Electronic Supplementary Information (ESI)

Stereoselective synthesis and reaction of gold(I) (Z)-enethiolate

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1. Method

General Information. IR spectra were recorded on JASCO FT/IR-420 spectrometers. 1H NMR and ¹³C NMR spectra were obtained on either a JEOL JNM-AL300, JNM-AL400, or Bruler AV400 spectrometers. Chemical shifts (d) are reported in parts per million (ppm) downfield from intermal Me₄Si. Mass spectra (MS) were obtained on either a Waters LCT Premier, or SHIMADZU Model GCMS-QP 505 spectrometer. Preparative thin layer chromatography (TLC) was carried out on precoated plates of silica gel (MERCK, silica gel F-254). Kiselgel 60 (MERCK, 230-400 mesh) was used for column chromatography. Melting points were determined with Yanaco micro melting points apparatus and are uncorrected.

Subsrate. (*Z*)- β -alkylvinylthioimidonium tetrafluoroborate 5 were prepared from corresponding (*E*)- β -alkylvinyl- λ^3 -iodanes according to a literature method.^{S1} AuCl-tetrahydrothiophene complex 6 was prepared by the reaction of tetrachloroauric acid and tetrahydrothiophene according to a literature method.^{S2} 1,3-Dimesityl-3,4,5,6-tetrahydropyrimidin-1-ium bromide was prepared according to a literature method as shown below.^{S3}



Synthesis of Gold Enethiolate 7. Gold (*Z*)-1-decenylthiolate (7a). To a stirred solution of gold(I) chloride-tetrahydrothiophene complex (6) (22.0 mg, 0.068 mmol) in THF (1.1 mL) was added (*Z*)-*S*-1-decenyl-*N*,*N*-(dimethyl)thiobenzimidonium tetrafluoroborate (5a) (26.9 mg, 0.068 mmol) and 5% aqueous solution of Na₂CO₃ (290 µL, 0.136 mmol) under argon, and the mixture was stirred for 4 h at room temperature. The reaction was added small portion of water and resulting yellow precipitate was collected by filtration, which was purified by repeated washing with methanol and then with ethyl acetate to give pure 7a (22.4 mg, 90%) as a yellow amorphous. Mp 93-95°C (decomposition); IR (Nujol) 2979, 2881, 2833, 1454, 1377, 721 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.74-6.03 (m, 1H), 5.83-5.50 (m, 1H), 2.54-2.10 (m, 2H), 1.50-1.19 (m, 12H), 0.89 (t, *J* = 6.6 Hz, 3H); HRMS (ESI, negative, THF:MeOH = 1:1) for monomer: calcd for C₁₀H₁₉S [(M-Au)⁻], 171.1207, found 171.1213; for dimer: calcd for C₂₀H₃₈AuS₂ [(M₂-Au)⁻] 539.2081, found 539.2034; for trimer: calcd for C₃₀H₅₇Au₂S₃ [(3M-Au)⁻] 907.2954, found

907.2971. Anal. Calcd for C₁₀H₁₉AuS • 1/4H₂O: C, 32.22; H. 5.27. Found: C, 32.25; H, 5.10.

Gold (*Z***)-3-phenyl-1-propenylthiolate (7b):** a yellow powder; mp 157-160 °C (decomposition); IR (neat) 3059, 3024, 1601, 1493, 1452, 1317, 1281, 1074, 1030, 750, 696 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.32-6.81 (m, 5H), 6.76-5.50 (m, 2H), 3.95-3.36 (m, 2H); HRMS (ESI, negative, CHCl₃:MeOH = 1:1) calcd for C₁₈H₁₈S₂Au [(2M-Au)⁻] 495.0516, found 495.0514. Anal. Calcd for C₉H₉AuS•1/2H₂O: C, 30.43; H. 2.84. Found: C, 30.14; H, 2.60.

Synthesis of Gold Enethiolate-triphenylphosphine Complex 8a. To a stirred solution of gold (Z)-1-decenylthiolate (7a) (20.0 mg, 0.054 mmol) in dichloromethane (2.0 mL) was added triphenylphosphine (14.2 mg, 0.054 mmol) under argon, and the mixture was stirred for 30 min at room temperature. The reaction mixture was washed with water and the aqueous layer was extracted with dichloromethane four times. Combined organic phase was dried over anhydrous Na₂SO₄, filtered, and concentrated under aspiratory vacuum to give (Z)-1-decenylthiolato(triphenylphosphine)gold (8a) (32.2 mg, 95%) as a colorless oil. IR (neat) 3053, 2924, 2850, 1587, 1479, 1435, 1346, 1309, 1182, 1101, 1028, 999, 746, 710, 692 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.59-7.39 (m, 15H), 6.62 (d, J = 9.1 Hz, 1H), 5.59 (dt, J = 9.1, 6.8 Hz, 1H), 2.44 (q, J = 6.8 Hz, 2H), 1.47-1.13 (m, 12H), 0.86 (t, J = 6.3 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 134.2 (d, ²*J*_{cp} = 13.7 Hz), 131.7, 131.6, (d, ⁴*J*_{cp} = 2.5 Hz), 129.7 (d, ¹*J*_{cp} = 55.4 Hz), 129.1 (d, ${}^{3}J_{cp} = 11.2 \text{ Hz}$), 123.4, 31.9, 29.6(2C), 29.4, 29.3, 28.7, 22.7, 14.1; HRMS (ESI, positive) calcd for C₂₈H₃₄AuNaSP [(M+Na)⁺] 653.1682, found 653.1686.

Synthesis of Gold **Enethiolate-NHC** Complex 9a. То а stirred solution of 1,3-diisopropylimidazolium chloride (7.5 mg, 0.04 mmol) in THF (2 mL) was added 1.63 M butyllithium (29 µl, 0.047 mmol) in hexane at -78 °C under argon, and the mixture was gradually warmed to room temperature for 3.5 h. The resulting NHC solution was then cooled to -30 °C and gold (Z)-1-decenylthiolate (7a) (10.2 mg, 0.028 mmol) was added in one portion. The mixture was gradually warmed to room temperature and stirred for 5 h. After addition of water, the mixture was extracted with diethyl ether four times. Combined organic phase was dried over Na₂SO₄, filtered, and concentrated under an aspiratory vacuum to give an oil, which was washed several times with hexane by decantaion at room temperature to give (Z)-1-decenylthiolato(1,3-diisopropylimidazol-2-ylidene)gold (9a) (18.6 mg, 89 %) as a colorless oil. IR (neat) 2972, 2924, 2852, 1462, 1431, 1415, 1394, 1371, 1211 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.96 (s, 2H), 6.60 (d, J = 8.5 Hz, 1H),

5.54 (dt, J = 8.5, 6.6 Hz, 1H), 5.07 (sept, J = 6.6 Hz, 2H), 2.43 (q, J = 6.6 Hz, 2H), 1.51-1.17 (m, 12H), 1.46 (d, J = 6.6 Hz, 12H), 0.87 (t, J = 7.3 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 180.2, 130.8, 124.9, 116.5, 53.2, 32.0, 29.7, 29.6, 29.5, 29.4, 28.4, 23.4, 22.7, 14.1; HRMS (ESI, positive) calcd for C₁₉H₃₅AuN₂NaS [(M+Na)⁺] 543.2084, found 543.2082.

Synthesis of Gold **Enethiolate-NHC** Complex 10a. To a stirred solution of 1,3-bis[1,3,5-(trimethyl)phenyl]-3,4,5,6-tetrahydropyrimidin-1-ium bromide (25.7 mg, 0.064 mmol) and gold (Z)-1-decenylthiolate (7a) (10.7 mg, 0.029 mmol) in THF (2 mL) was added 0.26 M lithium *tert*-butoxide (267 µl, 0.069 mmol) in THF at room temperature under argon, and the mixture was warmed to 45 °C and stirred at the temperature for 14 h. After addition of water, the mixture was extracted with hexane four times. Combined organic phase was dried over Na₂SO₄, filtered, and concentrated under an aspiratory vacuum to give an oil, which was washed several decantaion times with hexane by at room temperature to give (Z)-1-decenylthiolato {1,3-bis[1,3,5-(trimethyl)phenyl]-3,4,5,6-tetrahydropyrimidin-2-ylidene} gold (10a) (19.4 mg, 97 %) as colorless oil. Recrystallization from ethyl acetate-hexane at 4 °C yielded pure complex 10a as colorless prisms. mp 136-138 °C; IR (neat) 2922, 2852, 1608, 1520, 1479, 1402, 1375, 1344, 1308, 1207, 1032, 852, 754 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.90 (s, 4H), 5.48 (dt, J = 9.2 1.5 Hz, 1H), 5.11 (dt, J = 9.2, 7.0 Hz, 1H), 3.40 (t, J = 6.2 Hz, 4H), 2.32 (t, J = 6.2 Hz, 2H), 2.27 (s, 12H), 2.25 (s, 6H), 2.07 (q, J = 7.0 Hz, 2H), 1.29-1.15 (m, 12H), 0.85 (t, J = 7.0 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 201.8, 141.8, 138.0, 134.8, 129.6, 129.0, 125.8, 45.2, 31.9, 29.6, 29.6, 29.3, 27.8, 22.7, 21.1, 21.0, 18.0, 14.1; HRMS (ESI, positive) calcd for C₃₂H₄₇AuN₂NaS [(M+Na)⁺] 711.3023, found 711.2952. Anal. Calcd for C₃₂H₄₇AuN₂S•1/4H₂O: C, 55.43; H. 6.91; N, 4.04. Found: C, 55.22; H, 6.94; N, 4.02.

General Procedure for Alkylation of Gold (Z)-Enethiolates. A Typical Example (Table 1, entry 6). To a stirred solution of (Z)-1-decenylthiolato {1,3-bis[1,3,5-(trimethyl)phenyl] -3,4,5,6-tetrahydropyrimidin-2-ylidene}gold (10a) (14.6 mg, 0.021 mmol) in dichloromethane (2.0 mL) was added TEMPO (0.3 mg, 0.0021 mmol) and iodomethane (30.1 mg, 0.21 mmol) under argon, and the mixture was stirred at room temperature for 30 min. The reaction mixture was washed with water and the aqueous layer was extracted with dichloromethane four times. Combined organic phase was dried over anhydrous Na₂SO₄, filtered, and concentrated under aspiratory vacuum to give an oil, which was purified by silica gel column chromatography using hexane to give (Z)-1-decenyl methyl sulfide (**11a**) (3.7 mg, 95%) as a colorless oil. The *Z*:*E* ratio was determined by ¹H NMR of a crude oil. (Z)-**11a**: 54,55 ¹H NMR (400 MHz, CDCl₃) δ 5.86 (d, *J* = 9.1 Hz, 1H), 5.53 (dt, *J* = 9.1, 7.3 Hz, 1H), 2.27 (s, 3H), 2.11 (q, *J* = 7.3 Hz, 2H), 1.44-1.28 (m, 4H), 0.91 (t, *J* = 7.3 Hz, 3H).

(Z)-1-decenyl ethyl sulfide (11b) (Table 1, Entry 7):^{S4,S6} a colorless oil; ¹H NMR (400 MHz, CDCl₃) δ 5.91 (d, J = 9.1 Hz, 1H), 5.57 (dt, J = 9.1, 6.9 Hz, 1H), 2.67 (q, J = 6.9 Hz, 2H), 2.11 (q, J = 6.9 Hz, 2H), 1.44-1.20 (m, 15H), 0.88 (t, J = 6.9 Hz, 3H).

(Z)-1-decenyl isopropyl sulfide (11c) (Entry 9):^{S4} a colorless oil; IR (neat) 2958, 2924, 2854, 1607, 1462, 1243, 770 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.98 (d, J = 9.4 Hz, 1H), 5.58 (dt, J = 9.4, 7.0 Hz, 1H), 3.05 (sept, J = 6.6 Hz, 1H), 2.11 (dt, J = 7.0, 6.8 Hz, 2H), 1.42-1.21 (m, 12H), 1.31 (d, J = 6.6 Hz, 6H), 0.87 (t, J = 6.6 Hz, 3H).

(Z)-1-Decenyl allyl sulfide (11d) (Entry 12):^{S4} a colorless oil; ¹H NMR (400 MHz, CDCl₃) δ 5.87 (d, J = 9.1 Hz, 1H), 5.84 (ddt, J = 17.6, 10.6, 7.3 Hz, 1H), 5.58 (dt, J = 9.1, 6.9 Hz, 1H), 5.16 (d, J = 17.6 Hz, 1H), 5.11 (d, J = 10.6 Hz, 1H), 3.28 (d, J = 7.3 Hz, 2H), 2.11 (q, J = 6.9 Hz, 2H), 1.43-1.18 (m, 12H), 0.88 (t, J = 6.5 Hz, 3H).

(Z)-1-Decenyl benzyl sulfide (11e) (Entry 13):⁸⁴ a colorless oil; ¹H NMR (400 MHz, CDCl₃) δ 7.36-7.28 (m, 5H), 5.89 (d, *J* = 9.2 Hz, 1H), 5.55 (dt, *J* = 9.2, 7.0 Hz, 1H), 3.86 (s, 2H), 2.09 (q, *J* = 7.0 Hz, 2H), 1.39-1.16 (m, 12H), 0.88 (t, *J* = 6.6 Hz, 3H).

(Z)-1-Decenyl phenacyl sulfide (11f) (Entry 14):⁸⁴ a colorless oil; IR (neat) 3060, 2925, 2854, 1680, 1599, 1579, 1448, 1277, 1014, 688 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.97 (d, J = 7.3 Hz, 2H), 7.58 (t, J = 7.3 Hz, 1H), 7.47 (t, J = 7.3 Hz, 2H), 5.97 (d, J = 9.1 Hz, 1H), 5.65 (dt, J = 9.1, 6.9 Hz, 1H), 3.95 (s, 2H), 2.07 (q, J = 6.9 Hz, 2H), 1.37-1.14 (m, 12H), 0.87 (t, J = 6.9 Hz, 3H).

Bis((*Z*)-1-decenylthio)methane (13):⁸⁴ a colorless oil; ¹H NMR (400 MHz, CDCl₃) δ 6.06 (d, *J* = 9.1 Hz, 2H), 5.69 (dt, *J* = 9.1, 6.9 Hz, 2H), 3.87 (s, 2H), 2.12 (q, *J* = 6.9 Hz, 4H), 1.44-1.17 (m, 24H), 0.88 (t, *J* = 6.6 Hz, 6H).

Nucleophilic Substitution of 2,4-dinitrohalobenzene with Gold (Z)-Enethiolates. A Typical Example (Scheme 6, Reaction of 10a with 14b). To a stirred solution of (Z)-1-decenylthiolato $\{1,3-bis[1,3,5-(trimethyl)phenyl]-3,4,5,6-tetrahydropyrimidin-2-ylidene\}$ gol d (10a) (13.4 mg, 0.019 mmol) in THF (2.0 mL) was added TEMPO (0.3 mg, 0.002 mmol) and 2,4-dinitro-1-bromobenzene (4.8 mg, 0.019 mmol) under argon, and the mixture was stirred at room temperature for 24 h. The reaction mixture was washed with water and the aqueous layer was extracted with dichloromethane four times. Combined organic phase was dried over

anhydrous Na₂SO₄, filtered, and concentrated under aspiratory vacuum to give a yellow oil, which was purified by preparative TLC (hexane:diethyl ether = 6:4) to give (*Z*)-1-decenyl 2,4-dinitrophenyl sulfide (**15**) (5.6 mg, 88%) as yellow oil. The *Z*:*E* ratio was determined by ¹H NMR of a crude oil. (*Z*)-**15**:^{**S4**} ¹H NMR (400 MHz, CDCl₃) δ 9.09 (d, *J* = 2.6 Hz, 1H), 8.34 (dd, *J* = 8.8, 2.6 Hz, 1H), 7.66 (d, *J* = 8.8 Hz, 1H), 6.45 (dt, *J* = 9.1, 6.9 Hz, 1H), 6.19 (d, *J* = 9.1 Hz, 1H), 2.39 (q, *J* = 6.9 Hz, 2H), 1.51-1.41 (m, 2H), 1.38-1.17 (m, 10H), 0.87 (t, *J* = 6.6 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 145.4, 144.7, 143.9, 143.4, 127.7, 126.0, 120.7, 116.3, 30.9, 28.4, 28.3(2C), 27.9, 21.7, 13.2.

(*E*)-1-Decenyl 2,4-dinitrophenyl sulfide (15): yellow oil: IR (neat) 3105, 2925, 2854, 1595, 1523, 1456, 1340, 1304, 1051, 962, 918, 833, 735 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 9.09 (d, *J* = 1.8 Hz, 1H), 8.34 (dd, *J* = 7.7, 1.8 Hz, 1H), 7.65 (d, *J* = 7.7 Hz, 1H), 6.46 (dt, *J* = 14.7, 7.4 Hz, 1H), 6.14 (dt, *J* = 14.7, 1.5 Hz, 1H), 2.33 (dq, *J* = 7.4, 1.5 Hz, 2H), 1.56-1.18 (m, 12H), 0.90 (t, *J* = 6.6 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 148.5, 147.4, 144.3, 144.2, 128.3, 126.9, 121.6, 116.5, 33.5, 31.9, 29.3, 29.2(2C), 28.5, 22.7, 14.1; MS *m*/*z* (relative intensity) 338 (M⁺, 73 %), 239 (35), 183 (100), 137 (51), 83 (51); HRMS calcd for C₁₆H₂₂O₄N₂S (M⁺) 338.1300, found 338.1310.

Michael Addition of Gold (Z)-Enethiolate to Cycloalkenone. A Typical Example (Scheme 8, Reaction of 7a with 15b). To a stirred solution of gold (*Z*)-1-decenylthiolate (7a) (17.2 mg, 0.047 mmol) was added triphenylphosphine (24.7 mg, 0.094 mmol) in dichloromethane (3.5 mL) under argon, and stirred at room temperature for 10 min. After addition of lithium iodide (37.6 mg, 0.28 mmol) to the resulting (*Z*)-gold thiolate-PPh₃ complex solution, 2-cyclohexen-1-one (4.5 mg, 0.047 mmol) was added and stirred at room temperature for 24 h. The reaction mixture was washed with water and the aqueous layer was extracted with dichloromethane four times. Combined organic phase was washed with brine, dried over anhydrous Na₂SO₄, filtered, and concentrated under aspiratory vacuum to give an yellow solid, which was purified by column chromatography (dichloromethane-ethyl acetate 9:1) to give 3-[(*Z*)-1-decenylthio]cyclohexanone (**16b**) (9.6 mg, 76%) as a colorless oil;⁵⁴ IR (neat) 2925, 2854, 1715, 1457, 1313, 1220 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.92 (d, *J* = 9.2 Hz, 1H), 5.68 (dt, *J* = 9.2, 7.0 Hz, 1H), 3.21-3.11 (m, 1H), 2.73 (dd, *J* = 14.6, 3.6 Hz, 1H), 2.46-2.25 (m, 3H), 2.22-2.03 (m, 4H), 1.83-1.65 (m, 2H), 1.43-1.18 (m, 12H), 0.88 (t, *J* = 6.6 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 208.6, 132.8, 121.2, 48.2, 44.7, 40.9, 31.9, 31.6, 29.4, 29.3, 29.2(2C), 28.9, 24.1, 22.7, 14.1.

3-[(Z)-1-Decenylthio]cyclopentanone (16a): a colorless oil; IR (neat) 2925, 2854, 1747, 1608,

1457, 1403, 1274, 1248, 1157 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.94 (d, J = 9.2 Hz, 1H), 5.71 (dt, J = 7.0, 9.2 Hz, 1H), 3.61 (quint, J = 7.3 Hz, 1H), 2.62 (dd, J = 18.3, 7.3 Hz, 1H), 2.54-2.19 (m, 3H), 2.19-1.98 (m, 4H), 1.44-1.19 (m, 12H), 0.88 (t, J = 6.6 Hz, 3H); MS *m/z* (relative intensity) 254 (12%, M⁺), 191 (8), 171 (9), 96 (42), 82 (100): HRMS calcd for C₁₅H₂₆OS (M⁺) 254.1704, found 254.1718.

3-[(*Z***)-1-Decenylthio]cycloheptanone (16c**): a colorless oil; IR (neat) 2925, 2854, 1703, 1608, 1456, 1346, 1284, 1192, 935, 889, 723 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.93 (d, *J* = 9.2 Hz, 1H), 5.69 (t, *J* = 9.2, 7.3 Hz, 1H), 3.11 (tt, *J* = 9.2, 2.9 Hz, 1H), 2.87-2.73 (m, 2H), 2.61-2.44 (m, 2H), 2.20-2.07 (m, 3H), 2.03-1.63 (m, 4H), 1.60-1.49 (m, 1H), 1.42-1.20 (m, 12H), 0.88 (t, *J* = 7.3 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 211.6, 132.8, 122.0, 50.2, 44.0, 43.1, 37.4, 31.9, 29.4, 29.3, 29.2, 29.2, 29.0, 28.1, 23.9, 22.7, 14.1; MS *m/z* (relative intensity): 282 (13 %, M⁺), 171 (26), 138 (30), 111 (43), 96 (25), 83 (100), 81 (60): HRMS calcd for C₁₇H₃₀OS (M⁺) 282.2017, found 282.2057.

Job Plot. Equimolar solutions (0.17 M) of (*Z*)-gold thiolate **7a** and triphenylphosphine were prepared and mixed in various amount. ¹³C NMR spectra of the mixture were recorded at room temperature, and the complexation induced chemical shifts of an *ipso* carbon of triphenylphosphine were analyzed by the method developed by Newcomb and co-workers.⁸⁷

¹³C NMR Titration. A solution of triphenylphosphine (0.178 M) in CDCl₃ was prepared. Nine NMR tunes were each filled with the solution and with an adequate amount of (*Z*)-gold thiolate **7a** (0-2.2 equiv). ¹³C NMR spectra of the mixtures were recorded at room temperature. The curve-fittings of the chemical shift data of an *ipso* carbon of triphenylphosphine were carried out by a nonlinear least-squares method (Marquardt–Levenberg Algorithm) according to the reported equation^{S8} with use of SigmaPlot (Jandel Scientific, San Rafael, CA).

2. ESI-MS spectrum observed for gold (Z)-enethiolate 7a



Figure S1. ESI-MS spectrum observed for gold (Z)-enethiolate 7a in CH_2Cl_2 -MeOH (1:1). No monomeric gold enethiolate 7a species $[M-Au]^+$ was detected under the conditions.

3. Job plot for complexation of gole (Z)-enethiolate 7a with triphenylphosphine



 δ : ¹³C NMR chemcal shit of *ipso* carbon of PPh₃ X_{PPh3}: mol fraction of PPh₃

Figure S2. Job plot for complexation between gole (Z)-enethiolate **7a** and triphenylphosphine in CDCl₃ at rt.

4. Observed ¹³C NMR chemical shifts of PPh₃ (0.178 M) when titrated with 7a in CDCl₃ at rt



Figure S3. Observed ¹³C NMR chemical shifts of PPh₃ (0.178 M) when titrated with 7a in CDCl₃ at rt.

5. Isomerization of 8a in CDCl₃ at 23 °C in the dark.



Figure S4. Effect of additives on the configurational stability of gold (Z)-enethiolate-PPh₃ complex **8a** in CDCl₃ (0.01 M) at 23 °C in the dark. The filled blue circle means under argon. The filled pink circle means under air. The filled yellow circle means in the presence of TEMPO (0.1 equiv) under Ar.

References

- S1 M. Ochiai, S. Yamamoto, S. K. Sato, *Chem. Commun.* 1999, 1363.
- S2 R. Uson, A. Laguna, Organomet. Synth. 1986, 3, 324.
- S3 M. Mayr, K. Wurst, K. K.-H. Ongania, M. Buchmeiser, *Chem. Eur. J.* 2004, *10*, 1256.
- S4 M. Ochiai, M. Hirobe, K. Miyamoto, J. Am. Chem. Soc. 2006, 128, 9046.
- S5 K. Oshima, K, Shimoji, H. Takahashi, H. Yamamoto, H. Nozaki, J. Am. Chem. Soc. 1973, 95, 2694.
- S6 M. A. Khan, J. Chem. Res. Synopses, **1991**, 163.
- S7 M. T. Blanda, J. T. Horner, M. Newcomb, J. Org. Chem. 1989, 54, 4626.
- S8 N. Funasaki, S. Ishikawa, S, Neya, Bull. Chem. Soc. Jpn. 2002, 75, 719.









