## **Supporting Information**

## **Co-sensitization** Promoted Light Harvesting with a New Mixed-

Addenda Polyoxometalate  $[Cu(C_{12}H_8N_2)_2]_2[V_2W_4O_{19}]\cdot \frac{4}{4}H_2O$  in Dye-

## **Sensitized Solar Cells**

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bond	bond -length	bond-dist	$\left[\exp\left[\frac{r_0-r}{B}\right]\right]$	$BVS = \Sigma exp[(r_0 - r)/B]$
W(1)-O(2)	1.91	1.96	0.873	
W(1)-O(3)	1.91	2.296	0.351	
W(1)-	1.91	1.875	1.099	( 250
W(1)-O(7)	1.91	1.974	0.841	6.259
W(1)-O(8)	1.91	1.661	1.960	
W(1)-O(9)	1.91	1.863	1.135	
bond	bond -length	bond-dist	$[exp[(r_0-r)/B]]$	$BVS = \Sigma exp[(r_0 - r)/B]$
W(2)-O(1)	1.91	1.894	1.044	
W(2)-O(3)	1.91	2.314	0.335	
W(2)-O(4)#1	1.91	1.928	0.952	( 252
W(2)-O(7)#1	1.91	1.956	0.883	0.232
W(2)-O(9)	1.91	1.895	1.041	
W(2)-O(10)	1.91	1.654	1.997	
bond	bond -length	bond-dist	$[exp[(r_0-r)/B]]$	$BVS = \Sigma exp[(r_0 - r) / B]$
W(3)-O(1)	1.91	1.917	0.981	
W(3)-O(2)	1.91	1.886	1.067	
W(3)-O(3)	1.91	2.285	0.363	6.051
W(3)-O(4)	1.91	1.889	1.058	0.031
W(3)-O(5)	1.91	1.943	0.915	
W(3)-O(6)	1.91	1.721	1.667	
bond	bond -length	bond-dist	$[exp[(r_0-r)/B]]$	$BVS = \Sigma exp[(r_0 - r)/B]$
V(1)-O(2)	1.803	1.96	0.65	
V(1)-O(3)	1.803	2.296	0.263	
V(1)-O(5)#1	1.803	1.875	0.823	1 680
V(1)-O(7)	1.803	1.974	0.630	4.009
V(1)-O(8)	1.803	1.661	1.468	
V(1)-O(9)	1.803	1.863	0.850	
bond	bond -length	bond-dist	$\left[\exp\left[\left(r_{0}-r\right)/B\right]\right]$	BVS= $\Sigma exp[(r_0 - r)/B]$
V(2)-O(1)	1.803	1.894	0.782	
V(2)-O(3)	1.803	2.314	0.251	
V(2)-O(4)#1	1.803	1.928	0.713	4 (00
V(2)-O(7)#1	1.803	1.956	0.661	4.683
V(2)-O(9)	1.803	1.895	0.780	
V(2)-O(10)	1.803	1.654	1.496	

Table S1 Bond Valence Sums

bond	bond -length	bond-dist	$[exp[(r_0-r)/B]]$	BVS= $\Sigma exp[(r_0 - r)/B]$
Cu(1)-N(1)	1.763	2.115	0.386	
Cu(1)-N(2)	1.763	1.977	0.561	
Cu(1)-N(3)	1.763	2.088	0.415	2.339
Cu(1)-N(4)	1.763	1.960	0.587	
Cu(1)-O(7)	1.679	2.027	0.390	

Symmetry transformations used to generate equivalent atoms: #1 -x+1/2,-y+1/2,-z+1

Table S2 Comparison of M-O average bond distances of compound 1 with those in

anion	M-O <sub>c</sub> (Å)	M-O <sub>b</sub> (Å)	M-O <sub>t</sub> (Å)
$[W_6O_{19}]^{2-}$	2.331	1.922	1.694
$[V_2W_4O_{19}]^{4-}$	2.305	1.922	1.695
1	2.299	1.915	1.679

reported cluster anions.



Fig. S1 Packing arrangement of 1 viewed along c axis. The polyanions are represented with polyhedra: {CuON<sub>4</sub>}, yellow polyhedron; {W(V)O<sub>6</sub>}, blue octahedron; C (grey) and N (blue) ions are shown with thick sticks.



Fig. S2 The XPS spectra of C1s(1), Cu2p(2), W4f(3), V2p(4)



Fig. S3 The EDX diagram of 1



Fig. S4 The FT-IR spectra of 1(green), 1@TiO<sub>2</sub>/19P25(red) and pure P25(black)



Fig. S5 The UV-Vis spectrum of 1



Fig. S6 Thermogravimetric analysis (TGA) curve of 1



Fig. S7 The simulated and measured XRD patterns of 1



Fig. S8 XRD pattern of the 1@TiO<sub>2</sub>/19P25 composite



Fig. S9 The current voltage curves of DSSCs with  $1@TiO_2/nP25$  (n=5, 7, 19) and pure P25 electrode under AM 1.5 radiation (100 mW cm<sup>-2</sup>). The inset is the current voltage curves under dark condition.



## Fig. S10 Equivalent circuit used to fit the impedance measurements on the DSSCs

**Table S3** Fitted parameters and electron lifetime calculated from  $f_{max}$ 

Sample	$\frac{Rs(\Omega)}{\Omega}$	$R_{l}(\Omega)$	$R_2(\Omega)$	f <sub>max</sub> (Hz)	$\tau_e(\mathrm{ms})$
<mark>N719</mark>	<mark>33.2</mark>	<mark>37.9</mark>	<mark>5.83</mark>	<mark>56.3</mark>	<mark>2.83</mark>
<mark>1/N719</mark>	<mark>27. 1</mark>	<mark>34.9</mark>	<mark>5.17</mark>	<mark>27.7</mark>	<mark>5.75</mark>



**Fig. S11** The emission spectrum (Ex=320 nm) of the phen

Table 34 Select	eu bonu iengin	s (A) and bond angles (	
W(1)/V(1)-O(2)	1.961	W(3) -O(1)	1.917
W(1)/V(1)-O(3)	2.296	W(3)-O(2)	1.886
W(1)/V(1)-O(5)#1	1.875	W(3) -O(3)	2.285
W(1)/V(1)-O(7)	1.974	W(3) -O(4)	1.889
W(1)/V(1)-O(8)	1.661	W(3) -O(5)	1.943
W(1)/V(1)-O(9)	1.863	W(3) -O(6)	1.721
W(2)/V(2)-O(1)	1.894	Cu(1)-N(1)	2.115
W(2)/V(2)-O(3)	2.314	Cu(1)-N(2)	1.977
W(2)/V(2)-O(4)#1	1.928	Cu(1)-N(3)	2.088
W(2)/V(2)-O(7)#1	1.956	Cu(1)-N(4)	1.960
W(2)/V(2)-O(9)	1.895	Cu(1)-O(7)	2.027
W(2)/V(2)-O(10)	1.654		
N(1)-C(1)	1.332(18)	N(2)-C(12)	1.315(19)
N(1)-C(5)	1.369(18)	N(2)-C(7)	1.361(19)
N(3)-C(13)	1.323(18)	N(4)-C(24)	1.341(17)
N(3)-C(17)	1.364(18)	N(4)-C(18)	1.380(17)
O(8)-W(1)-O(2)	102.9(4)	O(1)-W(2)-O(3)	76.3(3)
O(8)-W(1)-O(3)	177.0(4)	O(1)-W(2)-O(4)#1	152.1(4)
O(8)-W(1)-O(7)	100.5(4)	O(1)-W(2)-O(7)#1	87.4(4)
O(8)-W(1)-O(9)	105.1(4)	O(1)-W(2)-O(9)	87.7(4)
O(8)-W(1)-W(2)	136.8(4)	O(1)-W(2)-W(1)	82.0(3)
O(7)-Cu(1)-N(3)	133.8(4)	O(7)-Cu(1)-N(1)	114.3(4)
N(2)-Cu(1)-O(7)	91.8(4)	N(4)-Cu(1)-O(9)	94.6(4)
N(2)-Cu(1)-N(1)	81.3(5)	N(4)-Cu(1)-N(2)	173.6(5)
N(2)-Cu(1)-N(3)	93.7(5)	N(4)-Cu(1)-N(3)	81.9(5)

Table S4 Selected bond lengths (Å) and bond angles (°) of 1

Symmetry transformations used to generate equivalent atoms: #1 -x+1/2,-y+1/2,-

z+1.