Electronic Supporting Information

for

A dark field light scattering platform for real-time monitoring the erosion of microparticles Co²⁺

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Fig. S1. SEM images of PpPD microparticles erode by Co^{2+} at different studied time points. $c_{Co^{2+}}$, 1×10⁻⁴ mol/L. The image scale bar corresponds to 1 µm.



Fig. S2. Study of the selectivity of Co^{2+} in the erosion of P*p*PD microparticles. Where I₀ and I were the dark field scattering intensity of P*p*PD before and after the reaction with the tested cation solution for 30 min.



Fig. S3 Dark-filed image of PpPD undergoing erosion by Co²⁺. The times at which the different images were captured are shown with latency time of 400/800 s. $c_{Co^{2+}}$, 1×10⁻⁴ mol/L. The image scale bar corresponds to 1 µm.



Fig. S4 Dark-filed image of PpPD undergoing erosion by Co²⁺. The times at which the different images were captured are shown with latency time of 400/800 s. $c_{Co^{2+}}$, 1×10⁻⁵ mol/L. The image scale bar corresponds to 1 µm.



Fig. S5 Dark-filed image of PpPD undergoing erosion by Co²⁺. The times at which the different images were captured are shown with latency time of 400/800 s. $c_{Co^{2+}}$, 1×10⁻⁶ mol/L. The image scale bar corresponds to 1 µm.



Fig. S6 Dark-filed image of PpPD undergoing erosion by Co²⁺. The times at which the different images were captured are shown with latency time of 400/800 s. $c_{Co^{2+}}$, 1×10^{-7} mol/L. The image scale bar corresponds to 1 µm.



Fig. S7 Dark-filed image of PpPD undergoing erosion by Co²⁺. The times at which the different images were captured are shown with latency time of 400/800 s. $c_{Co^{2+}}$, 1×10⁻⁸ mol/L. The image scale bar corresponds to 1 µm.

Table S1 Correlation coefficients and linear equations of the calibration graphs for Co²⁺.

Co ²⁺ concentration	Linear regression equation/µM	Correlation coefficient r
10 ⁻³ mol/L	F = 1.01 - 0.0229 c	0.9984
10-4 mol/L	F = 1.01 - 0.0155 c	0.9933
10 ⁻⁵ mol/L	F = 1.02 - 0.00861 c	0.9833
10 ⁻⁶ mol/L	F = 1.01 - 0.00749 c	0.9809
10-7 mol/L	F = 1.00 - 0.00567 c	0.9875
10 ⁻⁸ mol/L	F = 0.958 - 0.0239 c	0.9903