

Interfacial Interaction of Ag Nanoparticles with Graphene Oxide
supports for Improving NH₃ and NO Adsorption: a First-Principles
Study

Supporting Information

Shaobin Tang,^{*,†} Weihua Wu,[†] and Jianping Yu^{*,‡}

[†]Key Laboratory of Organo-Pharmaceutical Chemistry of Jiangxi Province, Gannan Normal University, Ganzhou 341000, China

[‡]School of life and environmental sciences, Gannan Normal University, Ganzhou 341000, China

**E-mail address:* tsb1980@xmu.edu.cn, yujianping402871@sina.com

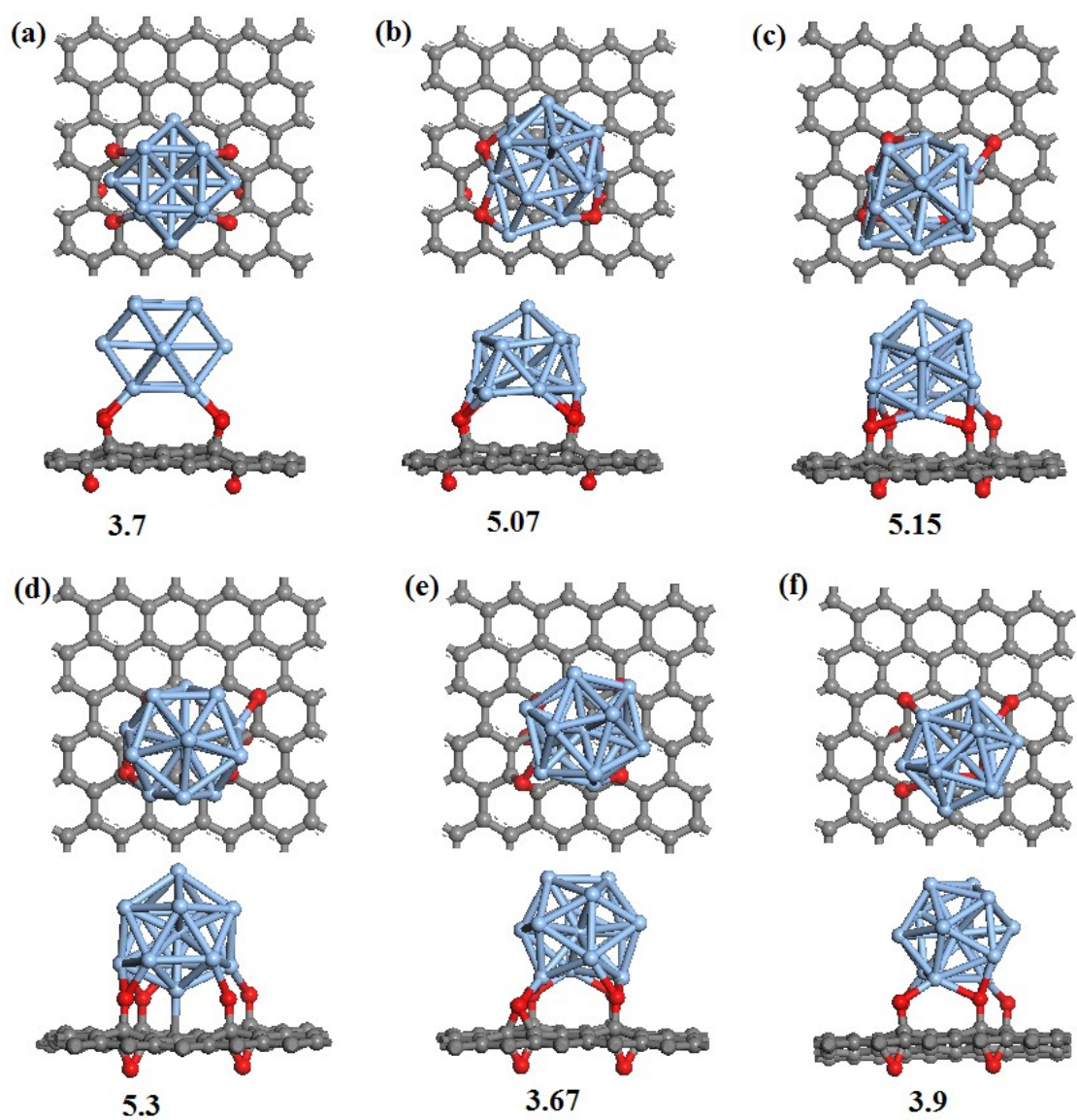


Figure S1. Top and side views of optimized structures and corresponding binding energies (in eV) for Ag_{13} /GOs with different direction of NPs relative to graphene surface: Ag_{13} with initial (a)-(c) cuboctahedron and (d)-(f) icosahedron structures.

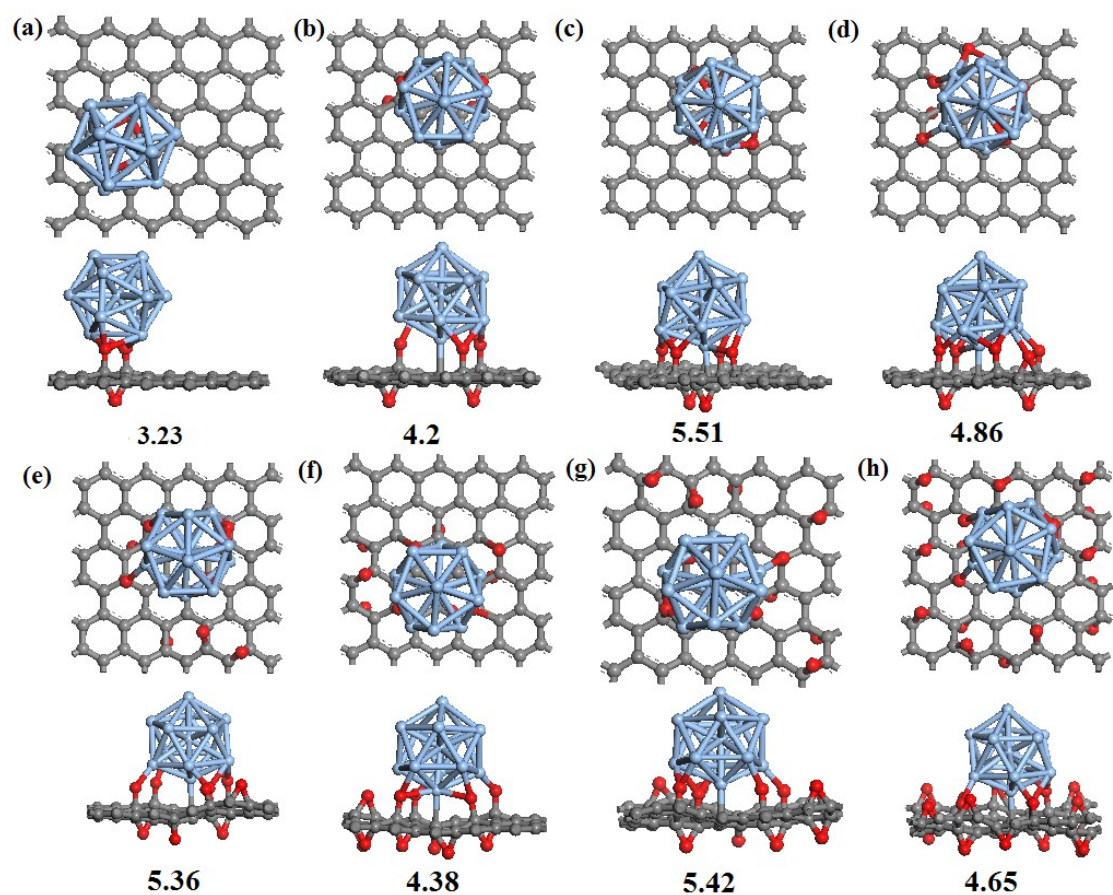


Figure S2. Top and side views of optimized structures and corresponding binding energies (in eV) for Ag_{13} /GOs with different oxidation level. (a) GO-30, (b) GO-50, (c) GO-60-2, (d) GO-70, (e) GO-90, (f) GO-120, and (h) GO-200.

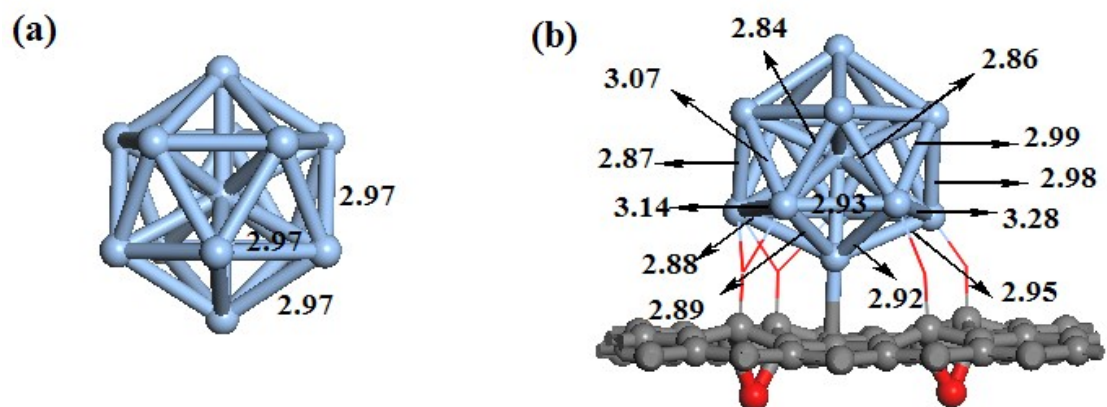


Figure S3. The changes in Ag–Ag bond length (in Å) of Ag₁₃ after deposition on GO-6O-1. The Ag–O and O–C bonds are shown with red solid line.

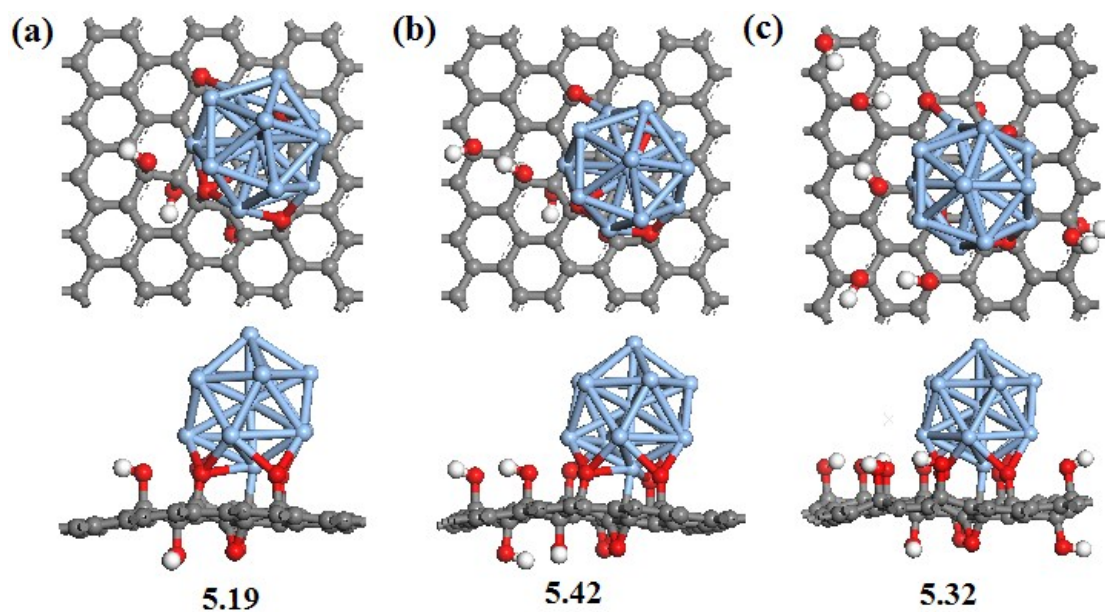


Figure S4. Top and side views of optimized structures and corresponding binding energies (in eV) for Ag_{13} deposited on GO-6O-1 with additional hydroxyl groups. (a) 2, (b) 4, and (c) 8 OH groups.

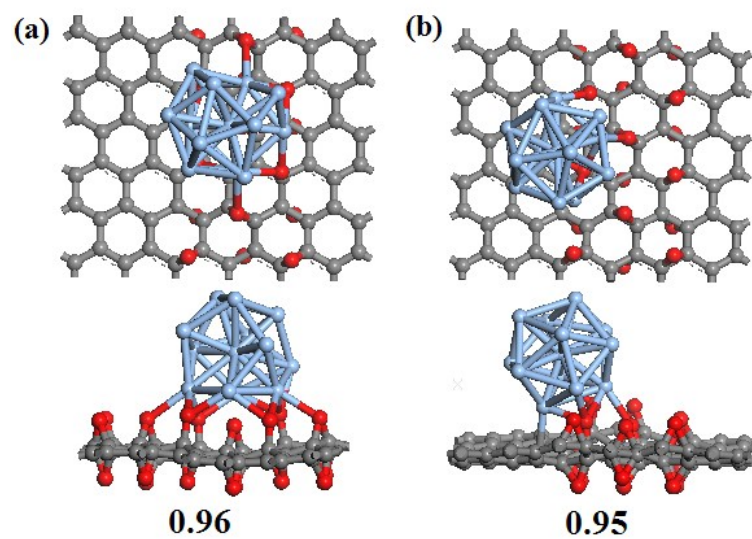


Figure S5. Top and side views of optimized structures and corresponding binding energies (in eV) for Ag₁₃/GO3 with two binding models.

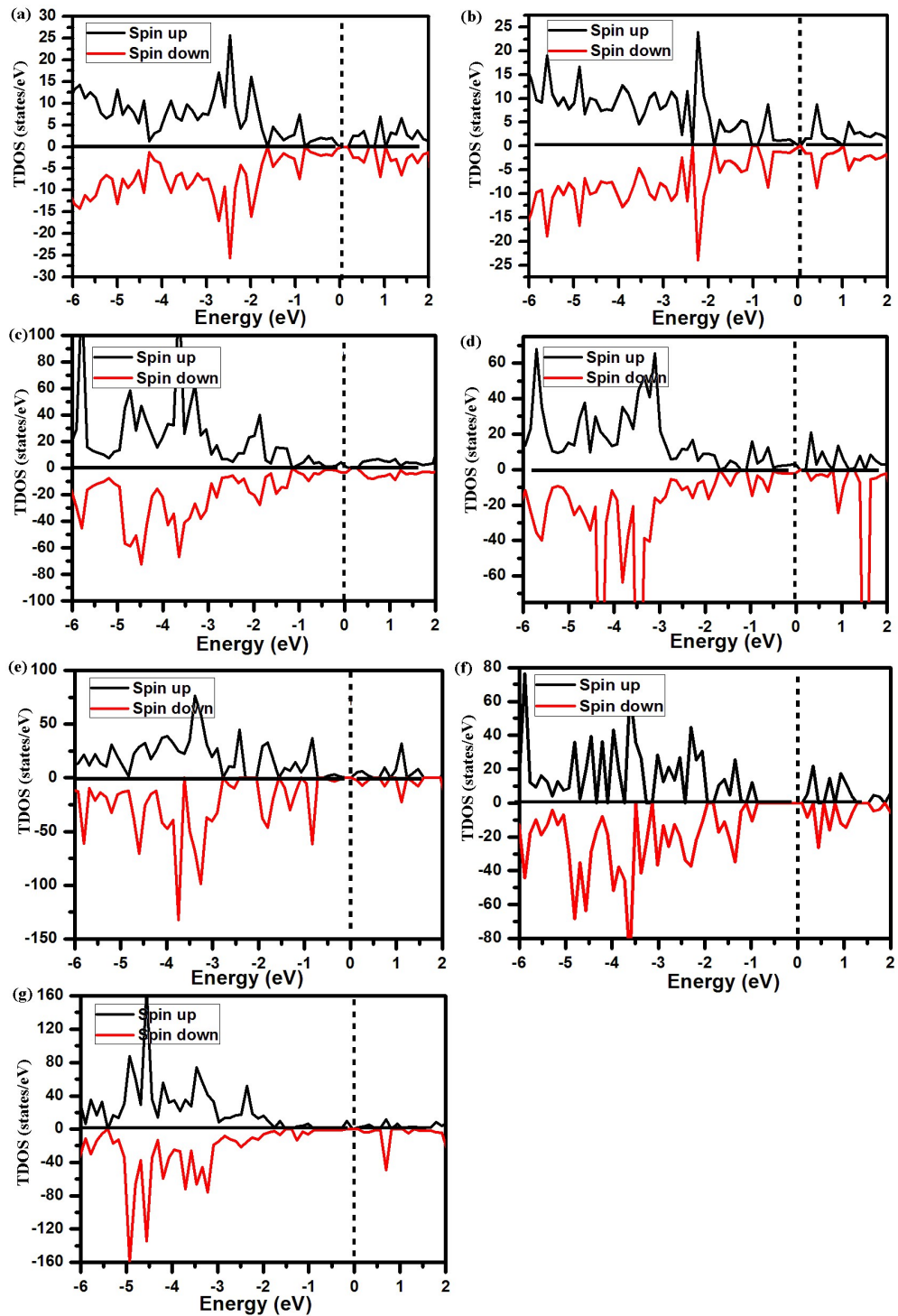


Figure S6. The spin-polarized TDOS of GOs and Ag₁₃-GOs hybrid with different oxygen concentration. (a) GO-6O-1, (b) GO-6O-2, (c) Ag₁₃/GO-6O-1, (d) Ag₁₃/GO-6O-2, (e) Ag₁₃/GO-14O-1, (f) Ag₁₃/GO-17O, and (g) Ag₁₃/GO₂. The Fermi level is set to 0.

Table S1. Summary of calculated results for NH₃ and NO adsorptions on selected GOs models: the adsorption energy (E_{ad}), the distance between adsorbate and binding site of GOs (d), and the charge transfers from the molecule to GOs (ΔQ).

substrate	adsorbate	binding site*	E_{ad} (eV)	d (Å)	ΔQ (e)
GO-6O-1	NH ₃	O	-0.13	2.49	0.04
	NO	No	-0.15	2.61	0.17
GO-14O-1	NH ₃	O	-0.18	2.43	0.06
GO-5O-2OH	NH ₃	H from OH	-0.56	1.76	0.09
	NH ₃	O from OH	-0.15	2.33	0.01
	NO	H from OH	-0.12	2.51	0.12

* Selected geometrical structures are shown in Figure S7.

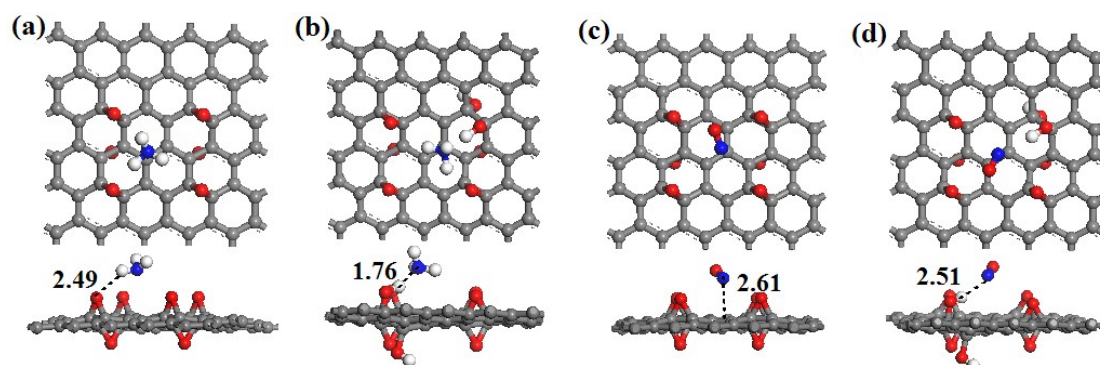


Figure S7. Top and side view of geometrical structures (distance in Å) for adsorption of NH₃ and NO on pristine GOs. (a, c) GO-6O-1 and (b, d) GO-5O-2OH.

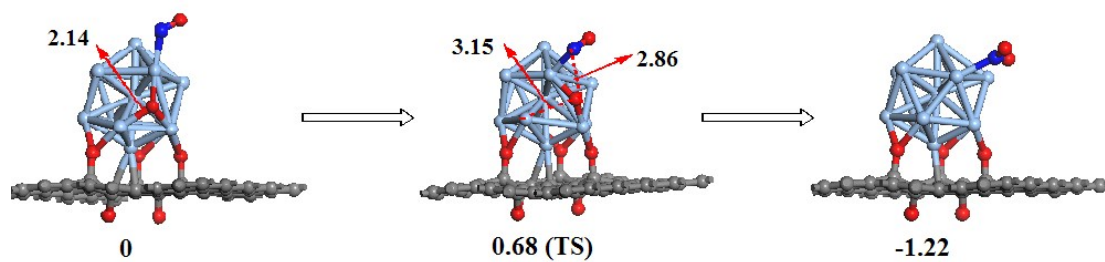


Figure S8. The geometrical structures (distance in Å) of initial state, transition state (TS), and final state for oxidation of adsorbed NO to NO₂ on Ag₁₃O/GO-50. All energies (in eV) are relative to initial state.

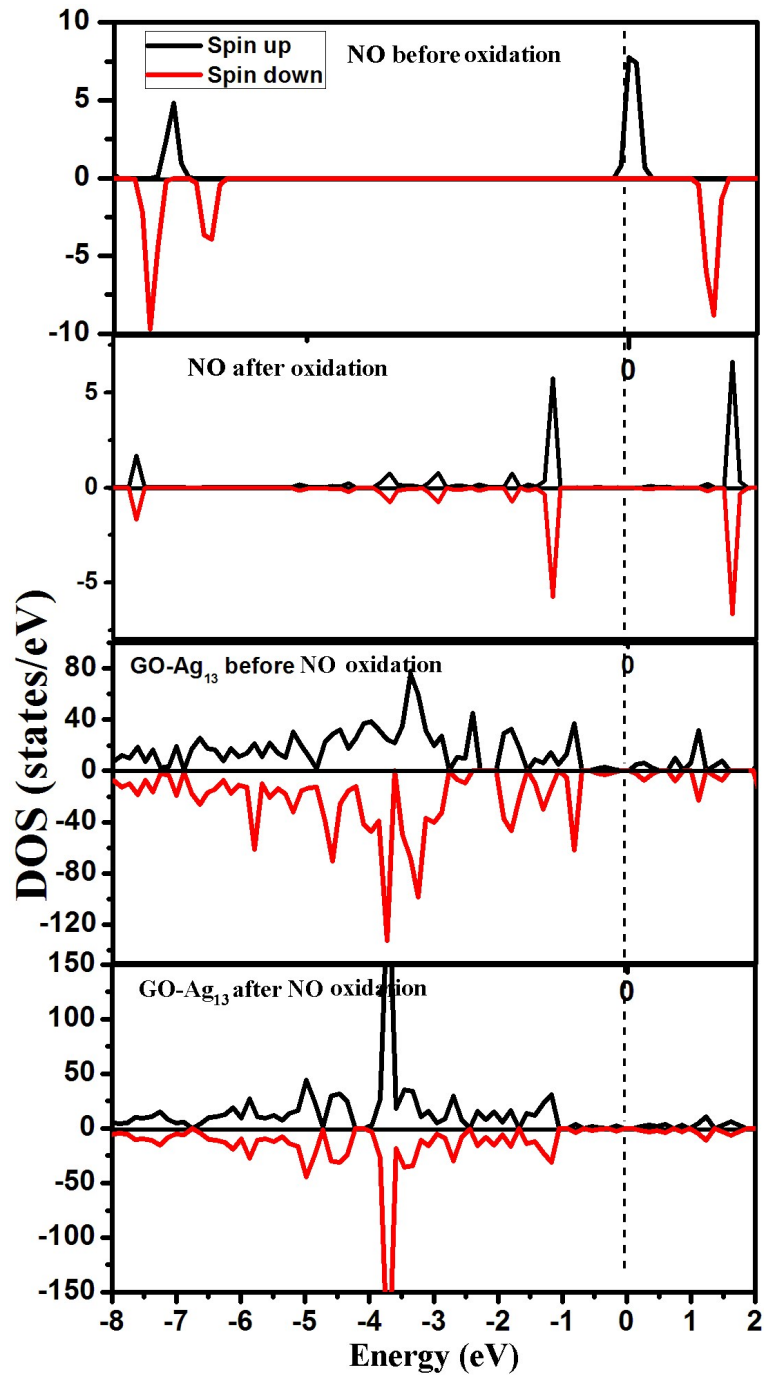


Figure S9. The spin-polarized TDOS and PDOS of NO and Ag₁₃/GO-14O-1 before and after NO oxidation on hybrid. The Fermi level is set 0.