Oxidative coupling of aniline and desulphurization over nitrogen rich mesoporous carbon

Reena Goyal,^a Deepa Dumbre,^b L. N. Sivakumar Konathala,^a Monica Pandey,^a and Ankur Bordoloi*^a

Figure S1



Figure S1 Wide angle XRD of WOx/MCNx (5).

Figure S2



Figure S2. (a-b) SEM images WOx/MCNx; c) EDAX analysis of the WOx/MCNx catalyst and (d-f) is the elemental mapping of carbon, nitrogen and tungsten respectively.



Figure S3. TPD pattern of WOx/MCNx (2.5 & 5).

Figure S4



Figure S4. Effect of H_2O_2 as a function aniline conversion and product selectivity. (**•**) conversion of DBT; (**•**) selectivity of azo-benzene; (**•**) selectivity of nitroso-benzene and (**v**) selectivity of azooxy-benzene.

Figure S3





Figure S5. GC-MS of DBTSO (screen shot).

Figure S6



Figure S6. Effect of H_2O_2 as a function of DBT conversion and product selectivity after 2h time-on-steam. (**n**) conversion of DBT; (**•**) selectivity of DBTS and (**\triangle**) selectivity of DBTSO.

Table S1

Catalyst	X _{BET} (m ² /g)	$X_{\rm P}$ (cc/g)	X _{Pd} (nm)	ICP	CHN ^[a]		XPS [a]	
					С	Ν	С	N
MNCx	357	0.61	3.8	-	62.7	13.38	65.8	14.02
WOx/MNCx (5)	327	0.36	3.5	262.50	61.4	14.7	64.1	12.05

Table S1. Physico-chemical	parameters of	WOx/	MNCx
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 $X_{\text{BET}} = \text{BET}$ surface area in m²/g; X_p = pore volume in cc/g; X_{pd} = pore diameter in nm; ICP is in (W/ppm); [a] in mole %.

Scheme S1



Scheme S1 Plausible reaction pathway for Oxidative desulphurisation of Dibenzothiophene over WOx/MCNx.