

Supplementary materials for the manuscript:

**Simultaneous voltammetric determination of aspartame and
acesulfame-K in food products using an anodically pretreated
boron-doped diamond electrode**

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Table S1 Figures of merit for the simultaneous determination of the AS and AK sweeteners using DPV or SWV coupled with an anodically pretreated BDD electrode.
Supporting electrolyte: 0.30 mol L⁻¹ H₂SO₄

	AS		AK	
	DPV	SWV	DPV	SWV
Linearity range (mol L⁻¹)	4.4×10^{-6} – 1.1×10^{-4}	5.3×10^{-6} – 1.1×10^{-4}	3.8×10^{-4} – 7.6×10^{-3}	5.2×10^{-4} – 7.6×10^{-3}
Sensitivity ($\mu\text{A mol}^{-1} \text{L}$)	3.5×10^5	3.3×10^5	4.3×10^4	4.4×10^4
Intercept (μA)	0.019	0.541	-22.0	-26.4
Correlation coefficient – R^2	0.998	0.999	0.997	0.996
LOD (mol L⁻¹)	1.60×10^{-6}	1.90×10^{-6}	2.80×10^{-4}	3.50×10^{-4}

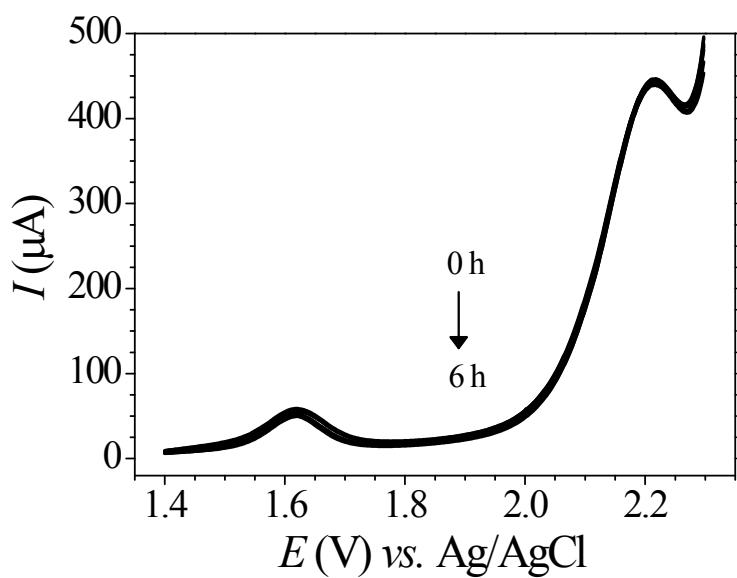


Fig. S1 SW voltammograms obtained every hour during a 6-hour period ($n = 7$) for a 46 $\mu\text{mol L}^{-1}$ AS and 3.7 mmol L^{-1} AK solution in 0.30 mol L^{-1} H_2SO_4 using an anodically pretreated BDD electrode. SWV conditions: $f = 30$ Hz, $a = 70$ mV, and $E_s = 4$ ms.

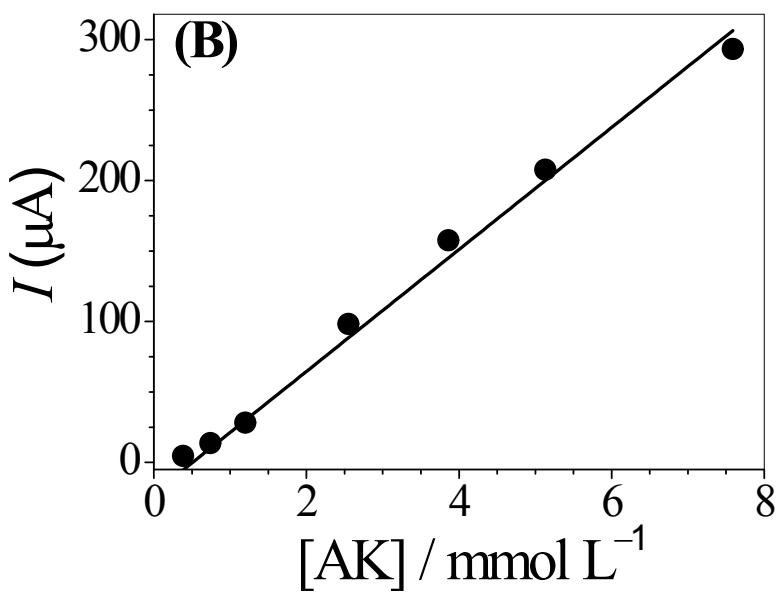
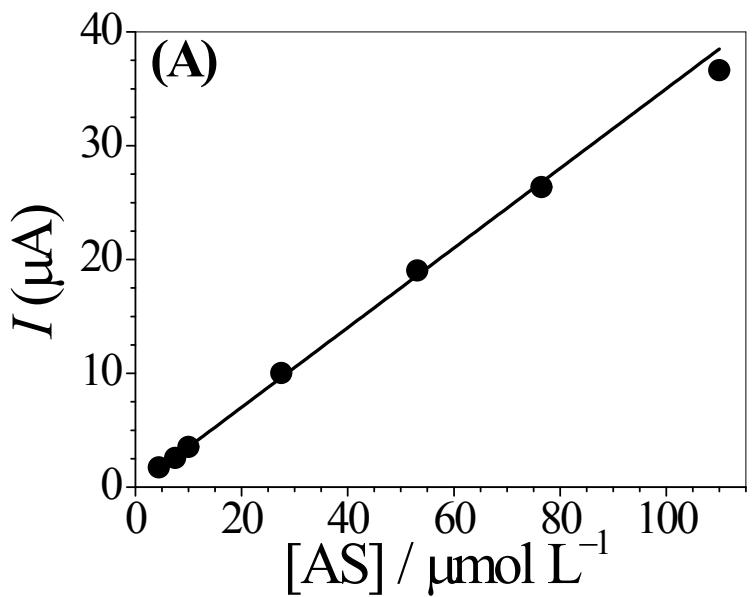


Fig. S2 Analytical curves for AS (A) and AK (B), obtained from the data shown in Fig. 4 in the paper.