Detection of glutathione utilizing a "off-on" fluorescent biosensor based on N-acetyl-L-cysteine capped CdTe QDs

Xuanping Tan^a, Jidong Yang^{*a b c}, Qin Li^d, Qiong Yang^c,

^a School of Chemistry and Chemical Engineering, Southwest University, Chongqing, 400715, China; ^b School of Chemistry and Environmental Engineering, Chongqing Three Gorges University, Wanzhou, Chongqing 404000, China; ^c School of Chemistry and Chemical Engineering, Yangtze Normal University, Chongqing, Fuling 408100, China; ^d Chongqing Medical and Health school, Fuling, Chongqing 408100, P. R. China;



Fig.S1 (A) TEM image; (B) UV-vis absorption (*curve* a) and fluorescence (*curve* b) spectra of asprepared NALC-capped CdTe QDs.



Fig.S2 Stern-Volmer curves for the NALC-CdTe QDs-Hg²⁺ solution system at three different temperatures (NALC-capped CdTe QDs, 1.425×10^{-4} mol·L⁻¹; Tris-HCl buffer solution, 0.5 mL, pH=7.4)

^{*}Corresponding author at: School of Chemistry and Chemical Engineering,

Southwest University, Chongqing, Beibei 400715, China. Tel.: +86 18996678896.

E-mail address: flyjd6400@163.com (J. Yang).



Fig S3 The effect of time on the NALC-capped CdTe QDs-Hg²⁺ system (A) and NALC-capped CdTe QDs-Hg²⁺-GSH system (B). (NALC-capped CdTe QDs, 1.425×10⁻⁴ mol·L⁻¹; Tris-HCl buffer solution, 0.5 mL, pH=7.4)



Fig S4 The effect of ionic strength on the NALC-capped CdTe QDs-Hg²⁺ system (*curve a*) and NALC-capped CdTe QDs-Hg²⁺-GSH system (*curve b*). (NALC-capped CdTe QDs: 1.425×10^{-4} mol·L⁻¹; HgCl₂: 24 µg·mL⁻¹; GSH: 64 µg·mL⁻¹; 0.5 mL Tris-HCl buffer solution (7.4)).