

Supporting Information

Investigation of ZnCo₂O₄-Pt hybrids with different morphologies towards catalytic CO oxidation

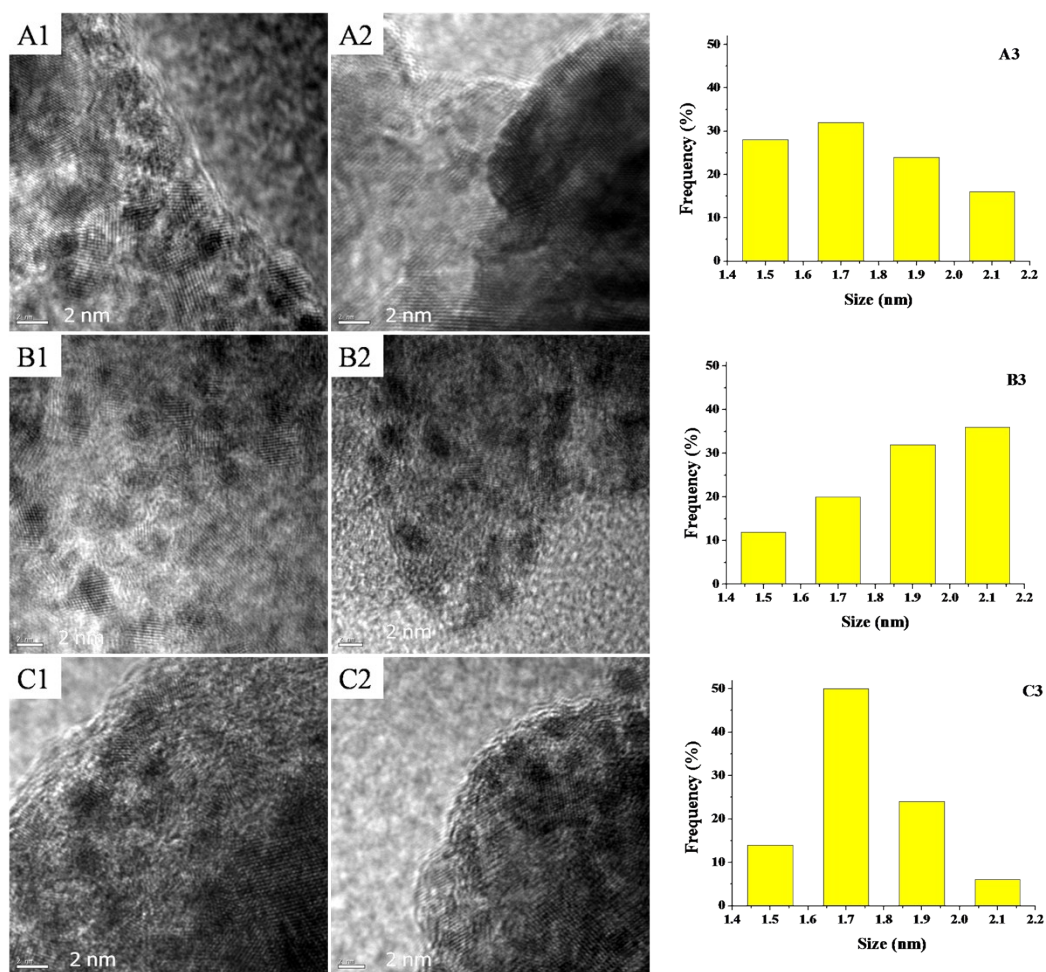
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FigureS1. TEM images of A: ZnCo₂O₄-Pt rods and size distribution of Pt NPs, B: ZnCo₂O₄-Pt plates and size distribution of Pt NPs, C: ZnCo₂O₄-Pt spheres and size distribution of Pt NPs.

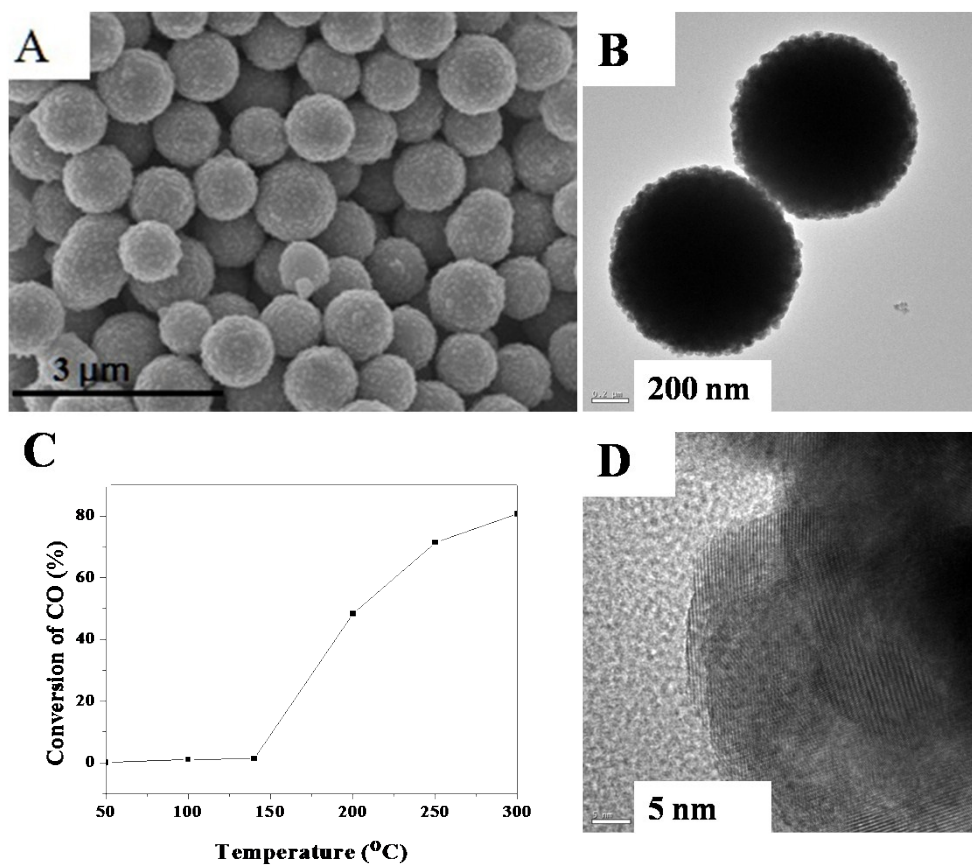
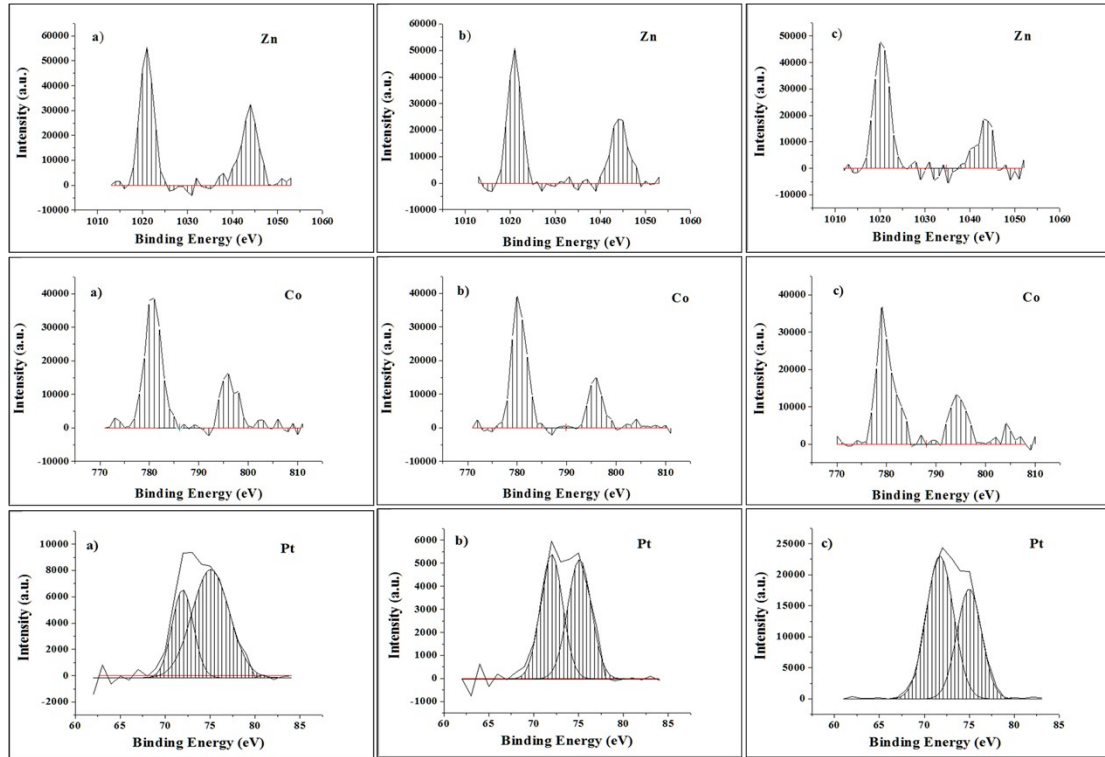


Figure S2. (A): SEM images of ZnCo₂O₄-Pt spheres without PVP; (B) and (D): TEM images of ZnCo₂O₄-Pt spheres without PVP; (C): CO conversion curves of ZnCo₂O₄-Pt spheres without PVP.



FigureS3. Normalized XPS spectra (a): ZnCo₂O₄-Pt rods, (b): ZnCo₂O₄-Pt plates, (C): ZnCo₂O₄-Pt spheres.

Sample	Pt (ev.s)	Zn (ev.s)	Co (ev.s)	Zn:Co:Pt
ZnCo ₂ O ₄ -Pt rod	62664	347777	244761	1.42 : 1.00 : 0.26
ZnCo ₂ O ₄ -Pt plate	33724	287218	200559	1.43 : 1.00 : 0.17
ZnCo ₂ O ₄ -Pt sphere	148706	259436	221544	1.17 : 1.00 : 0.67

Table S1. The integrals of the peaks of the three samples.

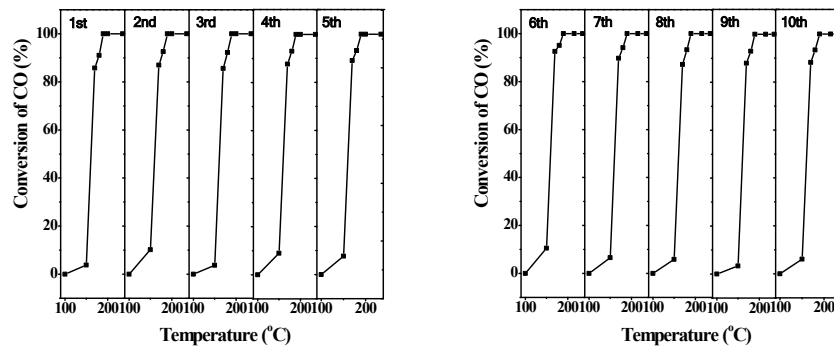


Figure S4. Cycling test of ZnCo₂O₄-Pt rod for CO conversion.

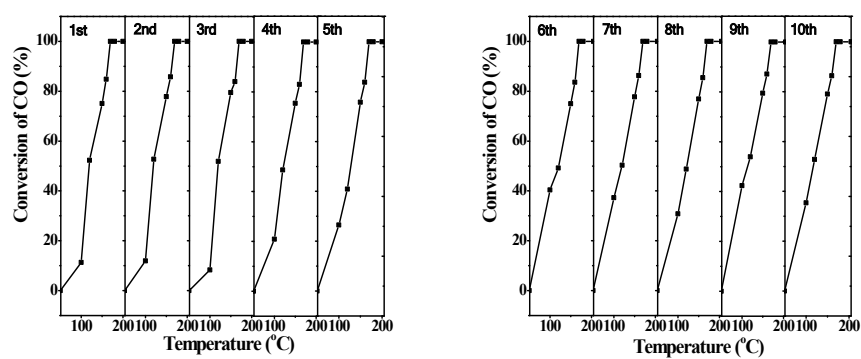


Figure S5. Cycling test of ZnCo₂O₄-Pt plate for CO conversion.