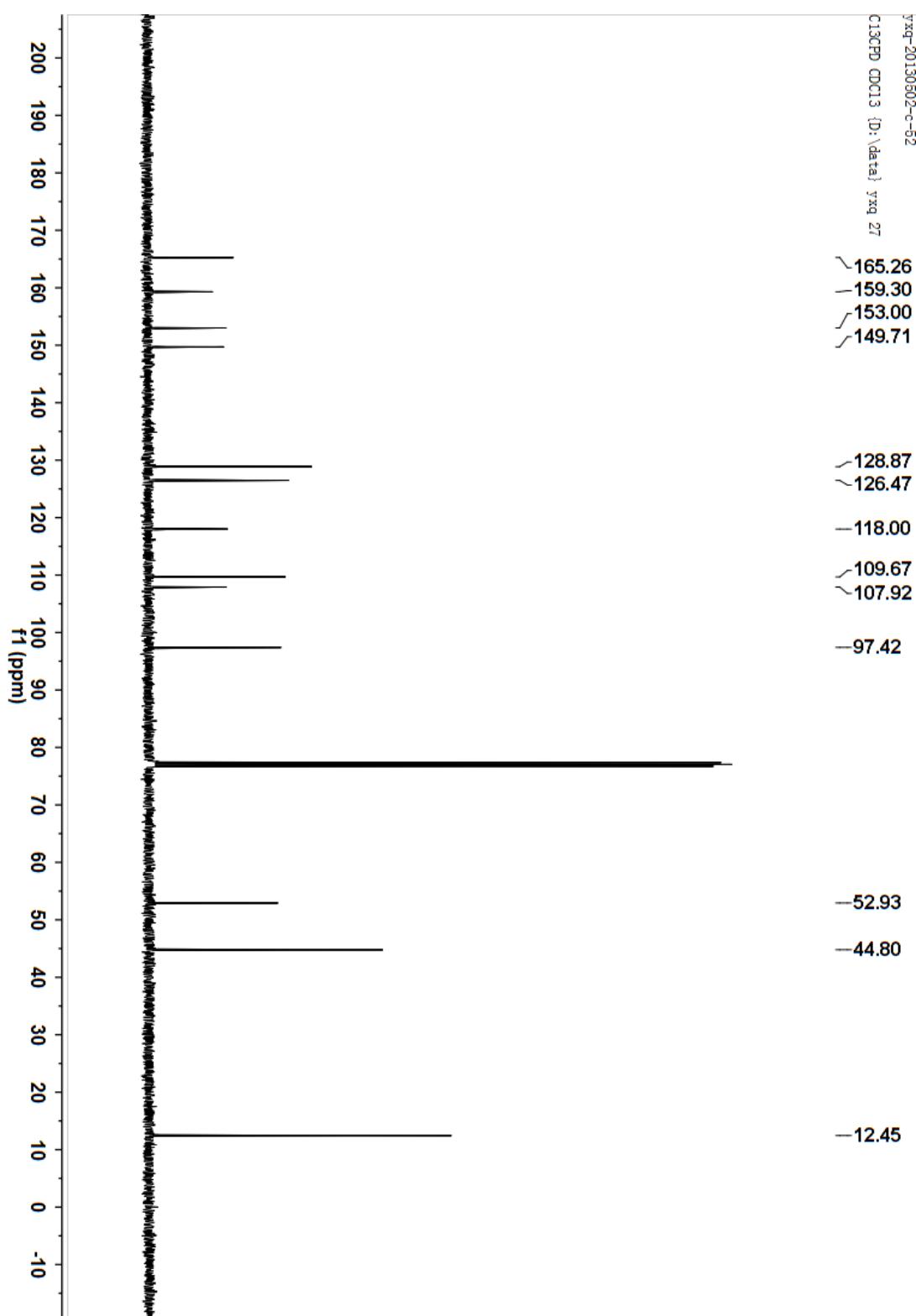
<sup>1</sup>H-NMR Spectrum of A-1 in CDCl<sub>3</sub> (400 MHz):



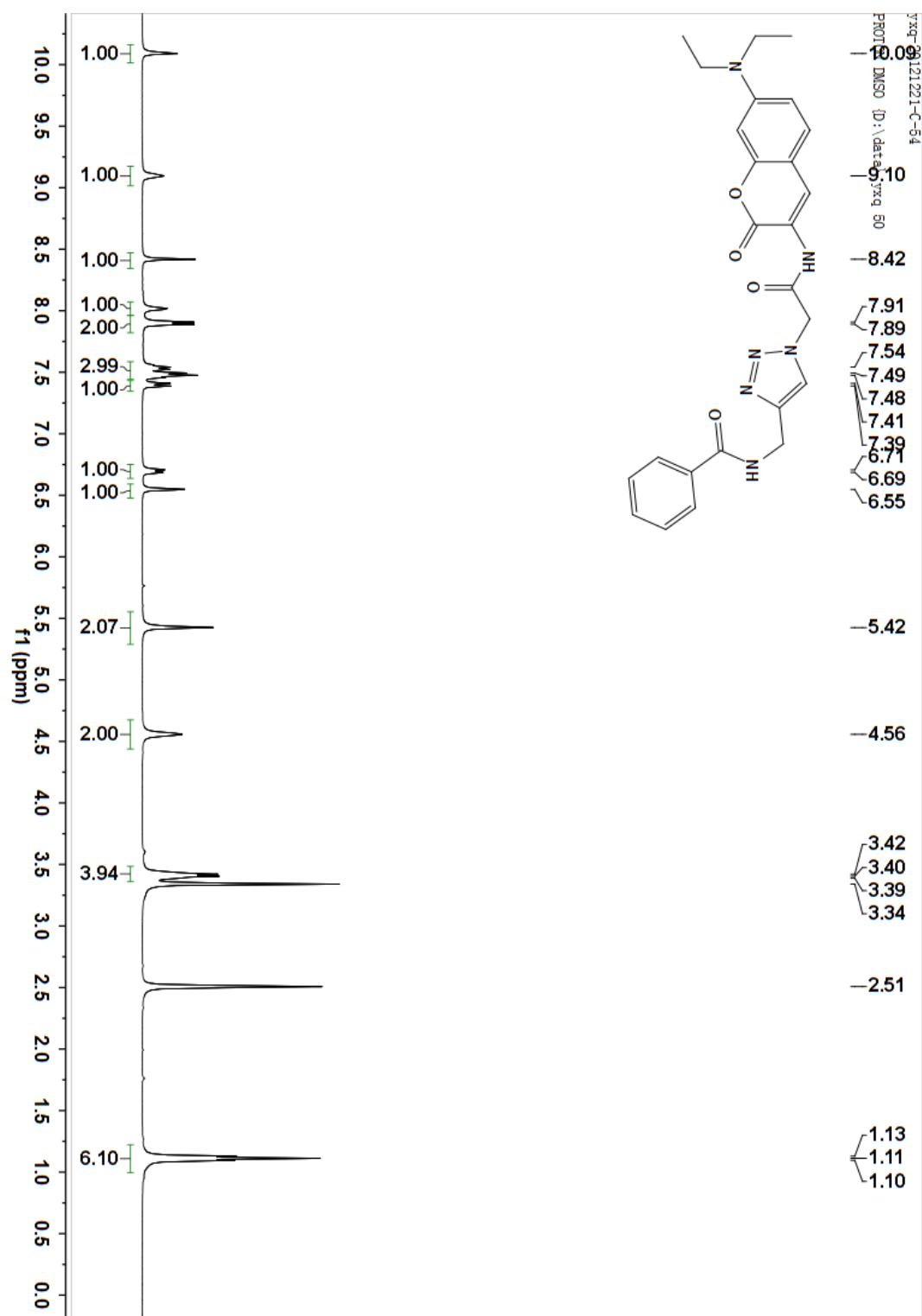
<sup>1</sup>H-NMR Spectrum of C-2 in CDCl<sub>3</sub> (400 MHz):



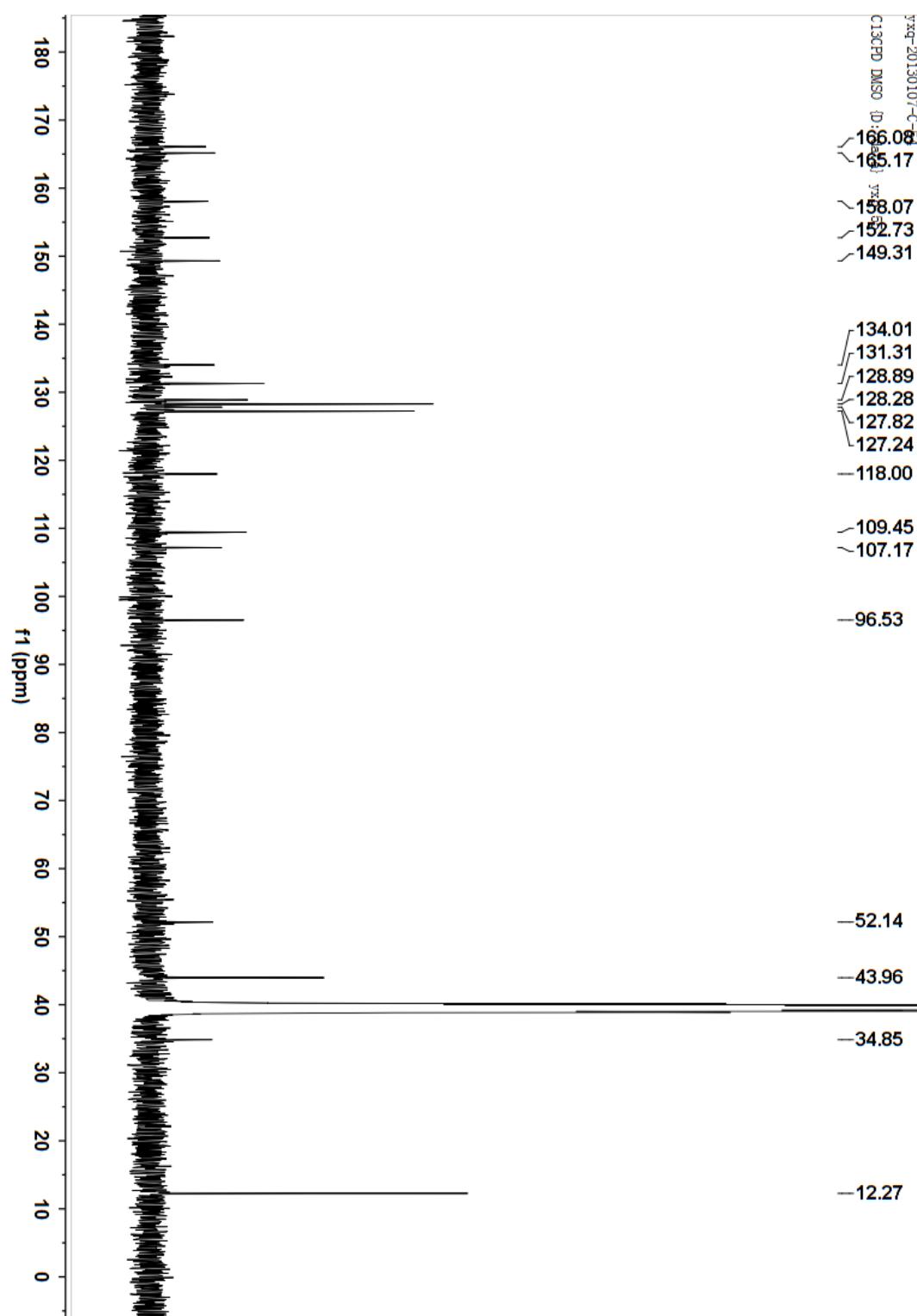
<sup>13</sup>C-NMR Spectrum of C-2 in CDCl<sub>3</sub> (100 MHz):



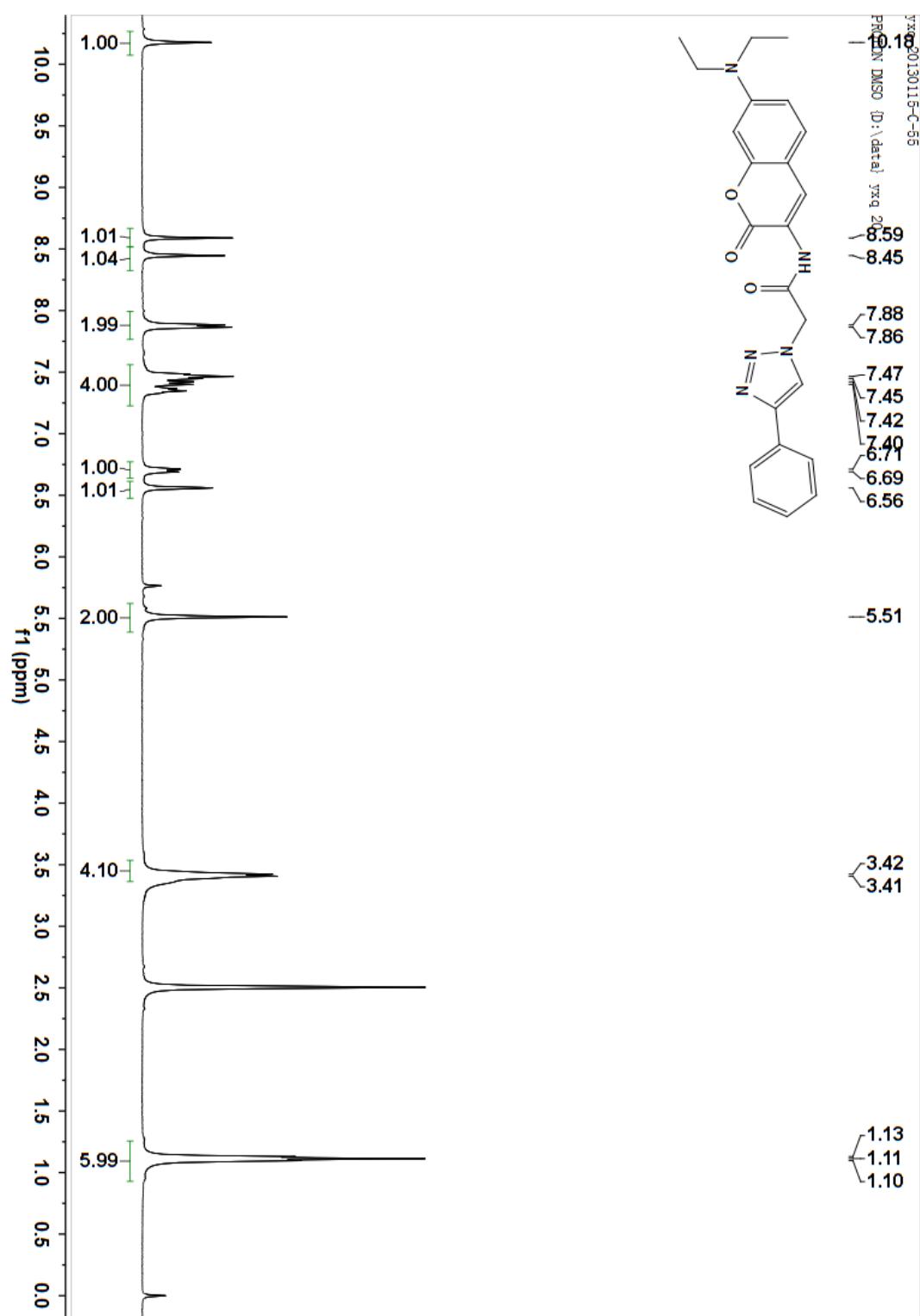
<sup>1</sup>H-NMR Spectrum of **PS-1** in DMSO-*d*<sub>6</sub> (400 MHz):



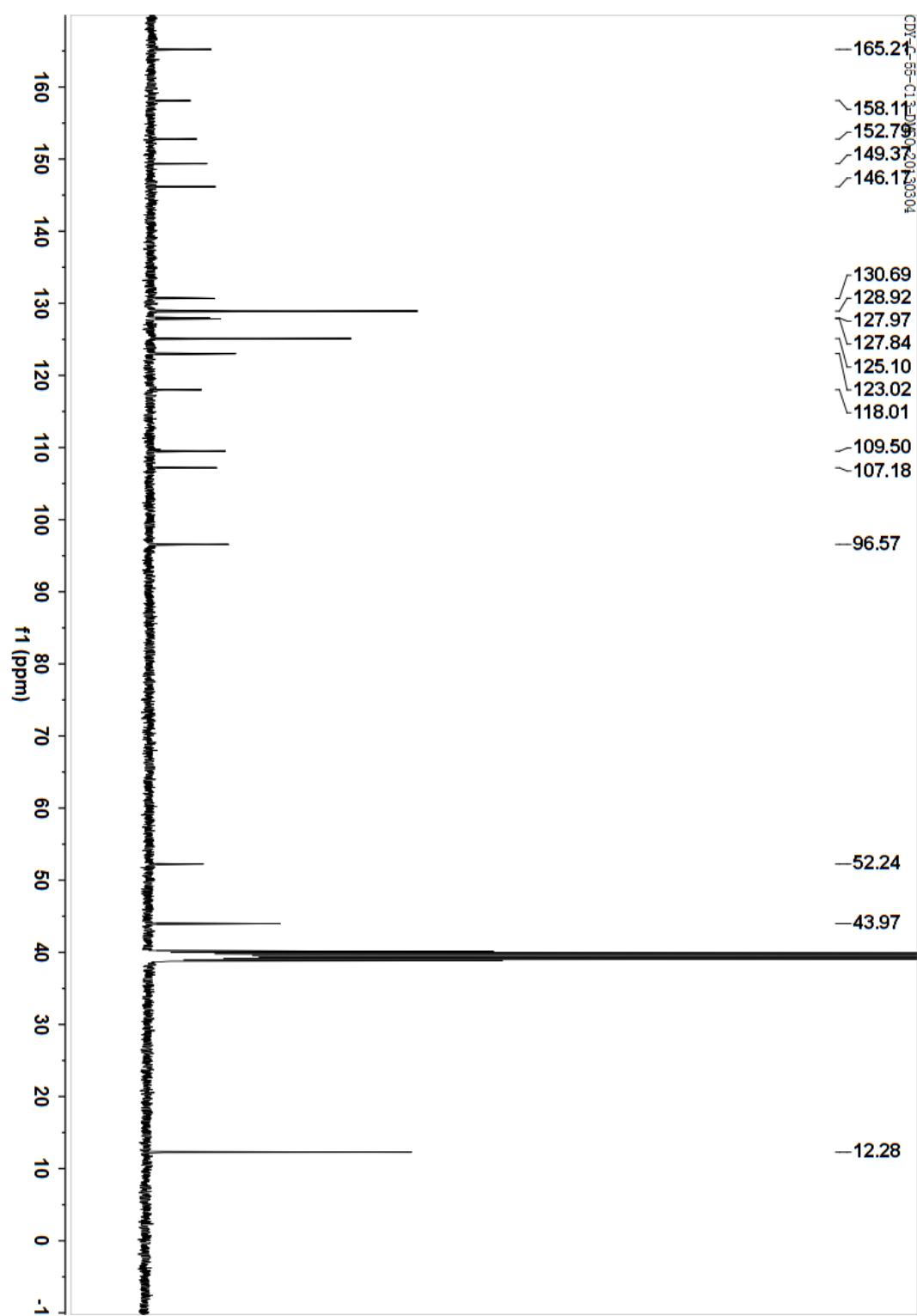
$^{13}\text{C}$ -NMR Spectrum of **PS-1** in  $\text{DMSO}-d_6$  (100 MHz):



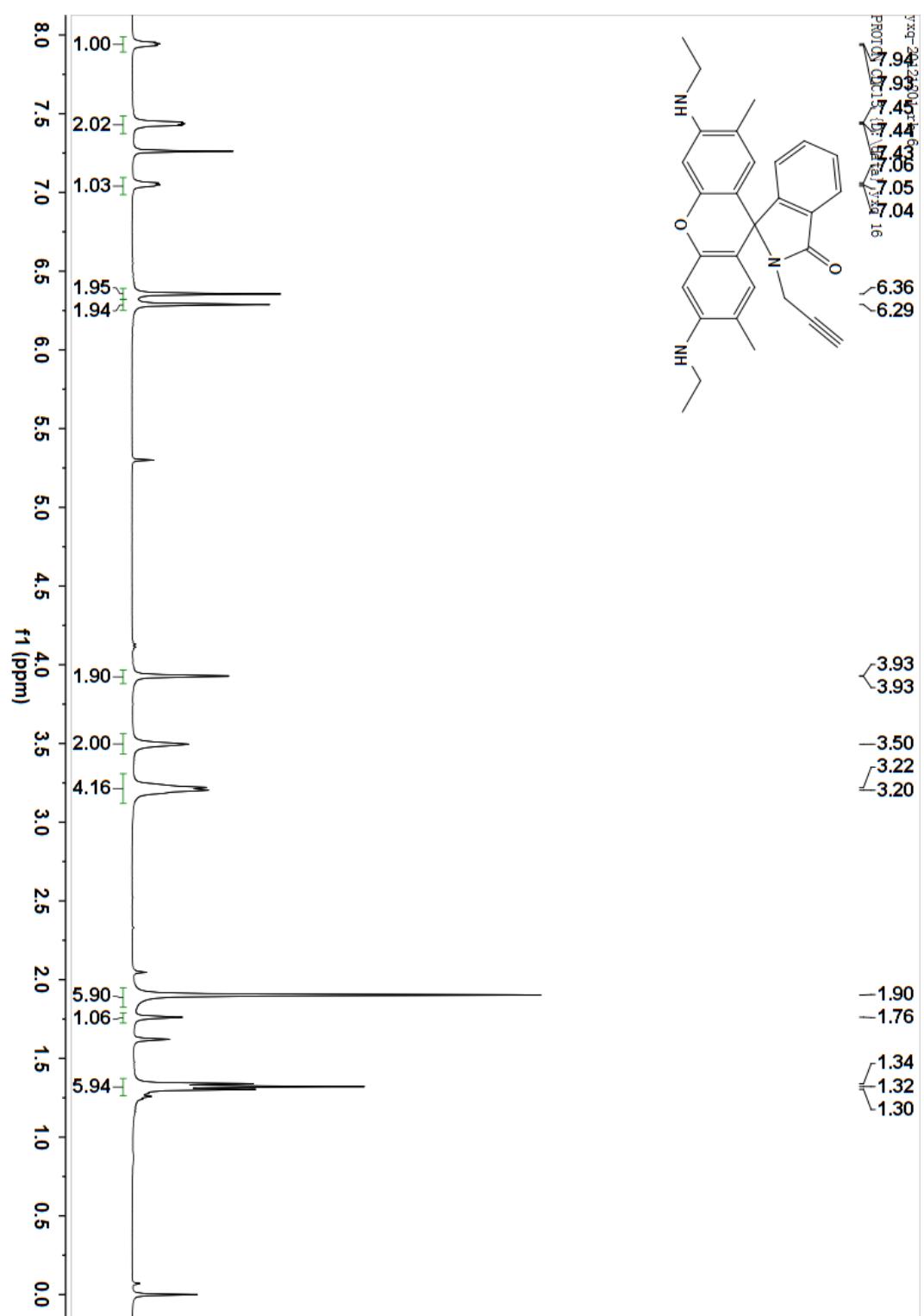
<sup>1</sup>H-NMR Spectrum of **PS-2** in DMSO-*d*<sub>6</sub> (400 MHz):



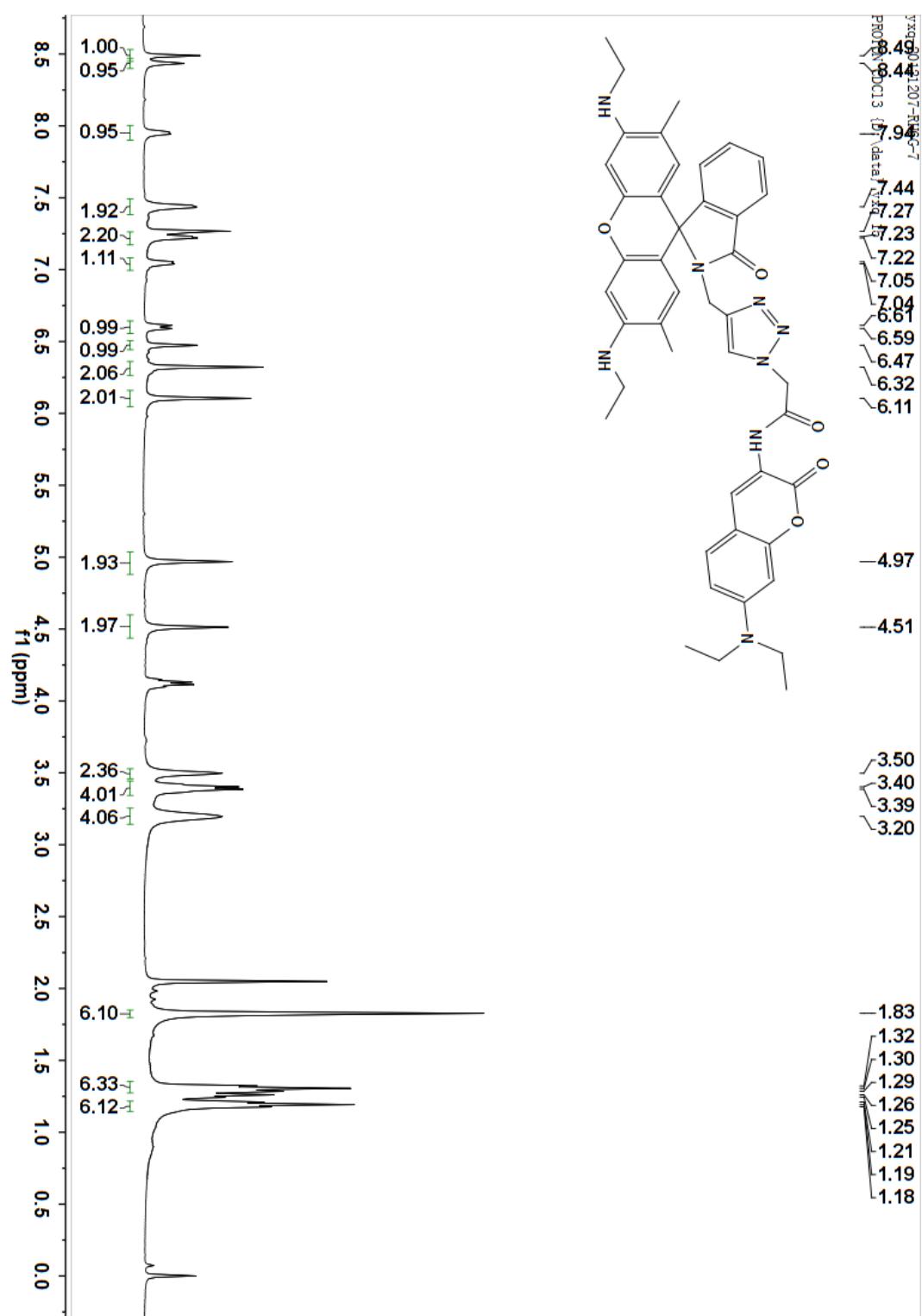
<sup>13</sup>C-NMR Spectrum of **PS-2** in DMSO-d<sub>6</sub> (100 MHz):



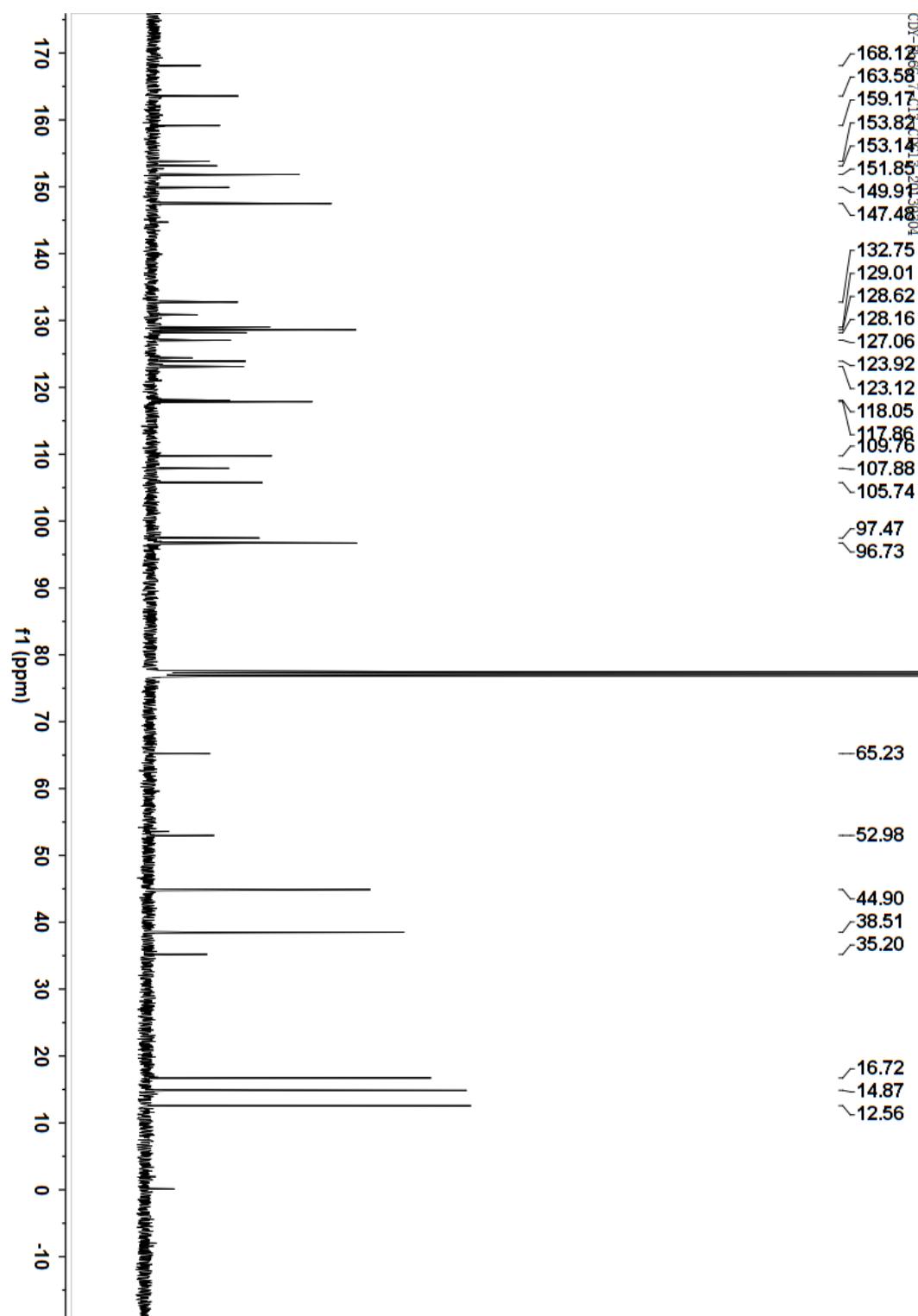
<sup>1</sup>H-NMR Spectrum of A-2 in CDCl<sub>3</sub> (400 MHz):



<sup>1</sup>H-NMR Spectrum of PS-3 in CDCl<sub>3</sub> (400 MHz):



$^{13}\text{C}$ -NMR Spectrum of **PS-3** in  $\text{CDCl}_3$  (100 MHz):

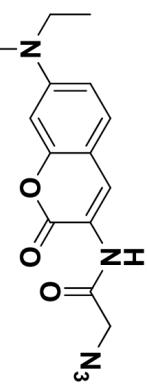


HRMS spectra of C-2

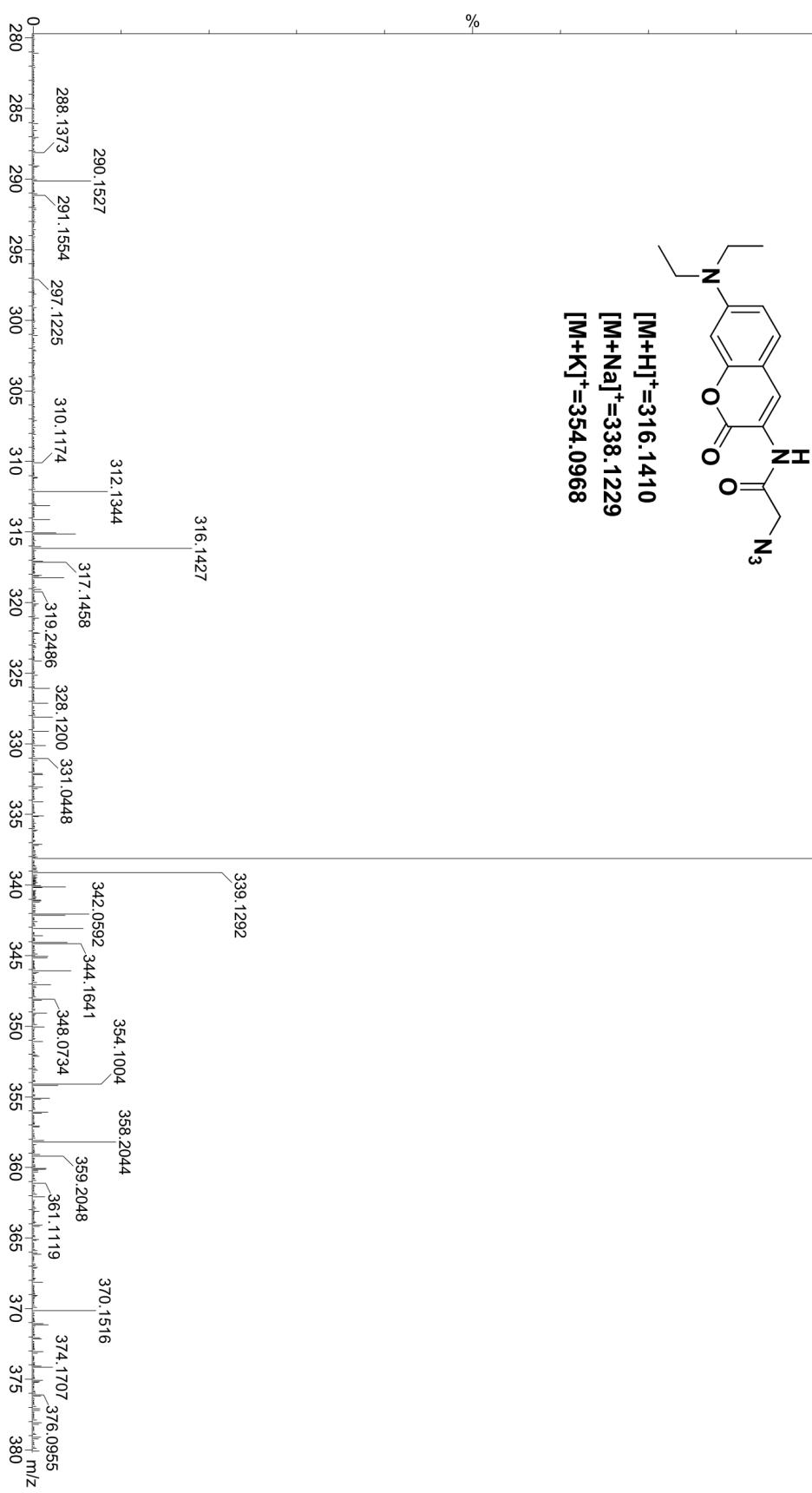
130502\_C\_52 34 (0.581).AM (Cen,4, 80.00, Ar,10000.0.0.0,0.70); Sm (\$G, 2x3.00); Cm (1.36)

338.1228

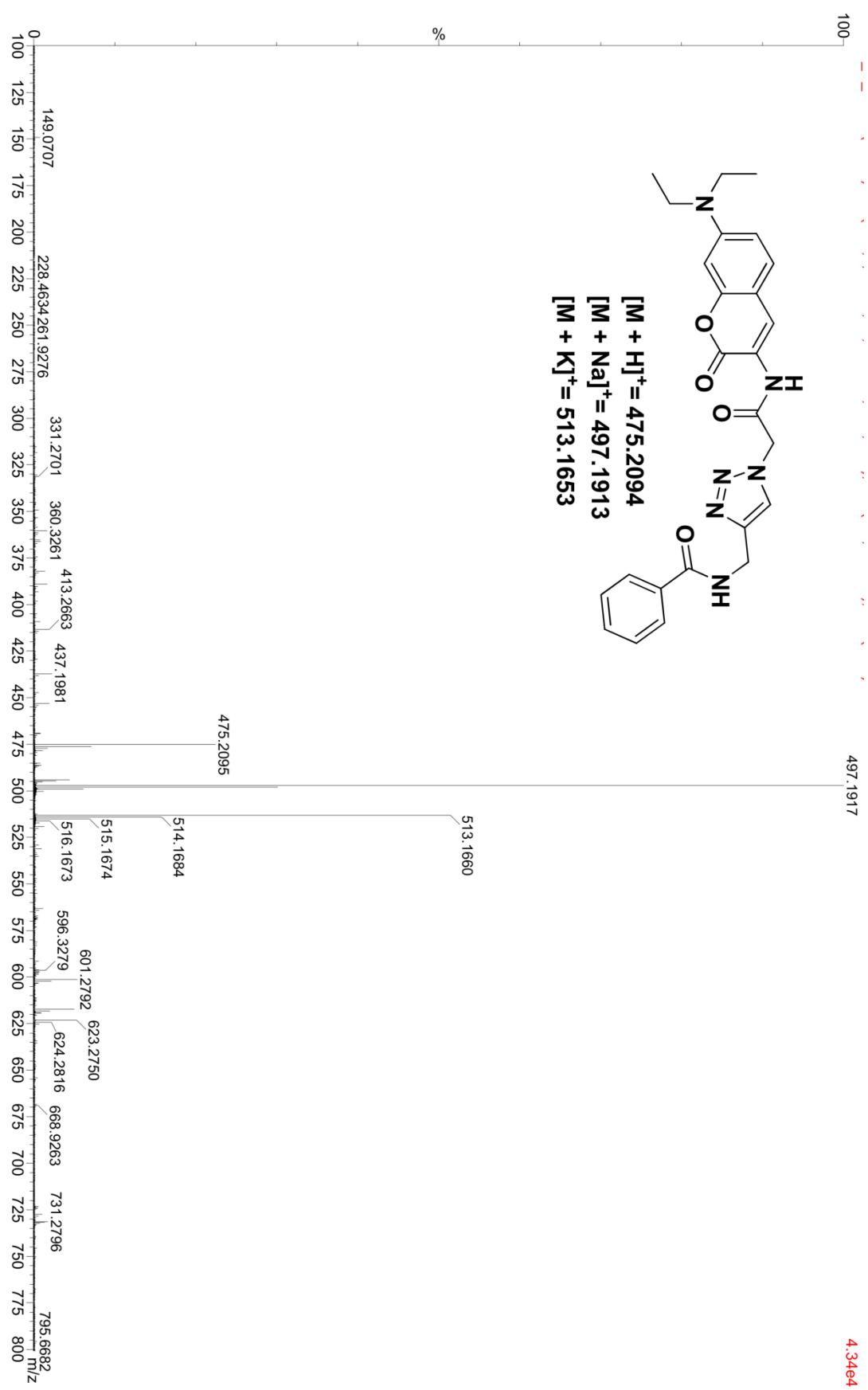
TOF MS ES<sup>+</sup>  
2.41e5



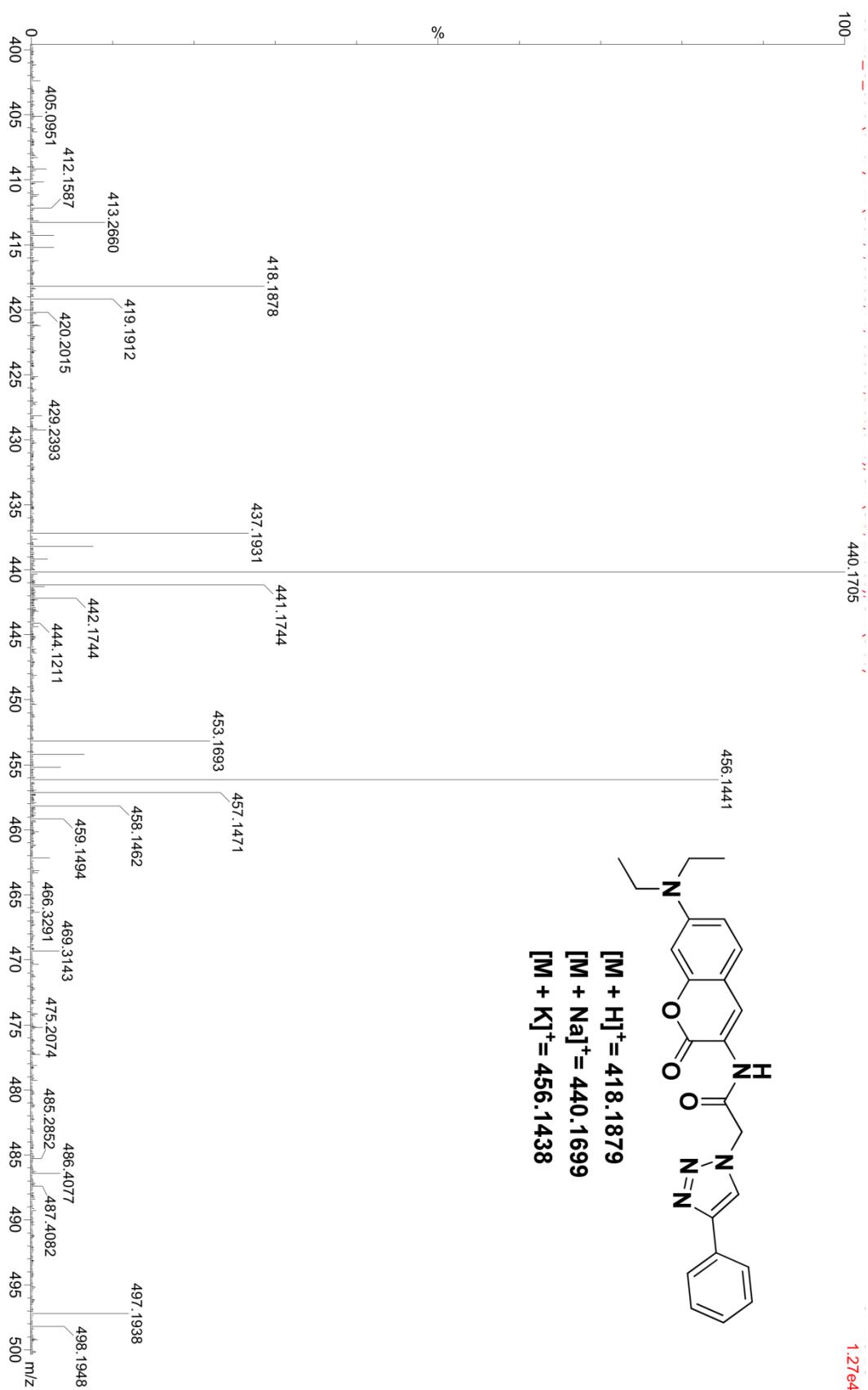
[M+H]<sup>+</sup>=316.1410  
[M+Na]<sup>+</sup>=338.1229  
[M+K]<sup>+</sup>=354.0968



HRMS spectra of PS-1



HRMS spectra of PS-2



HRMS spectra of PS-3

