

## Supplementary

### Gold nanoparticles supported on zirconium, tin and ruthenium oxides for reagentless electrochemical sensing of hydrogen peroxide

Fig. S1A. CV behaviors of RuO<sub>2</sub>-AuNP (curve a, b), ZrO<sub>2</sub>-AuNP (curve c, d) and SnO<sub>2</sub>-AuNP (curve e, f) modified GCEs in PBS measured at a scan rate 5 mVs<sup>-1</sup>

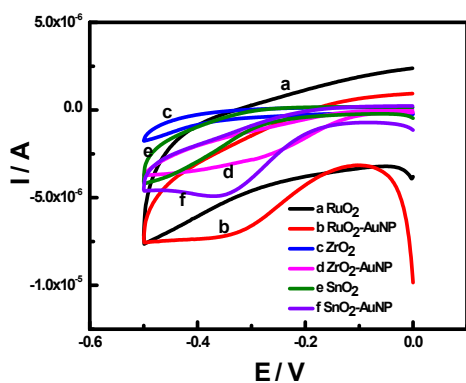


Fig. S1B. CV behaviors of RuO<sub>2</sub>-AuNP (curve a, b), ZrO<sub>2</sub>-AuNP (curve c, d) and SnO<sub>2</sub>-AuNP (curve e, f) modified GCEs in NaOH measured at a scan rate 5 mVs<sup>-1</sup>

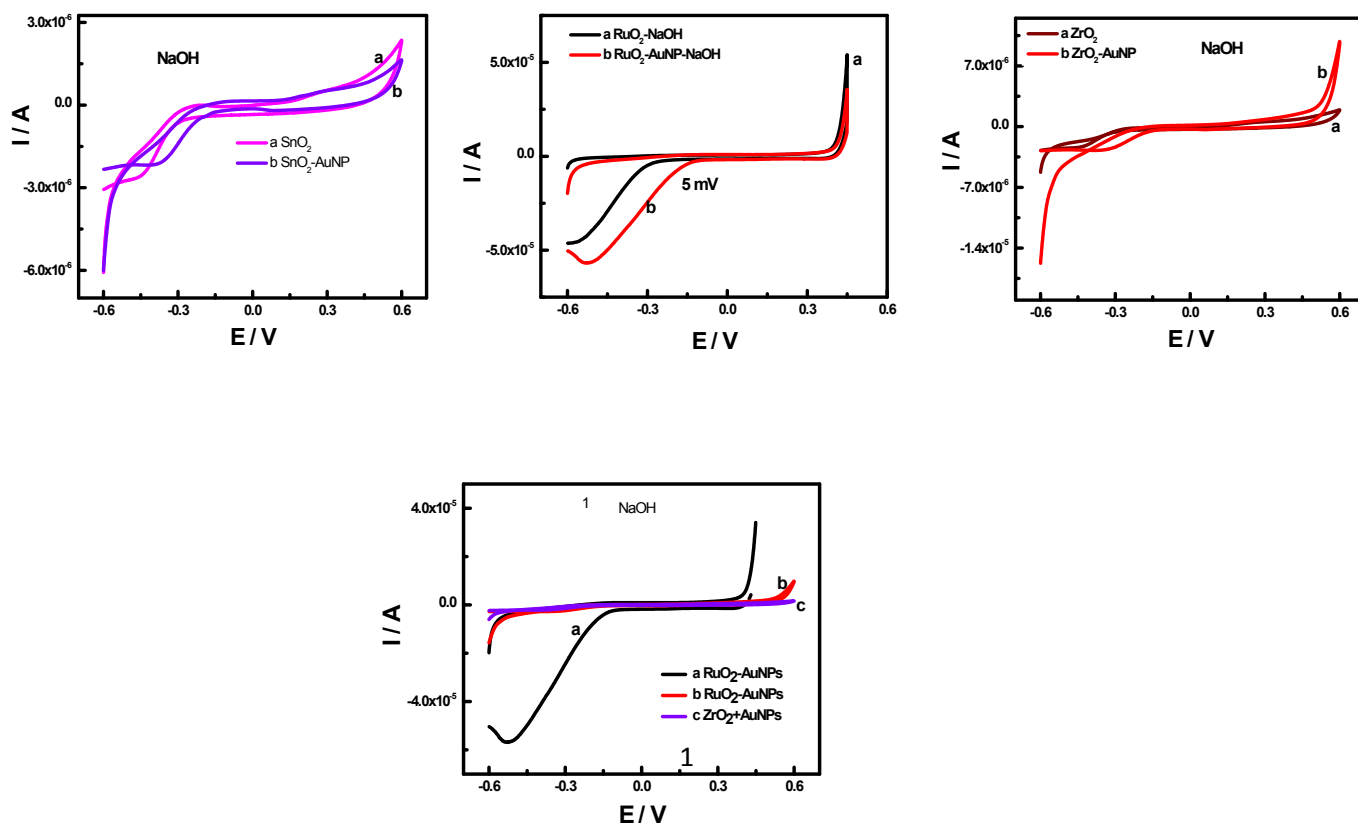


Fig. S1C. current versus pH plots (A) RuO<sub>2</sub>-AuNP , (B) ZrO<sub>2</sub>-AuNP and (C) SnO<sub>2</sub>-AuNP modified GCEs. (D). Effect of AuNP deposition time on the H<sub>2</sub>O<sub>2</sub> detection on the RuO<sub>2</sub> surface

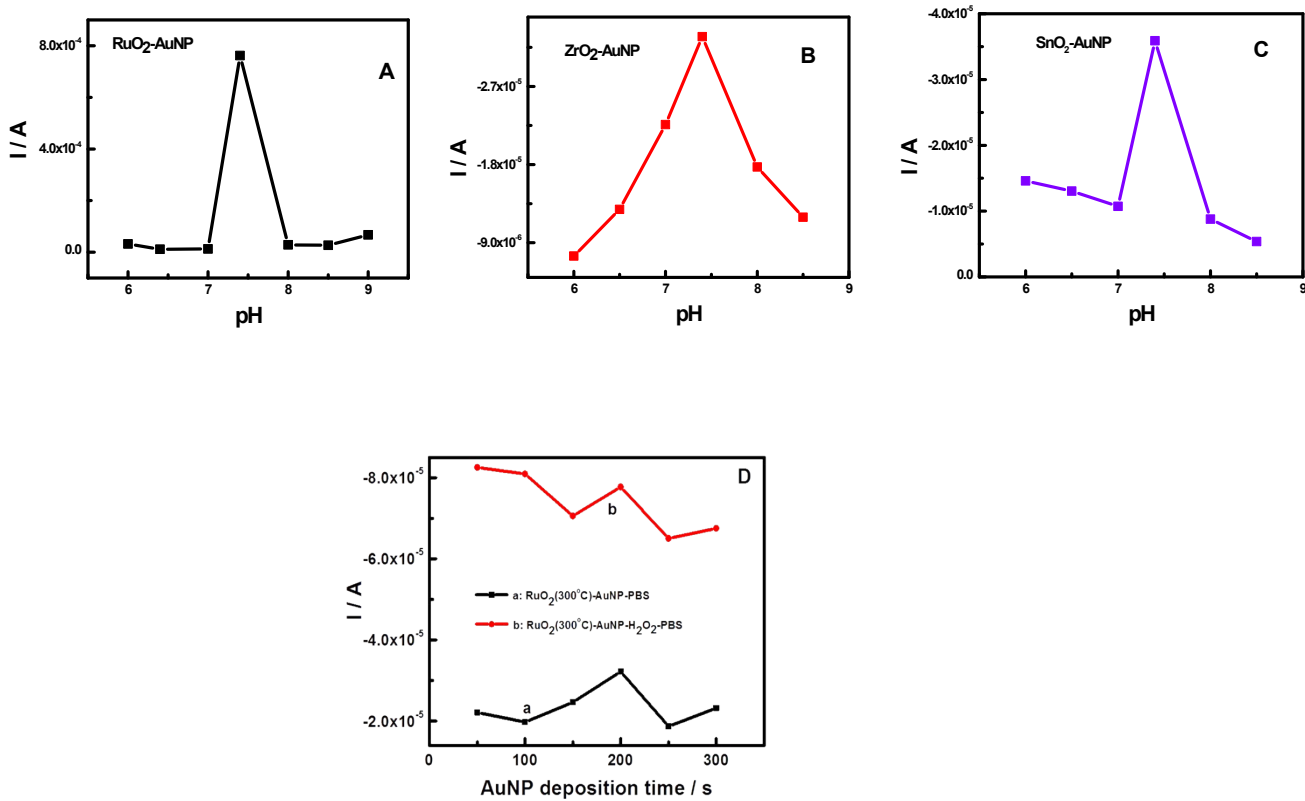
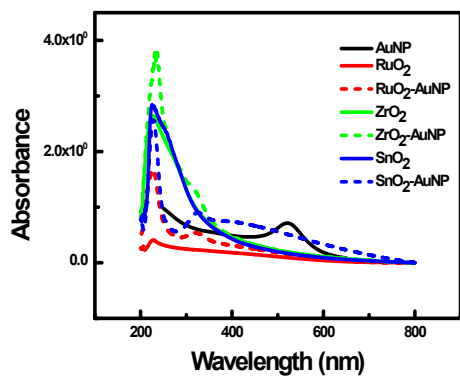


Fig. S2. UV – visible spectra of RuO<sub>2</sub> (curve a) , RuO<sub>2</sub>-AuNP (curve b), ZrO<sub>2</sub> (curve a), ZrO<sub>2</sub>-AuNP (curve b) SnO<sub>2</sub> (curve a) and SnO<sub>2</sub>-AuNP (Curve b)



Material	Band gap (eV)
RuO <sub>2</sub>	2.69
RuO <sub>2</sub> -AuNP	2.78
ZrO <sub>2</sub>	4.24
ZrO <sub>2</sub> -AuNP	3.14
SnO <sub>2</sub>	3.01
SnO <sub>2</sub> -AuNP	2.88

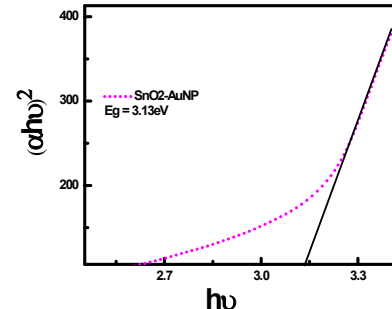
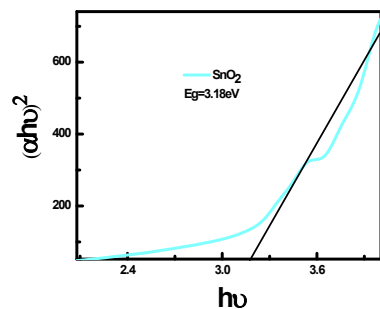
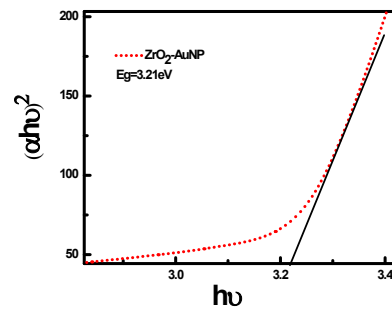
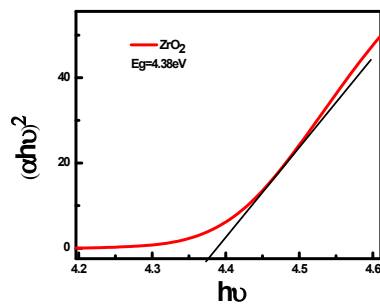
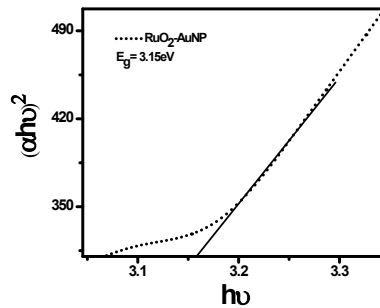
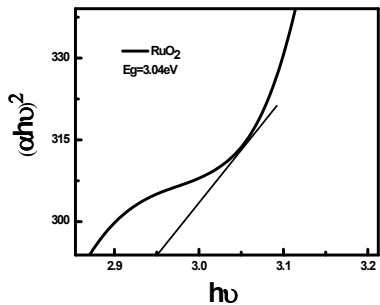


Fig. S3. CV behaviors of RuO<sub>2</sub>-AuNP (curve a,b), ZrO<sub>2</sub>-AuNP (curve c,d), SnO<sub>2</sub>-AuNP (curve e,f), modified GCEs in presence of 0.1 mM [Ru(NH<sub>3</sub>)<sub>6</sub>]<sup>3+</sup> in phosphate buffer measured at a scan rate 5 mVs<sup>-1</sup>. Solid lines indicate the CV behaviors in the absence (curves, a,c,e) and the dotted lines represent the CV behaviors (curves b, d, f) in the presence of 1 mM [Fe(CN)<sub>6</sub>]<sup>3-/4-</sup>. Inset : enlarged view of RuO<sub>2</sub> and RuO<sub>2</sub>-AuNP.

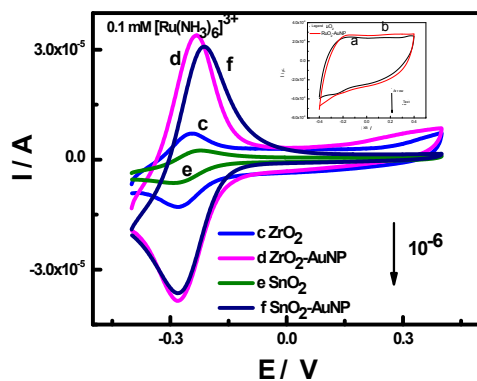


Fig.S4. FTIR spectra of RuO<sub>2</sub> (curve a), RuO<sub>2</sub>-AuNP (curve b), ZrO<sub>2</sub> (curve c), ZrO<sub>2</sub>-AuNP (curve d) SnO<sub>2</sub> (curve e) and SnO<sub>2</sub>-AuNP (Curve f)

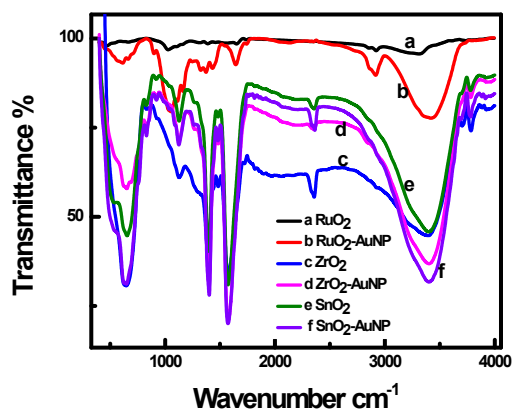


Table S1. XRD parameters for RuO<sub>2</sub>, ZrO<sub>2</sub>, SnO<sub>2</sub> and RuO<sub>2</sub>-AuNP, ZrO<sub>2</sub>-AuNP and SnO<sub>2</sub>-AuNP

RuO <sub>2</sub>	RuO <sub>2</sub> -AuNP	ZrO <sub>2</sub>	ZrO <sub>2</sub> -AuNP	SnO <sub>2</sub>	SnO <sub>2</sub> -AuNP
-	27.8 <b>(1 10)</b>	24.1 <b>(110)</b>	25.2 <b>(320)</b>	26.3 <b>(110)</b>	26.3 <b>(110)</b>
-	35.02 <b>(101)</b>	25.2 <b>(320)</b>	28.09 <b>(020)</b>	33.6 <b>(101)</b>	33.7 <b>(101)</b>
-	54.7 <b>(211)</b>	27.9 <b>(020)</b>	30.06 <b>(011)</b>	37.8 <b>(200)</b>	37.8 <b>(200)</b>
		30.6 <b>(011)</b>	31.4 <b>(120)</b>	51.5 <b>(211)</b>	51.7 <b>(211)</b>
		31.3 <b>(120)</b>	50.1 <b>(031)</b>	64.6 <b>(112)</b>	57.8 <b>(002)</b>
		35.1 <b>(101)</b>	54.0 <b>(131)</b>		64.6 <b>(112)</b>
		50.4 <b>(031)</b>			
		55.2 <b>(131)</b>			
		62.9 <b>(022)</b>			

Composite	Particle size / nm
RuO <sub>2</sub>	7.4
RuO <sub>2</sub> -AuNP	42.8
ZrO <sub>2</sub>	49
ZrO <sub>2</sub> -AuNP	60
SnO <sub>2</sub>	23
SnO <sub>2</sub> -AuNP	27

Fig. S5. EIS behaviors of RuO<sub>2</sub>-AuNP (curve a, b), ZrO<sub>2</sub>-AuNP (curve c, d) and SnO<sub>2</sub>-AuNP (curve e, f) modified GCEs in absence (curves, a, c, e) and presence of 1 mM H<sub>2</sub>O<sub>2</sub> (curves b, d, f) in phosphate buffer measured in the frequency range 100 kHz to 1 Hz at an applied potential -400 mV and amplitude 5 mV. While solid lines indicate EIS behaviors in the absence (curves, a,c,e) in the presence of 1 mM H<sub>2</sub>O<sub>2</sub> (curves b, d, f). Circles are the [R(QR)(QR)W] equivalent circuit fit data.

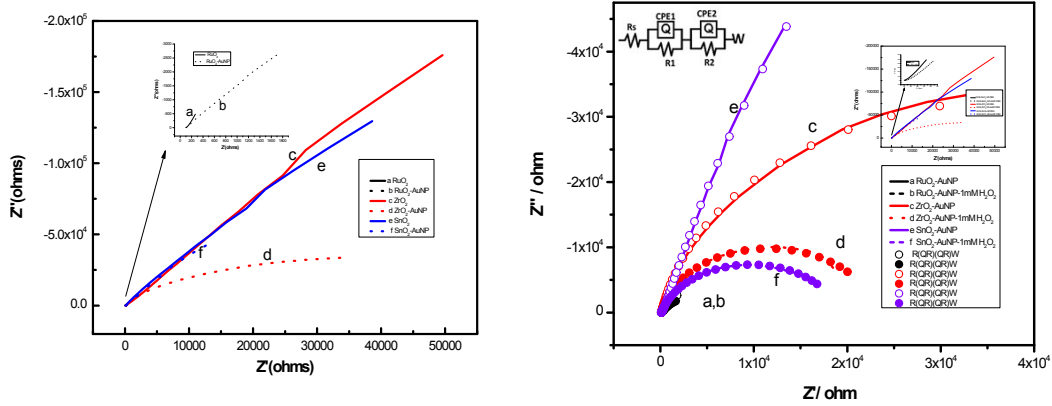


Table S2. Impedance parameters obtained for all three metal oxides in the absence and presence of AuNP fabricated in nafion film in the absence and presence of 1 mM H<sub>2</sub>O<sub>2</sub> in phosphate buffer. The circuit used is [R(QR)(QR)W].

Surfaces	R <sub>s</sub> (Ωcm <sup>-2</sup> )	Q <sub>CPE1</sub> (Fcm <sup>-2</sup> )	n1	R1(Ωcm <sup>-2</sup> )	Q <sub>CPE2</sub> (Fcm <sup>-2</sup> )	n2	R2 (Ω cm <sup>-2</sup> )	W (Ω s <sup>-1/2</sup> )	χ <sup>2</sup>
RuO <sub>2</sub>	52.500	4.14×10 <sup>-4</sup>	1.00	1.98×10 <sup>3</sup>	9.45×10 <sup>-9</sup>	0.98	1.96×10 <sup>2</sup>	2.50×10 <sup>-4</sup>	3.08×10 <sup>-4</sup>
RuO <sub>2</sub> -1mM H <sub>2</sub> O <sub>2</sub>	100.0	1.57×10 <sup>-7</sup>	0.68	1.12×10 <sup>3</sup>	4.89×10 <sup>-4</sup>	1.00	1.32×10 <sup>2</sup>	1.86×10 <sup>-4</sup>	4.90×10 <sup>-4</sup>
RuO <sub>2</sub> -AuNP	0.011	2.23×10 <sup>-6</sup>	0.48	1.08×10 <sup>2</sup>	8.25×10 <sup>-5</sup>	0.74	1.72×10 <sup>4</sup>	1.29×10 <sup>-4</sup>	2.48×10 <sup>-4</sup>
RuO <sub>2</sub> -AuNP-1mM H <sub>2</sub> O <sub>2</sub>	0.080	1.22×10 <sup>-4</sup>	0.78	4.38×10 <sup>3</sup>	6.14×10 <sup>-6</sup>	0.35	1.20×10 <sup>2</sup>	5.53×10 <sup>-4</sup>	2.44×10 <sup>-4</sup>
ZrO <sub>2</sub>	23.78	1.055×10 <sup>-8</sup>	1	1.16×10 <sup>2</sup>	1.226×10 <sup>-6</sup>	0.8314	4.9×10 <sup>15</sup>	0.000975	3.5×10 <sup>-4</sup>
ZrO <sub>2</sub> -1mM H <sub>2</sub> O <sub>2</sub>	0.01337	1.324×10 <sup>-6</sup>	0.8461	7.181×10 <sup>6</sup>	3.869×10 <sup>-8</sup>	0.8293	132.1	1.085×10 <sup>11</sup>	1.80×10 <sup>-4</sup>
ZrO <sub>2</sub> -AuNP	0.007587	2.735×10 <sup>-6</sup>	0.899	7.64×10 <sup>4</sup>	1.354×10 <sup>-6</sup>	0.5963	117.6	1.43×10 <sup>9</sup>	3.35×10 <sup>-4</sup>
ZrO <sub>2</sub> -AuNP-1mM H <sub>2</sub> O <sub>2</sub>	0.005672	6.406×10 <sup>-7</sup>	0.6435	119.8	2.736×10 <sup>-6</sup>	0.9023	2.25×10 <sup>4</sup>	0.00106	3.17×10 <sup>-4</sup>
SnO <sub>2</sub>	32.9	1.433×10 <sup>-6</sup>	0.868	1.283×10 <sup>6</sup>	2.144×10 <sup>-8</sup>	0.9561	74.08	2.029×10 <sup>8</sup>	2.03×10 <sup>-4</sup>
SnO <sub>2</sub> -1mM H <sub>2</sub> O <sub>2</sub>	22.54	3.013×10 <sup>-8</sup>	0.9043	90.58	1.388×10 <sup>-6</sup>	0.8727	3.821×10 <sup>5</sup>	3.284×10 <sup>6</sup>	1.31×10 <sup>-4</sup>
SnO <sub>2</sub> -AuNP	0.03902	3.19×10 <sup>-8</sup>	0.8478	114.6	4.321×10 <sup>-6</sup>	0.9013	3.954×10 <sup>5</sup>	0.0001306	9.5×10 <sup>-5</sup>
SnO <sub>2</sub> -AuNP-1mM H <sub>2</sub> O <sub>2</sub>	0.001266	3.3×10 <sup>-6</sup>	0.8335	1.915×10 <sup>4</sup>	1.686×10 <sup>-7</sup>	0.7239	124.2	6.211×10 <sup>19</sup>	2.42×10 <sup>-4</sup>



Fig. S6. (A) Comparative calibration curves of RuO<sub>2</sub>-AuNP, ZrO<sub>2</sub>-AuNP and SnO<sub>2</sub>-AuNP composites obtained for from chrono amperometry measurements made at -0.4 V for the successive addition of H<sub>2</sub>O<sub>2</sub>. Concentration range studied is 1×10<sup>-9</sup> to 30×10<sup>-3</sup> M.

(B) Comparative calibration curves of RuO<sub>2</sub> and RuO<sub>2</sub>-AuNP

(c) Comparative calibration curves of SnO<sub>2</sub> and SnO<sub>2</sub>-AuNP. Depicted for showing the superiority of metal oxide-AuNP over the lone metal oxide behavior

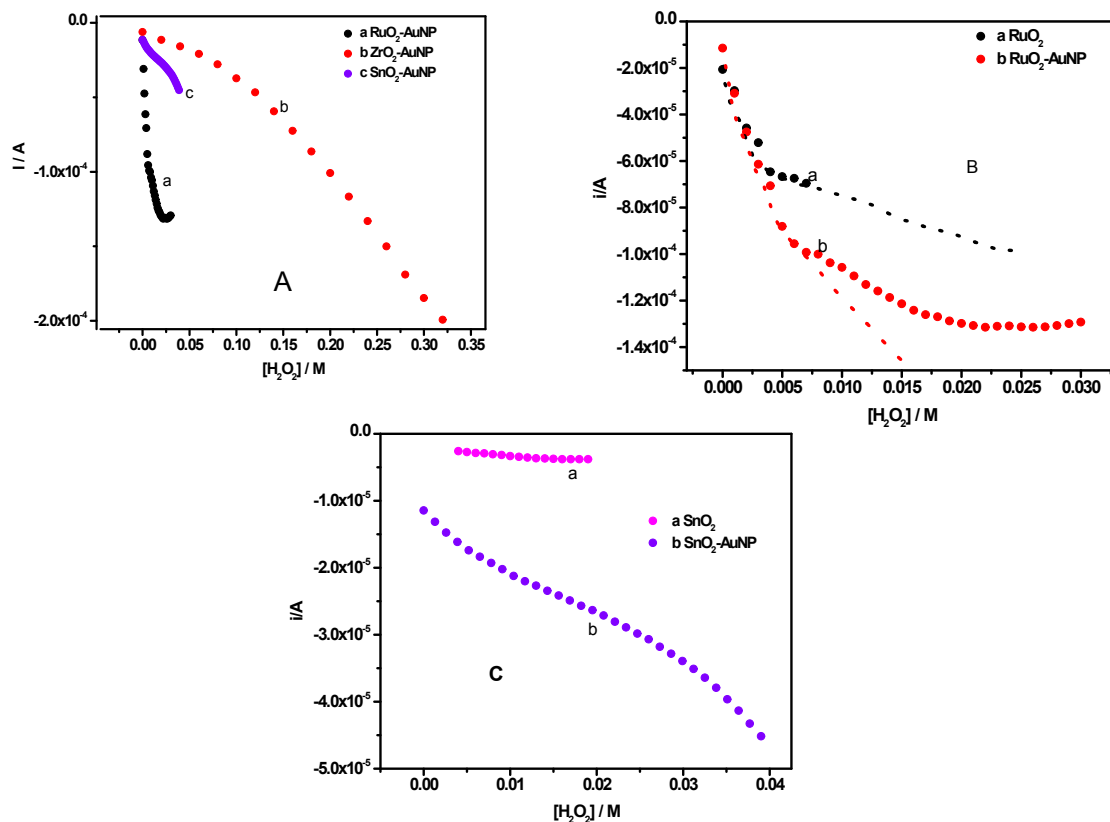


Fig. S7. CV behaviors of RuO<sub>2</sub>-AuNP (curve a,b), ZrO<sub>2</sub>-AuNP (curve c,d), SnO<sub>2</sub>-AuNP (curve e,f), modified GCEs in antiseptic (A) and commercial milk (B) in phosphate buffer measured at a scan rate 5 mVs<sup>-1</sup>. Solid lines indicate the CV behaviors in the absence (curves, a,c,e) and the dotted lines represent the CV behaviors (curves b, d, f) in presence of commercial samples

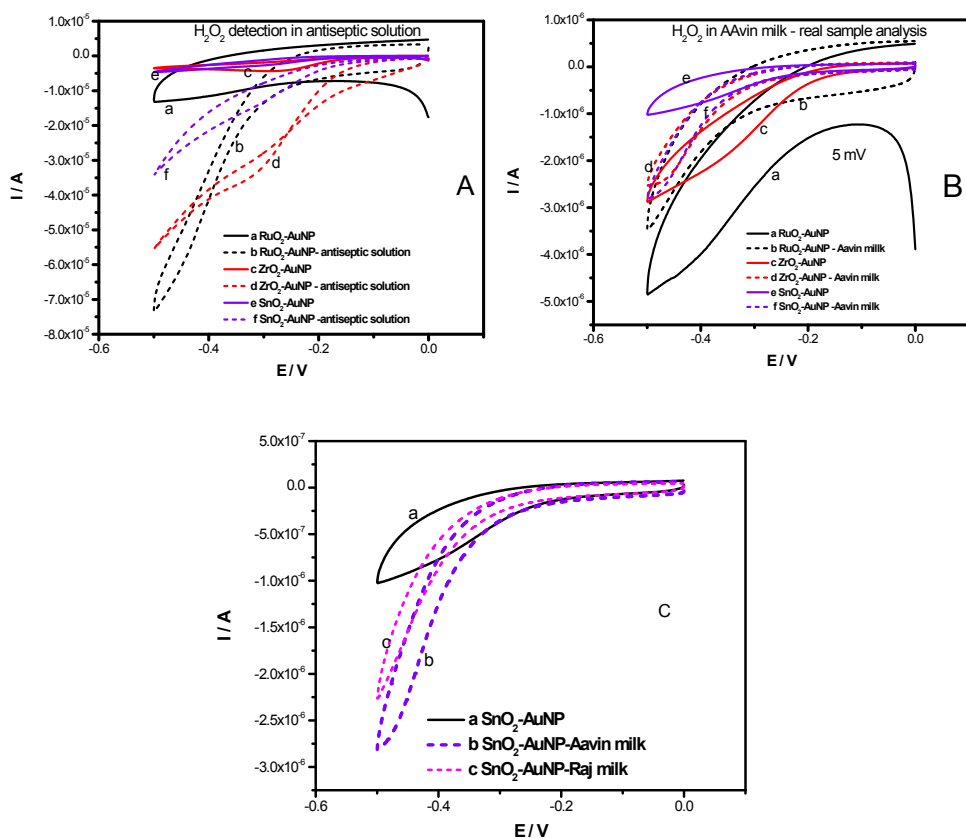


Table S3. Michealis-Menten kinetic parameters obtained for different metal oxide and metal oxide-AuNP used in this study for the reduction of H<sub>2</sub>O<sub>2</sub> in phosphate buffer

Surfaces	$K_m$ / M	$K_c$ / s	$\Gamma_t$ / moles cm <sup>-2</sup>
RuO <sub>2</sub>	$2.12 \times 10^{-3}$	$4.8 \times 10^4$	$1.42 \times 10^{-13}$
RuO <sub>2</sub> -AuNP	$10.1 \times 10^{-3}$	$1.18 \times 10^8$	$3.2 \times 10^{-17}$
ZrO <sub>2</sub>	$7.16 \times 10^{-5}$	$8.47 \times 10^3$	$1.38 \times 10^{-13}$
ZrO <sub>2</sub> -AuNP	1.48	$5.32 \times 10^6$	$9.53 \times 10^{-15}$
SnO <sub>2</sub>	$3 \times 10^{-3}$	$5.4 \times 10^3$	$5.9 \times 10^{-14}$
SnO <sub>2</sub> -AuNP	$2.3 \times 10^{-3}$	$2.9 \times 10^5$	$7.9 \times 10^{-15}$

Table S4. H<sub>2</sub>O<sub>2</sub> detection at different metal oxide /metal nanoparticle modified electrodes.

Electrode materials	Linear range and LOD	Effect of interferences with name	Real samples studied	Reference
AgNPs/ZnO/GCE	2μM to 5.5mM	Dopamine Ascorbic acid, Acetaminophen	--	QiWang.Microchim Acta (2010)169:361-365
AgNPs-MWCNT/Au electrode	0.05 to 17	AA,UA,	--	W. Zhao Talanta 80 (2009) 1029.
CPE/MWCNT/ RuO <sub>2</sub>	0.5 to 50mM	Glu,AA,UA	--	Ramin M.A. Biosensors and Bioelectronics (2012)
GCE/MnO <sub>2</sub> / Nafion	10.0×10 <sup>-6</sup> to 15.0×10 <sup>-5</sup>	UA.Dopamine,L-cysteine,AA	--	Li Zhen,Zhen Fang, Int.J.Electrochem. Sci.,4(2009)407-413
GC/MnO <sub>2</sub> /GO nano composite	5 to 600 μM	Glu, citric acid,cu <sup>2+</sup>	ClassicalKMnO <sub>4</sub> titration method	Limiao Li, Talanta 82(2010)1637-1641
CPE/NiO	1-110 μM	AA, UA	Diluted juice sample	Ying Mu, Biosensors and Bioelectronics 26(2011)2948-2952
GC/MWCNT/NiO	2.0×10 <sup>-4</sup> mol/L to 1.2 × 10 <sup>-2</sup> mol/L	AA, UA	Blood human serum sample	Shamsipur etal.Bioelectrochemistry 77(2010)120-124
GC/CuO/MWCNT	Upto 3.0mM	Dopamine, AA,UA,lactose, Citricacid	Blood human serum sample	Jiang Yang, Talanta 82(2010)25-33
Pt/CuO/nafion	0.15 μM to 9mM	AA,Dopamine, Glucose Ethanol, L-cysteine	--	Xiang –Min, Electroanalytical chemistry 612(2008) 157-163
GCE/β-MnO <sub>2</sub> nanorods	2.45 μM to 42.85 mM	L-glycine, glucose, L-glutamic acid,AA,UA, L-Ala	--	Ai-Jun Wang & Pei-Pei Zhang. Microchim Acta (2011) 175:31–37
GCE/ PPy-Pt hollow spheres nanocomposite.	3.5 μM to 9.9 mM	acetic acid, l-cysteine, AA,glucose, ethanol	--	Jingjing Li & Ruo Yuan. Microchim Acta (2010) 171:125–131
GCE/PTBO/ZrO <sub>2</sub> /GCNT/ H <sub>2</sub> O <sub>2</sub>	0.05mM to .25mM	--	Antiseptic solution	Int. J. Electrochem. Sci., 6 (2011) 4188 – 420
Hb/ZrO <sub>2</sub> chitosan/GCE/ H <sub>2</sub> O <sub>2</sub>	1.5 to 30.2M	--	--	Biosensors and Bioelectronics 19 (2004) 963–969
Au Electrode/ZrO <sub>2</sub> /Hb – H <sub>2</sub> O <sub>2</sub> sensor	1.75 μM to 4.9 mM	l-Cysteine Tyrosine Ethanol, Glucose, Acetic acid, Ascorbic acid		Microchim Acta (2010) 169:181–186
C@SnO <sub>2</sub> /PtNPs/GCE/ H <sub>2</sub> O <sub>2</sub>	1–170 μM	AA and UA	milk	Colloids and Surfaces B: Biointerfaces 101 (2013) 106–110
Au–chitosan/HRP/ITO/ H <sub>2</sub> O <sub>2</sub>	0.01 to 0.5 mM	--	--	Analytical Biochemistry 360 (2007) 288–293
AuNP arrays/ITO/ H <sub>2</sub> O <sub>2</sub>	1×10 <sup>-6</sup> –5×10 <sup>-4</sup> M	--	--	Analytica Chimica Acta 540 (2005) 299–306

HRP/nano-Ni-SnO <sub>2</sub> /GCE/ H <sub>2</sub> O <sub>2</sub>	1.0×10 <sup>-7</sup> to 3.0×10 <sup>-4</sup> M	--	--	Biosensors and Bioelectronics 36 (2012) 41–47
RuO <sub>2</sub> -Nafion-AuNP/GCE/ H <sub>2</sub> O <sub>2</sub>	0.1nM to 30mM	AA, UA and CA	Antiseptic solution	Present work
SnO <sub>2</sub> -Nafion-AuNP/GCE/ H <sub>2</sub> O <sub>2</sub>	1.3 mM to 39 mM	AA, UA and CA	Milk	Present work
ZrO <sub>2</sub> -Nafion-AuNP/GCE/ H <sub>2</sub> O <sub>2</sub>	1 nm to 1000 mM	AA, UA and CA	Milk	Present work