

Comparison of inclusion properties between *p-tert*-butylcalix[4]arene and
p-tert-butylthiacalix[4]arene towards primary alcohols in crystals

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Supporting Information

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I. Powder X-ray diffraction (PXRD) studies

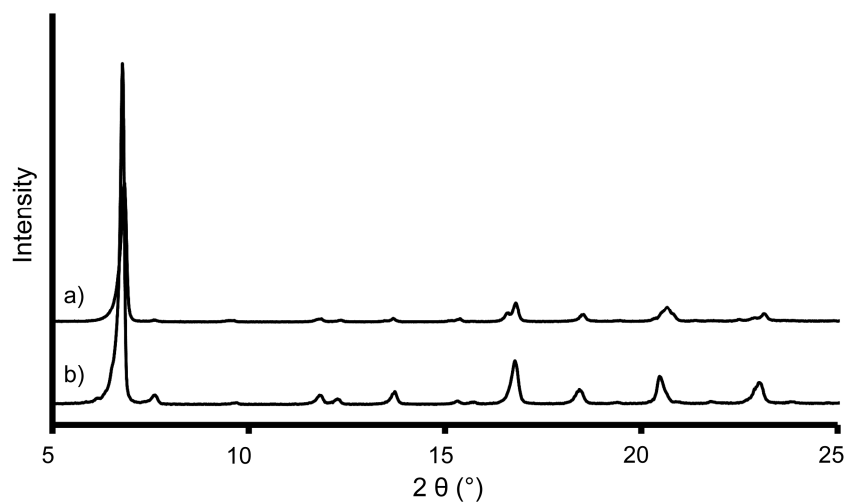


Figure S1. PXRD patterns of the inclusion crystals of compound **1** with PentOH (a) and HexOH (b) obtained by crystallization.

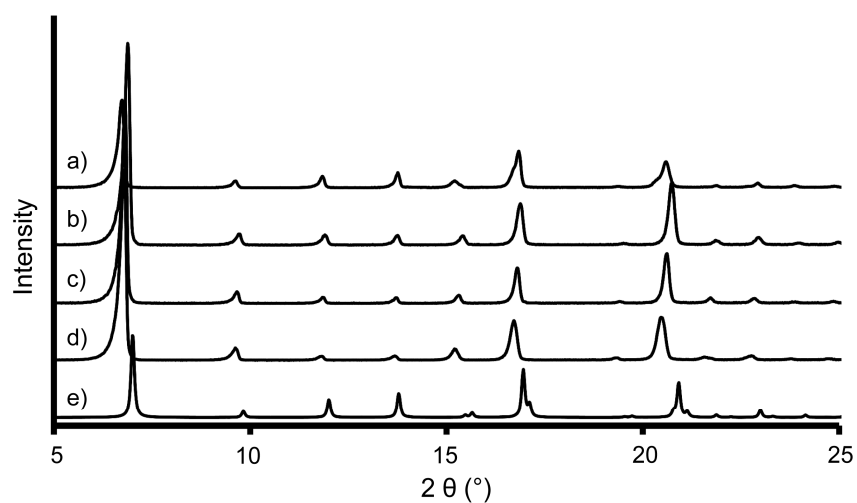


Figure S2. PXRD patterns of the inclusion crystals of compound **1** with MeOH (a), EtOH (b), PrOH (c) and BuOH (d) obtained by the suspension method, and a simulation from XRD data of **1**·EtOH measured at 100 K (e).

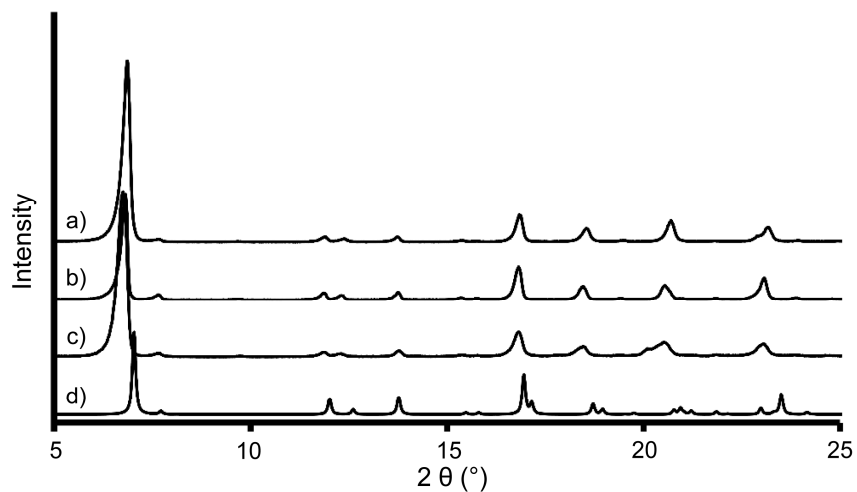


Figure S3. PXRD patterns of the inclusion crystals of compound **1** with PentOH (a), HexOH (b) and HeptOH (c) obtained by the suspension method, and a simulation from XRD data of $1_2 \cdot \text{PentOH}$ measured at 100 K (d).

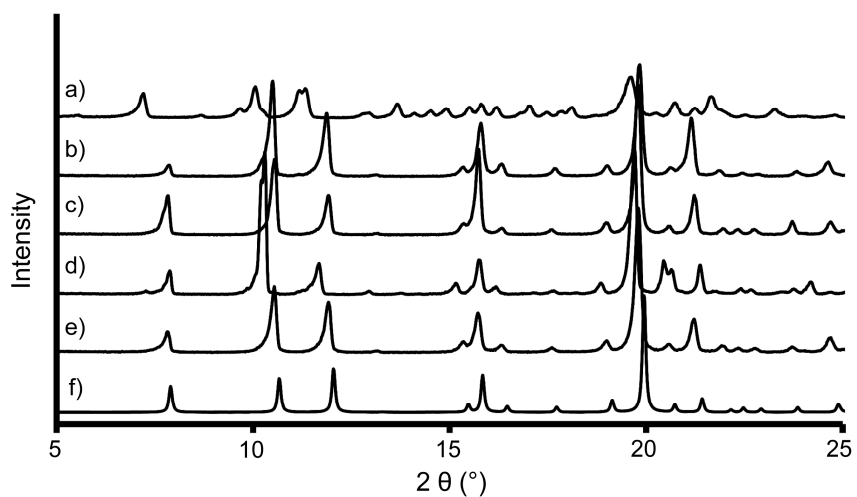


Figure S4. PXRD patterns of a crystalline powder of compound **2** (a) and its inclusion crystals with MeOH (b), EtOH (c) and PrOH (d) obtained by crystallization and with EtOH obtained by the suspension method (e), and a simulation from XRD data of $2 \cdot \text{EtOH}$ measured at 223 K (f).

II. Single crystal X-ray diffraction (XRD) studies for inclusion crystals of compound 1

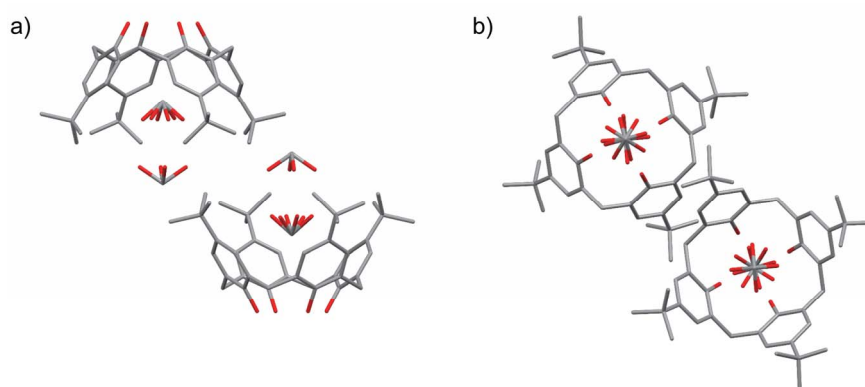


Figure S5. X-ray structure of 1·MeOH: Side view (a) and top view (b). Hydrogen atoms are omitted for clarity.

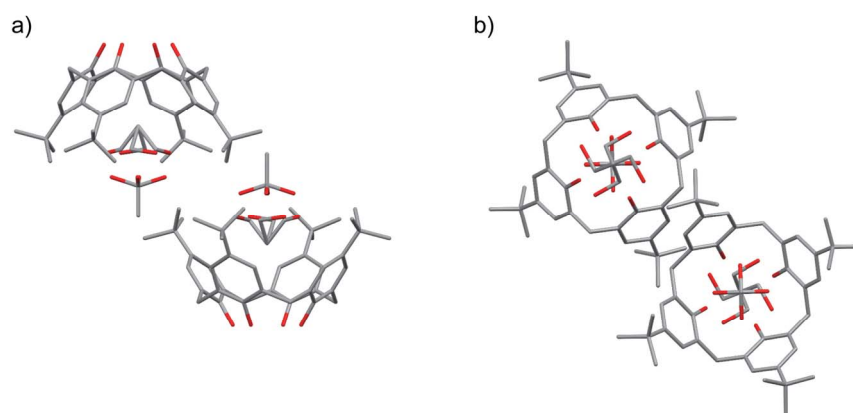


Figure S6. X-ray structure of 1·EtOH: Side view (a) and top view (b). Hydrogen atoms are omitted for clarity.

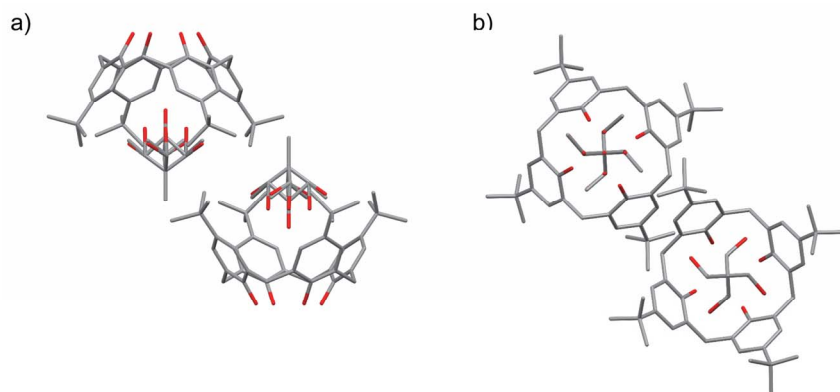


Figure S7. X-ray structure of **1·PrOH**: Side view (a) and top view (b). Hydrogen atoms are omitted for clarity.

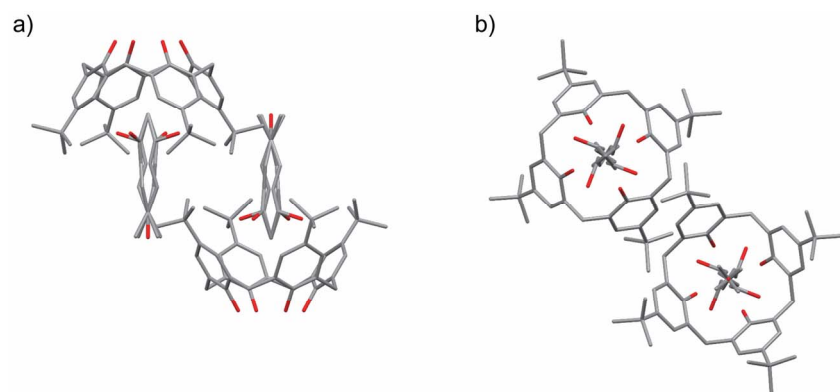


Figure S8. X-ray structure of **1·PentOH**: Side view (a) and top view (b). Hydrogen atoms are omitted for clarity.

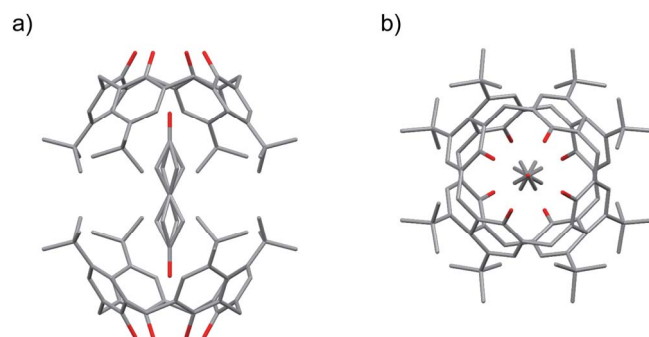


Figure S9. X-ray structure of **12-PentOH**: Side view (a) and top view (b). Hydrogen atoms are omitted for clarity.

III. Single crystal X-ray diffraction (XRD) studies for inclusion crystals of compound 2

The structures of inclusion crystals **2**·MeOH, **2**·EtOH and **2**·PrOH reported in a preliminary communication¹ were reanalyzed under the restraint of disordered *tert*-butyl groups and guest molecules. Crystallographic data have been deposited with Cambridge Crystallographic Data Centre: Deposition number CCDC 1050720–1050722.

Data for 2·MeOH. C₄₁H₅₂O₅S₄, fw = 753.07, tetragonal, *P4/nmm*, $a = 15.7827(7)$ Å, $b = 15.7827(7)$ Å, $c = 8.2365(7)$ Å, $V = 2051.7(2)$ Å³, $Z = 2$, $T = 223(2)$ K, 11188 reflections measured, 1332 independent reflections, 1190 reflections were observed ($I > 2\sigma(I)$), $R_1 = 0.0955$, $wR_2 = 0.2366$ (observed), $R_1 = 0.1011$, $wR_2 = 0.2559$ (all data).

Data for 2·EtOH. C₄₂H₅₄O₅S₄, fw = 767.09, tetragonal, *P4/nmm*, $a = 15.8115(11)$ Å, $b = 15.8115(11)$ Å, $c = 8.2875(12)$ Å, $V = 2071.9(4)$ Å³, $Z = 2$, $T = 223(2)$ K, 11205 reflections measured, 1341 independent reflections, 1107 reflections were observed ($I > 2\sigma(I)$), $R_1 = 0.0864$, $wR_2 = 0.2301$ (observed), $R_1 = 0.0962$, $wR_2 = 0.2557$ (all data).

Data for 2·PrOH. C₄₃H₅₆O₅S₄, fw = 781.12, tetragonal, *P4/nmm*, $a = 15.7986(9)$ Å, $b = 15.7986(9)$ Å, $c = 8.5474(10)$ Å, $V = 2133.4(3)$ Å³, $Z = 2$, $T = 223(2)$ K, 11373 reflections measured, 1378 independent reflections, 1174 reflections were observed ($I > 2\sigma(I)$), $R_1 = 0.0980$, $wR_2 = 0.2420$ (observed), $R_1 = 0.1063$, $wR_2 = 0.2630$ (all data).

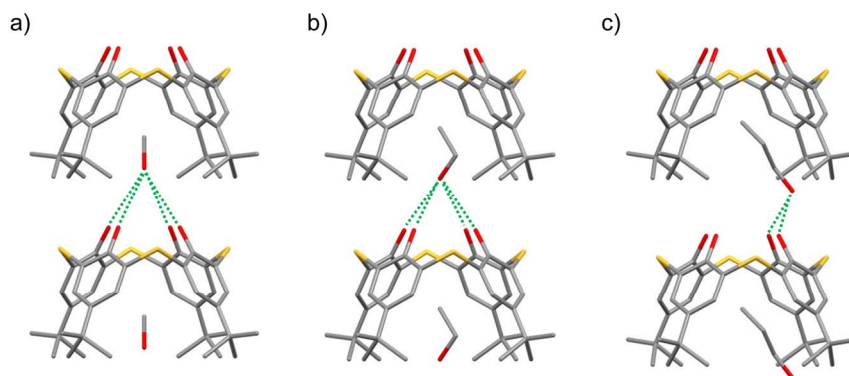


Figure S10. X-ray structures of **2·MeOH** (a), **2·EtOH** (b) and **2·PrOH** (c), showing partial packing structures along the *c*-axis. Hydrogen atoms and disordered atoms are omitted for clarity. Green dotted lines represent intermolecular hydrogen bonds; the O···O distances are 3.480 Å (a), 3.122 Å (b) and 2.688 Å (c), respectively. A CH- π interaction is observed between the terminal methyl group of an alcohol molecule and the benzene rings of the host molecule including the alcohol molecule in its cavity; the C···Ar distances are 3.842 Å (a), 3.685 Å (b) and 3.651 Å (c), respectively.

IV. Reference

- [1] N. Morohashi, S. Noji, H. Nakayama, Y. Kudo, S. Tanaka, C. Kabuto and T. Hattori, *Org. Lett.*, 2011, **13**, 3292.