

Supporting information.

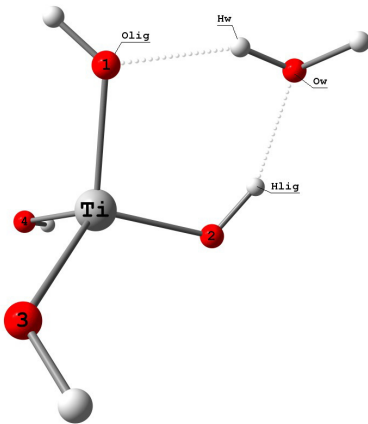
*To whom correspondence should be addressed

1 Water splitting by $\text{Ti}(\text{OH})_4$ in the ground state

To investigate if the reaction $\text{Ti}(\text{OH})_4 + \text{H}_2\text{O} \xrightarrow{h\nu} \text{Ti}(\text{OH})_3\text{HOH} \cdots \text{OH}^\bullet$ can proceed in the ground state, we begin by studying the ground state energy profiles.

The energy profiles along the $\text{H}^W\text{-O}^{lig}$ distance (see labels in Fig. S1) are shown in Fig. S2. The ground state PESs feature moderate activation energies of 11-16 kcal/mol. The calculated transition state barriers are 11 (BP86), 16 (B3LYP) and 14 (CCSD(T)) kcal/mol. Although the barriers are low, the reactions do not lead to the formation of a hydroxyl radical, instead, during the shortening of $\text{O}^{lig}\text{-H}^W$ distance, an H^{lig} atom from a neighbouring OH ligand moves to the water oxygen (O^W). This amounts to a relay of an H atom between the ligands of the $\text{Ti}(\text{OH})_4$ molecule resulting in an oxo-titanium compound $\text{TiO}(\text{OH})_2\text{H}_2\text{O}$ (see Fig. S3).

Fig. S1: Structure and the labelling of the atoms of the reactant complex $\text{Ti}(\text{OH})_4 \cdots \text{H}_2\text{O}$.

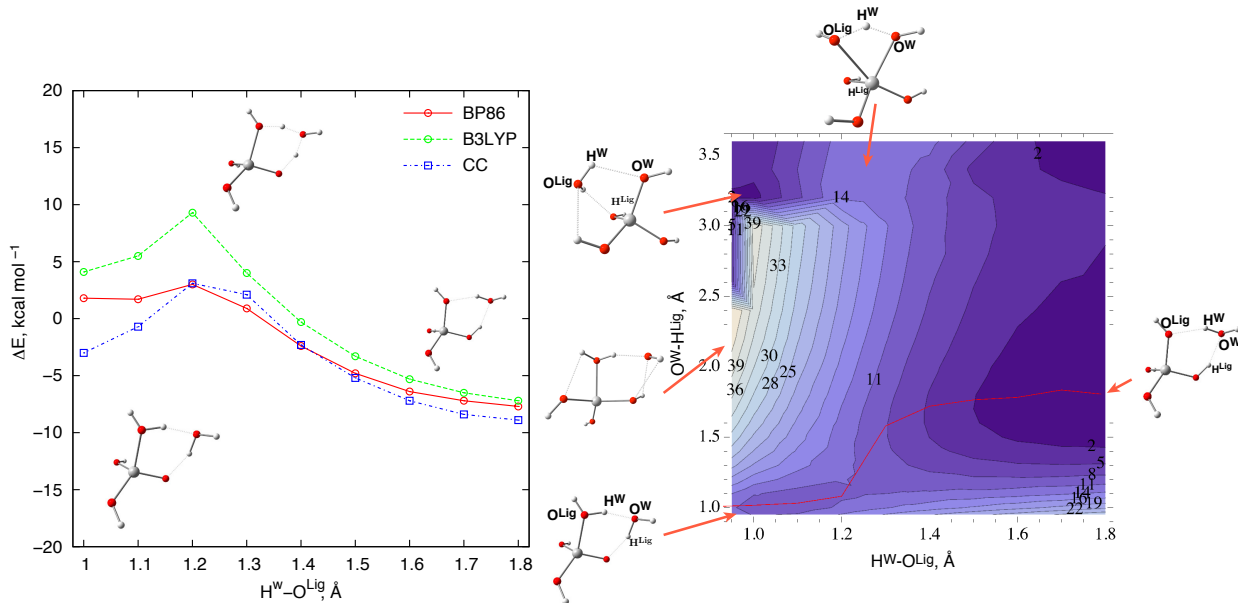


To prevent a spontaneous transfer of H^{lig} to the OH remnant of water and investigate the possibility of OH^\bullet radical formation in the ground state we have performed a two-dimensional PES scan whereby $\text{H}^W\text{-O}^{lig}$ and $\text{O}^W\text{-H}^{lig}$ coordinates have been fixed, while all the other coordinates have been optimised. The obtained potential energy surface (PES) is shown in Fig. S2 (right). The red curve on the plot shows the path taken in the one-dimensional energy scan along $\text{H}^W\text{-O}^{lig}$.

Fig. S2: Ground state PES (in kcal/mol).

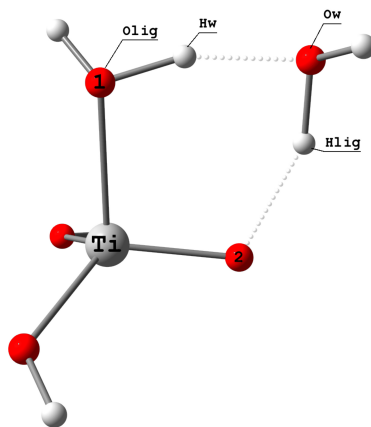
Left: energy profiles along the H^W-O^{Lig} distance relative to free $Ti(OH)_4$ and H_2O . The ground state reactions do not correspond to the process $Ti(OH)_4 + H_2O \xrightarrow{h\nu} Ti(OH)_3HOH \cdots OH^\bullet$, but rather to a relay of an H-atom from one OH ligand to the other via the water molecule, which results in an oxo-titanium species (see inset). Single point CCSD(T) calculations on the BP86 geometries are shown in the plot ('CC').

Right: 2D PES along the H^W-O^{Lig} and O^W-H^{Lig} coordinates. The two coordinates have been kept fixed, whereas all other coordinates have been optimised. The energies are with respect to the reactant complex energy. The red curve represents the projection of the 1D PES (BP86 in the left panel). The OH^\bullet radical formation is energetically prohibited (~ 40 kcal/mol).



At longer O^W-H^{Lig} (≥ 2.5 Å) the incident water coordinates to the Ti atom and the shortening of H^W-O^{Lig} leads to the transfer of H-atom from the coordinated water to the OH-ligand, whereby the nascent water ligand decoordinates from the titanium site. This results in a structure similar to the reactant complex at $H^W-O^{Lig} \sim 1.0$ Å, $O^W-H^{Lig} \sim 3.2$ Å. As can be seen that at O^W-H^{Lig} distances between 1.5 and 2.4 Å the energy rises to 40 kcal/mol. Only this region corresponds to the formation of an OH^\bullet radical. Therefore, in the ground state the OH^\bullet radical formation is energetically not feasible.

Fig. S3: Structure of $\text{TiO}(\text{OH})_2\text{H}_2\text{O}\cdots\text{H}_2\text{O}$ complex obtained upon shortening of $\text{O}^{lig}-\text{H}^w$ coordinate in the ground state.

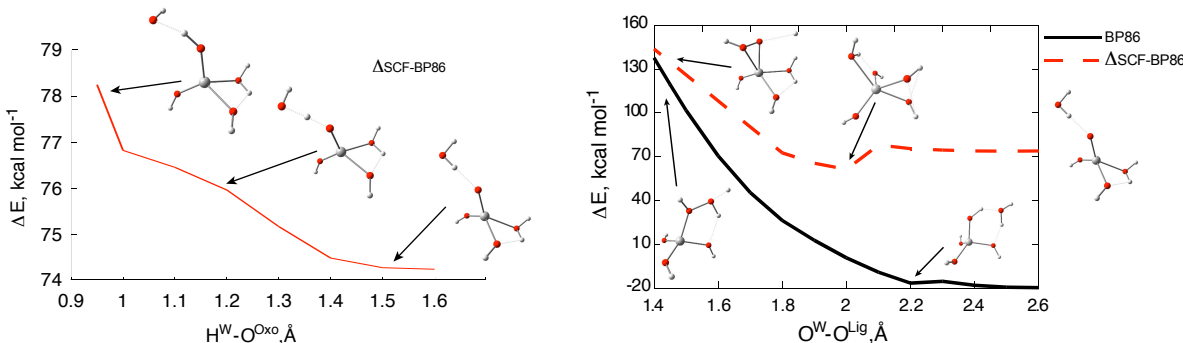


1.1 Reactivity of the oxo-titanium species $\text{TiO}(\text{OH})_2\text{H}_2\text{O}$.

We follow the same lines as in the $\text{Ti}(\text{OH})_4+\text{H}_2\text{O}$ study and first study the H-abstraction by the oxo-moiety of the oxo-titanium. The excited state PES ($\Delta\text{SCF-BP86}$) along the $\text{H}^W-\text{O}^{Oxo}$ distance is shown in Fig. S4 (left). There is no stable minimum on the curve, however, the energy penalty for the water OH bond breaking is small ~ 4 kcal/mol.

Fig. S4: Left panel: Excited state ($\Delta\text{SCF-BP86}$) energy profile for the water H-abstraction reaction by the $\text{TiO}(\text{OH})_2\text{H}_2\text{O}$.

Right panel: Ground (BP86) and excited ($\Delta\text{SCF-BP86}$) state energy profiles for the O-O bond formation reaction by the $\text{TiO}(\text{OH})_2\text{H}_2\text{O}$.



Additionally, we studied the nucleophilic attack of water on the oxo moiety. The energy curves of the $\text{O}^W-\text{O}^{Lig}$ bond formation reaction are shown in Fig. S4 (right). The ground state scan leads very quickly to the reproduction of the $\text{Ti}(\text{OH})_4+\text{H}_2\text{O}$ system and then, naturally, follows the same repulsive potential that was observed in the reaction with $\text{Ti}(\text{OH})_4$ (see main text). On the excited ($\Delta\text{SCF-BP}$) state a small barrier is surmounted after which the water molecule coordinates to Ti. Subsequent shortening of the O-O bond is energetically prohibited (≥ 60 kcal/mol).

Hereby we have found that the oxo-titanium species is not superior to the $\text{Ti}(\text{OH})_4$ system, but has similar traits. Namely, it is also capable of OH bond breaking in the excited state and is inactive in the ground state. Likewise, $\text{TiO}(\text{OH})_2\text{H}_2\text{O}$ is inert towards a nucleophilic attack of water on its oxo-moiety.

To conclude, the reaction $\text{Ti}(\text{OH})_4 + \text{H}_2\text{O} \xrightarrow{h\nu} \text{Ti}(\text{OH})_3\text{HOH} \cdots \text{OH}^\bullet$ can proceed either

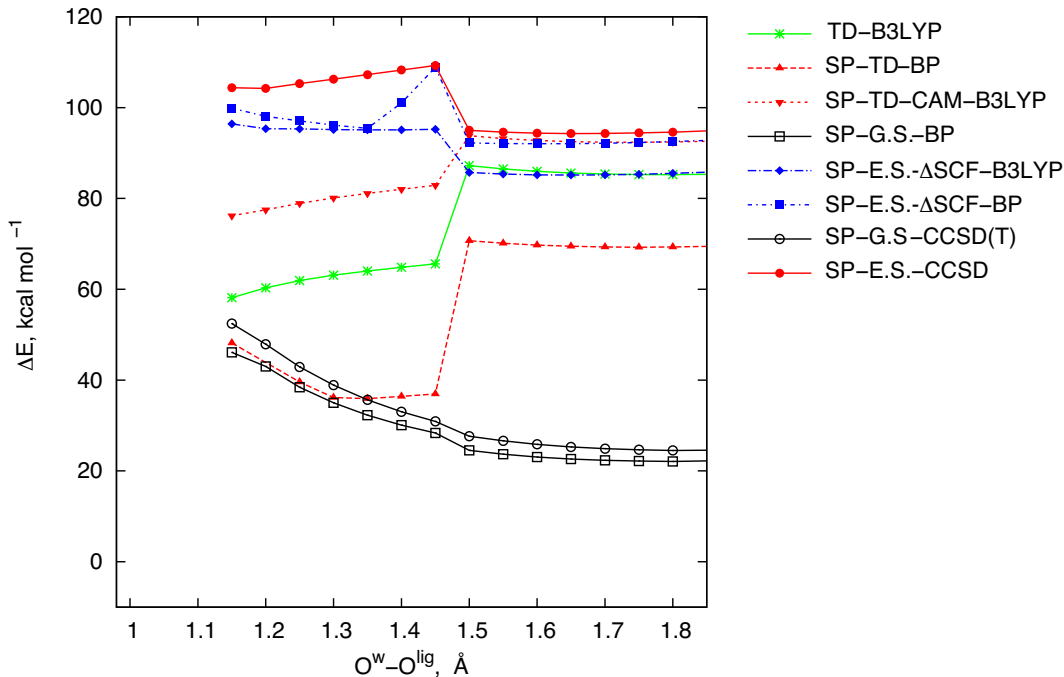
directly or via a precursor oxo-titanium species with comparable activation barriers. In each case the excitation is crucial for the generation of an OH^\bullet radical.

2 TDDFT treatment of the water H-abstraction reaction by $\text{Ti}(\text{OH})_4$.

Herein we investigate the excited state reaction of $\text{Ti}(\text{OH})_4$ with water using the TDDFT method for geometry optimisations. Energy profiles along a one-dimensional ($\text{H}^W - \text{O}^{Lig}$) relaxed TD-B3LYP scan are shown in Fig. S5. The single point energies using TD-DFT and ΔSCF with different functionals together with the excited CCSD and ground state CCSD(T) results have been plotted. The excited (S_1) state CCSD results were obtained by adding the CCSD(T) ground state energies and the CCSD excitation energies. It is evident that in the Franck-Condon region the TDDFT and ΔSCF curves are parallel to the CCSD ones. TDDFT performs better with hybrid functionals in agreement with our benchmark calculations in the main text and with the recent work of Berardo et al.^[1,2] Using the range-separated CAM-B3LYP brings TDDFT results very close to the CCSD values. At $\text{H}^W - \text{O}^{Lig} = 1.45 \text{ \AA}$ there is an abrupt drop in TDDFT curves. This is reflected in large geometric transformations, namely the ligand OH that initially donated an H-bond to water turns $\sim 180^\circ$ and now accepts an H-bond from the rotated water. After the drop water bridges two OH ligands. Although it is obvious that this sudden drop is due to the unfortunate reaction coordinate, it should be noted that the energy profiles from CCSD and ΔSCF methods do not show a lowering of energy, but a raise.

To make the interpretation clearer one should consider a more smooth reaction path, that does not involve such strong geometric transformations. To this end we fix the distance between the water oxygen and the OH ligand hydrogen (the one that will turn 180 degrees in the one-dimensional scan). The two dimensional relaxed scan along the $\text{H}^W - \text{O}^{Lig}$ and $\text{O}^W - \text{H}^{Lig}$ (Fig. S6) has been performed in the lowest singlet excited state using the TD-B3LYP/TZ2P approach. In contrast to the ΔSCF results (see main text) the TDDFT optimisation spontaneously leads to the isolation of the OH radical. The red curve on the 2D plot in Fig. S6 represents the path along the 1D- $\text{H}^W - \text{O}^{Lig}$ -scan (Fig. S5). The considerable

Fig. S5: Relaxed excited state TD-B3LYP/TZ2P scan along the H^W-O^{Lig} distance. Single point calculations using the TD-B3LYP optimised geometries are plotted.



drop on the one-dimensional plot is reflected in the turn of the red path towards large O^W-H^{Lig} values. Since the left part of this plot corresponds to the OH radical formation, we examine the cut at $O^W-H^{Lig} = 2.0 \text{ \AA}$, which gives a representative reaction coordinate without abrupt features.

The 1D plot along H^W-O^{Lig} with O^W-H^{Lig} distance fixed to 2.0 \AA is shown in Fig. S7. It is evident that before $H^W-O^{Lig} = 1.4 \text{ \AA}$ all the excited state curves, except TD-BP, are in a reasonable agreement and the best match with the CCSD results is seen with the TD-CAM-B3LYP and Δ SCF-BP results. However beyond the mentioned point the TDDFT results diverge. This behaviour is an indication of the artefact of the method. Notably, the Δ SCF-BP results follow the same trend as the CCSD ones even at shorter distances (when one water OH bond is broken and the OH radical is formed).

To investigate the source of the breakdown we have traced back the HOMO-LUMO gaps along the reaction coordinate. The values of the gaps together with the excitation

Fig. S6: Relaxed 2D excited state TD-B3LYP/TZ2P scan along the $H^W - O^{Lig}$ and $O^W - H^{Lig}$ distances. The energies are in kcal mol^{-1} , relative to the lowest energy point.

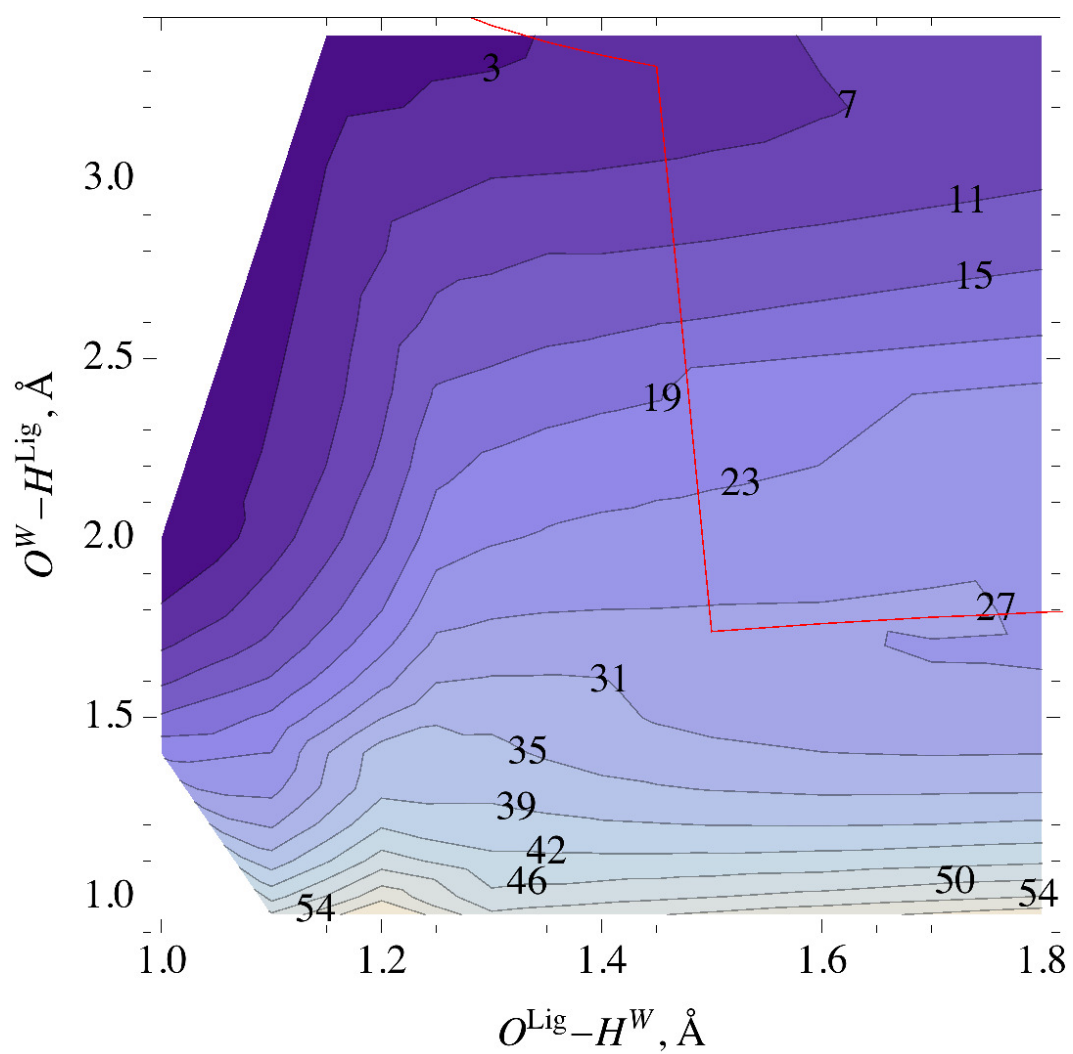
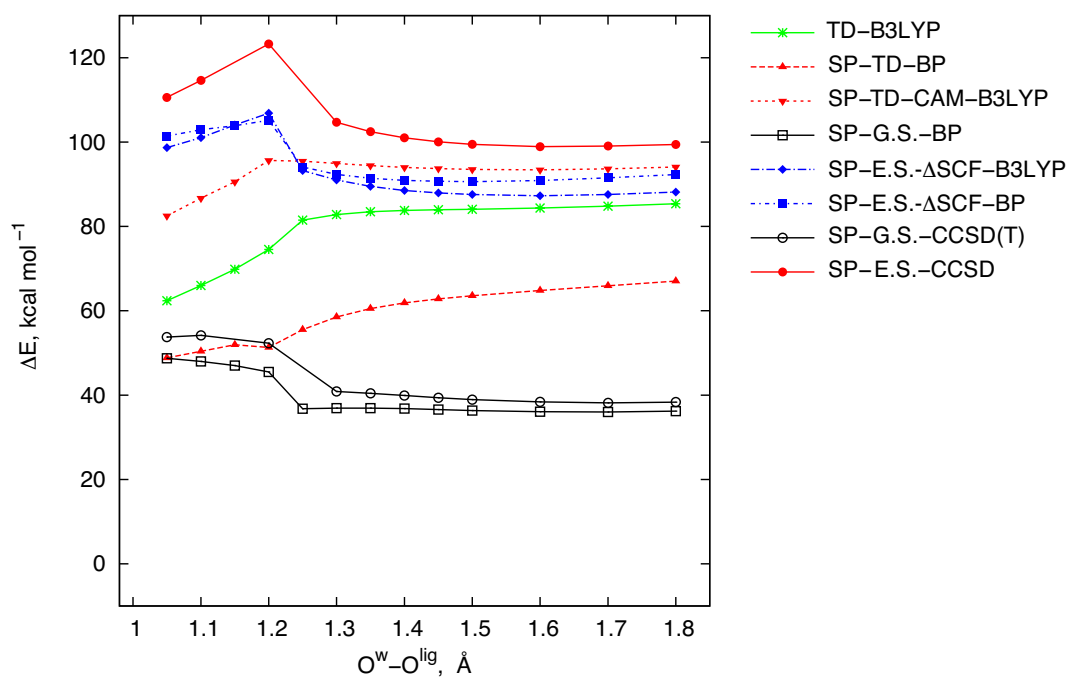


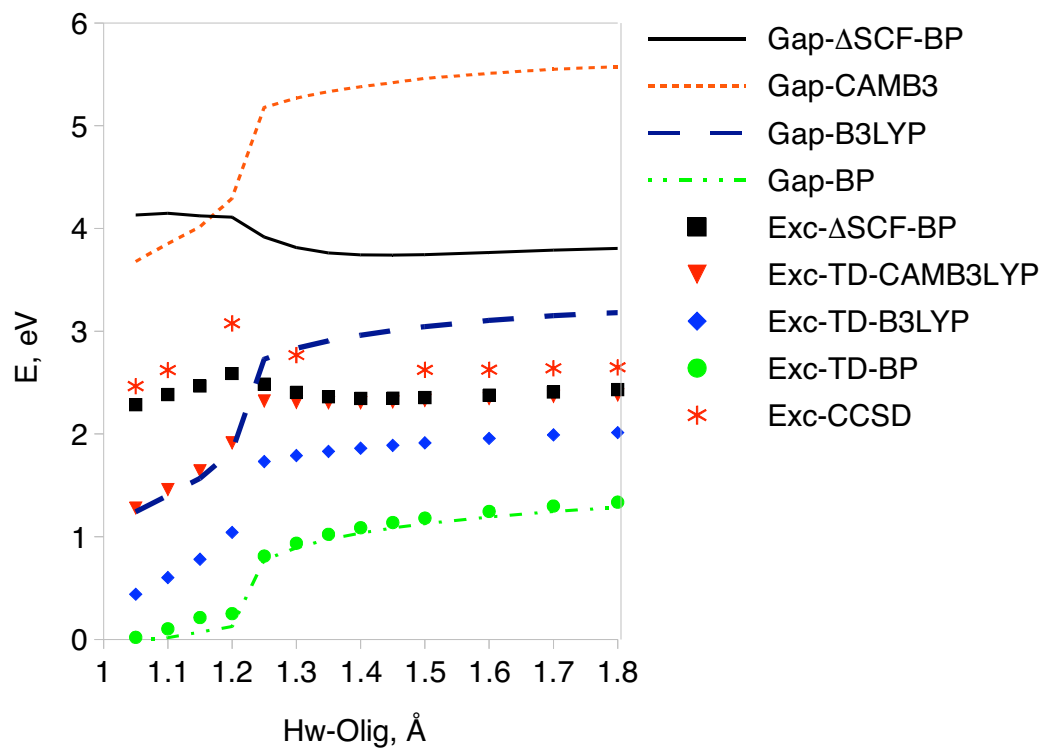
Fig. S7: Relaxed excited state TD-B3LYP/TZ2P scan along the $H^W - O^{Lig}$ distance with the $O^W - H^{Lig}$ distance fixed to 2.0 Å. Single point calculations using the TD-B3LYP optimised geometries are plotted.



energies are depicted in Fig. S8. It is evident that the TDDFT excitation energies follow the corresponding closed shell reference DFT gaps. At $H^W-O^{Lig} \leq 1.4 \text{ \AA}$ the gaps get smaller, until at a given point they tend to diminish. The CAM-B3LYP method has the largest gap and hence its excitation energies survive somewhat longer. But this does not salvage the method at too short distances.

We therefore prove that the TDDFT study of the water H-abstraction reaction by $Ti(OH)_4$ must be confined to the Franck-Condon region and is not generally applicable for this investigation. On the other hand, we have presented another strong evidence that the Δ SCF-BP approach gives very reasonable results.

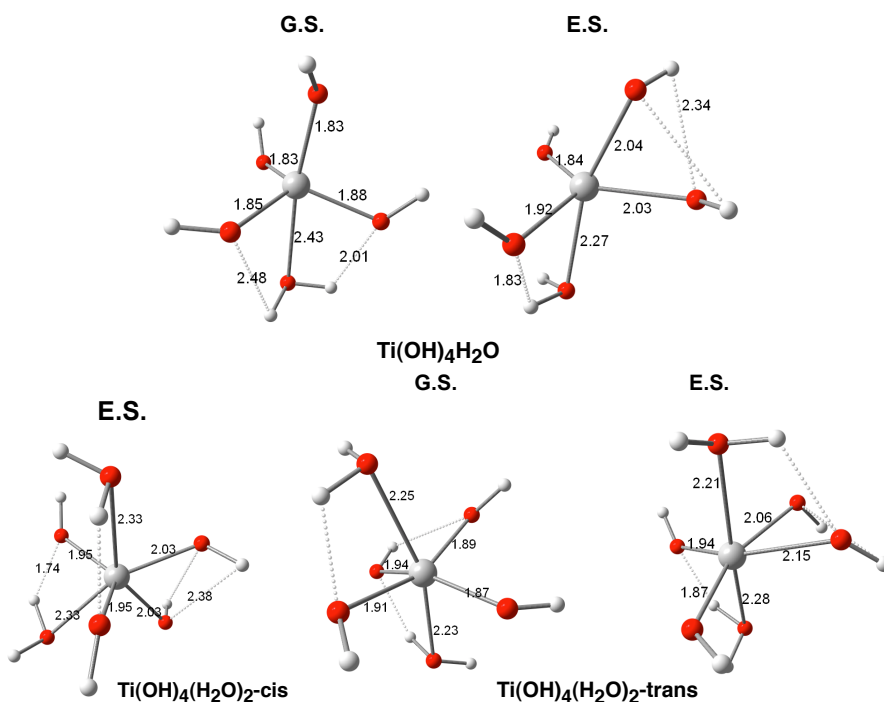
Fig. S8: HOMO-LUMO band gaps and the excitation energies at the geometries obtained by the TD-B3LYP relaxed scan along the H^W-O^{Lig} distance with the O^W-H^{Lig} distance fixed to 2.0 Å.



3 Stability and reactivity of $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{0,1,2}$ systems

The titanium atom in $\text{Ti}(\text{OH})_4$ is undercoordinated and in an aqueous solution can possibly afford one or two water molecules in its shell. The optimised structures of $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{1,2}$ are shown in Fig. S9. The corresponding energies relative to the free ground state $\text{Ti}(\text{OH})_4$ and water molecules are given in Tab. S1

Fig. S9: Structures of the coordinatively saturated $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{1,2}$ compounds.



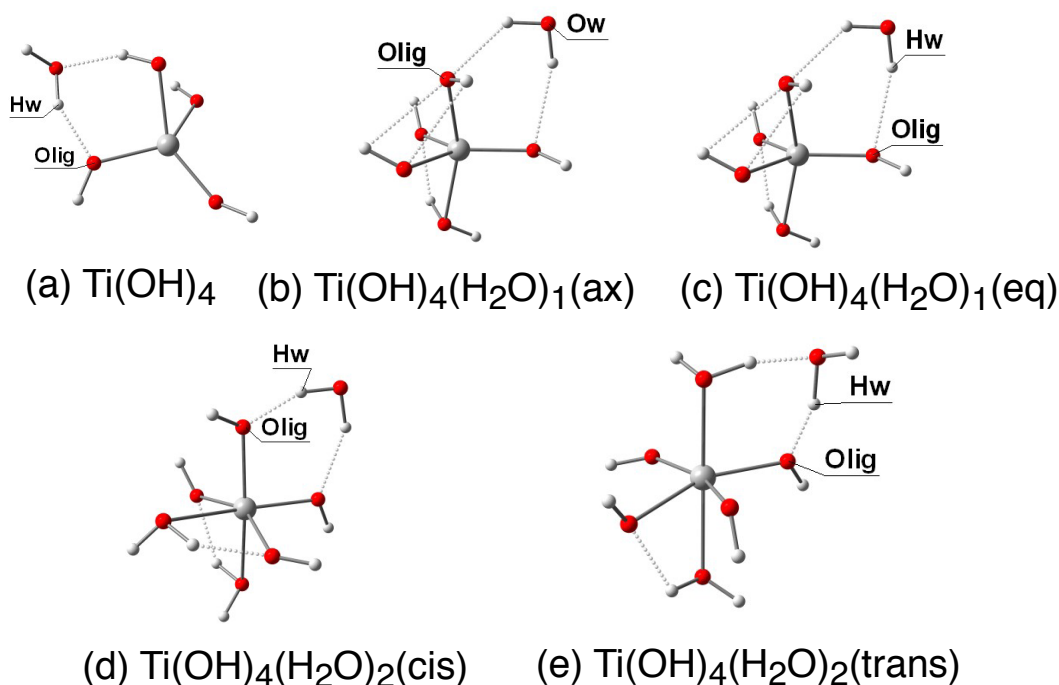
Tab. S1

Ground state			
	ΔE	ΔH	ΔG
$\text{Ti}(\text{OH})_4$	0.0	0.0	0.0
$\text{Ti}(\text{OH})_4\text{H}_2\text{O}$	-3.8	-2.1	8.4
$\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{2\text{-cis}}$	—	—	—
$\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{2\text{-trans}}$	4.7	7.9	29.4
Excited state			
	ΔE	ΔH	ΔG
$\text{Ti}(\text{OH})_4$	92.1	92.4	93.4
$\text{Ti}(\text{OH})_4\text{H}_2\text{O}$	80.4	82.5	93.8
$\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{2\text{-cis}}$	75.7	79.5	102.0
$\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{2\text{-trans}}$	74.6	78.2	98.1

In the ground state the $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{1,2}$ compounds are within 5 kcal/mol from $\text{Ti}(\text{OH})_4$. However, their free energies are considerably higher ($\sim 8\text{-}29$ kcal/mol). The major contribution comes from the entropic effect, which renders the $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{1,2}$ molecules unstable.

In the excited state, all $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{1,2}$ conformers are significantly lower in energy ($\sim 12\text{-}18$ kcal/mol) than $\text{Ti}(\text{OH})_4$. The entropic effect, however, raises their energies above $\text{Ti}(\text{OH})_4$. The single water-coordinated $\text{Ti}(\text{OH})_4\text{H}_2\text{O}$ becomes energetically accessible ($\Delta\Delta G=0.4$ kcal/mol), whereas the doubly-coordinated $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_2$ are less favored, being 5 (trans) and 9 (trans) kcal/mol higher than $\text{Ti}(\text{OH})_4$. The possibility of the additional coordination in the excited state can be explained by the geometry of $\text{Ti}(\text{OH})_4$, which features wider openings of the $\angle\text{OTiO}$ angles ($>109^\circ$) in the excited state.

Fig. S10: Structures of compounds $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_n \cdots \text{H}_2\text{O}$, $n=0$ (a); $n=1$ (b, c) and $n=2$ (d, e). The labels indicate the atoms, the distance between which is varied in the H-abstraction reactions.



From the above analysis it is seen that $\text{Ti}(\text{OH})_4$ is stable in the ground state, whereas in the excited state it can attach an additional water molecule. The reactivity of $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{1,2}$

Fig. S11: Relaxed potential energy scans along the H_w-O_{lig} coordinate. The energies are with respect to the reactant complex (RC). The corresponding structures are depicted in Fig. S5

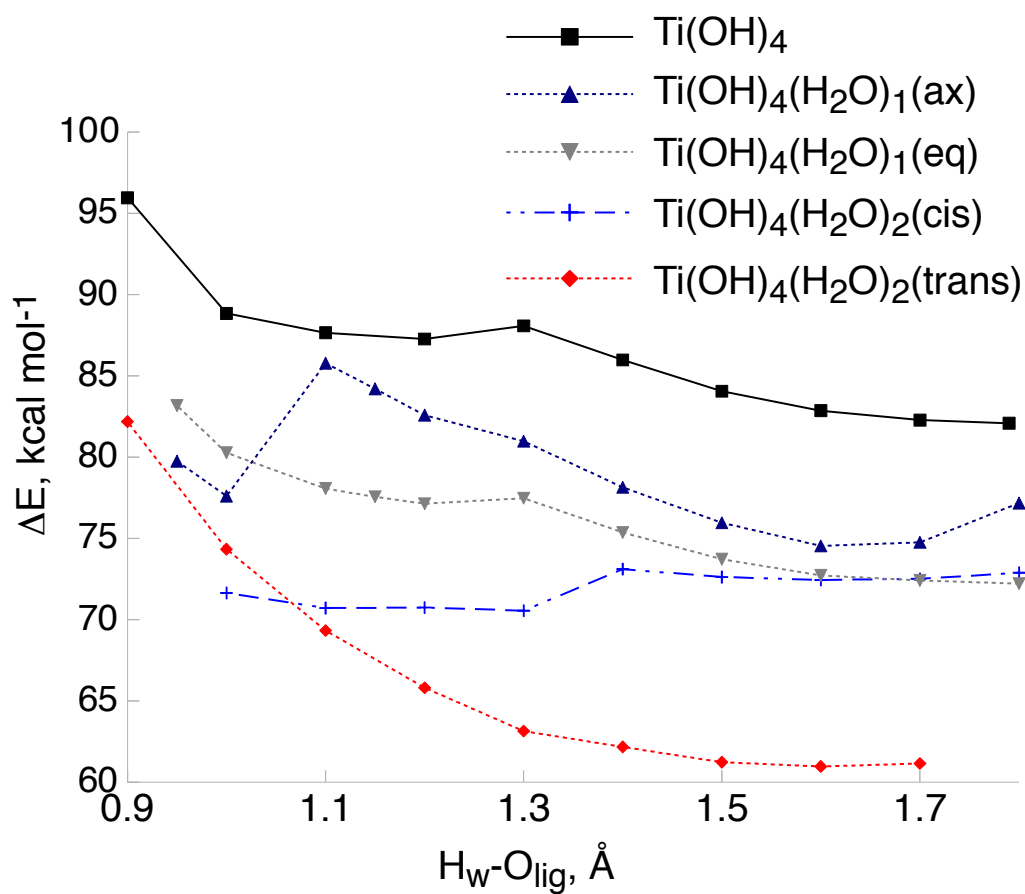
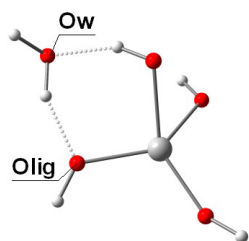
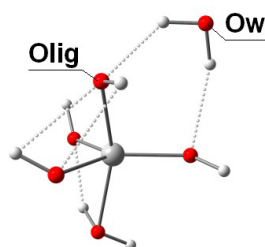


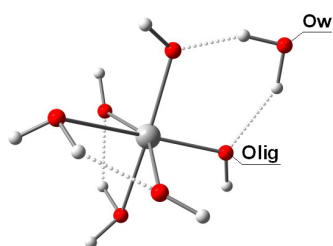
Fig. S12: Structures of compounds $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_n \cdots \text{H}_2\text{O}$, $n=0$ (a); $n=1$ (b) and $n=2$ (c, d). The labels indicate the oxygen atoms, the distance between which is shortened in the O-O bond formation reactions.



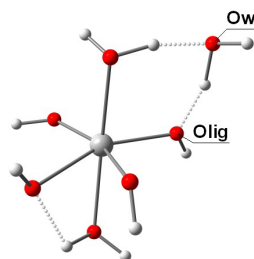
(a) $\text{Ti}(\text{OH})_4$



(b) $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_1(\text{ax})$

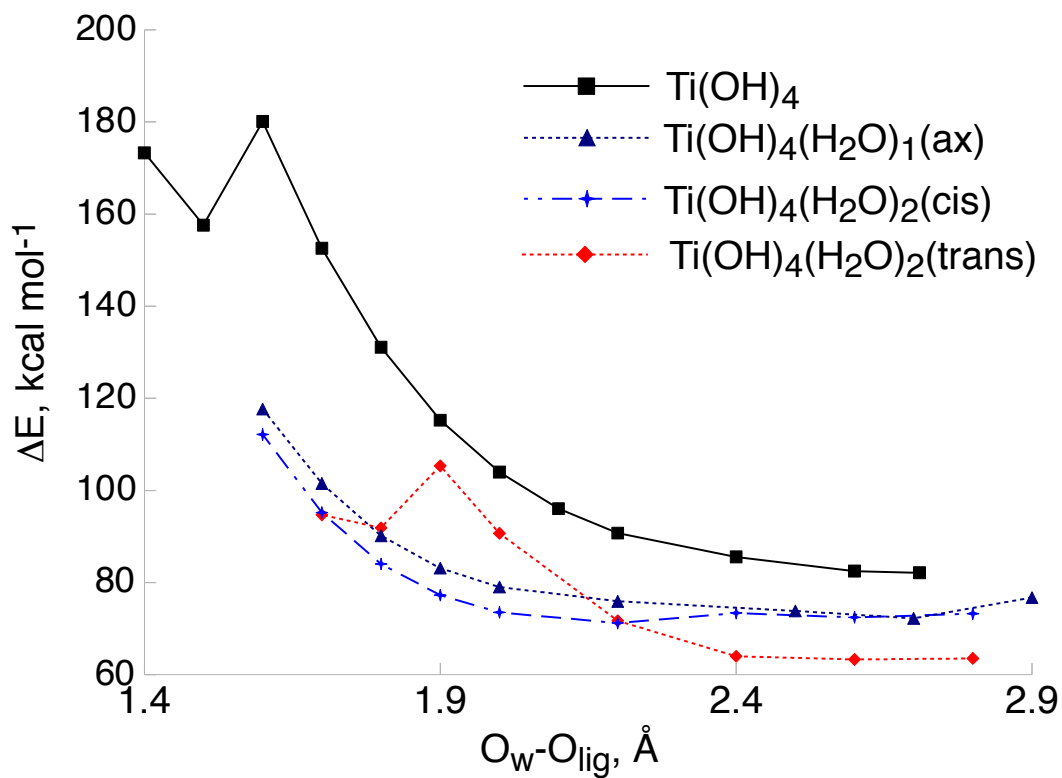


(c) $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_2(\text{cis})$



(d) $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_2(\text{trans})$

Fig. S13: Relaxed potential energy scans along the O_W - O_{lig} coordinate. The energies are with respect to the reactant complex (RC). The corresponding structures are depicted in Fig. S11



with water was found to be similar to that of $\text{Ti}(\text{OH})_4$ (see the SI for details of this study), therefore in the next sections we concentrate on the reactions driven by $\text{Ti}(\text{OH})_4$.

3.1 Water splitting and O-O bond formation reactions by $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{0,1,2}$.

As shown above, the $\text{Ti}(\text{OH})_4$ molecule is stable both in the ground and the lowest excited states in the gas phase. However, in a water solution, the titanium atom may take on more waters. To estimate the effect of this additional coordination on the reactivity, we studied the H-abstraction and the direct O-O bond formation reactions with different conformations of titanium tetrahydroxide. Since the free energies of the ground state $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{1,2}$ compounds are significantly higher than those of $\text{Ti}(\text{OH})_4$, (see Tab. S1) we only consider here the excited state reactions.

3.2 Water splitting.

The reactant complexes of the $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{0,1,2}$ species with a water molecule are illustrated in Fig. S11. In the compound $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_1 \cdots \text{H}_2\text{O}$ the water H-atom (Hw) can be transferred to either the oxygen (Olig) of the axial OH ligand (Fig. S11 b) or that of the equatorial ligand (Fig. S11 c). The system $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_2$ has cis and trans conformers (Fig. S11 d, e). The transition of Hw to the Olig of any of the OH ligands are equivalent.

From the plot in Fig. S11 it is evident, that the systems with only one extra water ligand behave similarly to the $\text{Ti}(\text{OH})_4$ molecule, although, the transfer of Hw to the axial Olig has a somewhat higher barrier (11.2 kcal/mol) than to the equatorial Olig (6.0 kcal/mol).

The Reactions of water with $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_2$ is more different. The H-abstraction by the cis conformer is virtually barrierless (~ 0.2 kcal/mol). The transfer of Hw atom to the Olig in the trans-conformer results in a simultaneous relay of an H-atom from the nearby water ligand to the remnant of water, similarly to the mechanism observed in the $\text{Ti}(\text{OH})_4 + \text{H}_2\text{O}$ reaction in the ground state (see the main text). To check if it is energetically feasible to form a hydroxyl radical, we constrained the OH bond lengths of the water ligand and carried

out the relaxed scan ($\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_2(\text{trans})$) in Fig. S11). It is seen, that there is an energy penalty of around 14 kcal/mol (corresponding to the Hw-Olig distance of ca. 0.98 Å) to the OH^\bullet radical formation, which is still reasonable.

3.3 Nucleophilic attack.

To study the possibilities of O-O bond formation between the oxygen of the incident water (Ow) and the oxygen of an OH ligand (Olig) relaxed scans along the Ow-Olig (see labels in Fig. S12) were performed. The corresponding PESs are shown in Fig. S13. In the system $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_1 \cdots \text{H}_2\text{O}$ only the approach to the Olig of the axial OH ligand is shown, since the attack at the equatorial OH leads quickly to the formation of $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_2$ (cis) compound. The relaxed scans for the other compounds are propagated until the systems break down to form compounds with one more or one less water ligand. It is evident, that in each case the energy barriers are too high (≥ 40 kcal/mol) for the reactions to proceed spontaneously.

From the above studies we infer that the reaction mechanisms observed for $\text{Ti}(\text{OH})_4 + \text{H}_2\text{O}$ do not change significantly when $\text{Ti}(\text{OH})_4$ is replaced by its coordinatively saturated species.

4 Cartesian coordinates of the key intermediates

4.1 Optimised geometries of the substrate molecules

$^1\text{H}_2$

BP86: -1.1723103812 CCSD(T): -1.1723103812

H 0 0 0.3745079

H 0 0 -0.3745079

$^2\text{OH}\bullet$

BP86: -75.6412823877 UCCSD(T): -75.6412822907

O 0.6539658 -3.51609 0.2697869

H 1.0026787 -2.6374483 -0.012261

$^2\text{OH}\bullet$ (B3LYP geom)

B3LYP: -0.34215934 UCCSD(T): -75.6414814125

O 0 0 2.211137

H 0 0 1.236025

$^2\text{OOH}_2$

BP86: -150.718502406 UCCSD(T): -150.7185006506

O 0.6552538 -3.517902 0.2699839

H 0.9759687 -2.6287963 -0.009253

O -0.6626378 -3.3532881 0.4826259

$^1\text{H}_2\text{O}$

BP86: -76.3369945521 CCSD(T): -76.3369945521

O -0.0172465 -1.761931 -0.3708988

H -0.0172465 -1.0001733 0.215041

H -0.0172465 -2.5236887 0.215041

$^1\text{H}_2\text{O}$ (B3LYP geom)

B3LYP: -0.62712589 CCSD(T): -76.3369645121

O -0.8809418 -2.3379923 0

H -0.158625 -1.7006825 0

H -0.4498049 -3.1993051 0

$^1\text{H}_2\text{O}_2$

BP86: -151.3648285999 CCSD(T): -151.3648285999

O -0.112877 -3.41114 -0.068109

H 0.6345318 -2.7920942 0.039207

O -0.054398 -4.1114638 1.2250637

H 0.2775799 -4.9827316 0.9322997

$^3\text{O}_2$

BP86: -150.1346120368 UCCSD(T): -150.1346130104

O 0 0 0.6129558

O 0 0 -0.6129558

4.2 Ground and excited state stationary structures of the reactant $\text{Ti}(\text{OH})_4(\text{R})$ and corresponding product (P) of H-abstraction, $\text{Ti}(\text{OH})_3\text{H}_2\text{O}$

$^1\text{Ti}(\text{OH})_4$ (G.S. geometry)

BP86: 1.95510464 CCSD(T): -1152.0073339806 CCSD-ex: 0.2258958

Ti 0 0 0.000176

O 0.408463 -1.412581 1.075608

O -1.413303 -0.405653 -1.07524

H -2.007516 -1.169882 -1.004803

O -0.408463 1.412581 1.075608

H -1.171167 2.007632 0.994781

O 1.413303 0.405653 -1.07524

H 2.007516 1.169882 -1.004803

H 1.171167 -2.007632 0.994781

$^1\text{Ti}(\text{OH})_4$ (E.S. geometry)

BP86: -1.8083389 CCSD(T): -1151.9013008537 CCSD-ex: 0.0582457

Ti 0.16401 0.001477 0.144557

O 0.125608 -0.2136959 1.9439815

O -1.0575427 1.0961497 -1.0524537

H -0.9783417 0.9552897 -2.0182074

O 1.8087695 0.2038519 -0.6249808

H 2.7254802 0.13333 -0.3199709

O -1.3745646 -1.0620837 -0.6309768

H -1.2823866 -1.4713476 -1.5123766

H -0.3682919 -0.071966 2.7632592

$^2\text{Ti}(\text{OH})_3\text{H}_2\text{O}$ (G.S. geometry)

BP86: -2.02504026 UCCSD(T): -1152.5582369267

Ti 0.011258 0.122962 -0.3630109

O -0.5660218 1.2708646 -1.6241495

O 1.5481376 -1.4127726 0.3352339

H 1.7540525 -0.9523247 1.1761237

O 0.8134068 0.9584047 1.1116827

H 0.6349768 1.8479385 1.4517396

O -1.0246787 -1.4285796 -0.2161949

H -1.9368275 -1.5443216 -0.5229479

H -0.7706748 1.9535025 -2.2740584

H 0.9078507 -2.1214604 0.5515708

4.3 Stationary structures of the reactant $\text{Ti}_2\text{O}_2(\text{OH})_4(\text{R})$ and corresponding products (P) of H-abstraction

^1R (G.S. geometry)

BP86: -2.86612811

Ti -0.464672 1.111454 -2.156558

Ti -0.205651 1.517553 0.510058

O 0.88664 1.148903 -0.928476

O -0.77554 -0.576186 -2.777927

H -0.921606 -0.883936 -3.686566

O -1.549877 1.531795 -0.728951

O -0.376847 2.382606 -3.444405

H -0.077611 3.294366 -3.571088

O 0.113078 3.104279 1.343924

H 0.135935 4.024666 1.041554
O -0.31762 0.163255 1.720332
H -0.516516 -0.776668 1.583174

²P-term (G.S. geometry)

BP86: -2.93609497

Ti -0.48619 1.25258 -2.118454
Ti -0.288815 1.771647 0.548633
O 0.872889 1.394961 -0.918675
O -0.607119 -0.451033 -2.789173
H -0.886621 -0.726758 -3.676413
O -1.617308 1.558463 -0.697754
O -0.544739 2.54681 -3.402167
O -0.47742 2.97876 1.920532
H -0.404075 3.878991 2.261684
O 1.336995 0.600726 1.532859
H 1.898577 0.571701 0.725531
H -0.36324 3.496208 -3.454035
H 1.833122 1.081001 2.220527

²P-br (G.S. geometry)

BP86: -2.95215123

Ti -0.367066 1.360822 -2.13813
Ti 0.080701 1.701205 0.561045
O 1.017515 1.918208 -0.989312
O 0.001383 -0.289421 -2.838538
H -0.095403 -0.655456 -3.730764

O -1.51205 1.221678 -0.512856
O -0.904177 2.651395 -3.298666
H -0.909474 3.611872 -3.416756
O -0.123574 3.246978 1.519949
H -0.013987 4.180567 1.285208
O 0.685251 0.308801 1.580903
H 1.24009 -0.443218 1.320519
H -2.421007 0.939901 -0.327726

4.4 Stationary structures of the reactant $\text{Ti}_3\text{O}_3(\text{OH})_6$ (R) and corresponding products (P) of H-abstraction

^1R (G.S. geometry)

BP86: -4.33549653

Ti -3.029843 0.724751 -2.517931
Ti -0.786833 2.054915 -0.401673
Ti -2.198565 3.967635 -2.767871
O -1.871205 0.755891 -1.133631
O -2.490652 -0.456257 -3.78318
H -1.693429 -1.002177 -3.866419
O -4.701294 0.315393 -1.915183
H -5.28566 -0.398902 -2.214934
O -3.040696 2.434249 -3.209294
O 0.957511 1.595983 -0.666289
H 1.632781 1.387648 -0.001522
O -1.14899 2.23787 1.36689
H -1.457091 2.992848 1.891798

O -1.141948 3.574875 -1.308229
O -1.154191 4.500066 -4.16235
H -1.067695 5.380227 -4.560699
O -3.394281 5.25454 -2.307946
H -4.363595 5.260897 -2.343842

²P-term (G.S. geometry)

BP86: -4.41505331

Ti -2.171894 0.387496 -1.594339
Ti -0.95242 3.450652 -1.585118
Ti -2.729929 2.547949 -4.122168
O -1.257995 1.809401 -0.855959
O -0.996948 -0.990761 -1.858921
H -0.963282 -1.832863 -1.378586
O -3.525828 -0.108943 -0.474719
H -4.457996 -0.296826 -0.66665
O -2.76789 0.946956 -3.228673
O 0.376593 3.073161 -3.155055
H 1.099771 2.434982 -3.035009
O -0.466476 4.924311 -0.612761
H -0.262115 5.128111 0.311643
O -2.291349 3.742633 -2.875105
O -1.086957 2.484945 -5.152074
H -0.967871 3.000475 -5.965828
O -4.189908 3.061006 -5.1062
H -4.893219 2.52281 -5.503109
H -0.108393 2.851462 -4.063239

²P-br (G.S. geometry)

BP86: -4.40565156

Ti -2.165694 0.414526 -1.704764

Ti -0.806715 3.389128 -1.657151

Ti -2.595885 2.602189 -4.157763

O -1.007966 1.68592 -1.052868

O -1.244311 -1.098163 -2.158768

H -0.704786 -1.666648 -1.586749

O -3.408331 0.084945 -0.410362

H -4.295763 -0.300278 -0.477182

O -2.93015 1.042628 -3.221681

O 0.984549 3.834014 -1.791019

H 1.552814 4.04069 -1.031005

O -1.774407 4.554262 -0.654271

H -2.445937 4.396145 0.028011

O -1.085129 3.745737 -3.58659

O -1.958005 2.100089 -5.789526

H -1.31635 1.429111 -6.073744

O -4.088462 3.631247 -4.19601

H -4.654629 3.974595 -4.90376

H -0.419483 4.187182 -4.149886

4.5 Stationary structures of the reactant $\text{Ti}_3\text{O}_4(\text{OH})_4$ (R) and corresponding products (P) of H-abstraction

$\text{Ti}_3\text{O}_4(\text{OH})_4$

Fout:510

¹R (G.S. geometry)

BP86: -3.74456233

Ti 1.795843 -0.527127 0.113294

O 0.003145 -0.145051 0.654719

O 1.755009 0.937814 -0.882033

Ti -0.019408 1.574099 -0.486783

O -0.079933 2.906146 0.455485

O 1.87057 -2.025212 -0.923994

Ti -1.784639 -0.580048 0.107091

O -1.872456 -2.194305 -0.717956

O -1.738677 0.83587 -0.955751

H -1.563521 -2.483316 -1.591428

O -3.001148 -0.48146 1.452923

H -3.437723 0.308333 1.812549

O 2.979308 -0.511091 1.482837

H 3.518629 0.186896 1.887471

H 2.318873 -2.866847 -0.740253

²P-term (G.S. geometry)

BP86: -3.83297526

Ti 0.694623 -0.307431 -0.809001

O -1.015314 0.070301 -0.186845

O 0.684098 1.109419 -1.850349

Ti -1.07269 1.816788 -1.320744
O -1.095138 3.224781 -0.481961
O 0.36771 -1.940303 -1.768896
Ti -2.873772 -0.225585 -0.758744
O -2.19029 -2.045259 -1.569536
O -2.787357 1.093889 -1.923852
H -1.16001 -2.118398 -1.69143
O -4.236045 -0.360953 0.452225
H -4.984812 0.182726 0.739178
O 2.015849 -0.425044 0.439882
H 2.734407 0.173722 0.697597
H 0.870056 -2.754217 -1.599692
H -2.602708 -2.48748 -2.329909

²P-br (G.S. geometry)

BP86: -3.80526073
Ti 0.890941 -0.270425 -0.850757
O -1.040366 -0.120038 -0.096426
O 0.677194 1.098848 -1.944529
Ti -1.06215 1.733964 -1.278786
O -1.124151 3.020659 -0.25086
O 0.979838 -1.832472 -1.803646
Ti -2.954741 -0.34639 -0.856844
O -3.159615 -1.989926 -1.626467
O -2.73806 0.965905 -2.005299
H -2.996325 -2.27024 -2.541372
O -4.087998 -0.00624 0.528676

H -4.515458 0.833668 0.765574
O 1.994187 -0.001751 0.569088
H 2.478786 0.786293 0.86479
H 1.33218 -2.708359 -1.579961
H -1.046935 0.026076 0.870155

²P-br2 (G.S. geometry)

BP86: -3.84225609

Ti 0.732264 -0.290431 -0.797977
O -1.060375 0.063838 -0.239269
O 0.716142 1.169992 -1.816152
Ti -1.037374 1.798941 -1.321495
O -1.108524 3.17417 -0.432528
O 0.701202 -1.813503 -1.834019
Ti -2.83877 -0.382385 -0.771131
O -2.813097 -2.000642 -1.593968
O -2.786474 1.107537 -2.093934
H -2.069767 -2.370287 -2.103757
O -4.1169 -0.245337 0.50938
H -4.274401 0.394646 1.222303
O 1.962183 -0.282305 0.539879
H 2.549304 0.4109 0.880784
H 1.211584 -2.623682 -1.67179
H -3.491374 1.418328 -2.682793

²P-oxo-term (G.S. geometry)

BP86: -3.86915208

Ti 2.272181 1.441074 4.722556
O 1.36981 2.939312 5.523875
O 1.555105 1.876337 3.118075
Ti 0.754739 3.477478 3.704617
O 1.370399 5.035884 2.983116
O 1.464244 -0.088789 5.38024
Ti -0.468652 3.045368 6.105264
O -0.814538 1.375222 6.727729
O -0.964522 3.301817 4.345945
H -0.369539 0.591244 6.342778
O -0.910792 4.364626 7.279839
H -1.209604 5.282463 7.194494
O 4.082183 1.509292 4.811702
H 4.798533 1.805397 4.22956
H 1.892001 -0.821453 5.852569
H 1.077669 5.946992 2.836793

4.6 Stationary structures of $\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_{1,2}$ compounds

$^1\text{Ti}(\text{OH})_4(\text{H}_2\text{O})$ (G.S. geometry)

BP86: -2.48051319

Ti -2.207238 0.005092 1.096729

O -4.564941 0.469224 1.424487

O -2.4222 0.7017 2.825799

O -0.396133 -0.263901 1.079951

O -2.456603 1.220692 -0.270069

H -4.638115 1.290802 0.905918

H -1.649118 0.739134 3.412622
H -2.616773 0.982422 -1.19784
H -4.376193 0.765403 2.339601
O -2.808833 -1.699308 0.808373
H 0.230804 0.27384 0.570023
H -2.176522 -2.422675 0.660087

$^1\text{Ti}(\text{OH})_4(\text{H}_2\text{O})$ (E.S. geometry)

BP86: -2.34631607
Ti -2.340861 0.149404 1.042985
O -4.557056 0.38867 1.495669
O -2.519718 0.731841 2.858853
O -0.477366 -0.683666 0.949763
O -2.342108 1.470394 -0.237313
H -4.887283 1.13343 0.962382
H -1.762035 0.987294 3.408307
H -2.15496 1.613344 -1.174802
H -4.280968 0.77387 2.368335
O -2.343225 -1.873298 0.918523
H -0.269135 -1.149566 0.11448
H -2.009189 -2.334723 1.71249

$^1\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_2$ -cis. (E.S. geometry)

BP86: -2.8732057
Ti -0.255399 -0.288314 0.05578
O 1.962175 -0.949438 0.319635
O -1.142741 1.754113 -0.620143

O -2.202976 -0.747503 0.384085
O 0.564851 0.659674 1.544108
O -0.629385 -2.242777 0.443359
O 0.162183 0.047587 -1.816936
H 1.951914 -0.271711 1.049963
H -0.707021 1.519493 -1.48536
H -2.636726 -1.239312 -0.340949
H 0.143644 1.371185 2.051661
H -0.857657 -2.43938 1.373395
H 0.731473 -0.511143 -2.369159
H 2.502072 -0.560909 -0.389882
H -0.609723 2.4778 -0.248514

$^1\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_2$ —trans. (G.S. geometry)

BP86: -2.98623008

Ti -2.563856 0.075386 1.036476
O -4.439801 1.287474 0.799341
O -2.436994 -1.08774 -0.425507
O -2.068902 0.812205 2.757981
O -0.441268 -0.496598 1.416826
O -2.144859 1.58981 -0.052537
H -3.864238 1.947963 0.32679
H -2.522042 -2.052309 -0.371295
H -2.552241 0.394046 3.491026
H -0.572646 -0.092361 2.319043
H -1.800762 1.395931 -0.940655
H -4.685897 1.670775 1.660177

O -3.715603 -1.084991 1.979284

H -0.283692 -1.44972 1.527182

H -4.440697 -1.54819 1.52927

$^1\text{Ti}(\text{OH})_4(\text{H}_2\text{O})_2$ —trans. (E.S. geometry)

BP86: -2.87491652

Ti -2.485325 0.288011 1.129951

O -4.408709 1.39597 0.624393

O -2.84824 -1.179738 -0.273917

O -2.487482 1.014139 2.850715

O -0.6026 -0.856683 1.31599

O -2.079104 1.653271 -0.185834

H -3.788589 1.899813 0.018897

H -3.789414 -1.278028 -0.514389

H -2.760416 0.557633 3.661235

H -0.136541 -0.573681 2.122479

H -1.36526 1.64026 -0.841327

H -4.550633 1.976389 1.394814

O -3.016257 -1.658517 1.885215

H -1.058118 -1.702146 1.543812

H -3.915175 -2.026679 1.816635

4.7 Intermediate structures of the reactions of H-abstraction and H₂ formation by Ti(OH)₄ in the gas phase

4.7.1 Intermediates of the H-abstraction reaction from H₂O

¹RC1 (G.S. geometry)

BP86: -2.48681932 CCSD(T): -1228.3595864051 CCSD-ex: 0.2137248

Ti 0.088915 0.047508 -0.167723

O -0.000363 0.9413647 -1.7537745

O 1.4751216 -1.2006157 -0.2122909

H 2.2236584 -1.1626137 -0.8298078

O 0.3587809 1.2810086 1.1527227

H -0.025352 1.2748506 2.0441144

O -1.3444006 -0.9774457 0.15497

H -1.1941857 -1.9611244 0.2308369

H -0.6129398 0.7687328 -2.4875003

O -0.156499 -3.463065 0.2040779

H 0.6261278 -2.8741202 0.071488

H -0.004872 -3.8971419 1.0594977

¹TS1 (G.S. geometry)

BP86: -2.46927152 CCSD(T): -1228.3377939416

Ti 0.076026 0.058953 -0.15122

O 0.172943 0.8343648 -1.8125645

O 1.4016846 -1.4403816 -0.019438

H 2.1863674 -1.5137066 -0.5857398

O 0.3147029 1.3108296 1.1708767

H -0.168237 1.3615636 2.0109154

O -1.2661816 -0.9665317 0.083583
H -0.9195827 -2.2543204 0.108963
H -0.4004179 0.6374118 -2.5706013
O -0.2620019 -3.1827121 0.083488
H 0.7165078 -2.4551353 0.044716
H -0.3203829 -3.612438 0.9537567

¹PC1 (G.S. geometry)

BP86: -2.47241204 CCSD(T): -1228.3490788672

Ti 0.053177 0.084607 -0.14515
O 0.1996179 0.8121978 -1.8330615
O 1.4626906 -1.4488716 -0.017451
H 2.1761554 -1.5364376 -0.6712128
O 0.3147479 1.3235416 1.1926767
H -0.2085519 1.4021266 2.0058814
O -1.2817066 -0.8985257 0.076225
H -0.8995797 -2.5032613 0.105363
H -0.3883699 0.6266358 -2.5825133
O -0.2678359 -3.3018321 0.077221
H 0.9133197 -2.3371073 0.029022
H -0.3859509 -3.7648589 0.9231867

¹RC1 (E.S. geometry)

BP86: -2.3436207 CCSD(T): -1228.2508579934 CCSD-ex: 0.0504252

Ti 0.2192179 0.09289 0.078467
O -0.6193368 0.3455469 -1.6949995
O 1.2634606 -1.3461656 0.7022238

O 0.7918718 1.7107995 0.6644978
O -1.4692146 -0.9883227 -0.098435
H -0.5320689 -0.3221269 -2.4005563
H 2.1765714 -1.2950776 1.0252207
H 0.5800728 2.5642933 1.0662437
H -1.1963757 -1.8800075 -0.4696799
O -0.045209 -3.2791681 -0.6913918
H 0.5982088 -2.7953422 -0.105261
H -0.2892689 -4.0832029 -0.2048739

¹TS1 (E.S. geometry)

BP86:-2.33577201 CCSD(T): -1228.2272168714 CCSD-ex: 0.0399999

Ti 0.110891 0.098136 -0.2770319
O -0.3460629 1.2367637 -1.6207535
O 1.4561256 -1.4341526 -0.3109159
O 0.4156929 0.8338798 1.3788666
O -1.2522686 -1.4244436 -0.3687239
H -0.2921569 1.4190336 -2.5692553
H 2.3694163 -1.3106536 -0.009248
H 0.3093119 1.7321865 1.7258395
H -1.0956047 -2.2102854 -0.9286437
O -0.055467 -2.9523282 0.7909758
H 0.8633688 -2.2345774 0.3061609
H -0.3682589 -2.5528883 1.6261795

¹PC1 (E.S. geometry)

BP86: -2.33286315

Ti 0.039261 0.087697 -0.2879679
O -0.5027149 1.1996277 -1.6181945
O 1.5611686 -1.3572126 0.153295
H 2.0238734 -1.0451547 0.9526197
O 0.7570078 1.0576457 1.1197287
H 0.4512089 1.8558095 1.5748016
O -1.2206037 -1.4037726 -0.2462149
H -1.3368216 -2.0019894 -1.0074267
H -0.2488339 1.8256985 -2.3096964
O -0.127301 -3.2354991 0.4499349
H 1.0330507 -2.1728334 0.3896309
H -0.6889288 -3.1390921 1.2443927

³RC1 (Triplet state B3LYP E.S. geometry)

B3LYP/def2-TZVPP: CCSD(T):-1228.2576590776 CCSD-ex: 0.0587616

Ti 0.2682119 0.079389 0.134199
O -0.4622249 -0.016943 -1.7298625
O 1.3993146 -1.3306616 0.5993588
H 2.2317614 -1.3882186 1.0745677
O 0.5809238 1.7787145 0.6916098
H 0.2666839 2.6815612 0.7621958
O -1.5558226 -0.7805768 0.02817
H -1.4640176 -1.7104425 -0.2613839
H -0.4462519 -0.8688808 -2.1901894
O -0.16248 -3.1442691 -0.8433678
H 0.5588328 -2.8000522 -0.2815309
H -0.2067479 -4.0920919 -0.6858638

³TS1 (Triplet state B3LYP E.S. geometry)

B3LYP/def2-TZVPP: -1229.248807186 CCSD(T): -1228.2511839274 CCSD-ex: 0.0699373

Ti -6.9806762 5.5346914 -1.6385668

O -7.0102635 6.1846392 -3.349375

O -5.5232295 4.1609294 -1.423784

H -4.5825752 4.2300967 -1.5884772

O -7.1801285 6.6259275 -0.1761553

H -7.3438912 7.5435689 0.0384098

O -8.2178613 3.8552238 -1.4433374

H -8.7773411 3.5852671 -2.1800849

H -6.883504 5.9526642 -4.2688269

O -6.8613528 2.2319886 -1.846054

H -6.069097 3.100089 -1.6253181

H -7.1671725 1.9035307 -0.9908534

³PC1 (Triplet state B3LYP E.S. geometry)

B3LYP/def2-TZVPP: CCSD(T): -1228.2453554022 CCSD-ex: 0.0357379

Ti 0.102832 0.147158 -0.112434

O -0.04424 0.6627858 -1.8623875

O 1.4966926 -1.4263226 -0.005797

H 2.3227653 -1.4829356 -0.4945299

O 0.2056209 1.5117086 1.1038457

H 0.2055379 1.6390255 2.0545154

O -1.0802487 -1.3286076 0.5264939

H -2.0097844 -1.3875356 0.2777399

H -0.3051439 0.3027029 -2.7125072

O -0.2617709 -3.2085431 -0.3462249

H 0.9345307 -2.2757304 -0.140396

H -0.5342818 -3.512196 0.5330279

4.7.2 Intermediates of the H-abstraction reaction from H₂O₂

¹RC2 (G.S. geometry)

BP86: -2.62931869 CCSD(T): -1303.3871536143 CCSD-ex: 0.2106941

Ti 0.052802 0.133655 -0.3063949

O 0.002966 0.9845407 -1.9137805

O 1.3881686 -1.1673927 -0.3048529

H 2.2432574 -1.1212107 -0.7614138

O 0.3661779 1.4070456 0.9596667

H 0.066386 1.4008706 1.8830565

O -1.4099076 -0.8436328 0.038801

H -1.3265846 -1.8261385 0.138833

H -0.6064238 0.8259268 -2.6528783

O -0.1968059 -3.3818331 -0.018862

H 0.5761808 -2.7523392 -0.065953

O 0.042854 -4.0378019 1.2758196

H 0.3109689 -4.9239386 0.9631997

¹TS2 (G.S. geometry)

BP86: -2.61191758 CCSD(T): -1303.3650469877 CCSD-ex: 0.1967089

Ti 0.023911 0.088171 -0.3201639

O -0.076098 1.0577687 -1.8709895

O 1.3967116 -1.3755506 -0.4340579

H 2.2611964 -1.3509916 0.006302
O 0.3832819 1.1745247 1.1100907
H -0.013985 1.1197087 1.9941954
O -1.2653166 -1.0110647 -0.087517
H -0.8825548 -2.2293314 0.032052
H -0.7063218 0.9257167 -2.5972583
O -0.1839899 -3.1705881 -0.045661
H 0.7866228 -2.3691483 -0.2279899
O 0.033181 -3.600744 1.3467926
H -0.3618539 -4.4923307 1.2940186

¹PC2 (G.S. geometry)

BP86: -2.61462357 CCSD(T): -1303.3757696246 CCSD-ex: 0.1978648
Ti 0.01387 0.148203 -0.2753039
O -0.059556 0.8813498 -1.9637494
O 1.4944926 -1.3231936 -0.3130259
H 2.1604534 -1.3773956 -1.0185477
O 0.3596429 1.4033836 1.0232037
H -0.051601 1.4415946 1.9014665
O -1.2379297 -0.9047297 0.086111
H -0.8267268 -2.4480523 0.108502
H -0.7067268 0.6588958 -2.6519963
O -0.1973959 -3.2403761 -0.089736
H 0.9916167 -2.2265324 -0.2389829
O -0.031632 -3.8842799 1.2211927
H -0.5803528 -4.6808147 1.0805757

¹PC2 (E.S. geometry)

BP86: -2.48210117 CCSD(T): -1303.2761622822 CCSD-ex: 0.0562142

Ti 0.05996 0.2303469 -0.3498379

O -0.4107959 1.1349367 -1.8749315

O 1.4909786 -1.2658266 0.03186

H 2.4211073 -1.2943616 -0.2356129

O 0.3890959 1.4141556 1.0018107

H 0.2181739 1.4351816 1.9550545

O -1.2551706 -1.2288237 -0.2648789

H -1.1690757 -2.1521484 -0.5859008

H -0.4227439 1.0083267 -2.8347652

O 0.044718 -3.3726921 0.00657

H 1.0836957 -2.2147684 0.093996

O -0.166119 -4.0325369 1.2633116

H 0.3910129 -4.8321806 1.1641287

4.7.3 Intermediates of the H-abstraction reaction from OOH•

²RC3 (G.S. geometry)

BP86: -2.47777203 UCCSD(T): -1302.7463197096

Ti 0.045529 0.105959 -0.2611779

O -0.2220479 0.8984257 -1.8763825

O 1.3458446 -1.2625736 -0.3969359

H 2.2191824 -1.1651317 -0.8091658

O 0.5806428 1.4070566 0.8947017

H 0.3910229 1.4706746 1.8444935

O -1.3284006 -0.8498288 0.3271329

H -1.2412087 -1.8410995 0.4875699
H -0.9174647 0.7130248 -2.5277673
O 0.5476218 -3.661789 0.2907879
H 0.9098187 -2.7307042 0.030238
O -0.7353268 -3.483164 0.6295608

²TS3 (G.S. geometry)

BP86: -2.43883683

Ti -0.141516 0.10489 -0.3098969
O 0.8597198 0.3840979 -1.8008665
O -0.127187 -1.6918225 0.063062
H 0.3649299 -2.6643983 -0.4040859
O 0.6239458 1.0015197 1.0743897
H 0.1927099 1.5354726 1.7604975
O -1.8470575 0.6507018 -0.5563798
H -2.6959912 0.1818369 -0.5668728
H 0.5728918 0.6666218 -2.6837652
O 0.5892398 -3.557637 1.2636496
H 0.102159 -2.3926243 0.9994907
O 0.8344238 -3.8147199 -0.027235

²PC3 (G.S. geometry)

BP86: -2.45373281

Ti -0.161825 0.158917 -0.3214569
O 0.8531198 0.3954479 -1.8139525
O -0.2136279 -1.6396165 0.016533
H 0.6279168 -3.1117761 -0.3975959

O 0.6137408 1.0258497 1.0739037
H 0.2089649 1.5950356 1.7473185
O -1.8378805 0.7922488 -0.5800408
H -2.7021602 0.3518299 -0.5703448
H 0.6222568 0.8550028 -2.6366323
O 0.6047948 -3.7883549 1.3178786
H -0.126748 -2.1232094 0.8801838
O 0.9630377 -3.9658949 0.037299

²PC3 (E.S. geometry)

BP86: -2.37253452 CCSD(T): -1302.6923227806 CCSD-ex: 0.030447

Ti -0.042155 0.7809948 -0.4340129
O 1.1853807 -0.2791309 -1.2805126
O -1.6501905 -0.3462249 0.5675008
H -1.8731915 -0.1981479 1.5014326
O 0.6969848 2.2012074 0.4548629
H 0.3877279 3.0804931 0.7156198
O -1.7110915 1.1537457 -1.3162166
H -1.8234125 1.6987455 -2.1091024
H 1.1553377 -0.8895588 -2.0315334
O 1.1463207 -5.5054925 1.2115907
H -2.3227403 0.119815 -0.006297
O 1.1516067 -6.6635661 0.8099348

4.7.4 Intermediates of the H₂ generation reaction

¹RC4 (G.S. geometry)

BP86: -4.13000603 CCSD(T): -2305.1732342035

Ti 5.6366064 -4.7470747 8.8346335

O 5.6569864 -6.0999743 10.0451242

O 6.5341012 -5.1512846 7.3602609

H 3.8959299 -5.8551624 5.8196344

O 3.8237969 -4.3937018 8.4509796

H 3.1781341 -5.0659626 8.7317135

O 6.3803712 -3.2297211 9.5257173

H 6.3467112 -2.9583292 10.4576021

H 6.0718773 -6.974177 9.9672502

H 6.3519452 -5.3500195 6.3988532

Ti 4.2745658 -4.7673757 4.5179237

O 6.0500963 -5.3083205 4.6583357

O 3.1023151 -3.7438349 5.9844033

H 3.3070171 -2.7978902 5.8328494

O 4.4592257 -2.9143492 4.1409218

H 4.5791797 -2.6746662 3.2056011

O 3.416451 -5.5582234 3.1414801

H 3.626754 -6.3563992 2.6316593

H 6.7486391 -4.7087757 4.3414468

H 3.3408241 -4.0045419 6.9311671

¹TS (G.S. geometry)

BP86: -4.12025973 CCSD(T): -2305.1609252041

Ti 5.5129615 -4.5510007 8.3099707

O 5.7889604 -5.0369596 10.0486272

O 6.1552423 -5.9945633 7.4151089

H 3.557447 -6.5110072 4.7274557
O 3.608974 -4.4971067 8.2891367
H 3.1511361 -4.7743397 9.0999884
O 6.3474172 -2.8906642 8.3113787
H 6.6676001 -2.5688723 9.1720894
H 6.1466663 -5.8975973 10.3225771
H 6.2850262 -6.1426143 6.4533572
Ti 4.3474238 -4.7978367 4.8133676
O 5.8542504 -5.8609244 4.5197927
O 2.6653423 -4.8879876 5.8650044
H 2.8915802 -4.7857647 6.8520081
O 5.1880405 -3.7648579 6.1500833
H 5.0938046 -2.8006742 6.2505432
O 3.9473469 -3.8430669 3.3469371
H 4.3468918 -3.8116429 2.4622443
H 6.6799731 -5.3643185 4.3614938
H 2.9445472 -5.9893593 5.3823735

¹PC4 (G.S. geometry)

BP86: -4.17743988 CCSD(T): -2305.2165705163
Ti 5.4936055 -4.7731027 8.6220156
O 6.0094003 -5.0764176 10.3329791
O 6.4006672 -5.7257214 7.4445249
H -0.4292369 -4.5511157 5.9652533
O 3.686282 -5.1499216 8.4259486
H 3.1183861 -5.5870124 9.0806035
O 5.7072584 -2.9946202 8.1131627

H 5.5295254 -2.2284984 8.6829906
H 6.7047721 -5.6594334 10.678941
H 6.2482162 -5.8866113 6.4523532
Ti 4.4589017 -4.4924117 4.6972807
O 5.7450764 -5.8705194 4.8933316
O 3.0402611 -4.9562796 5.7078234
H 3.0976301 -5.0640306 6.6913541
O 5.2879835 -3.0621001 5.3743295
H 5.4663865 -2.8289072 6.3224972
O 3.9045319 -4.3044608 2.9723172
H 4.1655598 -3.627895 2.3267763
H 5.5890354 -6.7592881 4.5329837
H 0.3065489 -4.6420807 5.8459344

4.8 Reactions of H-abstraction and H₂ formation by Ti(OH)₄ including explicit water solvent.

4.8.1 Intermediate structures of the reaction of H-abstraction from H₂O.

¹RC (G.S. geometry)

B3LYP-D3: -2070.291596619

Ti -11.691922 2.3047821 -0.9658204

O -11.5093503 1.2893642 0.5275147

O -12.4035801 3.895755 -0.534242

O -10.076827 2.4604333 -1.766013

O -12.9307011 1.4945579 -2.0686995

H -12.8125128 4.1625526 0.3435463

H -9.7140956 3.2404759 -2.2774199
H -12.7371119 0.7385298 -2.6844607
H -12.2359036 0.8089006 0.9691291
O -11.9029336 2.5169621 3.1749742
H -12.4890363 1.7474688 3.2356212
H -11.3425506 2.2968739 2.413835
O -13.236504 5.4780359 -2.7149295
H -14.1612865 5.195849 -2.843806
H -12.9389134 5.0232245 -1.9046121
O -8.4316605 0.3089417 -2.1279431
H -8.977125 1.1100722 -1.9494726
H -7.7319352 0.5917127 -2.7248447
O -15.4615034 1.916277 -1.6604227
H -14.4845117 1.7449149 -1.8559251
H -15.9318546 1.1394631 -1.9824616
O -13.5823353 4.2825839 1.7867779
H -13.0049969 3.7917464 2.4101988
H -14.3442351 3.7001785 1.6345961
O -13.7016812 0.4661323 2.0930594
H -14.0382919 -0.4227025 2.2412794
H -14.4523811 1.0110168 1.7721782
O -9.1628144 4.4691297 -3.2150676
H -9.9233654 4.8074569 -3.7557957
H -8.8106312 5.2322952 -2.745878
O -12.4352135 -0.6166478 -3.6351799
H -12.3000996 -0.4741585 -4.5773442
H -11.608729 -1.0565421 -3.3109956

O -15.8561907 4.4048792 -2.9648106
H -16.1783064 4.2463627 -3.8574736
H -15.7938741 3.5285817 -2.5389173
O -15.5479241 2.2323733 1.0489452
H -16.4484797 2.4176603 1.3345328
H -15.5686352 2.1583872 0.0627494
O -10.1173436 -1.6909238 -2.7834085
H -10.1050251 -2.2998814 -2.0387078
H -9.4583046 -0.9849936 -2.5683005
O -11.3138649 5.3455009 -4.5609043
H -12.0755409 5.4117901 -3.92424
H -11.6238768 4.8172447 -5.3021519

¹PC2 (G.S. geometry)

BP86: -2071.113082225
Ti -0.4279873 -0.0512456 -0.2963344
O 0.0402982 -1.6154416 -1.4683579
O 0.5088374 1.2538106 -0.7667909
O -2.2254716 0.3282888 -0.3563798
O 0.1786641 -0.5934645 1.4193295
H 2.1331102 0.9546625 -1.5702586
H -2.6196038 1.2464642 -0.5701092
H -0.1249596 -1.4655123 1.7816592
H 0.61571 -2.3260393 -0.9762324
O 1.3647753 -0.9659369 -3.5462141
H 1.8362459 -1.6520103 -4.0452475
H 0.5299027 -1.4153046 -2.3754022

O 0.7661967 3.8583972 0.0940344
H 1.2513179 3.6862133 0.9382694
H 0.6984534 2.958473 -0.3187193
O -4.0024122 -1.517011 0.3389677
H -3.3721622 -0.7785759 0.0546846
H -4.5890725 -1.1204002 1.0025307
O 2.4539425 0.2082748 2.262868
H 1.5013221 -0.1076617 1.9954474
H 2.6060849 -0.1219674 3.1642442
O 2.8940255 0.4683878 -1.9848371
H 2.0673943 -0.4117573 -3.0506996
H 3.2206103 -0.1064436 -1.2542986
O 1.5199326 -3.2594026 -0.1346239
H 0.9896404 -3.3936197 0.6914534
H 2.2812785 -2.674718 0.1087989
O -3.2629506 2.6836114 -0.8392499
H -2.6867036 3.4350471 -0.4758349
H -3.3503215 2.8711405 -1.7880715
O -0.3486037 -3.261527 1.9556097
H -0.3065734 -3.7053466 2.8179354
H -1.2771126 -3.4731087 1.5756666
O 2.0907532 3.0436574 2.4739
H 1.469114 3.075865 3.2193037
H 2.2746928 2.0802404 2.3455984
O 3.5208026 -1.27933 0.2771402
H 4.4664803 -1.4674455 0.3960585
H 3.248284 -0.701756 1.0545301

O -2.7217146 -3.7168872 0.9990817
H -2.7612226 -4.2627368 0.1971092
H -3.232603 -2.8672672 0.7731609
O -1.7409333 4.659643 0.0393724
H -0.7600115 4.3987286 0.1158042
H -1.9819952 5.0660236 0.8860542

¹RC (E.S. geometry)

(E.S.) BP86: -2070.966545925
Ti -11.7294182 2.1294284 -1.0825951
O -11.5256781 2.2432969 0.9309065
O -12.7199688 3.6649186 -0.3567331
O -10.106845 2.1945991 -1.9317864
O -13.0501928 1.0880011 -1.917891
H -13.3881179 3.8691766 0.4171922
H -9.6244765 3.000122 -2.3238876
H -12.9251664 0.3506069 -2.5980028
H -12.22649 1.6467413 1.3473056
O -12.0438841 3.9275797 3.1164118
H -11.9736362 3.3206761 3.8717797
H -11.6290174 3.4374296 2.3608334
O -13.1038738 5.5582304 -2.3112179
H -13.9080725 5.0850556 -2.6488581
H -12.860284 5.0601244 -1.4918681
O -8.6519859 0.0426617 -2.6510767
H -9.1772251 0.8487798 -2.3574827
H -8.0057723 0.3787644 -3.2930988

O -15.5005727 1.7041805 -1.5515684
H -14.5100853 1.4089984 -1.7071832
H -16.0464289 1.0252518 -1.9831934
O -14.3018062 4.1939426 1.5999035
H -13.6390029 4.1261247 2.3416038
H -14.8950975 3.4114307 1.6570465
O -13.4038877 0.6025253 2.0445841
H -13.2999363 -0.2974317 1.6933519
H -14.3029427 0.9050497 1.7439107
O -8.9395568 4.2794683 -3.0229111
H -9.6793865 4.8934722 -3.3379626
H -8.4161027 4.8126548 -2.4025905
O -12.7743882 -0.8918537 -3.6707374
H -12.8536093 -0.6237729 -4.6005968
H -11.8463785 -1.2866087 -3.598617
O -15.3131855 4.0495968 -3.1791092
H -15.1088676 3.6429462 -4.0370789
H -15.4349206 3.2822158 -2.565435
O -15.7119222 1.7554379 1.1217388
H -16.6239461 1.5862013 1.4076084
H -15.7125604 1.7473276 0.1109847
O -10.2972259 -1.8422328 -3.5470424
H -10.140639 -2.6164454 -2.9833701
H -9.6517107 -1.1349731 -3.2249797
O -10.8990949 5.9089501 -3.7509114
H -11.7656476 5.7748747 -3.2465179
H -11.1373427 5.987533 -4.6873994

¹PC1 (E.S. geometry)

BP86: -2070.95711932

Ti 0.6353292 -0.1343987 -0.2979544

O 1.6101306 -1.5496752 0.5555496

O 0.7895829 1.0301856 1.3589016

O 1.3135926 0.1053327 -2.0213027

O -1.2799513 -0.0616049 -0.4121139

H 0.7881453 0.6523378 2.3487189

H 1.556467 1.0057651 -2.3907056

H -1.7991593 -0.5071929 -1.1488306

H 1.0570342 -2.3411049 0.7930763

O 3.0209842 -1.3553084 2.8038262

H 3.2659775 -2.2540321 3.0773584

H 2.6313064 -1.4506083 1.8771093

O -0.0759746 3.3830202 1.0938744

H -1.004964 3.3003743 1.4933769

H 0.6116556 2.0392178 1.3646968

O 1.2134172 -1.6913012 -3.9854538

H 1.2948534 -1.01203 -3.2406526

H 1.5046939 -1.2313868 -4.7890752

O -2.7130823 0.1454896 1.611767

H -2.1066004 0.0548051 0.7146251

H -3.5989116 -0.1639479 1.3565315

O 0.8797113 -0.0091193 3.6552342

H 1.7037271 -0.5723822 3.5155264

H 0.1126583 -0.6326372 3.6502783

O -0.1819147 -3.4668907 1.5354344
H -0.8248012 -3.7679431 0.8726701
H -0.6937359 -2.8727168 2.1410201
O 1.8695881 2.598018 -2.9172297
H 1.3101184 3.2179391 -2.3669026
H 2.7855639 2.8807956 -2.7621123
O -2.7667243 -1.2816741 -2.2950891
H -3.2198717 -0.6472155 -2.8739854
H -2.2008149 -1.8346322 -2.9236263
O -2.5245479 2.8405158 2.0436871
H -3.2719463 3.2451584 1.5743306
H -2.6394789 1.8512669 1.9310349
O -1.4344772 -1.5811116 3.2023815
H -1.9789861 -1.8263238 3.968268
H -1.9812601 -0.9267072 2.6557783
O -1.2572572 -2.6423051 -4.0119369
H -1.1643655 -3.5998748 -3.8861352
H -0.3124994 -2.2790515 -4.0349887
O 0.3236759 4.2647651 -1.4159603
H 0.1353626 3.9846859 -0.4710377
H -0.5426256 4.4388841 -1.8181186

4.8.2 Intermediates of the H₂ formation reaction.

¹RC (G.S. geometry)

B3LYP-D3: -4141.694131806

Ti 0.0830695 -0.5628762 -2.5430944

O -1.0280617 -2.147581 -1.8451465
O -0.6715571 1.0417062 -2.0554645
O 1.9160996 -0.8320643 -2.2698995
O -0.5037284 -1.0016301 -4.3664686
H -1.6120351 1.1276884 -1.7987148
H 2.654997 -0.20196 -2.408049
H 0.1434867 -0.6730614 -5.0266342
H -1.6041763 -2.6172507 -2.5164726
O -2.440753 -1.4817646 0.3063498
H -2.8837676 -2.023917 0.9715517
H -1.5201286 -2.0686847 -0.989421
O 0.3529749 3.0377466 -3.7586243
H -0.1698848 2.8738477 -4.5759829
H -0.0388507 2.4605552 -3.0716423
O 3.7714491 -1.9198188 -4.2291448
H 2.9991518 -1.9124384 -3.6378932
H 4.2636867 -1.1348282 -3.9419727
O -2.5708806 0.110533 -5.538886
H -1.7731203 -0.290994 -5.0645212
H -2.6333447 -0.3579084 -6.379144
O -3.380749 0.6292887 -1.339601
H -3.2406842 -0.1534367 -0.7730709
H -3.8163979 0.2855387 -2.1421756
O -2.42507 -2.9486651 -3.9258795
H -1.7450964 -2.4130297 -4.3866045
H -3.2054587 -2.3711975 -3.8696169
O 4.0990026 0.6989807 -2.9843961

H 3.682396 1.3761354 -3.5714467
H 4.6325031 1.1674715 -2.3351177
O 1.2638419 0.297566 -6.1227468
H 1.8585348 0.9004632 -5.6373816
H 1.8507831 -0.3675026 -6.5483442
O -0.7773693 2.288636 -6.1753462
H -0.1051013 1.6263442 -6.4184477
H -1.571649 1.7588467 -5.9975491
O -4.3226495 -0.6842484 -3.6681694
H -5.2545207 -0.6623948 -3.9084684
H -3.8184919 -0.2832441 -4.4191986
O 3.0738229 -1.6077927 -6.8521936
H 2.8939697 -2.4551634 -7.2697151
H 3.3655615 -1.8059912 -5.9308251
O 2.6730349 2.2768307 -4.6156793
H 1.8098647 2.6283724 -4.2150039
H 3.0302844 2.9800925 -5.1676983
H -1.9564565 -0.7932881 0.8457717
Ti 0.2009533 0.4642218 2.6092447
O 0.448102 -1.5529767 1.9802057
O 1.9452341 0.9439986 2.0860667
O -1.5330939 0.4833424 1.8333904
O -0.1140795 -0.1238003 4.4579212
H 2.7192903 0.5135534 2.5080023
H -2.1562448 1.1941047 1.534616
H -0.7962053 0.271651 5.0160068
H 0.4865606 -2.2269576 2.6970771

O 2.5403391 -1.2599786 0.3923044
H 2.3752763 -1.2726438 -0.575607
H 1.1968498 -1.6851317 1.3411131
O 0.1894464 2.6172664 3.3404378
H 0.2693664 2.6825694 4.3311305
H 1.0687701 2.827576 2.9861758
O -3.2097856 -1.1583934 3.3063624
H -2.6012925 -0.505691 2.8865254
H -4.0712949 -0.7278939 3.3234063
O 1.5495692 0.0628447 6.490238
H 1.0432934 -0.0064101 5.6331516
H 0.8533923 -0.1968409 7.1226524
O 3.9164067 -0.8149006 3.0240462
H 3.7983904 -1.4108475 2.2720266
H 3.6424429 -1.3040058 3.8208275
O 0.3633785 -2.7697408 4.4150233
H 0.1276079 -1.8215248 4.5713527
H 1.2480081 -2.8598161 4.8067908
O -3.3779176 2.2418664 1.001337
H -2.8920571 3.0857821 1.0500423
H -3.5057797 2.0065221 0.0675049
O -1.1297595 -0.3311721 7.2965793
H -1.6392886 -0.3064413 8.1121459
H -1.5299709 -1.0612825 6.7561449
O 0.5408814 2.674632 6.0160552
H -0.235656 2.6692035 6.5869759
H 1.0524571 1.8803476 6.2797599

O 2.9799819 -2.0346392 5.4616543
H 3.6505652 -2.4555246 6.0090913
H 2.6133975 -1.2829404 5.9850702
O -2.2013059 -2.2657665 5.7149759
H -1.4785796 -2.8331416 5.4038726
H -2.593512 -1.8945864 4.8977437
O -1.5709895 4.2230551 1.7213854
H -1.0177481 3.7117624 2.3382061
H -1.7103577 5.0827242 2.1291362
H 2.452745 -0.3245967 0.6589567

¹IN1 (G.S. geometry)

B3LYP-D3: -4141.745893772
Ti 0.1083221 -0.9979986 -2.24721
O -1.0607635 -2.7195958 -1.7702385
O -0.837799 -0.2894737 -0.6074639
O 1.906898 -1.3889986 -2.2321578
O -0.437931 -0.0393156 -3.7827659
H -1.8124005 -0.2769074 -0.741699
H 2.6431603 -0.8906148 -1.8045436
H 0.2809301 0.5583506 -4.1076848
H -1.581136 -2.9842484 -2.5602063
O -2.8177845 -2.6927414 0.3494636
H -2.9555619 -3.4957698 0.8613437
H -1.6424637 -2.7649655 -0.9717092
O 0.1284398 2.4438506 -1.1234878
H -0.304682 2.8595692 -1.915677

H -0.0739043 1.4881008 -1.1887928
O 4.0573763 -1.0159316 -4.0800174
H 3.2199669 -1.3596971 -3.7191494
H 4.4906601 -0.6540052 -3.2904318
O -2.5143077 1.2668245 -4.6280228
H -1.7152486 0.7312081 -4.3161019
H -2.4039463 1.3847694 -5.5781358
O -3.6297419 -0.5165428 -1.1328528
H -3.698802 -1.4145777 -0.7644946
H -3.9329038 -0.555981 -2.0606516
O -2.2840895 -2.6195781 -4.2298367
H -1.5973954 -2.0221957 -4.5611896
H -3.0812901 -2.0621151 -4.1679325
O 4.068922 0.0281599 -1.3368127
H 3.6914593 0.9267951 -1.3843701
H 4.2992547 -0.1460293 -0.4050453
O 1.453673 1.8624675 -4.4164338
H 1.9987947 2.0194812 -3.6213537
H 2.0956233 1.5835544 -5.1117076
O -0.7670179 3.3749136 -3.492731
H -0.0304355 3.024503 -4.0283102
H -1.5503763 2.8898928 -3.7985868
O -4.2739287 -0.5966969 -3.844776
H -5.1839465 -0.5067861 -4.1446501
H -3.7630981 0.1779657 -4.1941303
O 3.4445672 0.8612021 -5.9423247
H 3.446393 0.4695861 -6.8202926

H 3.7381588 0.1544364 -5.3155954
O 2.7214554 2.4166767 -1.901675
H 1.8333268 2.4768607 -1.4706843
H 3.1204802 3.2912137 -1.8418669
H -2.5545511 -1.9830949 1.0036308
Ti -0.3852756 -0.1383827 1.391131
O -0.0701884 -2.2822757 1.7804226
O 1.4227365 -0.0021014 0.8099848
O -2.2312809 -0.5257274 1.821255
O 0.1183454 0.1846391 3.3038246
H 2.2719215 0.2032298 1.2181757
H -3.0410038 0.0064566 1.6243661
H -0.6189085 0.3975516 3.8891203
H 0.2638142 -2.425361 2.6953096
O 2.154267 -2.5955125 0.3067305
H 2.1814877 -2.7751026 -0.6452377
H 0.6232553 -2.6723055 1.1998919
O -0.7491662 1.9998498 1.3332251
H -0.3457218 2.4788932 2.1119518
H -0.4143925 2.4053215 0.4934107
O -2.6761574 -0.6467904 4.4776976
H -2.5221443 -0.7032956 3.4997917
H -3.5617009 -0.2862303 4.5818516
O 2.3171727 1.3818271 4.1597572
H 1.4999084 0.957222 3.7674914
H 1.9952024 1.650549 5.035602
O 4.2805146 -0.9278703 1.3103115

H 3.8403136 -1.768093 1.1053638
H 4.2918816 -0.873796 2.2848273
O 1.0684444 -2.0425066 4.335284
H 0.7798354 -1.1493167 4.0017537
H 2.0378963 -1.9955258 4.3292352
O -4.5330229 0.792819 1.2070822
H -4.3093512 1.742631 1.2439193
H -4.4689504 0.5357673 0.2716483
O 0.1761245 1.9643886 5.9632365
H -0.0111102 2.4341881 6.7817292
H -0.1057239 1.0206637 6.1150522
O 0.252927 3.3317362 3.4305294
H -0.0128598 2.9194926 4.2759874
H 1.2023277 3.136306 3.4006809
O 3.8082956 -0.858171 4.0769021
H 4.5075141 -0.9029487 4.7365909
H 3.3602883 0.0201565 4.1905385
O -0.5916009 -0.5917165 6.2837377
H 0.0319101 -1.2488843 5.9304562
H -1.4073275 -0.7137117 5.7542583
O -3.301725 3.2695123 1.3098485
H -2.4215662 2.8515693 1.3568428
H -3.3098799 3.9554644 1.9842568
H 1.9603818 -1.6264136 0.3470741

¹IN2 (G.S. geometry)

B3LYP-D3: -4141.805845031

Ti -0.1999327 -1.3715932 -2.0368604
O 0.0387469 -3.157054 -2.393204
O -0.945587 0.1291942 -0.8443582
O 1.5775995 -1.0078204 -2.0296754
O -0.6444966 -0.4111498 -3.6030148
H -1.9155257 0.206243 -0.7424463
H 2.1966053 -0.2642911 -1.8442112
H 0.0328446 0.2559674 -3.8626223
H -0.7230438 -3.7409494 -2.4658043
O -3.3766017 -2.5984688 0.3097406
H -3.7075415 -3.3766862 0.7657686
H -0.0840438 -1.7644287 -0.1661044
O -0.2342801 2.6669214 -1.2603943
H -0.728217 2.9958383 -2.0499526
H -0.4035547 1.6915148 -1.266795
O 3.6379713 -1.0571693 -4.0908736
H 2.8269972 -1.4051134 -3.6874887
H 3.966823 -0.4503137 -3.4038961
O -2.7089718 0.8057191 -4.5569051
H -1.9100434 0.2809863 -4.2108124
H -2.6865152 0.7332331 -5.5168336
O -3.9043449 0.1321449 -0.3215325
H -3.9663228 -0.8123856 -0.0918857
H -4.3187959 0.1875554 -1.1990086
O -2.3284246 -2.047933 -2.0467442
H -2.7031665 -2.4082724 -1.2039937
H -3.0433827 -1.5206924 -2.4562131

O 3.7205811 0.6515441 -1.8032007
H 3.3777877 1.5675811 -1.8077606
H 4.0907856 0.4629503 -0.9231683
O 1.0273994 1.6630547 -4.3999744
H 1.6080151 2.0659015 -3.7284026
H 1.6159374 1.3478734 -5.124898
O -1.2556087 3.1965589 -3.7489372
H -0.4743508 2.8070704 -4.1865268
H -1.9690142 2.5767154 -3.974573
O -4.4412936 -0.4595946 -3.0140939
H -5.2073597 -0.8432683 -3.4512008
H -3.9450858 0.0744238 -3.6879396
O 2.8299359 0.5740015 -6.1140138
H 2.6395077 0.0859408 -6.9203963
H 3.1924766 -0.0791001 -5.4671685
O 2.2650341 2.9284235 -2.1590668
H 1.3817403 2.8846014 -1.6998844
H 2.5026538 3.8578677 -2.2310899
H -2.7144384 -2.1151764 0.9202661
Ti -0.1101577 -0.4162451 1.0262559
O 0.8225171 -2.2753341 1.7397556
O 1.4889869 0.2506253 0.5575229
O -1.8527477 -1.0798757 1.6321878
O 0.264033 0.0569797 2.9533345
H 2.4556807 0.1387718 0.6397558
H -2.4827299 -0.3676177 1.8778133
H -0.5155884 -0.0151253 3.5194539

H 1.2917923 -2.216429 2.6021124
O 2.7410573 -2.7651235 -0.219806
H 2.8806491 -3.6291112 -0.6219984
H 1.4544082 -2.629832 1.0802002
O -0.970904 1.6547717 1.2401671
H -0.6937854 2.0702177 2.0830399
H -0.7217869 2.2484561 0.5034473
O -2.1998881 -1.2478043 4.3504765
H -1.9880193 -1.4884808 3.4232843
H -2.9790182 -0.6825008 4.2564709
O 2.1938085 1.8069419 3.6682381
H 1.475821 1.2551749 3.2619367
H 1.8913033 1.8722669 4.590273
O 4.1476287 -0.4709708 0.7267111
H 3.9358213 -1.4066937 0.5647923
H 4.3164198 -0.3524929 1.684515
O 1.9122196 -1.5894233 4.166537
H 1.3609085 -0.8559607 3.7724624
H 2.8277639 -1.2692572 4.1204788
O -3.8730551 0.6420057 2.5446731
H -3.6394987 1.5784865 2.7110815
H -4.2542646 0.648169 1.6522204
O 0.3309382 1.7564706 5.8377921
H 0.2138778 2.1609401 6.7030687
H 0.1638014 0.7749958 5.9604755
O -0.4745454 3.1476087 3.5853708
H -0.3482035 2.61712 4.4036659

H 0.4086726 3.4914578 3.3986716
O 4.204468 0.0518609 3.444987
H 4.9977973 0.3161822 3.921134
H 3.5473013 0.7960998 3.5368265
O -0.0718868 -0.8416035 6.1098111
H 0.6709968 -1.3085088 5.6876889
H -0.8678676 -1.093949 5.5985405
O -3.1440703 3.3675036 2.601564
H -2.8843438 3.2931285 1.6750091
H -2.2999112 3.4987375 3.0671396
H 2.3126294 -2.218288 -0.9196211

¹PC (G.S. geometry)

B3LYP-D3: -4141.862071568
Ti -0.2904284 -0.6075144 -2.2833354
O 0.7398288 -2.1678651 -2.6423586
O -0.9988805 -0.1596376 -0.5500712
O 1.1330002 0.3924038 -2.1799231
O -1.1343402 0.1968471 -3.6989939
H -1.9821471 -0.1288335 -0.5209194
H 2.0811719 0.6803487 -2.2447527
H -0.5550082 0.8983043 -4.1500268
H 0.2262439 -2.9529043 -2.8667692
O -1.7544781 -3.2385446 0.1386289
H -0.7973918 -3.2751221 0.299102
H 0.6761214 -4.987299 -0.5898366
O 0.0425366 2.9000054 -0.5780714

H -0.7428336 2.6602494 -1.1238558
H 0.4908771 2.0644073 -0.3093661
O 3.1192439 -1.2126255 -4.0400262
H 2.2839237 -1.5173437 -3.6387055
H 3.4171443 -0.4635052 -3.4966026
O -3.5770522 0.0584455 -4.7572388
H -2.6329375 0.094679 -4.4542112
H -3.5611394 -0.1268833 -5.700465
O -3.6828994 0.3667621 -0.6901719
H -4.1219619 0.5116553 0.1659938
H -4.2326628 -0.2015549 -1.2600432
O -1.9079621 -2.1272418 -2.2547777
H -1.915844 -2.6013089 -1.368335
H -2.8398829 -1.9199331 -2.4721743
O 3.6306 1.1494123 -2.3353143
H 3.3544375 2.0834766 -2.364749
H 3.8662857 0.9067125 -1.4126663
O 0.4060858 2.017284 -4.770218
H 0.8992382 2.4736505 -4.0641745
H 1.0780944 1.5256232 -5.2918582
O -2.2679064 2.4649862 -2.020783
H -2.071412 1.9914651 -2.8407315
H -2.8772592 1.8700341 -1.5464752
O -4.5301813 -1.3085866 -2.7266697
H -5.2607429 -1.9147852 -2.8804463
H -4.3497161 -0.8190704 -3.5746984
O 2.2820868 0.4664529 -6.0692737

H 2.1493872 0.0291454 -6.9150098
H 2.6144271 -0.2173989 -5.445075
O 1.8628318 3.3562416 -2.5815044
H 1.2190118 3.2830585 -1.8313054
H 1.9954416 4.298402 -2.7282598
H -2.0265726 -2.5375361 0.7726577
Ti -0.1438872 -0.3410285 1.3837221
O 0.7004176 -1.9181059 1.02104
O 1.2063608 0.8744591 0.822764
O -1.9053608 -1.0410404 1.8469892
O 0.3031926 -0.1169874 3.182602
H 2.0148308 0.4723962 0.4354266
H -2.698924 -0.4822902 1.8255313
H -0.3414995 -0.4890256 3.8131284
H 1.3943833 -2.2512776 1.6163335
O 3.1128437 -2.5126192 -1.1850396
H 3.6283272 -2.5762389 -1.9995274
H 1.2124427 -4.4644365 -0.5514919
O -1.2199605 1.8076856 1.6735149
H -0.9335356 2.0966847 2.5648564
H -0.8330582 2.4261188 1.014807
O -1.7465328 -1.4276736 4.6018182
H -1.9196586 -1.5607149 3.6435779
H -2.3547968 -0.7109272 4.8508858
O 2.6033338 1.772273 3.1127314
H 2.0331558 1.5728052 2.3384356
H 2.0834406 1.4940714 3.885701

O 3.6668011 -0.1426156 0.029083
H 3.5212017 -1.0580631 -0.3090468
H 4.112431 -0.1988084 0.8972815
O 2.3843096 -1.8736676 3.4920304
H 1.7675225 -1.1163728 3.3957287
H 3.2637969 -1.4931467 3.3283776
O -3.8815213 1.0974292 1.9486969
H -3.9891638 1.0617524 2.9128946
H -3.0286543 1.5580403 1.8150755
O 1.1803519 1.1623817 5.624675
H 1.5707707 1.4795181 6.445222
H 1.0645982 0.1751862 5.7294737
O -0.8653054 2.4550606 4.3878621
H -0.1173335 1.9636247 4.8155586
H -0.7275091 3.3863622 4.5916649
O 4.4746642 -0.0634274 2.6907455
H 5.3493537 0.1665133 3.0178095
H 3.8632808 0.6960834 2.9113814
O 0.8714976 -1.487194 5.8985337
H 1.4836886 -1.8844741 5.2504778
H -0.0238036 -1.7169821 5.5932471
O -3.2254503 1.051758 4.7737727
H -3.731679 1.3522344 5.5353967
H -2.43273 1.6295636 4.7165837
H 2.1897587 -2.4379583 -1.5012898

References

- [1] Berardo, E.; Hu, H.-S.; Shevlin, S. a.; Woodley, S. M.; Kowalski, K.; Zwijnenburg, M. A. *J. Chem. Theor. Comp.* **2014**, *10*, 1189–1199.
- [2] Berardo, E.; Hu, H.-S.; van Dam, H. J. J.; Shevlin, S. A.; Woodley, S. M.; Kowalski, K.; Zwijnenburg, M. A. *J. Chem. Theor. Comp.* **2014**, *10*, 5538–5548.