

Supplementary Information

A Zwitterionic Triphosphenium Compound as a Tunable Multifunctional Donor

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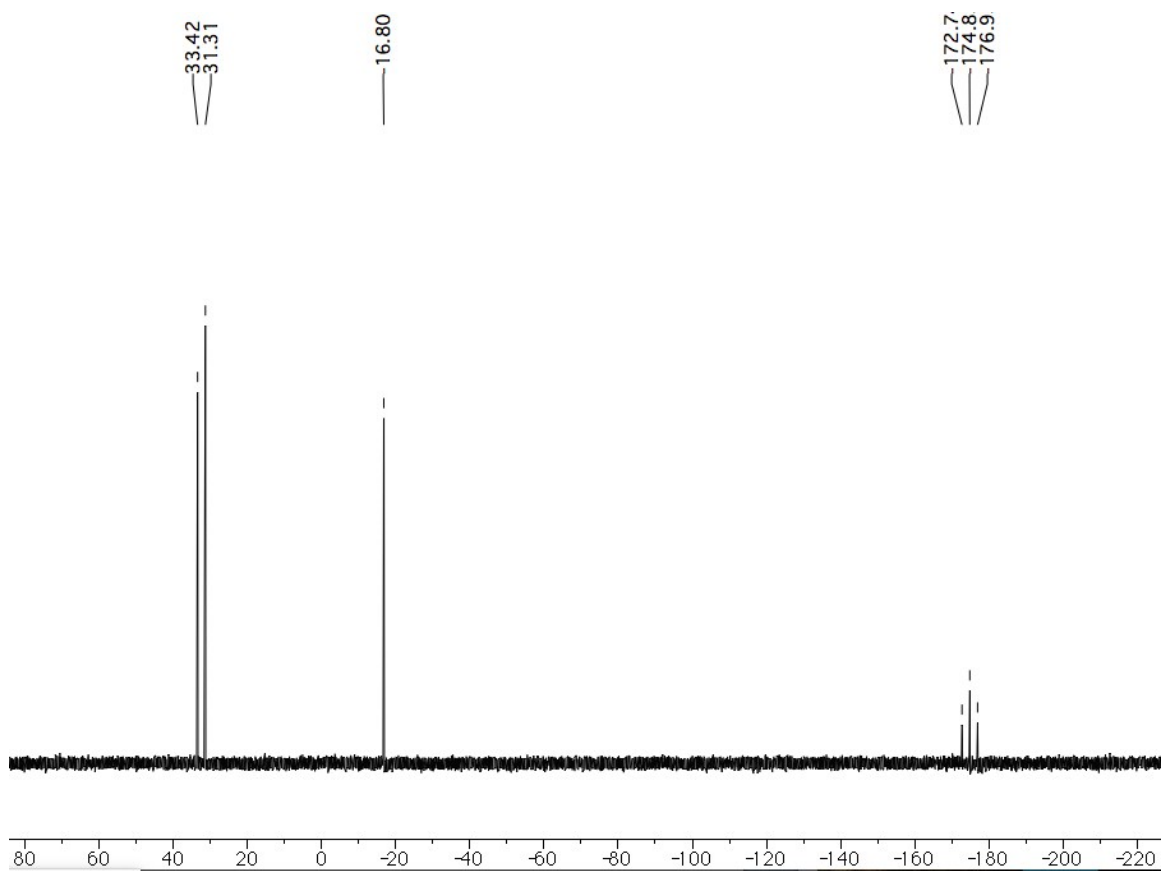


Fig 1. ^{31}P $\{^1H\}$ NMR of **1**

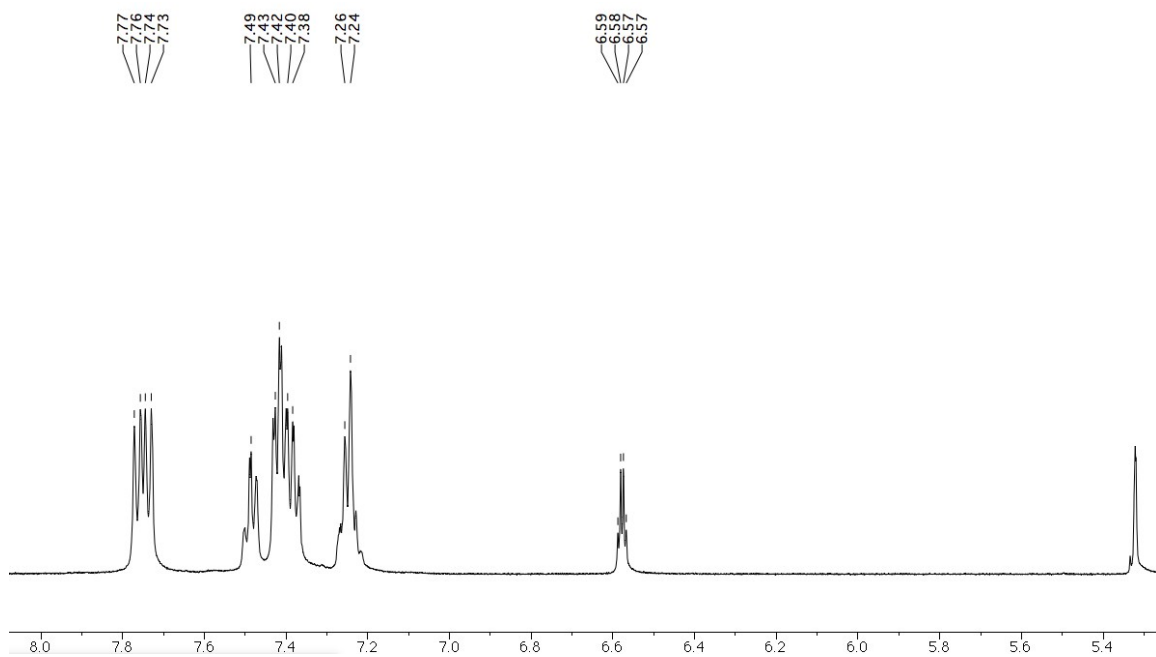


Fig. 2 ^1H NMR of 1

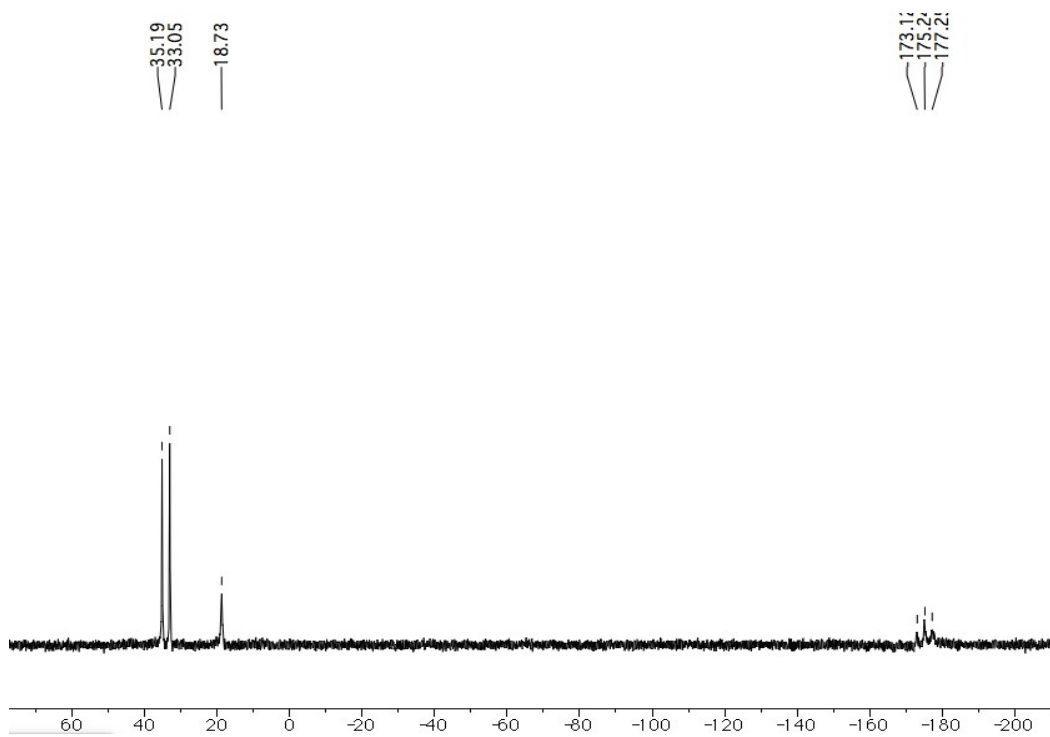


Fig. 3 ^{31}P $\{^1\text{H}\}$ NMR of 2

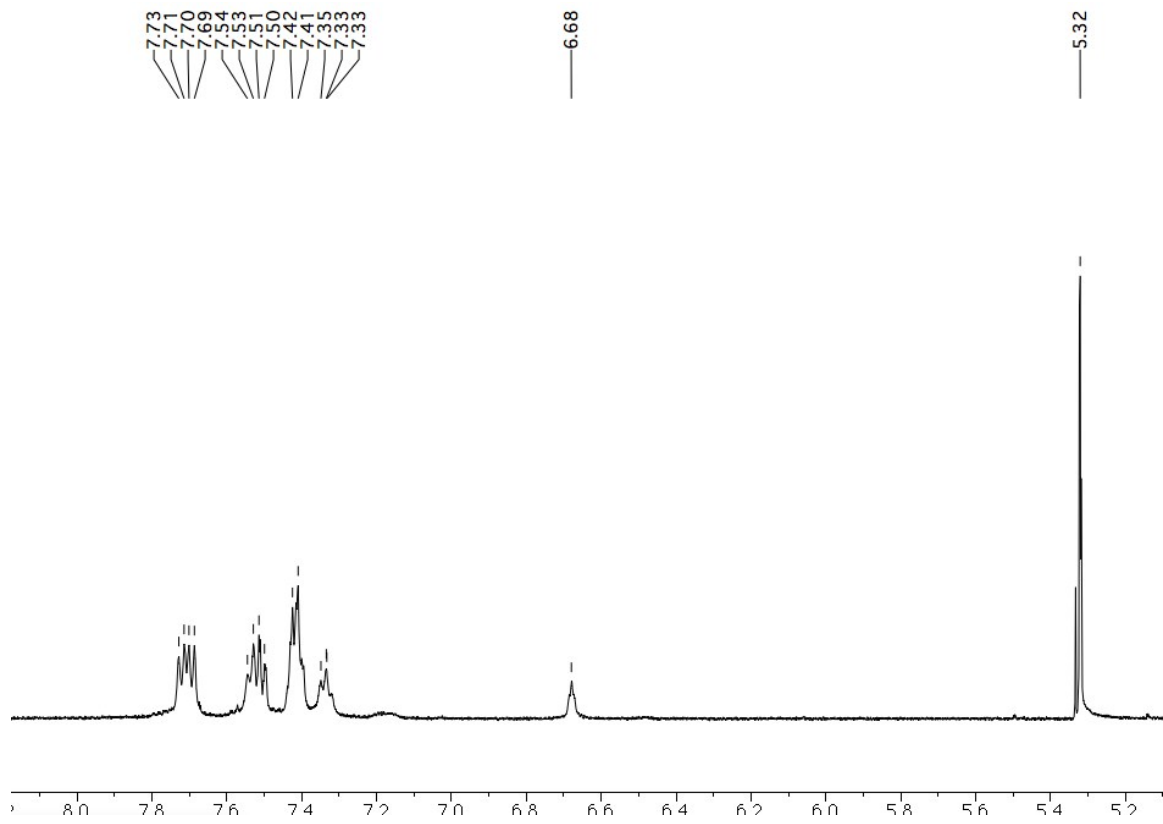


Fig. 4 ^1H NMR of **2**

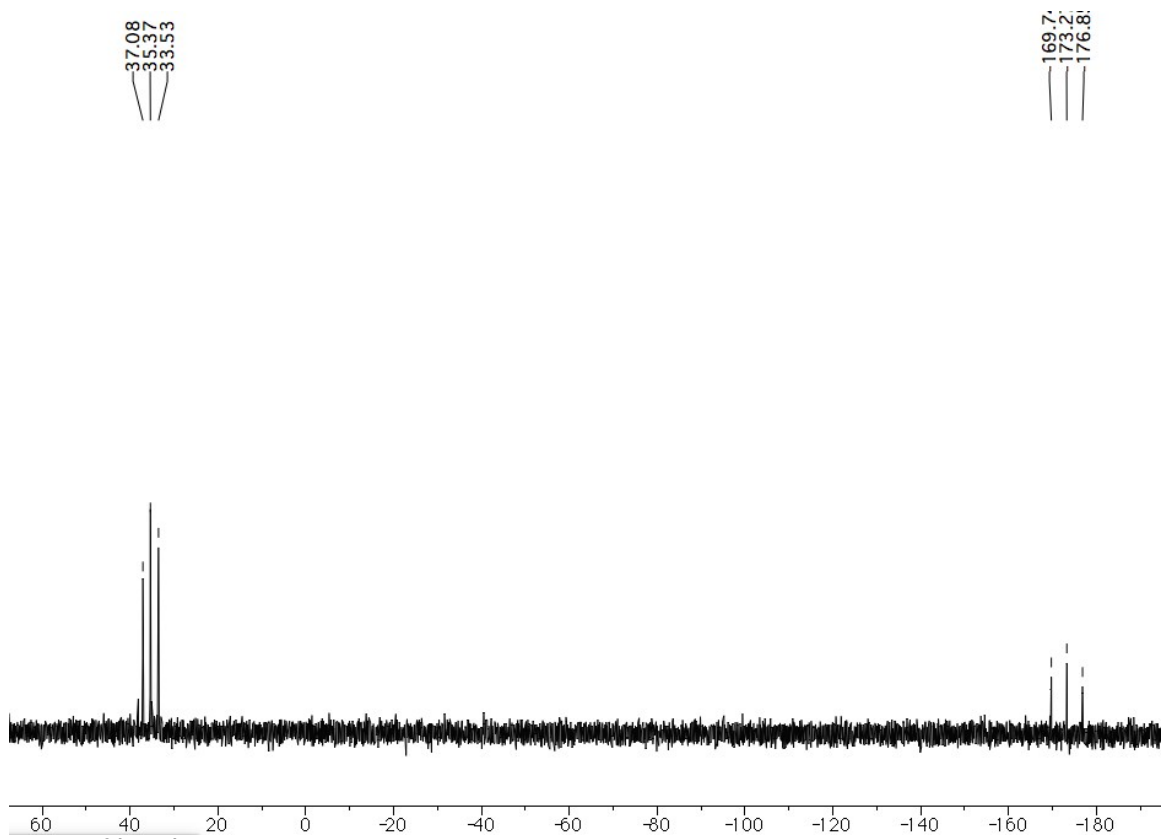


Fig. 5 ^{31}P $\{^1\text{H}\}$ NMR of 3

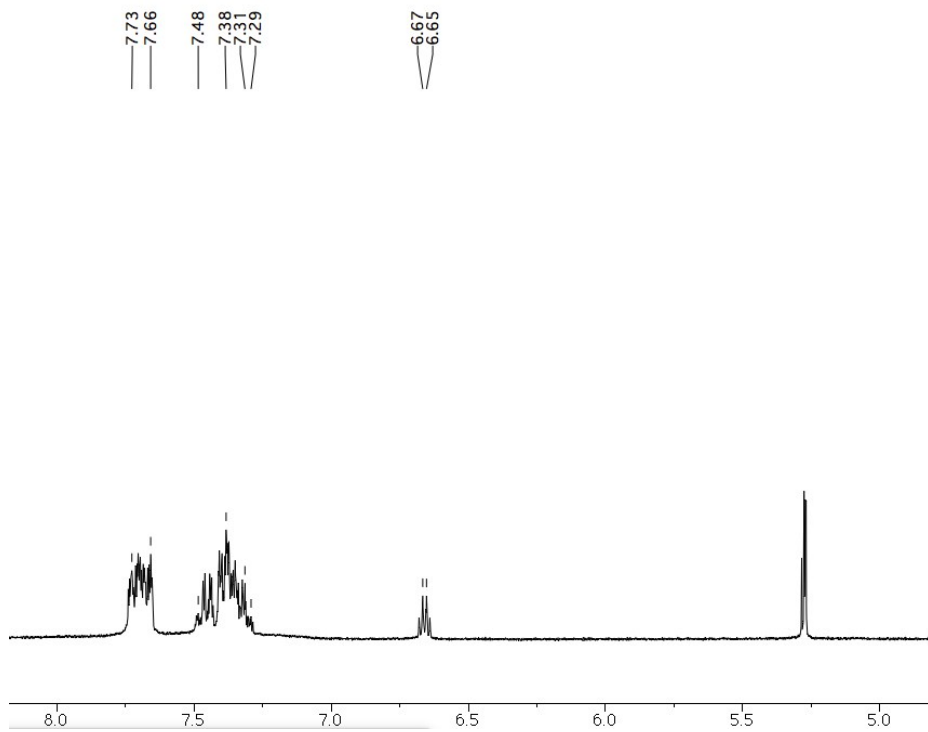


Fig. 6 ^1H NMR of 3

Experimental Procedure NMR Reactions with Triflic Acid

To a Schlenk flask containing **1** (0.200g, 0.39 mmol) in dichloromethane, was added 1 stoichiometric equivalent of triflic acid via syringe (0.058g, 0.39 mmol). There is no colour change upon addition, however the reaction can be tracked by ^{31}P NMR; after stirring for five minutes the reaction is complete. $^{31}\text{P}\{^1\text{H}\}$ NMR: δ 35.17 (d, $^1J_{\text{P-P}}=443$ Hz), -4.2 (s), -178.0 (t, $^1J_{\text{P-P}}=437$ Hz). ^{31}P NMR: δ 35.2 (d, $^1J_{\text{P-P}}=445$ Hz), -4.15 (d, $^1J_{\text{P-H}}=500$ Hz), -178.0 (t, $^1J_{\text{P-P}}=445$ Hz). ^1H NMR: δ 8.60 (d, $^1J_{\text{P-H}}=498$ Hz, 1H, P-H), 7.8-7.4 (m, 30H, Ph), 6.82 (*pseudo-q*, $^2J_{\text{P-H}}=3.5$ Hz, $^3J_{\text{P-H}}=3.5$ Hz, 2H, C_5H_2).

To an NMR tube containing **1** in CD_2Cl_2 , was added one drop of triflic acid syringe. There is no colour change upon addition, and due to the instability of this molecule, however the NMR spectra were collected immediately upon addition. $^{31}\text{P}\{^1\text{H}\}$ NMR: δ 35.17 (d, $^1J_{\text{P-P}}=443$ Hz), -4.2 (s), -178.0 (t, $^1J_{\text{P-P}}=437$ Hz). ^{31}P NMR: δ 35.2 (d, $^1J_{\text{P-P}}=445$ Hz), -4.15 (d, $^1J_{\text{P-H}}=500$ Hz), -178.0 (t, $^1J_{\text{P-P}}=445$ Hz). ^1H NMR: δ 11.2 (b, $^1J_{\text{P-H}}$ unresolved, 1H, P-H), 8.60 (d, $^1J_{\text{P-H}}=500$ Hz, 1H, P-H), 8.8-7.1 (m, 30H, Ph), 6.82 (*pseudo-q*, $^2J_{\text{P-H}}=3.5$ Hz, $^3J_{\text{P-H}}=3.5$ Hz, 2H, C_5H_2). Excess triflic acid is visible in the spectrum.

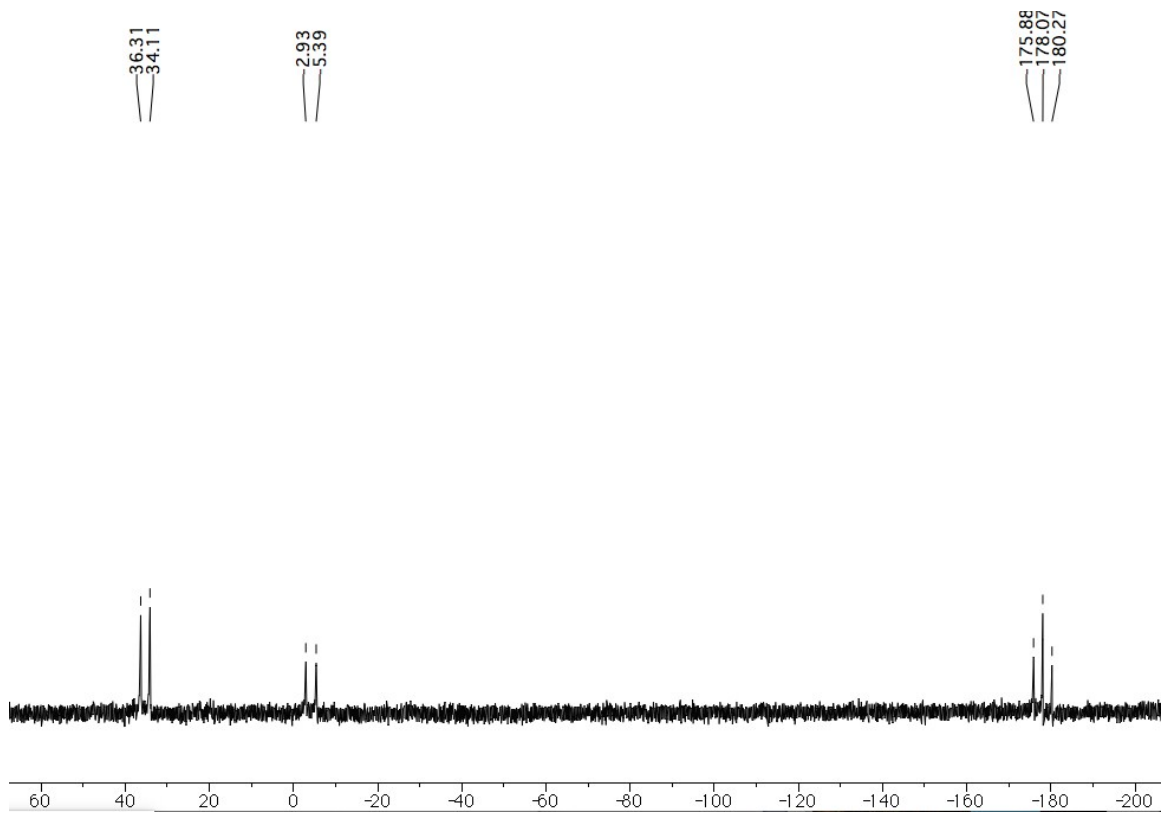


Fig. 7 ^{31}P NMR of the reaction with 1 eq. HOTf

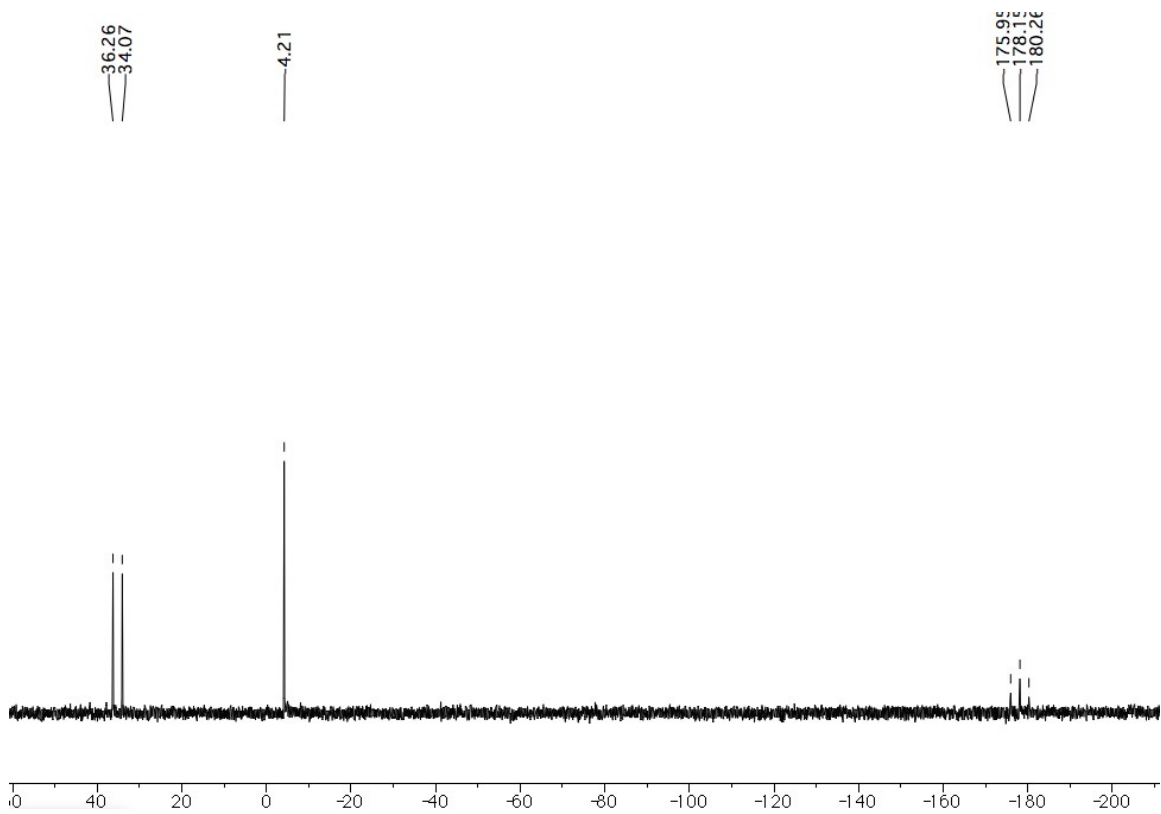


Fig. 8 ^{31}P $\{^1\text{H}\}$ NMR of the reaction with 1 eq. HOTf

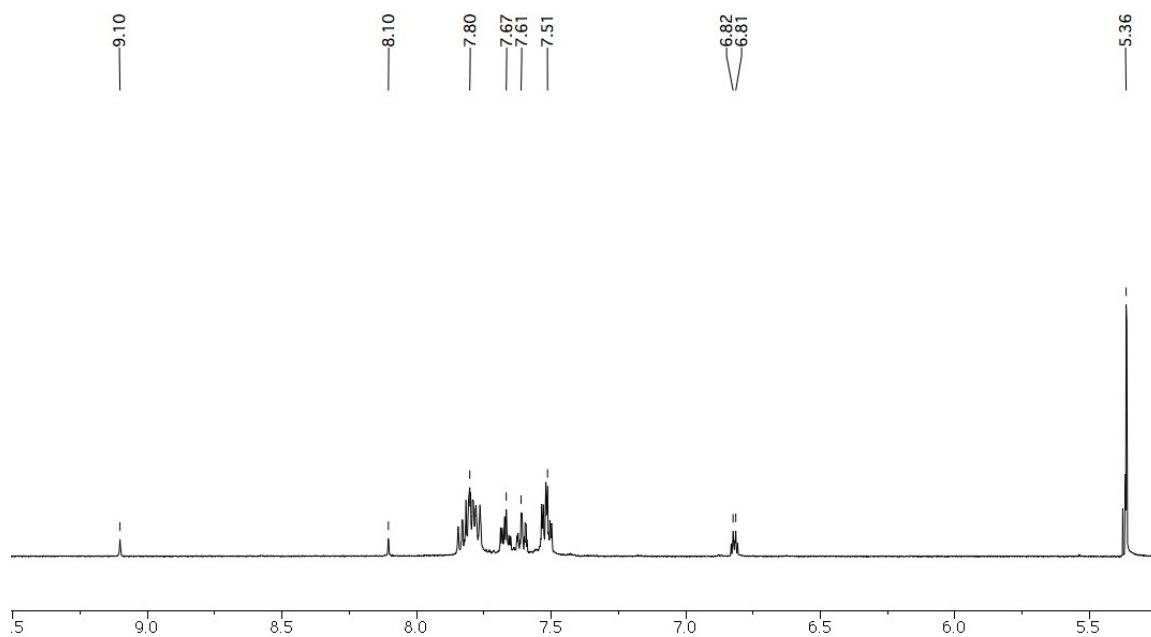


Fig. 9 ^1H NMR of the reaction with 1 eq. HOTf

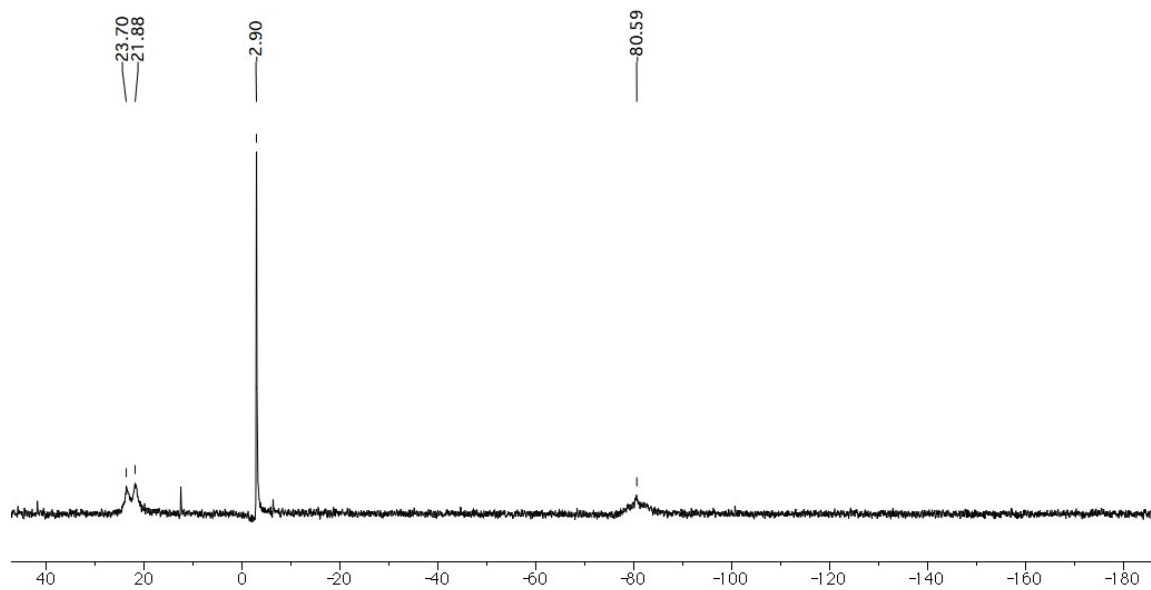


Fig. 10 ^{31}P $\{^1\text{H}\}$ NMR of the reaction with 2 eq, HOTf

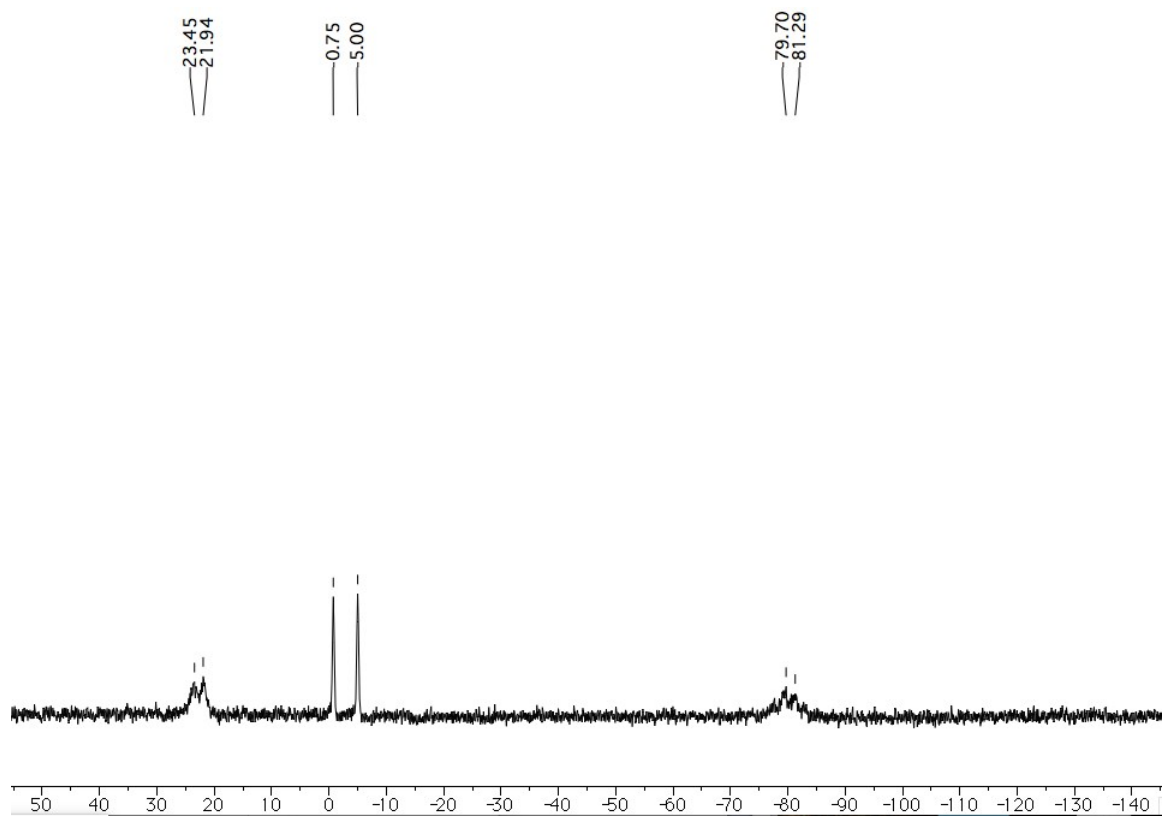


Fig. 11 ^{31}P NMR of the reaction with 2 eq. HOTf

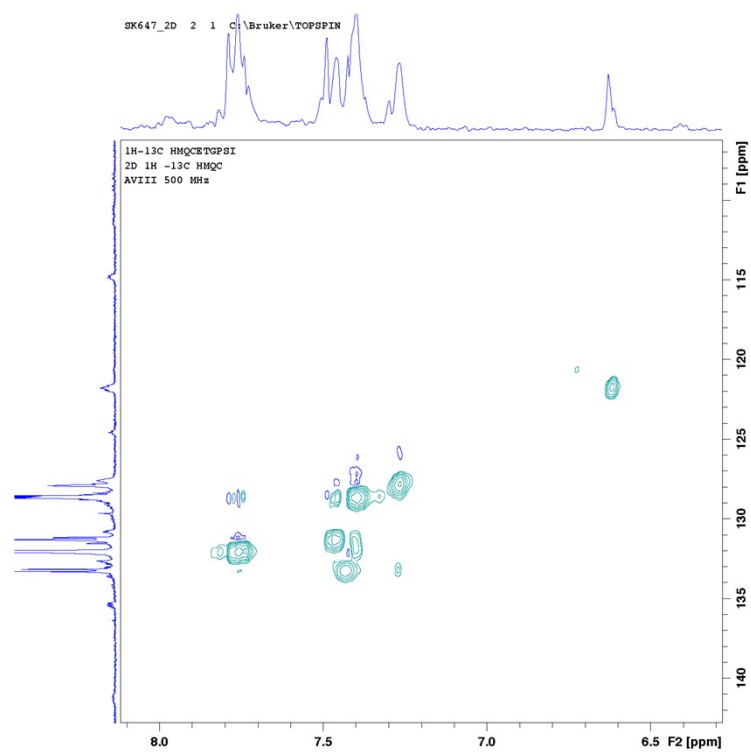


Fig. 12. HMQC of 1.

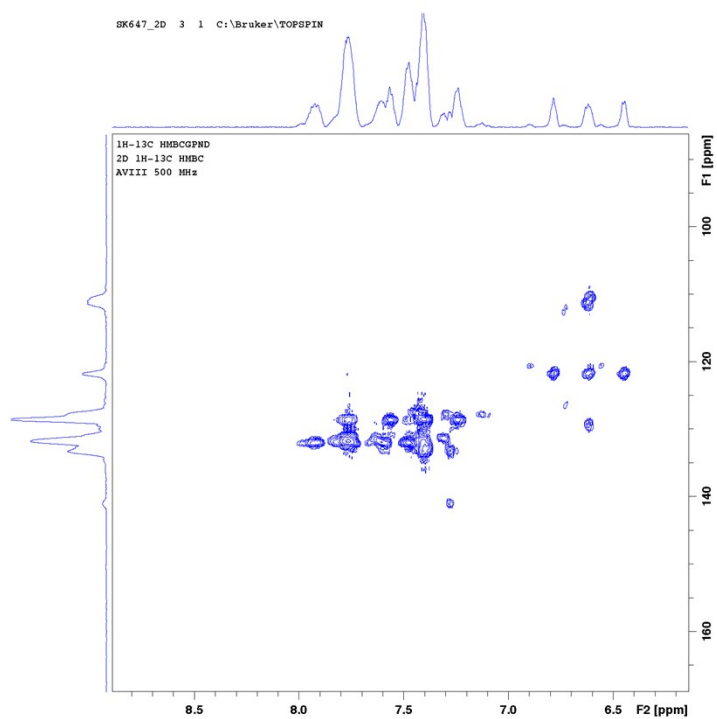


Fig. 13. HMBC of 1.

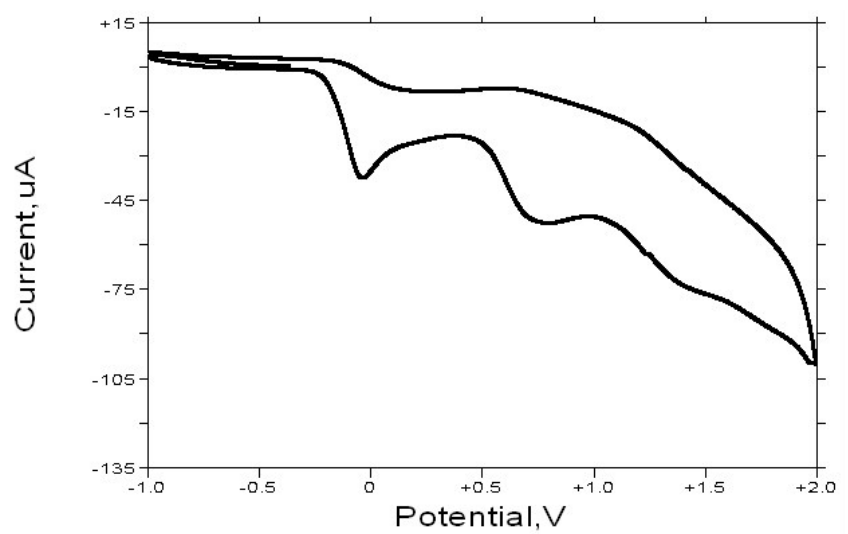


Fig. 14 Cyclic Voltammogram of $K^+(Ph_2P)_3(C_5H_2)$

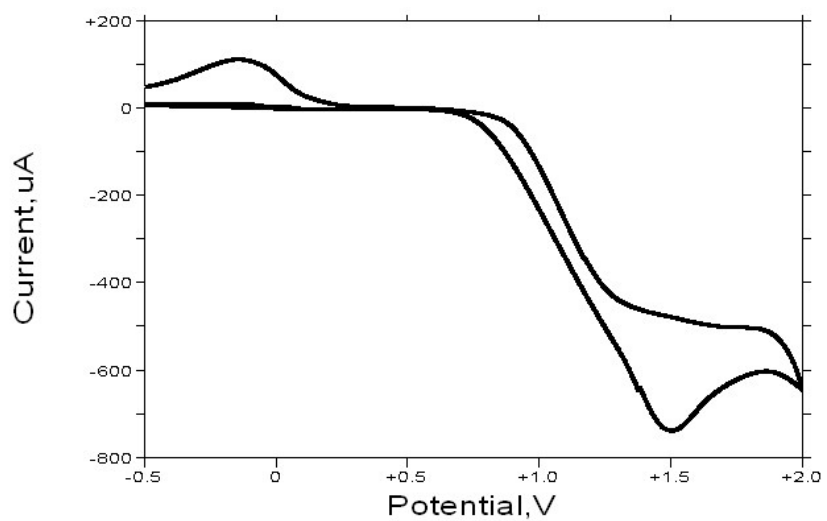
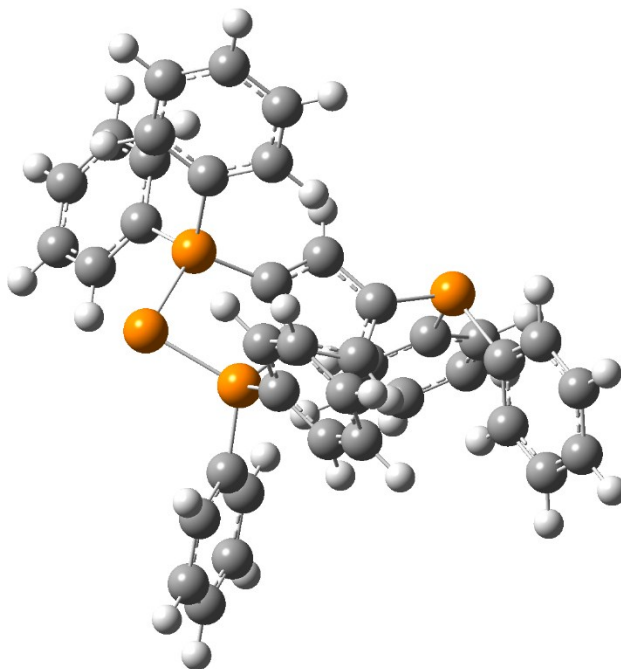


Fig. 15 Cyclic Voltammogram of $[dppeP][Br]$

Computational Investigations

All of the computational investigations were performed using the **Compute Canada Shared Hierarchical Academic Research Computing Network (SHARCNET)** facilities (www.sharcnet.ca) with the Gaussian09¹ program suites. Geometry optimizations have been calculated using density functional theory (DFT), specifically implementing the M062X method² ENREF_60 in conjunction with the TZVP basis set³ for all atoms. The geometry optimizations were not subjected to any symmetry restrictions and each stationary point was confirmed to be a minimum having zero imaginary vibrational frequencies. Pictures of the optimized structures were prepared using Gaussview 3.0.⁴ Population analyses were conducted using the Natural Bond Orbital (NBO)⁵ implementation included with the Gaussian package. Plots of molecular orbitals and electrostatic potentials were generated at the M062X/TZVP level of theory and examined using ADF2014.⁶⁻⁸ Summaries of the optimized structures, including electronic energies and Cartesian components for each of the atoms, are detailed in the sections below.

$P(Ph_2P)_2C_5H_2PPh_2$ model



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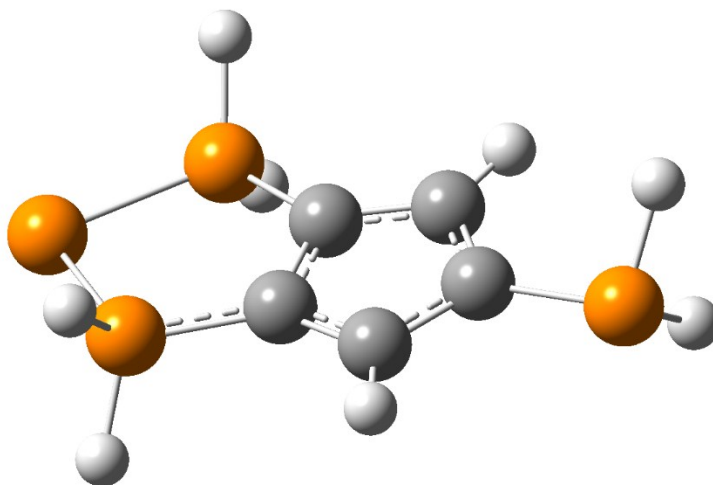
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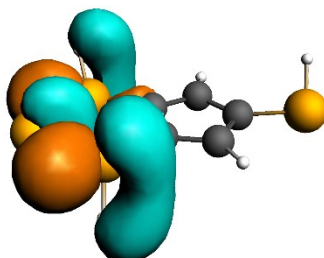


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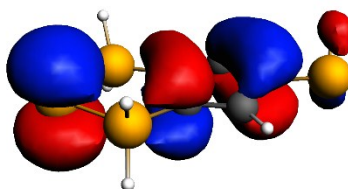
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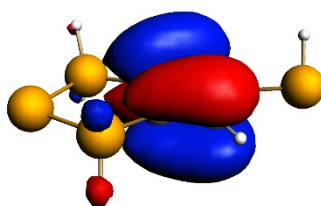
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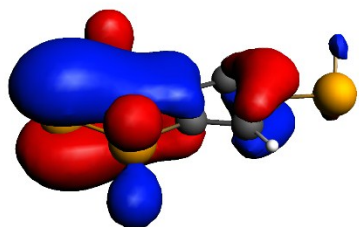
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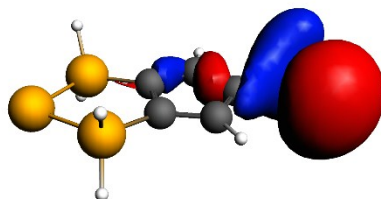
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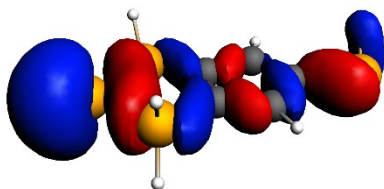
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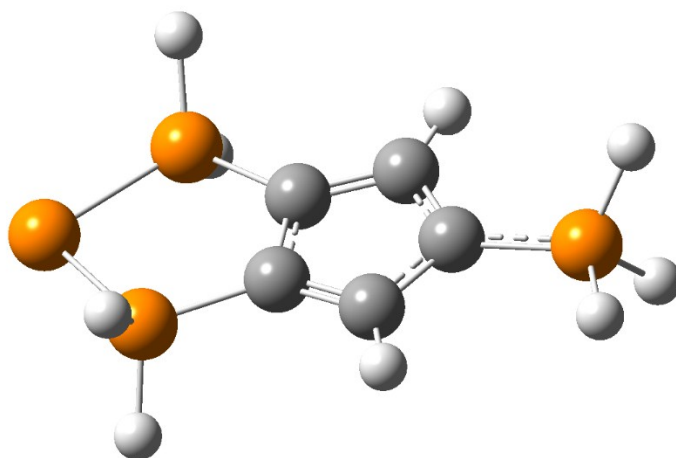
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HOMO-4



[P(H₂P)₂C₅H₂PH₃] cation model



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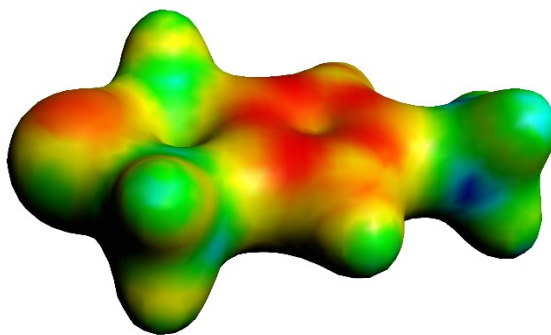
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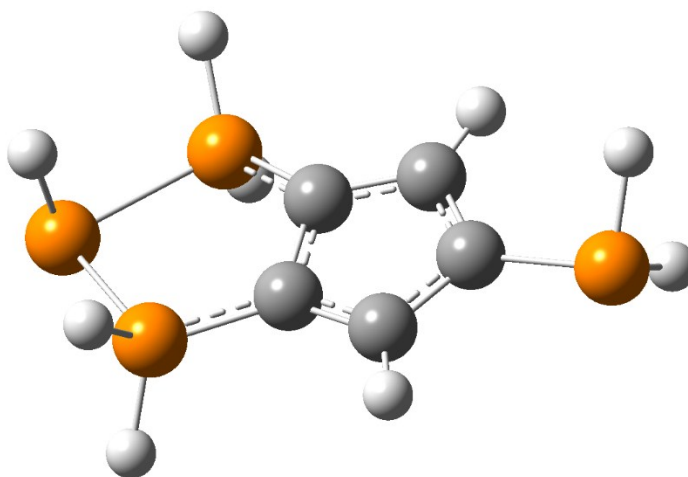
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Electrostatic Potential



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,0.1992868\Quadrupole=6.1550404,5.0750513,-11.2300917,2.8912412,-1.954

2039,-0.1425096\PG=C01 [X(C5H9P4)]\ \@

Zero-point correction= 0.119568 (Hartree/Particle)

Thermal correction to Energy= 0.130569

Thermal correction to Enthalpy= 0.131514

Thermal correction to Gibbs Free Energy= 0.079551

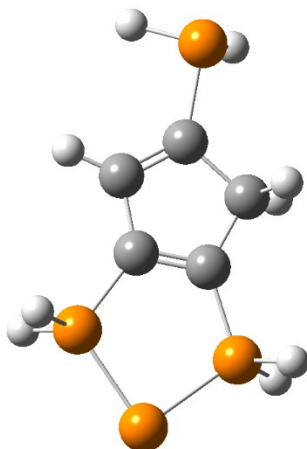
Sum of electronic and zero-point Energies= -1560.806840

Sum of electronic and thermal Energies= -1560.795838

Sum of electronic and thermal Enthalpies= -1560.794894

Sum of electronic and thermal Free Energies= -1560.846856

[P(H₂P)₂C₅H₃PH₂] cation model (isomer 1)



1\1\GINC-SAW298\FOpt\RM062X\TZVP\C5H9P4(1+)\CMACD\05-Oct-2015\0\#\#
M06

2x/TZVP scf=tight opt=tight freq pop=(full,nbo6read) test\Optimizatio
n of H[P(Cp(PH₂)₃)] axial H-Cp 1 cation\1,1\P,-1.6573767634,1.4964453
423,0.0826449333\P,-3.1963932504,-0.012038042,-0.0854294412\P,-1.61487
75328,-1.4911837593,-0.1446451613\C,-0.0418629671,0.6829551371,0.10433
66645\C,-0.0357583888,-0.6608403619,0.011124922\C,1.3155357884,1.19970
82333,0.1912295297\C,1.3871309557,-1.1343844416,0.0371747532\C,2.16999
33962,0.1540032175,0.1424048273\H,1.5782739816,2.2442141558,0.27503378
33\H,1.6529966931,-1.6982419399,-0.8630935111\P,3.9870403022,0.1613416
336,0.3922898022\H,-1.6936712611,-2.4694233193,0.8634701338\H,-1.57431
0623,-2.2727346498,-1.3148615593\H,-1.7248901712,2.3015035693,1.234275
9465\H,-1.6148793866,2.4488646072,-0.9524899108\H,4.1874502519,1.49608
78269,-0.0316028514\H,4.3119686938,-0.4230914997,-0.8571139358\H,1.586
3890414,-1.7860895693,0.894298415\Version=EM64L-G09RevD.01\State=1-A\
HF=-1560.9508734\RMSD=4.873e-09\RMSF=4.770e-06\Dipole=-0.3932632,-0.08
8672,-0.3122051\Quadrupole=4.0761962,7.4207321,-11.4969283,0.3395187,-

1.8775741,1.7538689\PG=C01 [X(C5H9P4)]\ \@

Zero-point correction= 0.122695 (Hartree/Particle)

Thermal correction to Energy= 0.133552

Thermal correction to Enthalpy= 0.134497

Thermal correction to Gibbs Free Energy= 0.083735

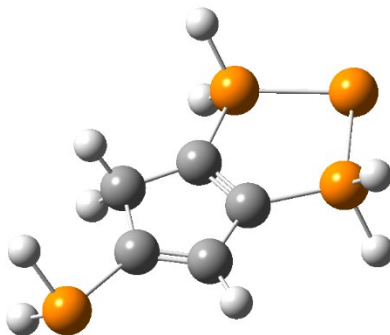
Sum of electronic and zero-point Energies= -1560.828178

Sum of electronic and thermal Energies= -1560.817321

Sum of electronic and thermal Enthalpies= -1560.816377

Sum of electronic and thermal Free Energies= -1560.867139

[P(H₂P)₂C₅H₃PH₂] cation model (isomer 2)



1\1\GINC-SAW125\FOpt\RM062X\TZVP\C5H9P4(1+)\CMACD\05-Oct-2015\0\#\#
M06

2x/TZVP scf=tight opt=tight freq pop=(full,nbo6read) test\Optimizatio
n of H[P(Cp(PH2)3)] axial H-Cp 2 cation\1,1\P,-1.6570074937,1.4941585
307,0.00099662\P,-3.1975829368,-0.0276299867,-0.0160552714\P,-1.616929
5979,-1.5030626743,-0.0037773246\C,-0.0484931906,0.6966021164,0.000201
9572\C,-0.0227919317,-0.6493252539,-0.0006559181\C,1.3617962983,1.2064
084297,0.0079674713\C,1.3555312493,-1.1304640877,0.0071636609\C,2.1780
165092,-0.0624572162,0.0118611488\H,1.5758518137,1.8208606092,0.888903
9561\H,1.6550865111,-2.1689990719,0.009743349\P,4.0218827755,-0.189836
9816,0.0229888283\H,-1.6085268561,-2.3799562481,1.0966480876\H,-1.6040
987332,-2.3864634878,-1.0986011947\H,-1.6956726498,2.3690925604,1.1027
322026\H,-1.6928407511,2.3838540544,-1.0885285003\H,4.2165457295,0.805
2480461,1.0164175336\H,4.2236637915,0.7185445484,-1.0480172262\H,1.585
7500728,1.821047163,-0.8703415401\Version=EM64L-G09RevD.01\State=1-A\
HF=-1560.9501681\RMSD=8.025e-09\RMSF=4.721e-06\Dipole=-0.6536597,0.465
3677,-0.0076242\Quadrupole=2.7243097,6.9712329,-9.6955426,3.4817233,-0

.0855672,0.0880039\PG=C01 [X(C5H9P4)]\ \@

Zero-point correction= 0.122571 (Hartree/Particle)

Thermal correction to Energy= 0.132571

Thermal correction to Enthalpy= 0.133516

Thermal correction to Gibbs Free Energy= 0.086432

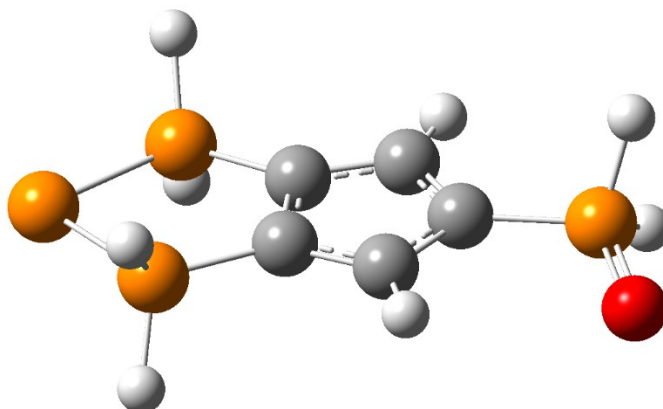
Sum of electronic and zero-point Energies= -1560.827597

Sum of electronic and thermal Energies= -1560.817597

Sum of electronic and thermal Enthalpies= -1560.816652

Sum of electronic and thermal Free Energies= -1560.863736

$P(H_2P)_2C_5H_2P(O)H_2$ model



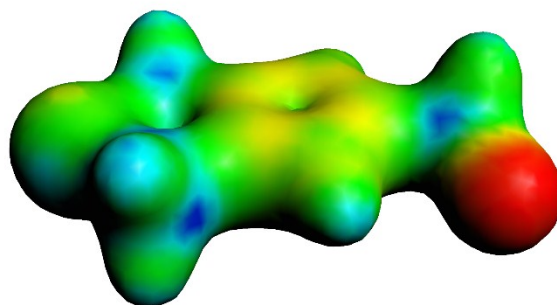
1\1\GINC-SAW38\FOpt\RM062X\TZVP\C5H8O1P4\CMACD\30-Sep-2015\0\#\nM062x/

tzvp scf=tight opt freq pop=(full,nbo6read) test\Optimization of [P(C
p(PH2)3)O] oxide zwitterion\n0,1\P,1.5720136101,-1.5047476679,0.103060
528\P,3.1317816474,-0.0065309858,-0.0499954657\P,1.5762892135,1.500368
8889,-0.1054719111\C,0.0022106159,-0.7092817966,0.0531139729\C,0.00292
66588,0.7095650747,-0.0874680846\C,-1.3204513485,-1.1413393489,0.12583
38262\C,-1.3195864067,1.149742013,-0.08344668\C,-2.1289210931,0.004410
8995,0.0401119461\H,-1.686207871,-2.1504952751,0.2405345903\H,-1.65786
86266,2.1729317904,-0.158336008\P,-3.9148003993,-0.093450795,0.0645168
973\H,1.6827302546,2.4332720273,0.945775461\H,1.7100764991,2.351730036
1,-1.2207462352\H,1.694610671,-2.3028324272,1.257354524\H,1.6860018958
, -2.483180993,-0.9045503797\H,-4.33042071,0.2666507784,-1.2399262146\H
, -4.3114878385,1.0612254417,0.7851694933\O,-4.4694475226,-1.4004853306
,0.5509437199\nVersion=EM64L-G09RevD.01\State=1-A\HF=-
1635.8169409\RMS

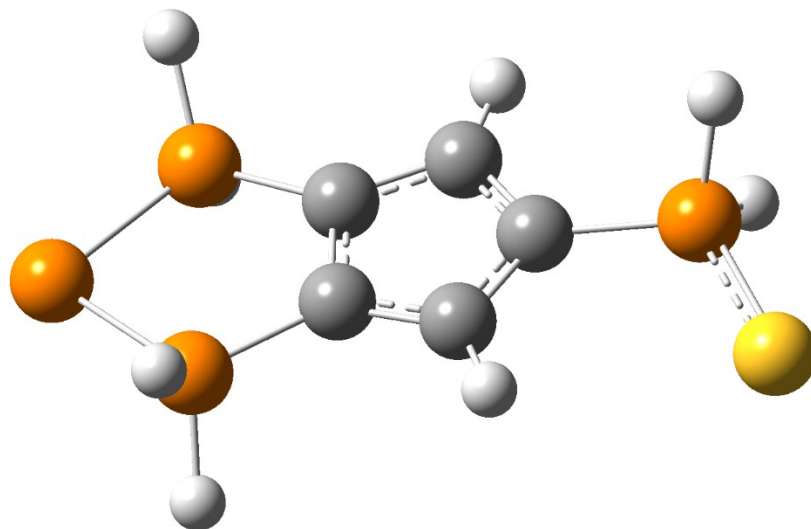
D=4.185e-09\RMSF=9.538e-06\Dipole=2.2067803,1.2804909,-0.4572542\Quadrupole=-12.567451,11.939073,0.628378,-11.048655,4.2320712,0.5000082\PG=C01 [X(C5H8O1P4)]\ \@

Zero-point correction=	0.116179 (Hartree/Particle)
Thermal correction to Energy=	0.127604
Thermal correction to Enthalpy=	0.128549
Thermal correction to Gibbs Free Energy=	0.076062
Sum of electronic and zero-point Energies=	-1635.700762
Sum of electronic and thermal Energies=	-1635.689337
Sum of electronic and thermal Enthalpies=	-1635.688392
Sum of electronic and thermal Free Energies=	-1635.740879

Electrostatic Potential



$P(H_2P)_2C_5H_2P(S)H_2$ model



1\1\GINC-SAW52\FOpt\RM062X\TZVP\C5H8P4S1\CMACD\30-Sep-2015\0\#\#
M062x/

TZVP scf=tight opt freq pop=(full,nbo6read) test\Optimization of [P(C
p(PH2)3)S] sulfide zwitterion\0,1\P,1.5764938673,-1.5020115536,0.1316
096122\P,3.1324709796,-0.0114940303,-0.1004414587\P,1.5784696998,1.495
6349919,-0.1624992807\C,0.0063158026,-0.7091279407,0.0668513195\C,0.00
45823378,0.7047759739,-0.1108887661\C,-1.3144740919,-1.141255903,0.167
0157396\C,-1.3172322476,1.1444205685,-0.1058608492\C,-2.1228016975,0.0
007889146,0.0590120512\H,-1.6800331925,-2.1475152259,0.3073561126\H,-1
.659412196,2.1636358432,-0.2085464594\P,-3.9164738804,-0.0502483411,0.
1048271364\H,1.6929408171,2.4491228402,0.8690824953\H,1.6991658592,2.3
238493521,-1.2953455365\H,1.7107169696,-2.2532645535,1.3157200683\H,1.
6817100549,-2.5188429131,-0.8380555896\H,-4.3227595396,0.4368176413,-1
.1576896988\H,-4.2594625138,1.05839169,0.9124212715\S,-4.7255017386,-1
.7641574243,0.6204824224\Version=EM64L-G09RevD.01\State=1-A\HF=-1958.
7800991\RMSD=5.164e-09\RMSF=2.314e-06\Dipole=2.400365,1.396644,-0.3982

164\Quadrupole=-12.3586199,12.3594267,-0.0008068,-10.3448633,3.4019045
,-0.478375\PG=C01 [X(C5H8P4S1)]\ \@

Zero-point correction= 0.114515 (Hartree/Particle)

Thermal correction to Energy= 0.126205

Thermal correction to Enthalpy= 0.127149

Thermal correction to Gibbs Free Energy= 0.073787

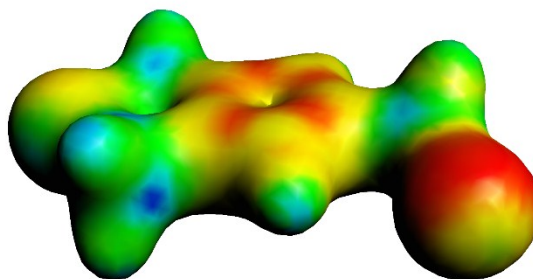
Sum of electronic and zero-point Energies= -1958.665584

Sum of electronic and thermal Energies= -1958.653894

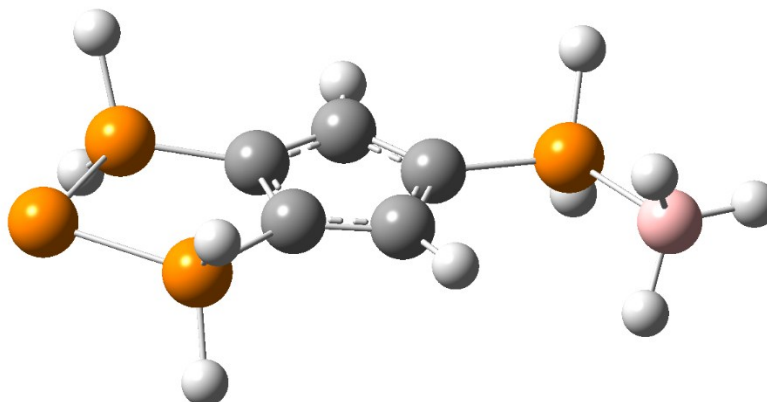
Sum of electronic and thermal Enthalpies= -1958.652950

Sum of electronic and thermal Free Energies= -1958.706312

Electrostatic Potential



$P(H_2P)_2C_5H_2P(BH_3)H_2$ model



1\1\GINC-SAW76\FOpt\RM062X\TZVP\C5H11B1P4\CMACD\30-Sep-2015\0\#\nM062x

```
/TZVP scf=tight opt freq pop=(full,nbo6read) test\Optimization of [P(\n\nCp(PH2)3)BH3] borane adduct zwitterion\0,1\P,-0.0211255995,0.12430273\n87,0.4347055102\P,2.0702603814,-0.0380793259,-0.1122550652\P,2.0650764\n141,2.0966615462,-0.4808350585\C,-0.530370996,1.8008101381,0.269931576\n6\C,0.4678846058,2.7432503936,-0.1142284433\C,-1.7251060657,2.48809483\n41,0.4724690705\C,-0.118436908,4.0052449421,-0.1668561758\C,-1.4687266\n254,3.8436419943,0.2000435927\H,-2.678809156,2.0828192119,0.7760568946\n\H,0.3655841599,4.9323729669,-0.4355964379\P,-2.7464902002,5.098909739\n9,0.3315368994\H,3.0641773204,2.7672371613,0.25203319\H,2.4517080017,2\n.4171162464,-1.7973922662\H,-0.2703647977,-0.3612128791,1.7342941172\H\n,-0.8414315942,-0.7309140211,-0.3262561327\H,-2.674939242,5.8086795438\n,-0.8835869426\H,-2.2064258421,6.0759135192,1.193186138\B,-4.512184159\n1,4.4428114144,0.8158443096\H,-4.7863990062,3.673357371,-0.0713462881\H\nH,-5.2058769383,5.4260820304,0.8521159943\H,-4.3393857028,3.9188452539
```

,1.8885709372\\Version=EM64L-G09RevD.01\\State=1-A\\HF=-
1587.2218363\\RMS

D=8.775e-09\\RMSF=4.610e-06\\Dipole=2.6297648,-0.6822138,-0.5201269\\Quad
rupole=-13.0313784,10.6719963,2.359382,12.3324354,1.2010692,-4.5052192

\\PG=C01 [X(C5H11B1P4)]\\@

Zero-point correction= 0.142949 (Hartree/Particle)

Thermal correction to Energy= 0.155733

Thermal correction to Enthalpy= 0.156678

Thermal correction to Gibbs Free Energy= 0.101559

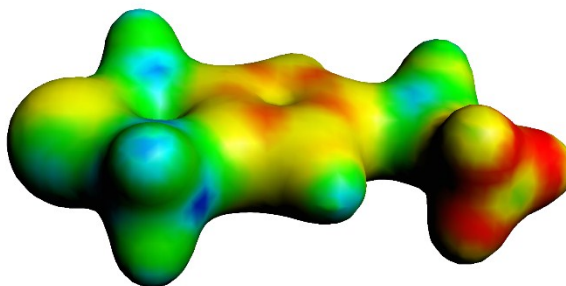
Sum of electronic and zero-point Energies= -1587.078887

Sum of electronic and thermal Energies= -1587.066103

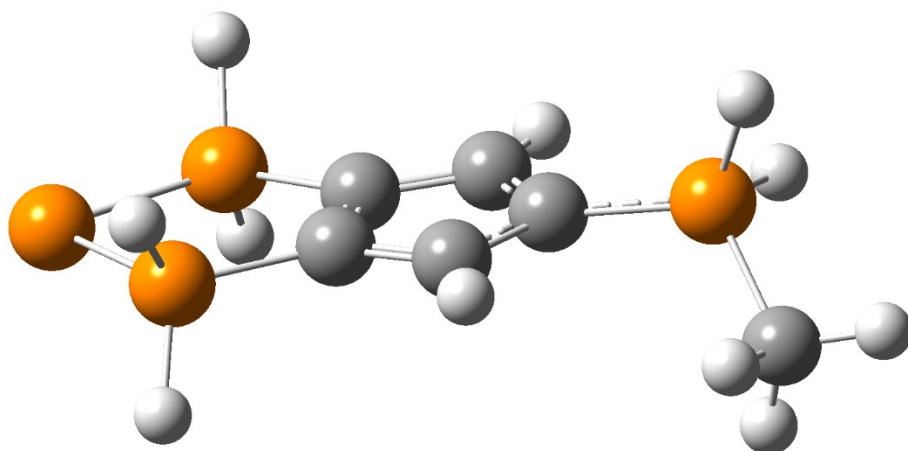
Sum of electronic and thermal Enthalpies= -1587.065159

Sum of electronic and thermal Free Energies= -1587.120277

Electrostatic Potential



P(H₂P)₂C₅H₂P(CH₃)H₂ model



1\1\GINC-SAW215\FOpt\RM062X\TZVP\C6H11P4(1+)\CMACD\01-Oct-2015\0\#\# M0

62x/TZVP scf=tight opt freq pop=(full,nbo6read) test\Optimization of
[P(Cp(PH2)3)CH3] methylated cation\1,1\P,-0.01099758,0.0292929103,0.0
880868764\P,2.1487051881,0.0235622883,-0.0553030857\P,2.1001565343,2.1
877296921,-0.0654592066\C,-0.5884300879,1.7128545837,0.0417863298\C,0.
405201168,2.726724817,-0.0352484907\C,-1.8386535938,2.3096255362,0.078
9630601\C,-0.218377563,3.9642974018,-0.0462433502\C,-1.6077217749,3.70
52986701,0.0261019359\H,-2.7948416795,1.8112959057,0.1332698907\H,0.25
37406718,4.9339377646,-0.0962993542\P,-2.8429198941,4.9421547655,-0.00
68760705\H,2.7959242928,2.7812479117,1.0050693195\H,2.7543797222,2.749
9434526,-1.1782817118\H,-0.5116106568,-0.5998592542,1.2429037083\H,-0.
6365101888,-0.7099043002,-0.9330517734\H,-3.50012133,5.0679713626,-1.2
430847931\H,-2.2274614008,6.186572351,0.1798666135\C,-4.1390556854,4.7
352127724,1.2387497866\H,-4.6250649249,3.7726961205,1.0829103538\H,-4.
8751211034,5.5328631972,1.1476958046\H,-3.6826020638,4.7524268709,2.22

68795767\\Version=EM64L-G09RevD.01\State=1-A\HF=-
1600.2700727\RMSD=5.4

34e-09\RMSF=6.047e-06\Dipole=-2.186539,2.0032121,0.1910243\Quadrupole=
11.7375502,7.5869848,-19.324535,-12.0985067,-2.7982975,0.525314\PG=C01

[X(C6H11P4)]\ \@

Zero-point correction= 0.150971 (Hartree/Particle)

Thermal correction to Energy= 0.162374

Thermal correction to Enthalpy= 0.163318

Thermal correction to Gibbs Free Energy= 0.112023

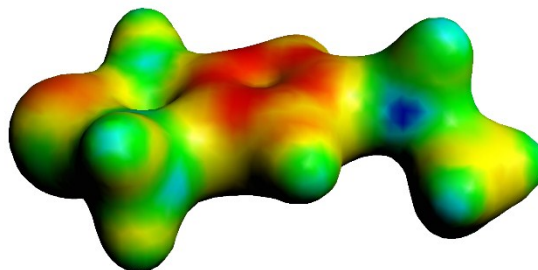
Sum of electronic and zero-point Energies= -1600.119102

Sum of electronic and thermal Energies= -1600.107699

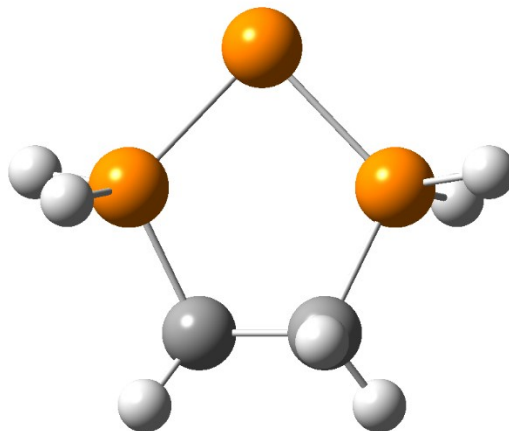
Sum of electronic and thermal Enthalpies= -1600.106754

Sum of electronic and thermal Free Energies= -1600.158049

Electrostatic Potential



[P(H₂PCH₂CH₂PH₂)] cation model

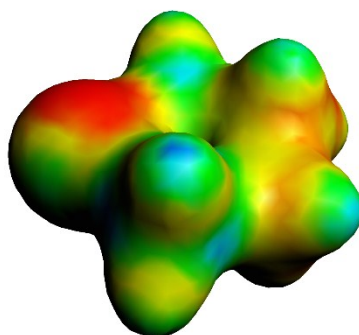


1\1\GINC-SAW79\FOpt\RM062X\TZVP\C2H8P3(1+)\CMACD\02-Jun-2015\0\#\nM062

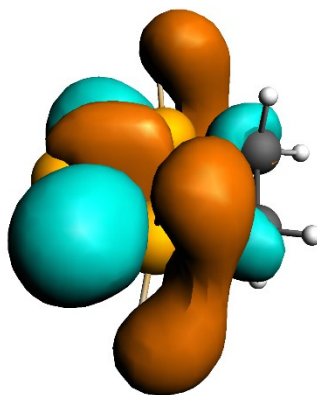
```
x/TZVP scf=tight opt freq pop=(full,nbo6read) test\Optimization of [P
(dHpe)] cation\1,1\P,0.0000417394,-2.6793916749,-0.0001176252\P,1.398
1617574,-1.1226426049,-0.5163736556\P,-1.3980511753,-1.122671065,0.516
3085372\C,0.5308135091,0.5020920684,-0.5486435446\C,-0.530715773,0.502
0731751,0.5486045091\H,-1.2494406531,1.3100975502,0.4108666387\H,-0.08
26574128,0.6001755001,1.538378314\H,0.0827546957,0.6002057584,-1.53841
58681\H,1.2495276798,1.3101230931,-0.4108878402\H,-2.0044590682,-1.291
6926864,1.7724753102\H,-2.5052707011,-0.9807451649,-0.3423063083\H,2.5
053339555,-0.9807338347,0.3423073979\H,2.0046369368,-1.2916171942,-1.7
725127252\Version=EM64L-G09RevD.01\State=1-A\HF=-
1104.7489336\RMSD=3.
672e-09\RMSF=2.864e-05\Dipole=0.0000085,1.3694838,0.0000623\Quadrupole
=5.7988971,-2.434916,-3.3639811,0.0001105,-4.5007131,0.0000522\PG=C01
[X(C2H8P3)]\@\n
```

Zero-point correction=	0.098120 (Hartree/Particle)
Thermal correction to Energy=	0.103877
Thermal correction to Enthalpy=	0.104822
Thermal correction to Gibbs Free Energy=	0.068202
Sum of electronic and zero-point Energies=	-1104.650813
Sum of electronic and thermal Energies=	-1104.645056
Sum of electronic and thermal Enthalpies=	-1104.644112
Sum of electronic and thermal Free Energies=	-1104.680731

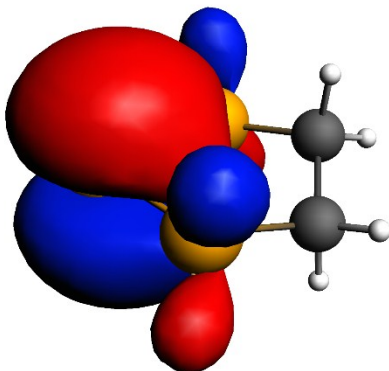
Electrostatic Potential



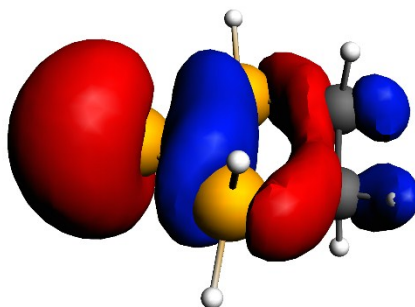
LUMO



HOMO



HOMO-1



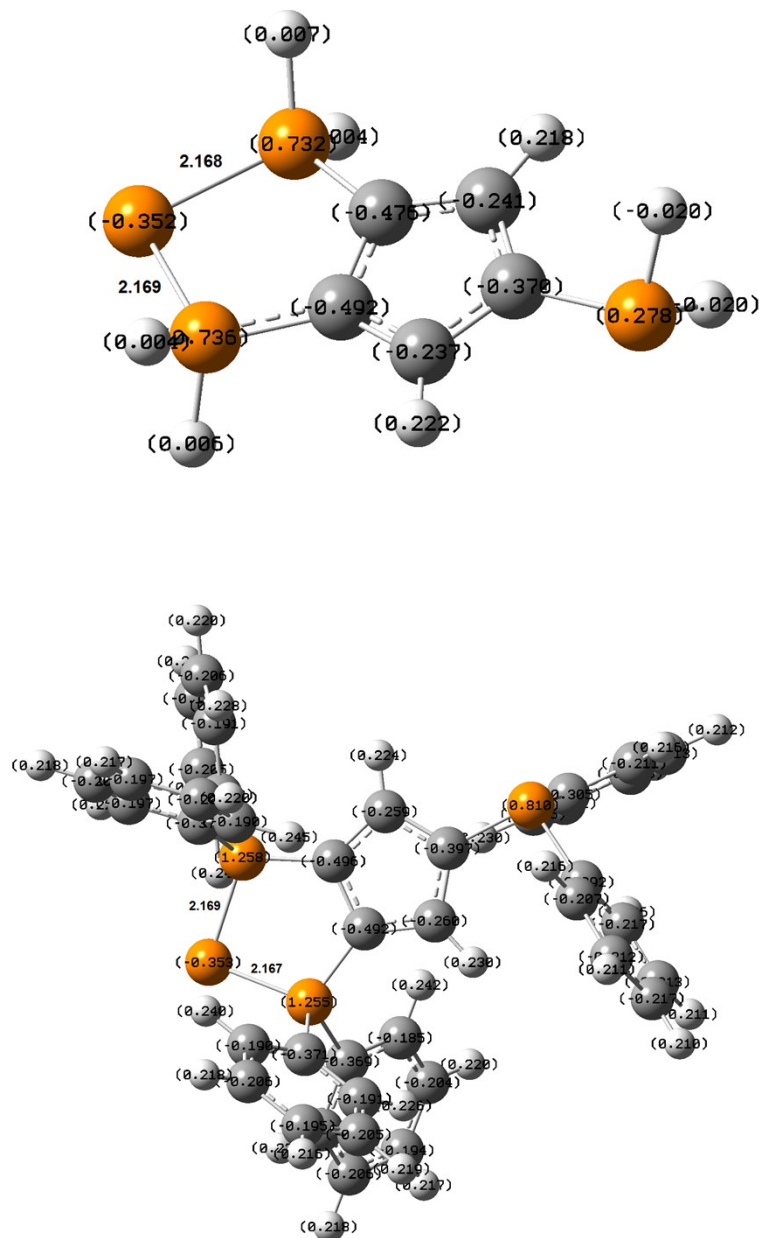


Fig. 16 NBO Charges and selected bond lengths of 1' (above) and 1 (below)
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1. M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. Montgomery, J. A., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S.

- S. Iyengar, J. Tomasi, M. Cossi, N. Rega, N. J. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski and D. J. Fox, *Gaussian09, Revision D01*, (2009) Gaussian, Inc., Wallingford CT.
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