

**A novel polynoborn-base chemosensor fluorescence sensing of Zn²⁺
and Cd²⁺ and subsequent pyrophosphate in absolute aqueous solution**

Pei-Shui Yao,[†] Zhanghua Liu,[‡] Jin-Zhu Ge,[†] Yong Chen,[‡] Qian-Yong Cao^{†,*}

[†]Department of Chemistry and [‡]Institute for Advanced Study, Nanchang University,
Nanchang 330031, P. R. China

Fax: (+86)791-83969386; E-mail: cqyong@ncu.edu.cn

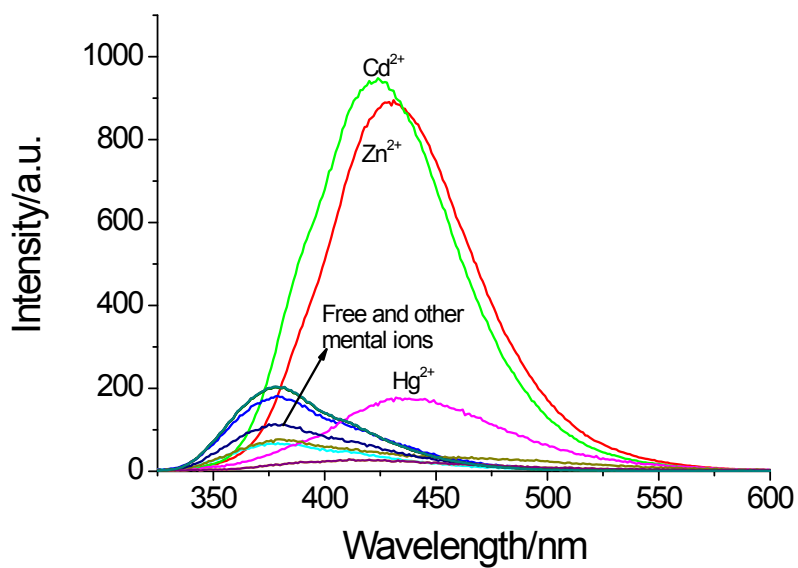


Figure S1. Emission spectra of **1** upon the addition of various metal ions (2 equiv) in CH₃CN solution

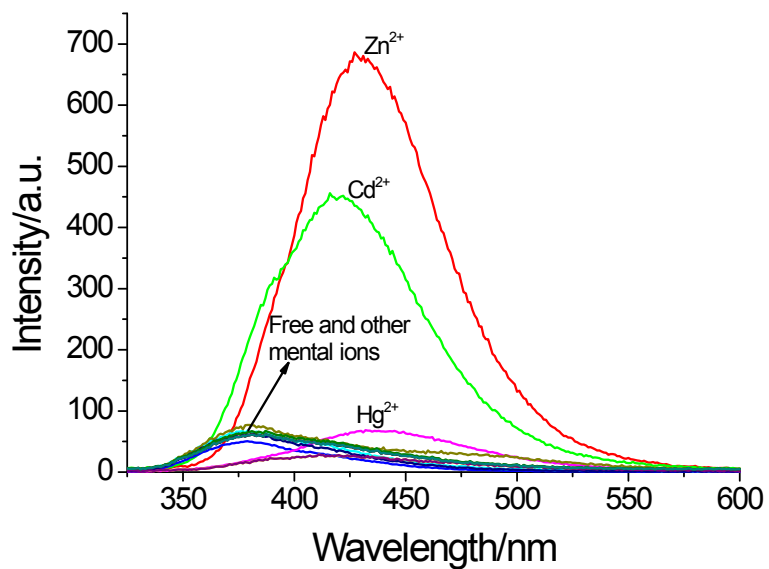


Figure S2 Emission spectra of **P1** upon the addition of various metal ions (2 equiv) in CH₃CN solution.

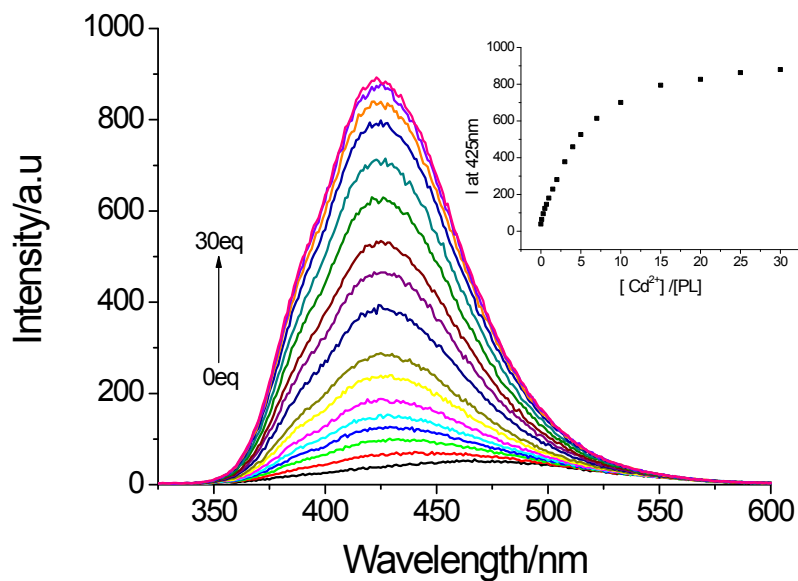


Figure S3. Emission titration of **P1** (20 mM) upon the addition of various Cd²⁺ ions in Tris-HCl solution (pH = 7.4). Inset: emission intensity at 425 nm upon titration with Cd²⁺.

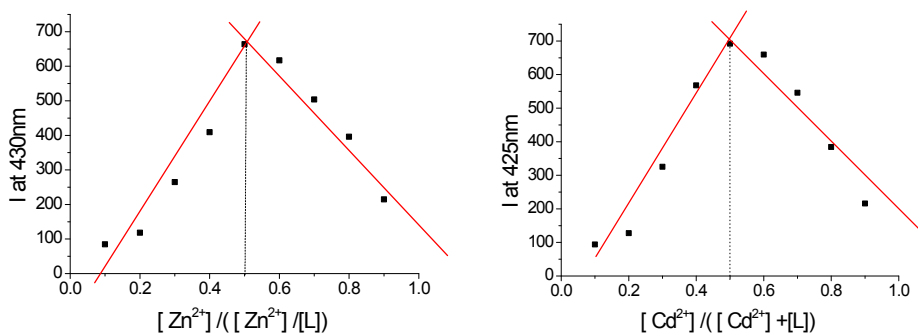


Figure S4. Job plots of **P1** with Zn²⁺ (left) and Cd²⁺ (right).

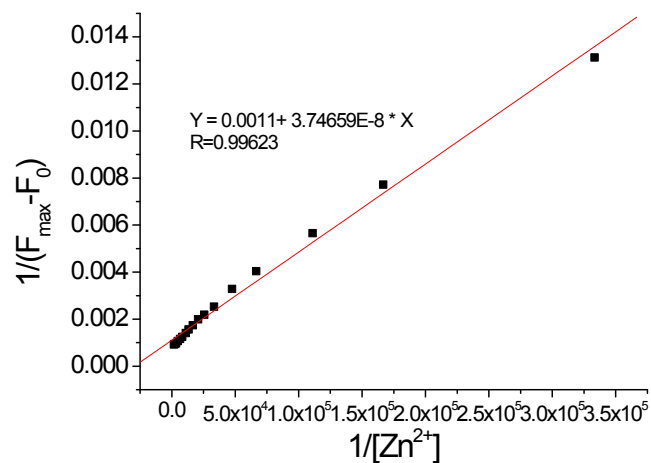


Figure S5. The calculated binding constants of **P1-Zn²⁺**.

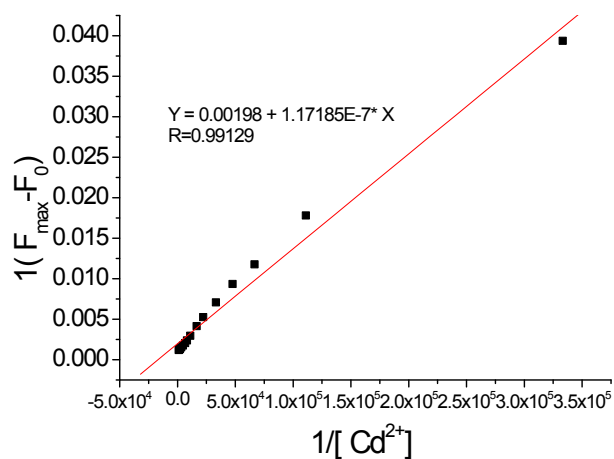


Figure S6. The calculated binding constants of **P1-Cd²⁺**.

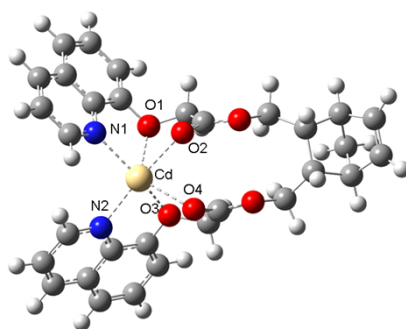


Figure S7. Optimized conformation of **1**-Cd²⁺ by DFT calculation. Selected distances (Å): Cd–N1, 2.320; Zn–N2, 2.328; Zn–O1, 2.478; Zn–O2, 2.366; Zn–O3, 2.512; Zn–O4, 2.324.

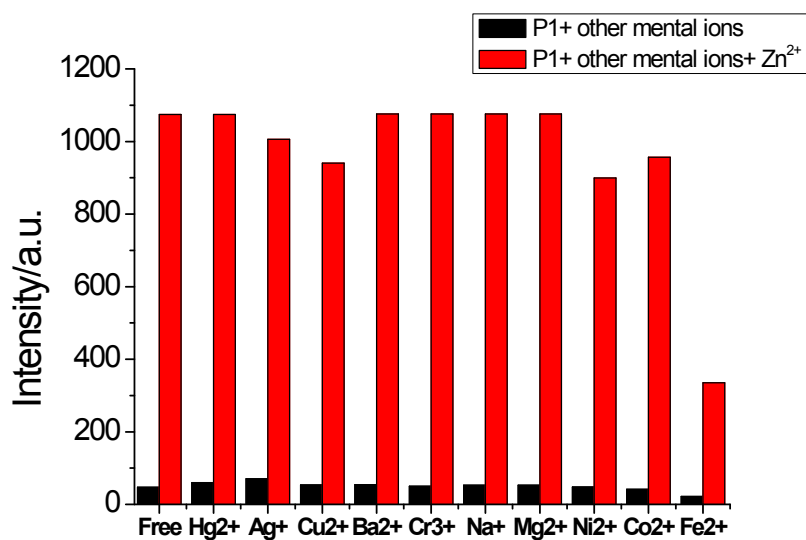


Figure S8. Competition experiments

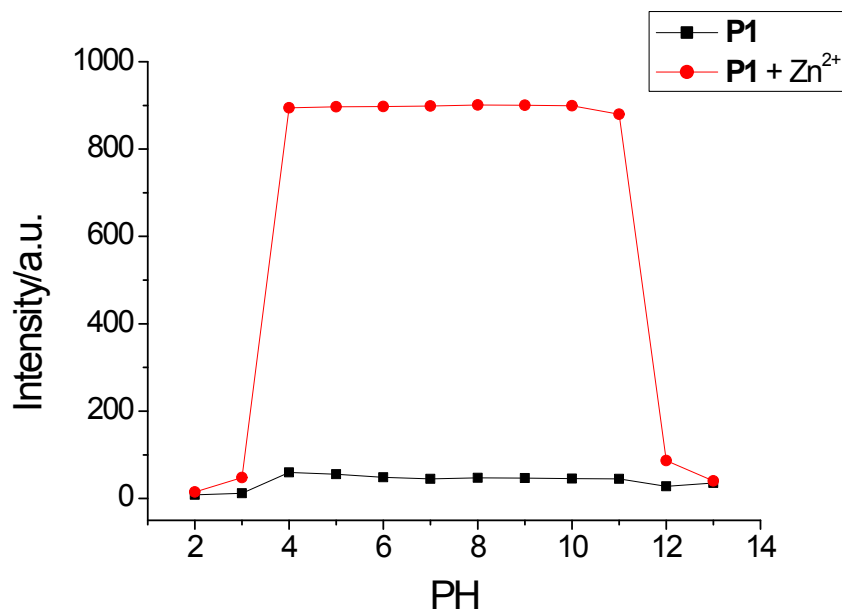


Figure S9. Fluorescence intensities of **P1** (20 μm) at 425 nm before and after the additional of Zn^{2+} (5 equiv) at various pH values in H_2O solution.

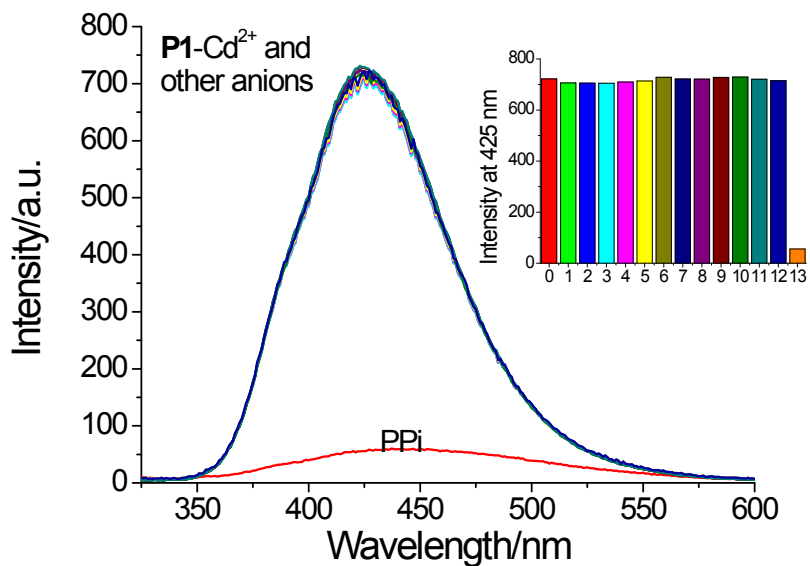


Figure S10. The emission spectra of **P1-Cd²⁺** upon the addition of various anions (20 equiv) in aqueous solution. Inset shows the emission intensity at 425 nm upon the addition of various anions (0-**P1-Cd²⁺**, 1- F^- , 2- Cl^- , 3- Br^- , 4- I^- , 5- NO_3^- , 6- CH_3COO^- , 7- CO_3^{2-} , 8- HSO_4^- , 9- Pi , 10- ATP , 11- AMP , 12- ADP , 13- PPI).

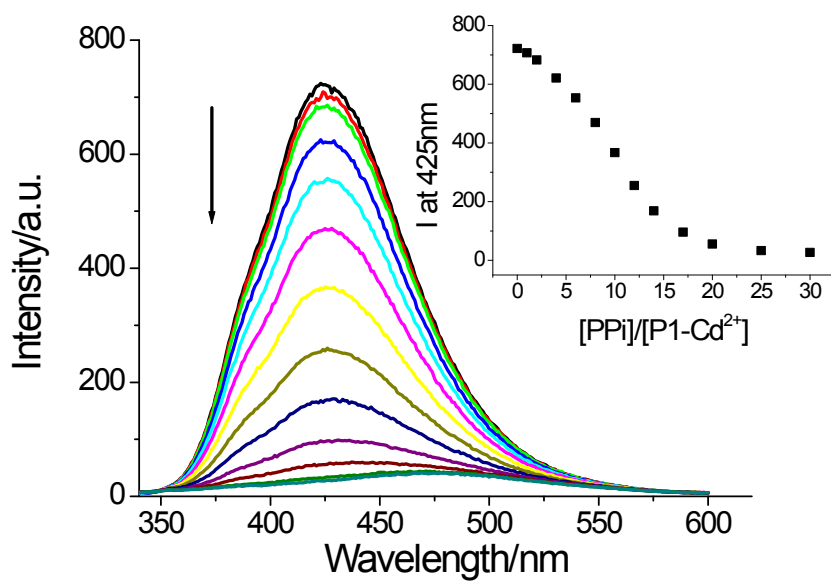


Figure S11. The emission titration of **P1-Cd²⁺** upon the addition of PPI in 100% aqueous solution

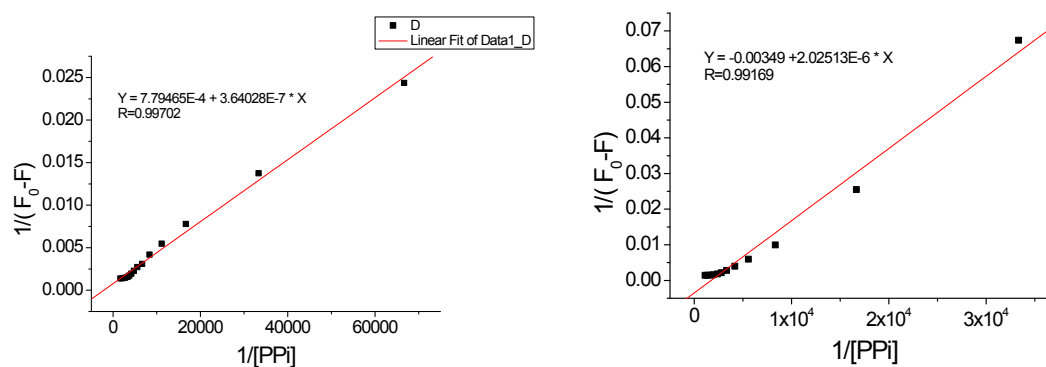


Figure S12. Benesi-Hildebrand plot of for evaluation of binding constants of PPI with **P1-Zn²⁺** (left) and **P1-Cd²⁺** (right) with the 1:1 binding stoichiometry in aqueous (PH-7.4) solution.

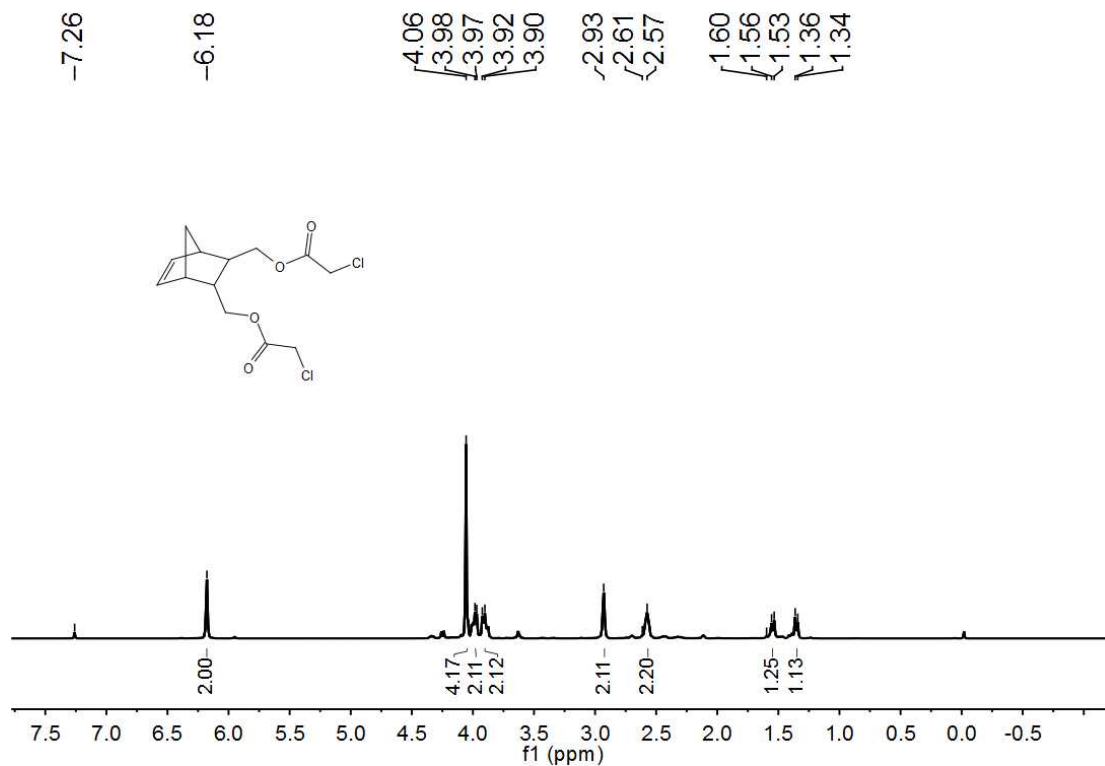


Figure S13. ¹H-NMR of **4** in CDCl₃ solution

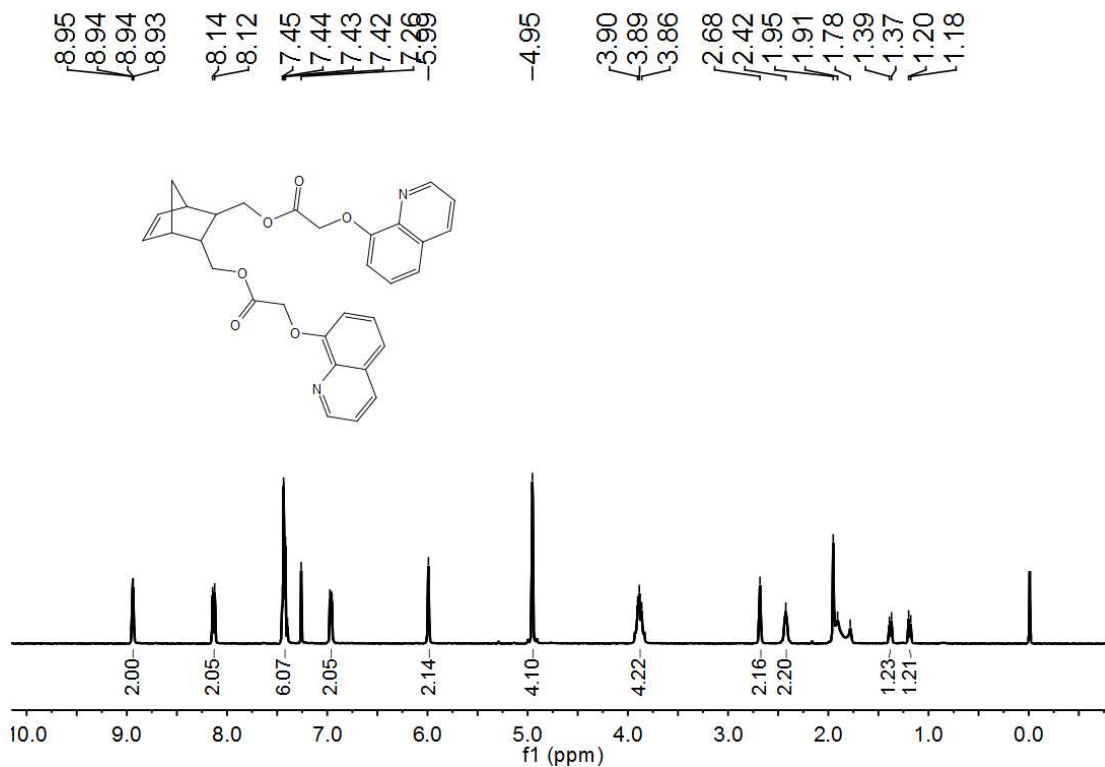


Figure S14. ¹H-NMR of **1** in CDCl₃ solution

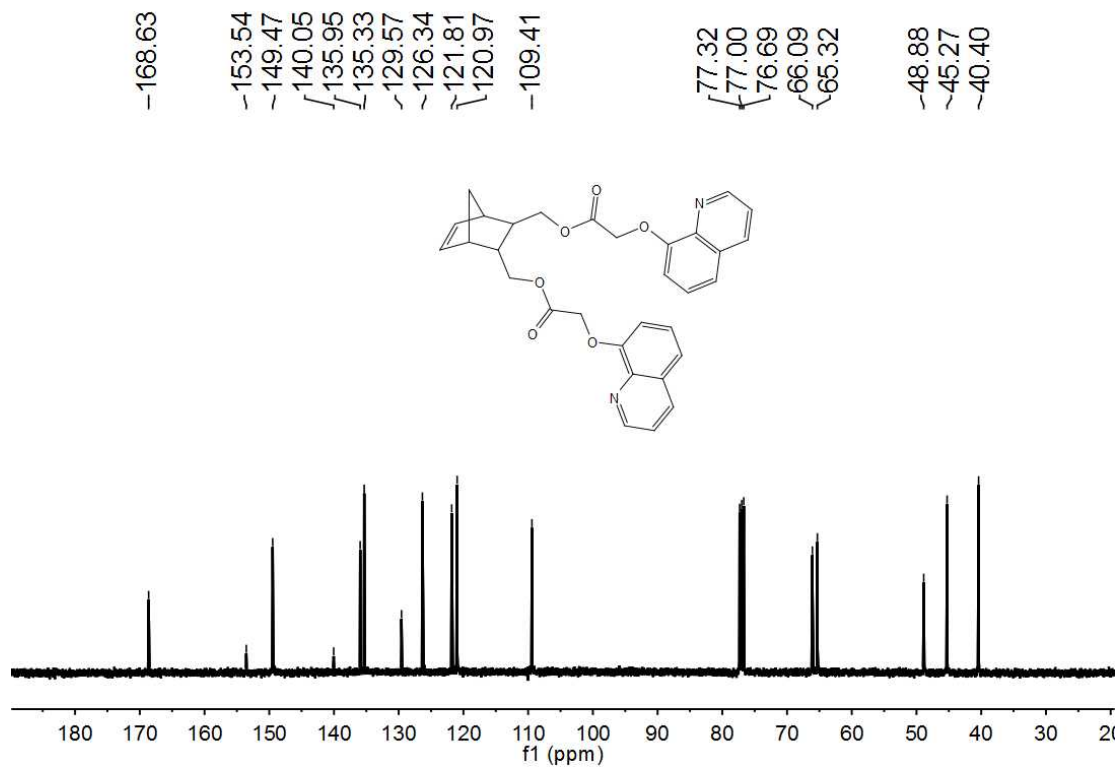


Figure S15. ¹³C-NMR of **1** in CDCl₃ solution.

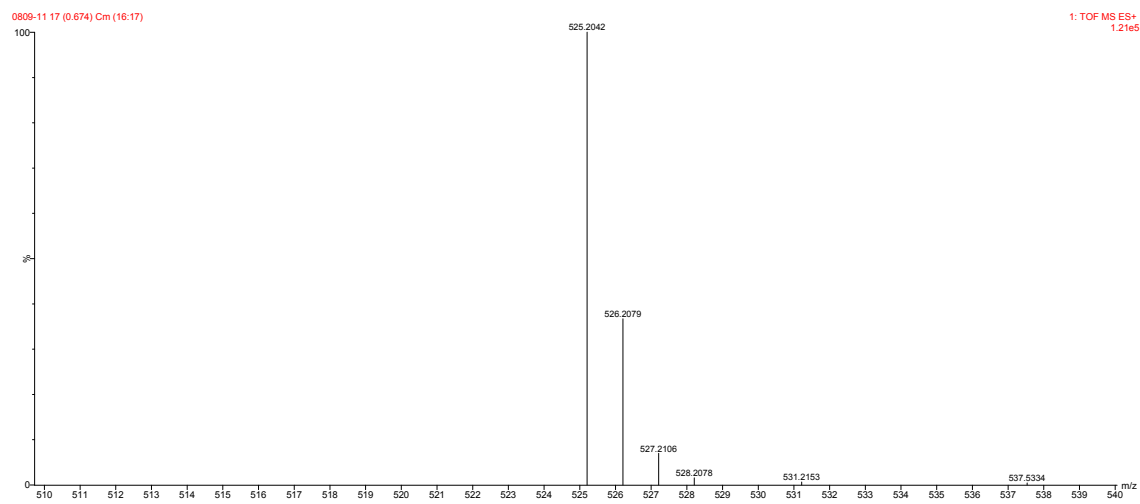


Figure S16. ESI-MS of **1**.

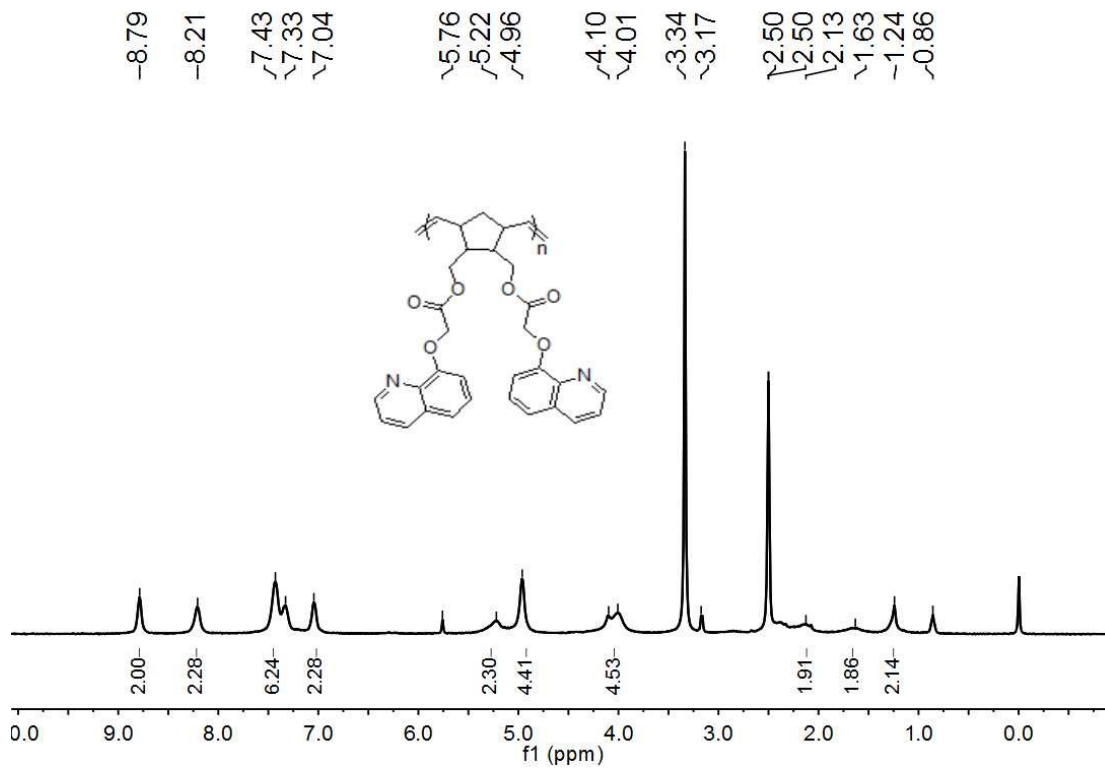


Figure S17. $^1\text{H-NMR}$ of **P1** in DMSO.