

1        **Simple colorimetric detection of Cr (III) in the aqueous solutions by the as synthesized**  
2                    **citrate capped gold nanoparticles and development of a paper based assay**

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10                    Supplementary information

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24 **Results:**

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26 **Table S1. The effect of pH on UV-visible intensity measured at a range of Cr (III)**

27 **concentrations [range:  $10^{-3}$  M -  $10^{-6}$  M], with a mixing ratio of 700  $\mu$ l of AuNP: 500 $\mu$ l of Cr**

28 **(III)**

pH	Different Cr (III) concentrations			
	$10^{-3}$ M	$10^{-4}$ M	$10^{-5}$ M	$10^{-6}$ M
3	0.716 $\pm$ 0.006	0.578 $\pm$ 0.004	0.432 $\pm$ 0.012	0.32 $\pm$ 0.004
4	0.317 $\pm$ 0.002	0.382 $\pm$ 0.007	0.192 $\pm$ 0.005	0.062 $\pm$ 0.002
5	0.467 $\pm$ 0.008	0.412 $\pm$ 0.005	0.451 $\pm$ 0.007	0.129 $\pm$ 0.012

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69 **Table S2. Selectivity of the analytical process measured using UV-vis spectrophotometer:**  
70 **The AuNPs at pH 4 interacted with different metal ions Cr (III), Cr (VI), Cd (II), Co (II),**  
71 **Hg (II), Mg (II), Mn (II), Ni (II), Fe (III), Zn (II), Cu (II), Al (II), Pb (II) of  $10^{-3}$ M**  
72 **concentrations at pH 3 with a mixing ratio of 700  $\mu$ l of AuNP: 500 $\mu$ l of metal ion.**

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Heavy metals	$A_{714}/A_{526}$			Mean	Std Deviation	Standard Error
	R1	R2	R3			
Cr (III)	0.716	0.714	0.718	0.716	0.002	0.001
Cr (VI)	0.053	0.054	0.056	0.054	0.001	0.0007
Cd (II)	0.052	0.059	0.051	0.054	0.004	0.002
Co (II)	0.081	0.08	0.085	0.082	0.002	0.001
Hg (II)	0.072	0.075	0.072	0.073	0.001	0.000
Mg (II)	0.081	0.082	0.084	0.082	0.001	0.000
Mn (II)	0.051	0.051	0.053	0.051	0.0015	0.0005
Ni (II)	0.059	0.054	0.054	0.055	0.002	0.001
Fe (II)	0.074	0.07	0.071	0.071	0.002	0.001
Al (II)	0.058	0.059	0.057	0.058	0.001	0.0005
Cu (II)	0.06	0.062	0.06	0.060	0.001	0.0006
Pb (II)	0.184	0.183	0.185	0.184	0.001	0.0005
Zn (II)	0.021	0.02	0.024	0.021	0.002	0.001

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80 **Table S3. Interfering effect of Pb (II) on the analytical process measured using UV-vis**  
81 **spectrophotometer: The AuNPs at pH 4 interacted with Pb (II) of  $10^{-3}$  M concentrations at**  
82 **pH 3 with a mixing ratio of 700  $\mu$ l of AuNP: 500 $\mu$ l of metal ions.**

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Heavy metal	$A_{714}/A_{526}$										Mean	SD	SE
	R1	R2	R3	R4	R5	R6	R7	R8	R9	R10			
Pb (II)	0.184	0.187	0.178	0.18	0.189	0.178	0.182	0.188	0.185	0.187	0.183	0.004	0.001

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86 **Table S4. Selectivity of the analytical process measured using UV-vis spectrophotometer:**  
87 **The AuNPs at pH 4 interacted with Cr (III) in the artificial mixture of heavy metals [Cr**  
88 **(VI), Cd (II), Co (II), Hg (II), Mg (II), Mn (II), Ni (II), Fe (III), Zn (II), Cu (II), Al (II), Pb**  
89 **(II) ] at  $10^{-3}$  M concentrations with a mixing ratio of 700  $\mu$ l AuNP to 500 $\mu$ l metal ions.**

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Artificial mixture of Heavy metals	$A_{714}/A_{526}$					Mean	SD	SE
	R1	R2	R3	R4	R5			
	0.637	0.655	0.667	0.642	0.648	0.649	0.011	0.005

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102 **Table S5. Color intensity in the Paper based assay measured using Adobe photoshop CS6:**  
103 **Cr (III) response in the artificial mixture of heavy metals [Cr (VI), Cd (II), Co (II), Hg (II),**  
104 **Mg (II), Mn (II), Ni (II), Fe (III), Zn (II),Cu (II),Al (II), Pb (II)] at $10^{-3}$  M concentrations,**  
105 **with a mixing ratio of 7  $\mu$ l AuNP to 3 $\mu$ l metal ions**

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Artificial mixture of Heavy metals	R1	R2	R3	R4	R5	Mean	SD	SE
	43	40	48	36	45	42	4.61	2.06

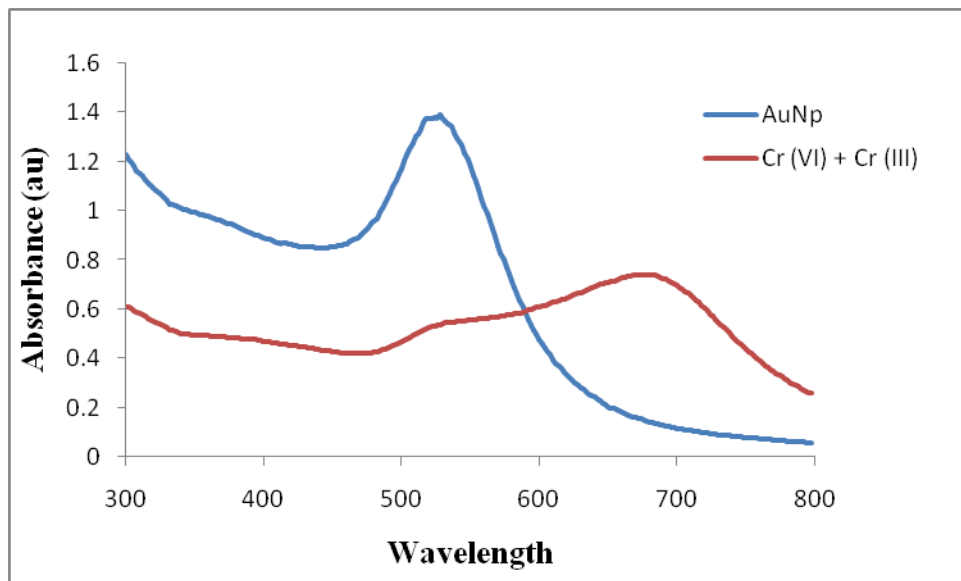
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125 **Figure S1. Selectivity of the process in mixture of Cr (III) and Cr (VI): The AuNPs at pH 4**  
126 **interacted with the mixture of Cr(III) and Cr (VI) of  $10^{-3}$ M concentrations at pH 3 with a**  
127 **mixing ratio of 700  $\mu$ l of AuNP: 500 $\mu$ l of metal ions.**

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