Surfactant ModulatedAggregation Induced Enhancement of Emission (AIEE) A Simple Demonstration toMaximize Sensor Activity

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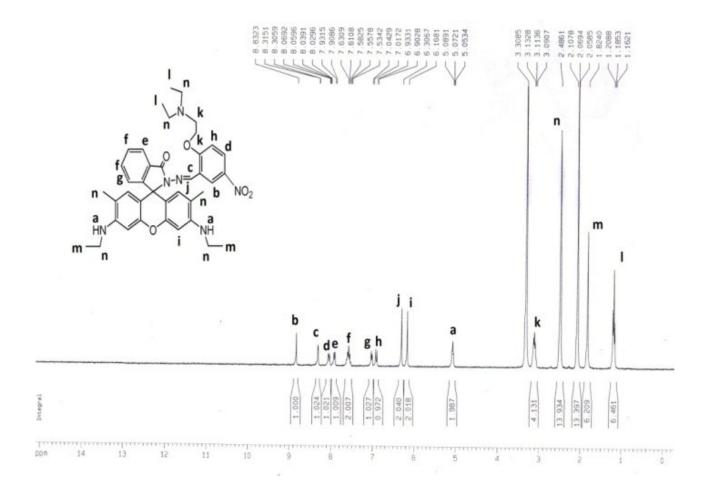


Fig. S1¹H NMR spectra of L³ in CDCl₃.

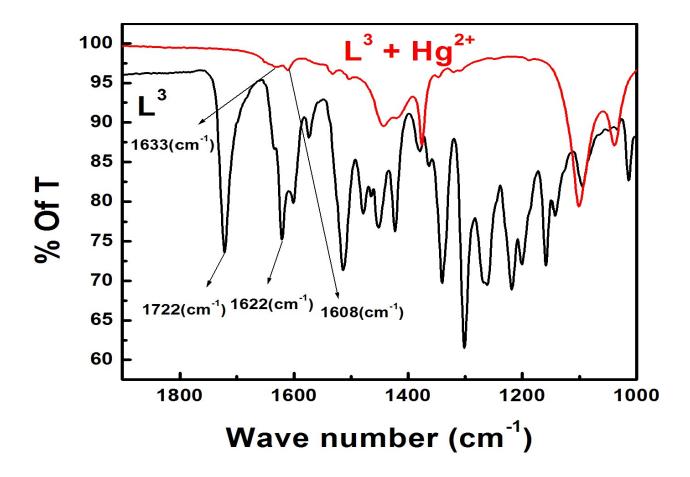


Fig. S2 IR spectra of L³ and L³-Hg²⁺ recorded

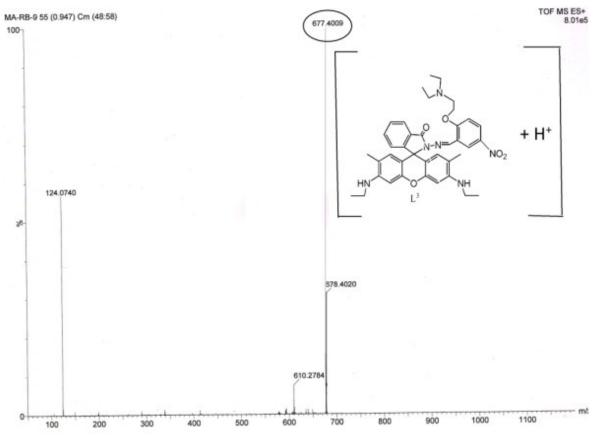


Fig. 3 Mass spectra of L^3

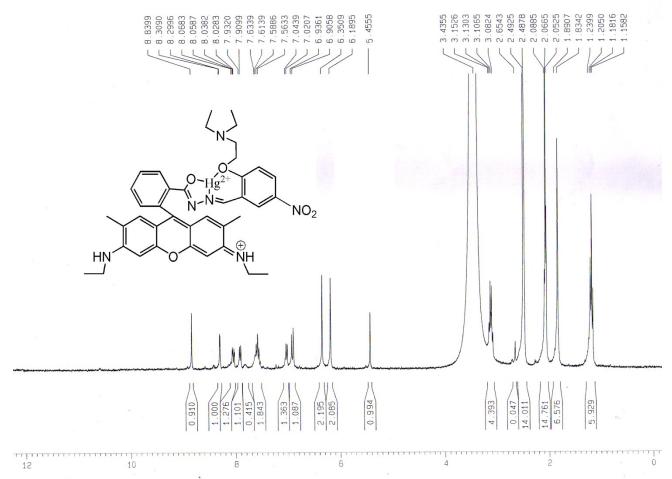


Fig. S4 ¹H-NMR spectra of L³-Hg²⁺ in DMSO-d₆

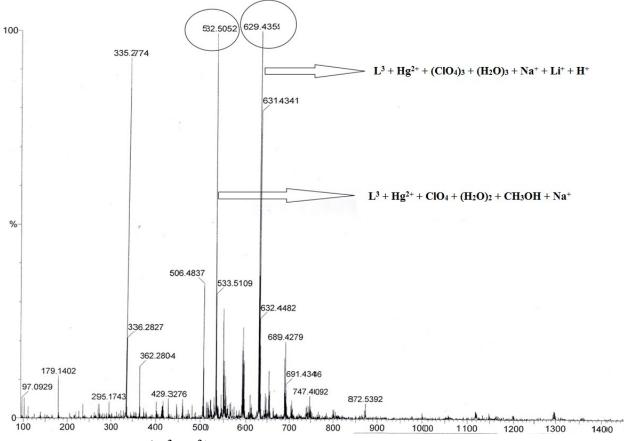


Fig. S5 Mass spectra of L^3 -Hg $^{2+}$

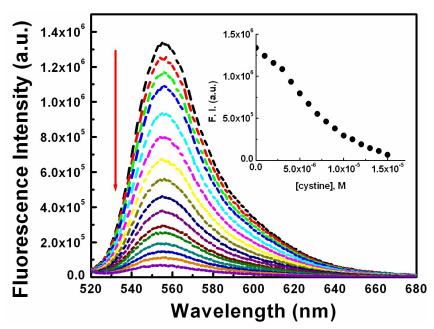


Fig. S6 Fluorescence quenching titration of L^3 -Hg²⁺ (10 μ M) complex by cystine. Inset is the plot of FI vs. [cystine].

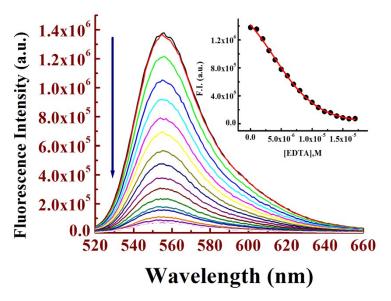


Fig. S7 (a) EDTA titrationin absence of SDS. Conditions are: $[L^3] = [Hg^{2+}] 10\mu M$.

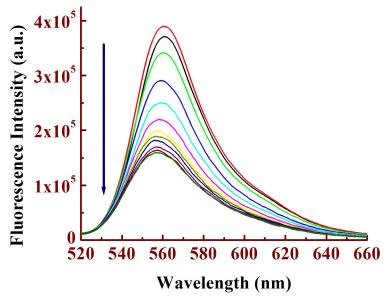


Fig. S7 (b) EDTA titration in presence of 5mM of SDS. Conditions are: $[L^3] = [Hg^{2+}] 10\mu M$.

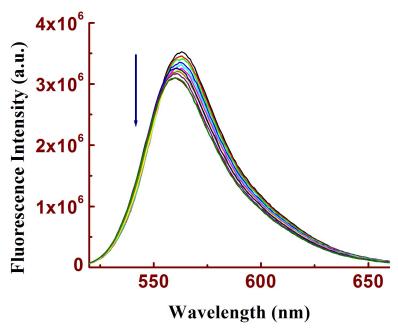


Fig. S7 (c) EDTA titration in presence of 9mM of SDS. Conditions are: $[L^3] = [Hg^{2+}] 10\mu M$.

Quantum Yield Determination:

Fluorescence quantum yields (Φ) were estimated by integrating the area under the fluorescence

curves with the equation:
$$\Phi_{sample} = \frac{OD_{std}}{OD_{sample}} \times \frac{A_{sample}}{A_{std}} \times \Phi_{std}$$

where, A is the area under the fluorescence spectral curve and OD is optical density of the compound at the excitation wavelength. The standard used for the measurement of fluorescence quantum yield was rhodamine $6G\Phi_{std}$ =0.94 in CH₃OH).

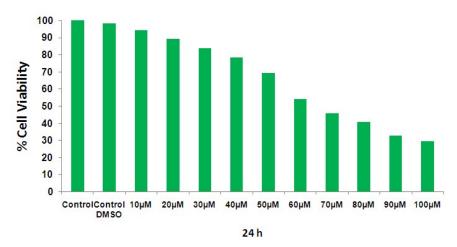


Fig. S8Cell viability assay performed by using ligand L^3

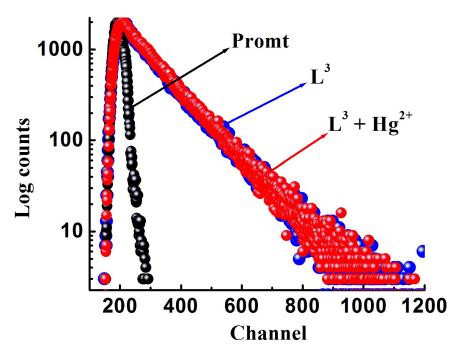


Fig. S9(a) Fluorescence decay curves of free L^3 ligand (10 μ M) and in presence of Hg^{2+} (1 equivalent) in H_2O at 25 °C.

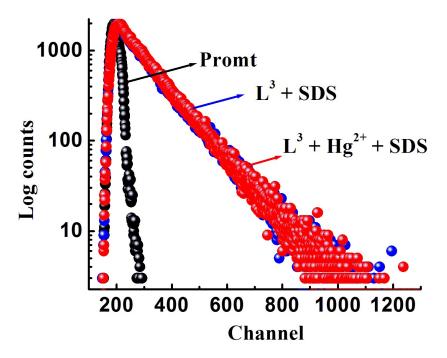


Fig. S9(b) Fluorescence decay curves of free L^3 ligand (10 μ M) and in presence of Hg²⁺ (1 equivalent) in H₂O in presence of SDS of 9 mM concentration at 25 °C.

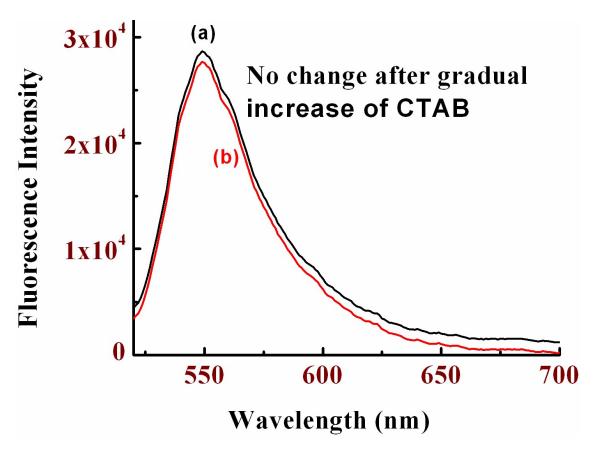


Fig. S10 Dependence of fluorescence intensity on [CTAB]. (a) 10 μ M L³-Hg²+only; (b) 10 μ M L³-Hg²+in presence of 10 mM CTAB.