#### **Supporting information**

# Highly Sensitive and Selective Detection of Silver (I) in Aqueous Solution with Silver(I)-specific DNA and Sybr Green I

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#### **Experimental Section**

SG (10000×) was purchased from invitrogen inc. When in use, it was diluted to  $250 \times$  with DMSO, then to  $10 \times (1.96 \times 10^{-6} \text{ M})$  with water to make a stock solution.  $1.625 \times 10^{-8}$  M SSO was first incubated for 15 min with different amount of Ag<sup>+</sup> in 10 µL of 100 mM MOPS buffer containing 500 mM NaNO<sub>3</sub>, pH 6.90 and one amount of MilliQ–H<sub>2</sub>O to 100µL reaction solution. Then 5 µLof 10×SG was added to the solution. After incubating for 2 min,the mixture was added one amount MilliQ–H<sub>2</sub>O to 800µL detection solution and used for the fluorescence study at ex497nm.

To study the quenching of  $Ag^+$ ,  $Cu^{2+}$  and  $Hg^{2+}$  to SG, 2 µL  $Ag^+(10 \text{ mM})$ ,  $Cu^{2+}(10 \text{ mM})$  and  $Hg^{2+}(10 \text{ mM})$  were respectively added into 10 µL SG(100×) for fluorescence study. The fluorescence spectra were shown in Fig. S1, indicating that metal ions such as  $Ag^+$ ,  $Cu^{2+}$  and  $Hg^{2+}$  could quench the fluorescence of SG.



**Fig. S1** The quenching of  $Ag^+$ ,  $Cu^{2+}$  and  $Hg^{2+}$  to SG

We used SG-SSO as a probe and illustrated the formation of  $Ag^+$ -hairpin structure by measuring fluorescence intensity at 530 nm every minute. As shown in Fig. S2, the  $Ag^+$ -hairpin structure forms immediately once the silver ion is added.



**Fig. S2** Fluorescence intensity change against time after adding Ag<sup>+</sup> into SG–SSO.



**Fig. S3** The fluorescence spectra of solutions containing SSO with different concentration of  $Ag^+(0, 1, 5, 10, 30, 50, 80, 100, 500 \text{ nM})$  [SSO] =  $1.625 \times 10^{-8}$  M and [SG]= $1.225 \times 10^{-7}$  M. A buffer of 100 mM MOPS, 500 mM NaNO<sub>3</sub>, pH 6.90 was used.



**Fig. S4** The fluorescence spectra of SSO in the absence or presence of 500 nM Ag<sup>+</sup>, 20 mM K<sup>+</sup>, 50 mM Na<sup>+</sup>, 100 mM Li<sup>+</sup>, 10 mM Ca<sup>2+</sup>, 10 mM Mg<sup>2+</sup>, 100  $\mu$ M Cu<sup>2+</sup>, 100  $\mu$ M Cd<sup>2+</sup>, 100  $\mu$ M Pb<sup>2+</sup>, 100  $\mu$ M Hg<sup>2+</sup>, respectively. [SSO]=1.625×10<sup>-8</sup> M and [SG]= 1.225×10<sup>-7</sup> M. Mixed sample (K<sup>+</sup>, Na<sup>+</sup>, Li<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, Cu<sup>2+</sup>, Zn<sup>2+</sup>, Cd<sup>2+</sup> and Pb<sup>2+</sup>, each 1  $\mu$ M) and real sample (river water) in the absence or presence of Ag<sup>+</sup> (500 nM) were detected.

## Table S1 Comparison between the current method and other methods by using

Methods	Linear range	LOD	Ref.
A graphene-based fluorescent nanoprobe for silver(I) ions	Not indicated	5 nM	a
Un-labeled C-rich ssDNA probe and controlled assembly of	10 to 500 nM	1.3 nM	b
MWCNTs			
Oligonucleotide-immobilized oscillator	Not indicated	10 nM	с
Oligonucleotide-based fluorogenic probe	50 to 700 nM	32 nM	d
Nucleic acid functionalized CdSe/ZnS quantum dots	Not indicated	1 µM	e
DNA SWCNT-based fluorescent sensor	0–150 nm	1 nM	f
Light scattering technique of DNA-functionalized AuNPs	200-9000 nM	50 nM	g
Silver(I)-specific DNA and Sybr Green I	1nM to 100 nM	1nM	Current
			work

### $C-Ag^+-C$ interaction

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