## **Supplementary Information:**

## Significant differences of monooxotungsten(IV) and dioxotungsten(VI) benzenedithiolates containing two intramolecular NH····S hydrogen bonds from molybdenum analogues

Taka-aki Okamura,\* Yui Omi, Manami Fujii, Miki Tatsumi and Kiyotaka Onitsuka

Department of Macromolecular Science, Graduate School of Science, Osaka University, Toyonaka, Osaka 560-0043, Japan

E-mail: tokamura@chem.sci.osaka-u.ac.jp

**Table S1** Crystallographic data for  $(NEt_4)_2[W^{IV}O(1,2-S_2-3-t-BuNHCOC_6H_3)_2]$  (**1-W**),  $(NEt_4)_2[W^{IV}(1,2-S_2-3-t-BuNHCOC_6H_3)_3] \cdot H_2O$  (**3**·H<sub>2</sub>O), and  $(NEt_4)_2[\{W^{VI}O_2(1,2-S_2-3-t-BuNHCOC_6H_3)\}_2(\mu-O)] \cdot 2(Et_2O)$  (**4**·2(Et\_2O))

	<b>1-W</b>	$3 \cdot H_2O$	$4 \cdot 2(Et_2O)$
empirical formula	$C_{38}H_{66}N_4O_3S_4W$	$C_{49}H_{81}N_5O_4S_6W$	$C_{46}H_{86}N_4O_9S_4W_2$
formula weight	939.04	1180.39	1335.12
color	orange	red	yellow
crystal system	orthorhombic	orthorhombic	monoclinic
a, Å	8.865(2)	21.3866(5)	17.9606(14)
b, Å	16.289(3)	18.6641(4)	8.7047(8)
<i>c</i> , Å	29.929(7)	29.515(2)	18.3212(14)
$\beta$ , deg	90	90	91.438(5)
$V, Å^3$	4321.8(16)	11781.4(9)	2863.5(4)
space group	$P2_{1}2_{1}2_{1}$	$Pna2_1$	$P 2_1/c$
Z	4	8	2
$D_{\text{calc}}, \text{g/cm}^3$	1.443	1.331	1.548
F(000)	1936	4896	1348
$\mu$ (MoK $\alpha$ ), mm <sup>-1</sup>	2.905	2.216	4.210
Scan type	ω	ω	ω
$2\theta_{\rm max}$ , deg	50	50	55
No. of Reflections unique	7323	20684	6532
No. Variables	454	1172	295
residuals; $R1^a$ (I > 2 $\sigma$ (I)), w $R2^b$ (all data)	0.0685, 0.1540	0.0781, 0.1503	0.0293, 0.0437
GOF	1.199	1.113	0.858

 Table S2 Raman bands of monooxo-tungsten and molybdenum complexes

Complex	$\nu(M=O)/cm^{-1}$	References
(NEt4)[W <sup>V</sup> O(SC <sub>6</sub> F <sub>5</sub> ) <sub>4</sub> ]	980	This work
(NEt4)[W <sup>V</sup> O(SPh)4]	943 <sup><i>a</i></sup>	12
(NEt4)[Mo <sup>V</sup> O(SC <sub>6</sub> F <sub>5</sub> )4]	982 <sup><i>a</i></sup>	30
(NEt4)[Mo <sup>v</sup> O(SPh)4]	936	38
$(NEt_4)_2[W^{IV}O(SC_6F_5)_4]$	949	This work
(HNEt3)2[Mo <sup>IV</sup> O(SC6F5)4]	927 <sup><i>a</i></sup>	30
<sup>a</sup> ID are a stres		

<sup>*a*</sup>IR spectra.

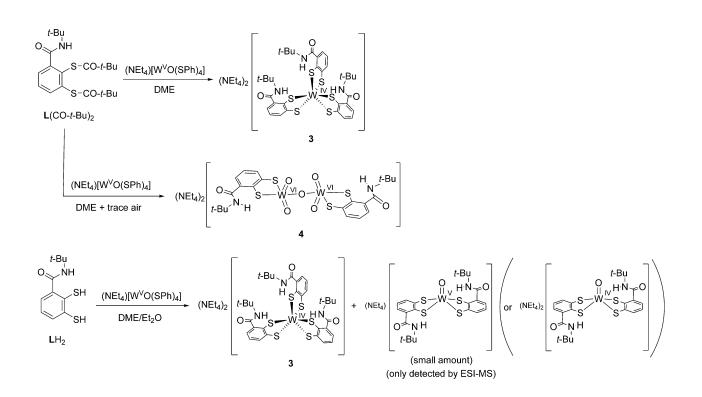
	4-H (dd)	6-H (dd)	5-H (t)	NH (br s)	SH (br s)	<i>t</i> -Bu (s)
$LH_2$	7.42	7.23	7.05	6.85	5.00	1.41
$(NEt_4)[LH]$	7.72	7.33	6.57	11.78	9.01	1.43
$(\operatorname{NEt}_{4})_{2}[\mathbf{L}]$	7.78	7.04	6.58	12.21		1.45

**Table S3** Chemical shifts and proposed structures of  $LH_2$  and the deprotonated forms in acetonitrile- $d_3$ 

Table S4 Contributions of NH $\cdots$ S hydrogen bonds to W and Mo complexes

			W	Mo <sup>e</sup>
1-M	X-ray	$\Delta (M=O)^a/Å$	-0.012(9)	-0.001(6)
		$\Delta (M-S)^{a}/Å$	-0.028(4)0.001(4)	-0.040(2)0.003(2)
			(mean -0.012(4))	(mean -0.019(4))
	IR	$\Delta v (\text{NH})^b / \text{cm}^{-1}$	-191	-198
		$\Delta v (M=O)^{a}/cm^{-1}$	+15	+13
	CV	$\Delta E_{1/2}$ <sup><i>a</i></sup> /V	+0.15	+0.12
2-M	IR	$\Delta v (NH)^d / cm^{-1}$	-206	-209
	Raman	$\Delta v_{s}(M=O)^{c}/cm^{-1}$	+9	+13
		$\Delta v_{as}(M=O)^{c/cm^{-1}}$	+13	+11

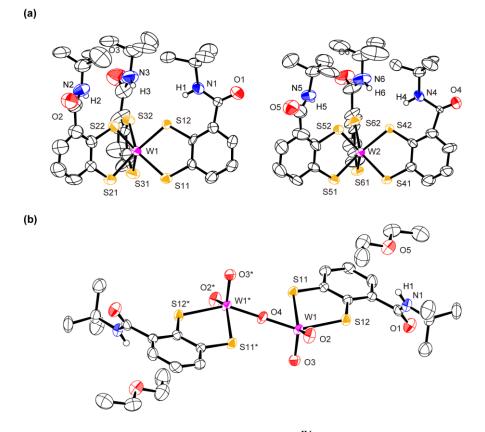
<sup>*a*</sup>Difference of the values between **1-M** and (NEt<sub>4</sub>)<sub>2</sub>[ $M^{IV}O(bdt)$ ] (**5-M**). (W=O, 1.727(2) Å; W–S, 2.372(4) Å for (NEt<sub>4</sub>)<sub>2</sub>[ $W^{IV}O(bdt)_2$ ] (**5-W**):<sup>14</sup> Mo=O, 1.699(6) Å; Mo–S, 2.388(2) Å for (NEt<sub>4</sub>)<sub>2</sub>[Mo<sup>IV</sup>O(bdt)<sub>2</sub>] (**5-Mo**).<sup>50</sup>) <sup>*b*</sup>See Table 3. <sup>*c*</sup>Difference of the values between **2-M** and (NEt<sub>4</sub>)<sub>2</sub>[ $M^{VI}O_2(bdt)_2$ ] (**6-M**). <sup>*d*</sup>See Table 4. <sup>*e*</sup>Ref. 29.



Scheme S1 Reaction of  $(NEt_4)[W^VO(SPh)_4]$  with LH<sub>2</sub> or L(CO-*t*-Bu)<sub>2</sub> according to the molybdenum analogue.

$$(\mathsf{NEt}_4)[\mathsf{W}^{\mathsf{V}}\mathsf{O}(\mathsf{SPh})_4] \xrightarrow{\mathsf{C}_6\mathsf{F}_5\mathsf{SH}} (\mathsf{NEt}_4)[\mathsf{W}^{\mathsf{V}}\mathsf{O}(\mathsf{SC}_6\mathsf{F}_5)_4] \xrightarrow{\mathsf{NEt}_4\mathsf{BH}_4} (\mathsf{NEt}_4)_2[\mathsf{W}^{\mathsf{I}\mathsf{V}}\mathsf{O}(\mathsf{SC}_6\mathsf{F}_5)_4]$$

Scheme S2 Preparation of (NEt<sub>4</sub>)<sub>2</sub>[W<sup>IV</sup>O(SC<sub>6</sub>F<sub>5</sub>)<sub>4</sub>].



**Fig. S1** Molecular structures (anion parts) of (a)  $(NEt_4)_2[W^{IV}(1,2-S_2-3-t-BuNHCOC_6H_3)_3]$  (**3**) and (b)  $(NEt_4)_2[\{W^{VI}O_2(1,2-S_2-3-t-BuNHCOC_6H_3)\}_2(\mu-O)]$  (**4**) in the crystal. Two molecules of **3** were found in the asymmetric unit.

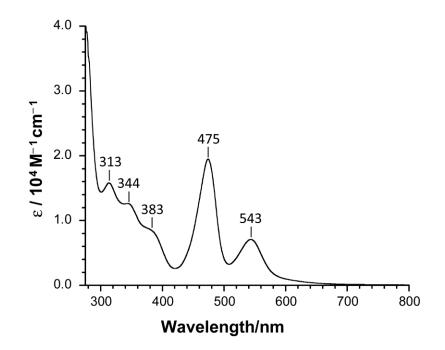
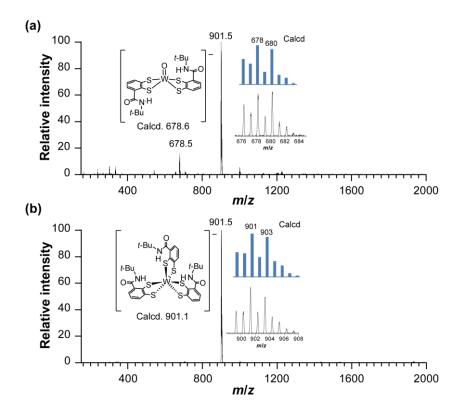
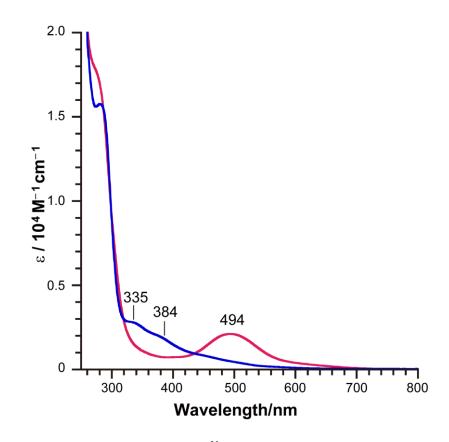


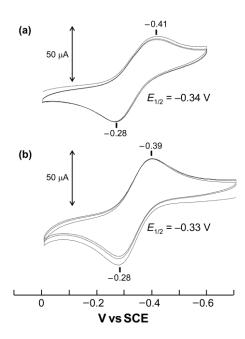
Fig. S2 UV-vis spectrum of  $(NEt_4)_2[W^{IV}(1,2-S_2-3-t-BuNHCOC_6H_3)_3]$  (3) in DMF.



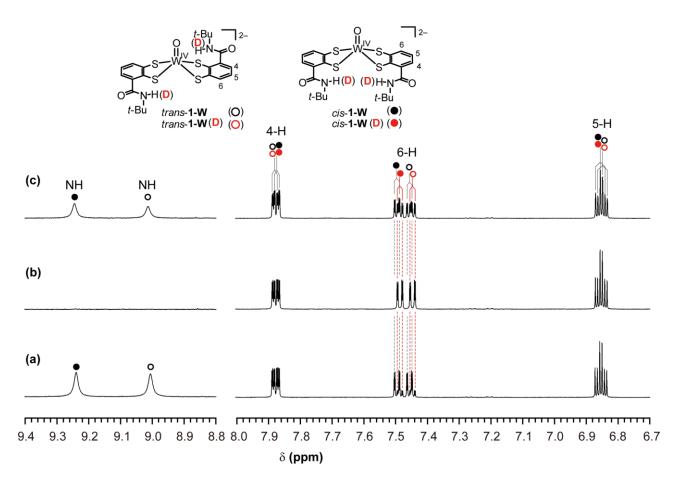
**Fig. S3** ESI-MS spectra of (a) the products in DME/Et<sub>2</sub>O shown in Scheme S1, where the main peak was  $(NEt_4)_2[W^{IV}(1,2-S_2-3-t-BuNHCOC_6H_3)_3]$  (3) and a small amount of  $(NEt_4)[W^{VO}(1,2-S_2-3-t-BuNHCOC_6H_3)_2]$  was found, and (b) pure 3. The enlarged spectra were obtained in the zoom scan mode, accompanying simulated isotope patterns.



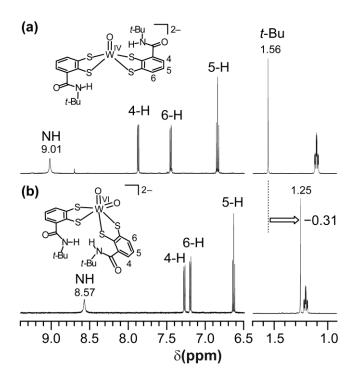
**Fig. S4** UV-vis spectra of  $(NE_4)[W^VO(SC_6F_5)_4]$  (red line) in acetonitrile and  $(NE_4)_2[W^{IV}O(SC_6F_5)_4]$  (blue line) in DMF.



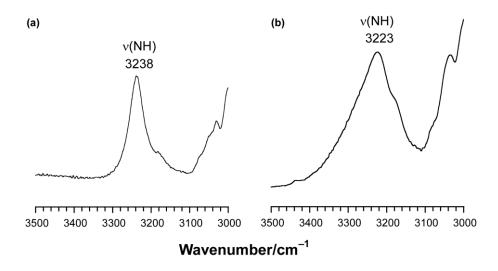
**Fig. S5** Cyclic voltammograms of (a)  $(NE_4)[W^VO(SC_6F_5)_4]$  and (b)  $(NE_4)_2[W^{IV}O(SC_6F_5)_4]$  in DMF.



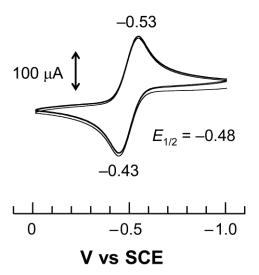
**Fig. S6** <sup>1</sup>H NMR spectra (measured at 30 °C) of (a) (NEt<sub>4</sub>)<sub>2</sub>[W<sup>IV</sup>O(1,2-S<sub>2</sub>-3-*t*-BuNHCOC<sub>6</sub>H<sub>3</sub>)<sub>2</sub>] (**1-W**, *trans/cis* = 1:1) containing a minute amount of NEt<sub>4</sub>BH<sub>4</sub> in acetonitrile-*d*<sub>3</sub>, (b) after heating at 50 °C for 30 min., and (c) after addition of a trace amount of H<sub>2</sub>O. The content of (NEt<sub>4</sub>)<sub>2</sub>[W<sup>IV</sup>O(1,2-S<sub>2</sub>-3-*t*-BuNDCOC<sub>6</sub>H<sub>3</sub>)<sub>2</sub>] (**1-W**(ND) ) was increased from (a) 22% to (b) 93%, and then back to about 50% by the addition of H<sub>2</sub>O. The 4-H and 5-H signals of **1-W** and **1-W**(ND) were found at identical chemical shifts for each isomer.



**Fig. S7** <sup>1</sup>H NMR spectra of (a)  $(NEt_4)_2[W^{IV}O(1,2-S_2-3-t-BuNHCOC_6H_3)_2]$  (**1-W**) and (b)  $(NEt_4)_2[W^{VI}O_2(1,2-S_2-3-t-BuNHCOC_6H_3)_2]$  (**2-W**) in acetonitrile-*d*<sup>3</sup> at 30 °C. The aromatic region was enlarged along vertical axis.



**Fig. S8** IR spectra of (a)  $(NEt_4)_2[W^{IV}O(1,2-S_2-3-t-BuNHCOC_6H_3)_2]$  (1-W) and (b)  $(NEt_4)_2[W^{VI}O_2(1,2-S_2-3-t-BuNHCOC_6H_3)_2]$  (2-W) in the solid state.



**Fig. S9** Cyclic voltammogram of  $(NEt_4)_2[W^{IV}O(1,2-S_2-3-t-BuNHCOC_6H_3)_2]$  (1-W) in DMF.