

Electronic Supplementary Information for

B-site ordered double perovskite  $\text{LaBa}_{1-x}\text{Sr}_x\text{ZnSbO}_6$  ( $0 \leq x \leq 1$ ):  
 $\text{Sr}^{2+}$ -doping-induced symmetry evolution and structure-  
luminescence correlations

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Table S1 Lattice parameters of Eu<sup>3+</sup> doped samples obtained by Le-Bail fitting.

Formula	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	<i>V</i> (Å <sup>3</sup> )	$\beta$ (°)
La <sub>0.95</sub> Eu <sub>0.05</sub> BaZnSbO <sub>6</sub>	5.7093	5.7092	8.0735	263.16	89.893
La <sub>0.95</sub> Eu <sub>0.05</sub> Ba <sub>0.8</sub> Sr <sub>0.2</sub> ZnSbO <sub>6</sub>	5.6977	5.6943	8.0576	261.42	90.138
La <sub>0.95</sub> Eu <sub>0.05</sub> Ba <sub>0.6</sub> Sr <sub>0.4</sub> ZnSbO <sub>6</sub>	5.6908	5.6850	8.0632	260.86	90.219
La <sub>0.95</sub> Eu <sub>0.05</sub> Ba <sub>0.4</sub> Sr <sub>0.6</sub> ZnSbO <sub>6</sub>	5.6798	5.6740	8.0276	258.71	90.011
La <sub>0.95</sub> Eu <sub>0.05</sub> Ba <sub>0.2</sub> Sr <sub>0.8</sub> ZnSbO <sub>6</sub>	5.6739	5.6702	8.0219	258.08	90.040
La <sub>0.975</sub> Eu <sub>0.025</sub> SrZnSbO <sub>6</sub>	5.6564	5.6567	7.9991	255.94	89.976
La <sub>0.95</sub> Eu <sub>0.05</sub> SrZnSbO <sub>6</sub>	5.6566	5.6569	7.9977	255.92	89.995
La <sub>0.92</sub> Eu <sub>0.08</sub> SrZnSbO <sub>6</sub>	5.6537	5.6552	7.9941	255.59	90.006

 Table S2 Atomic coordinates, isotropic thermal displacement factors and site occupancies for LaBa<sub>1-x</sub>Sr<sub>x</sub>ZnSbO<sub>6</sub> (*x* = 0.2, 0.4, 0.6, 0.8).

LaBa <sub>0.8</sub> Sr <sub>0.2</sub> ZnSbO <sub>6</sub>	<i>x</i>	<i>y</i>	<i>z</i>	Occupancy	<i>B</i> <sub>eq</sub> (Å <sup>2</sup> )
Ba/La/Sr	0.499(1)	0	0.2495(4)	0.4/0.5/0.1	0.014(1)
Zn	0	0	0	1	0.014(1)
Sb	0	0	0.5	1	0.011(3)
O1	0.063(2)	0	0.256(3)	1	0.018(1)
O2	0.275(2)	0.236(2)	0.971(1)	1	0.018(1)

LaBa <sub>0.6</sub> Sr <sub>0.4</sub> ZnSbO <sub>6</sub>	<i>x</i>	<i>y</i>	<i>z</i>	Occupancy	<i>B</i> <sub>eq</sub> (Å <sup>2</sup> )
Ba/La/Sr	0.498(1)	0	0.2492(3)	0.3/0.5/0.2	0.014(1)
Zn	0	0	0	1	0.014(1)
Sb	0	0	0.5	1	0.011(1)
O1	0.066(2)	0	0.251(3)	1	0.022(1)
O2	0.264(3)	0.246(2)	0.031(1)	1	0.022(1)

LaBa <sub>0.4</sub> Sr <sub>0.6</sub> ZnSbO <sub>6</sub>	<i>x</i>	<i>y</i>	<i>z</i>	Occupancy	<i>B</i> <sub>eq</sub> (Å <sup>2</sup> )
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Ba/La/Sr	0.001(1)	0.5140(3)	0.2519(7)	0.2/0.5/0.3	0.009(1)
Zn	0	0	0	1	0.005(1)
Sb	0	0	0.5	1	0.003(1)
O1	0.081(4)	0.991(3)	0.262(5)	1	0.005(3)
O2	0.759(6)	0.275(5)	0.007(4)	1	0.005(3)
O3	0.277(5)	0.231(5)	0.954(3)	1	0.005(3)
LaBa <sub>0.2</sub> Sr <sub>0.8</sub> ZnSbO <sub>6</sub>	<i>x</i>	<i>y</i>	<i>z</i>	Occupancy	<i>B</i> <sub>eq</sub> (Å <sup>2</sup> )
Ba/La/Sr	0.005(2)	0.5202(2)	0.253(1)	0.1/0.5/0.4	0.008(1)
Zn	0	0	0	1	0.006(1)
Sb	0	0	0.5	1	0.0043(9)
O1	0.090(3)	0.995(2)	0.253(4)	1	0.003(2)
O2	0.759(5)	0.272(4)	0.005(3)	1	0.003(2)
O3	0.293(4)	0.228(4)	0.954(3)	1	0.003(2)

Table S3 Selected interatomic distances (Å) and bond angles (°) for LaBa<sub>1-x</sub>Sr<sub>x</sub>ZnSbO<sub>6</sub> (*x* = 0.2, 0.4, 0.6, 0.8). Calculated bond valence sums (BVS) for Zn<sup>2+</sup> and Sb<sup>5+</sup> are also given.

<b>LaBa<sub>0.8</sub>Sr<sub>0.2</sub>ZnSbO<sub>6</sub></b>			
La/Ba/Sr-O2 ×2	2.61(1)	Zn-O1 ×2	2.11(2)
La/Ba/Sr -O2 ×2	2.79(1)	Zn-O2 ×4	2.08(1)
La/Ba/Sr -O2 ×2	2.94(1)	BVS	2.09
La/Ba/Sr -O2 ×2	3.10(1)	Sb-O1 ×2	1.98(2)
La/Ba/Sr -O1 ×1	2.8(2)	Sb-O2 ×4	1.99(1)
La/Ba/Sr -O1 ×2	2.87(2)	BVS	5.33
La/Ba/Sr -O1 ×1	3.43(2)	Zn-O1- Sb	158.9
		Zn-O2- Sb	165.3

**LaBa<sub>0.6</sub>Sr<sub>0.4</sub>ZnSbO<sub>6</sub>**

La/Ba/Sr -O2 ×2	2.57(1)	Zn-O1 ×2	2.09(2)
La/Ba/Sr -O2 ×2	2.81(1)	Zn-O2 ×4	2.08(1)
La/Ba/Sr -O2 ×2	2.91(1)	BVS	2.16
La/Ba/Sr -O2 ×2	3.12(1)	Sb-O1 ×2	1.99(2)
La/Ba/Sr -O1 ×1	2.49(2)	Sb-O2 ×4	1.99(1)
La/Ba/Sr -O1 ×2	2.86(2)	BVS	2.25
La/Ba/Sr -O1 ×1	3.23(2)	Zn-O1- Sb	159.7
		Zn-O2- Sb	163.9

**LaBa<sub>0.4</sub>Sr<sub>0.6</sub>ZnSbO<sub>6</sub>**

La/Ba/Sr-O1	2.39(2)	Zn-O1 ×2	2.15(4)
La/Ba/Sr-O1	2.75(2)	Zn-O2 ×2	2.08(3)
La/Ba/Sr-O1	3.00(2)	Zn-O3 ×2	2.09(3)
La/Ba/Sr-O1	3.31(2)	BVS	2.04
La/Ba/Sr-O2	2.75(3)	Sb-O1 ×2	1.97(4)
La/Ba/Sr-O2	2.76(3)	Sb-O2 ×2	1.95(3)
La/Ba/Sr-O2	2.85(3)	Sb-O3 ×2	2.02(3)
La/Ba/Sr-O2	3.01(3)	BVS	5.44
La/Ba/Sr-O3	2.49(3)	Zn-O1- Sb	154.0
La/Ba/Sr-O3	2.71(3)	Zn-O2- Sb	171.5
La/Ba/Sr-O3	2.95(3)	Zn-O3- Sb	156.8
La/Ba/Sr-O3	3.28(3)		

**LaBa<sub>0.2</sub>Sr<sub>0.8</sub>ZnSbO<sub>6</sub>**

La/Ba/Sr-O1	2.30(2)	Zn-O1 ×2	2.09(3)
La/Ba/Sr-O1	2.73(2)	Zn-O2 ×2	2.06(3)
La/Ba/Sr-O1	3.02(2)	Zn-O3 ×2	2.14(2)
La/Ba/Sr-O2	2.73(3)	BVS	2.08

La/Ba/Sr-O2	2.80(3)	Sb-O1 ×2	2.04(3)
La/Ba/Sr-O2	2.83(3)	Sb-O2 ×2	1.96(3)
La/Ba/Sr-O2	2.99(3)	Sb-O3 ×2	1.97(2)
La/Ba/Sr-O3	2.46(3)	BVS	5.28
La/Ba/Sr-O3	2.77(3)	Zn-O1- Sb	151.3
La/Ba/Sr-O3	2.87(3)	Zn-O2- Sb	172.4
La/Ba/Sr-O3	3.34(2)	Zn-O3- Sb	154.7

Table S4 Lattice parameters of  $\text{LaBa}_{1-x}\text{Sr}_x\text{ZnSbO}_6$  ( $0 \leq x \leq 1$ ) obtained by Rietveld refinement.

Formula	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	<i>V</i> (Å <sup>3</sup> )	$\beta$ (°)
$\text{LaBaZnSbO}_6$	5.7253(2)	5.7066(3)	8.0689(3)	263.63(2)	90.00(4)
$\text{LaBa}_{0.8}\text{Sr}_{0.2}\text{ZnSbO}_6$	5.7247(1)	5.6960(1)	8.0533(2)	262.602(9)	90.01(1)
$\text{LaBa}_{0.6}\text{Sr}_{0.4}\text{ZnSbO}_6$	5.7132(1)	5.6807(1)	8.0318(2)	260.67(1)	90.035(6)
$\text{LaBa}_{0.4}\text{Sr}_{0.6}\text{ZnSbO}_6$	5.6945(2)	5.6731(2)	8.0225(3)	259.17(2)	89.99(2)
$\text{LaBa}_{0.2}\text{Sr}_{0.8}\text{ZnSbO}_6$	5.6750(2)	5.6667(2)	8.0126(2)	257.67(1)	89.961(6)
$\text{LaSrZnSbO}_6$	5.6586(2)	5.6577(2)	7.9964(2)	256.00(2)	90.00(2)

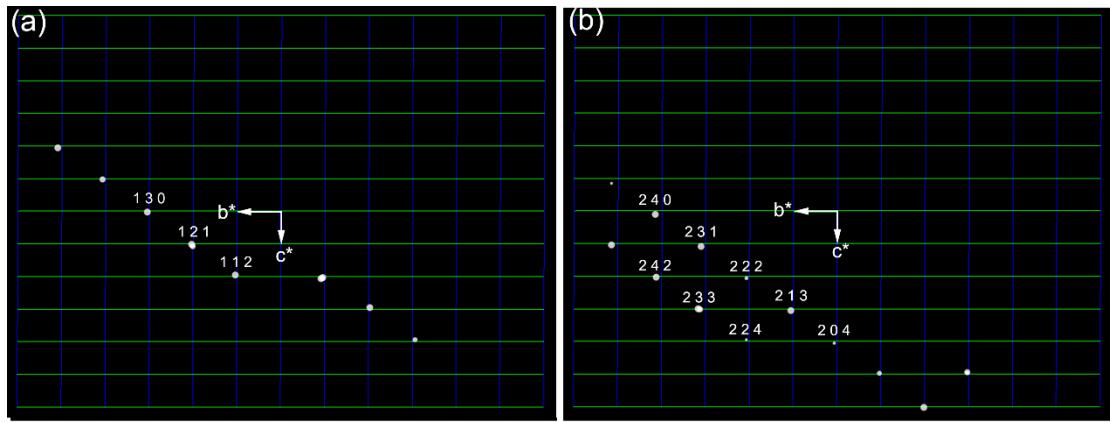


Fig. S1 Selected views in the reciprocal space corresponding to (a)  $1kl$  and (b)  $2kl$  planes for  $\text{LaBaZnSbO}_6$ . The reflection condition is consistent with  $h + k + l = 2n$ .

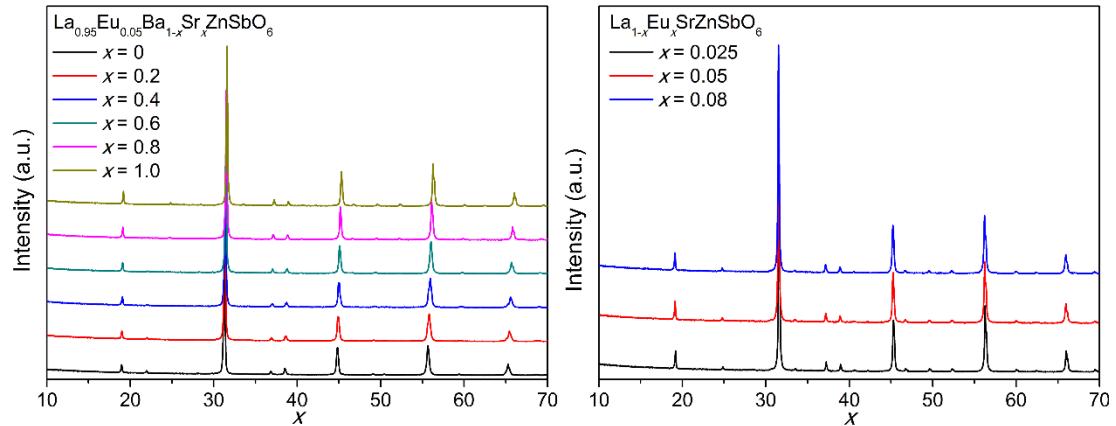
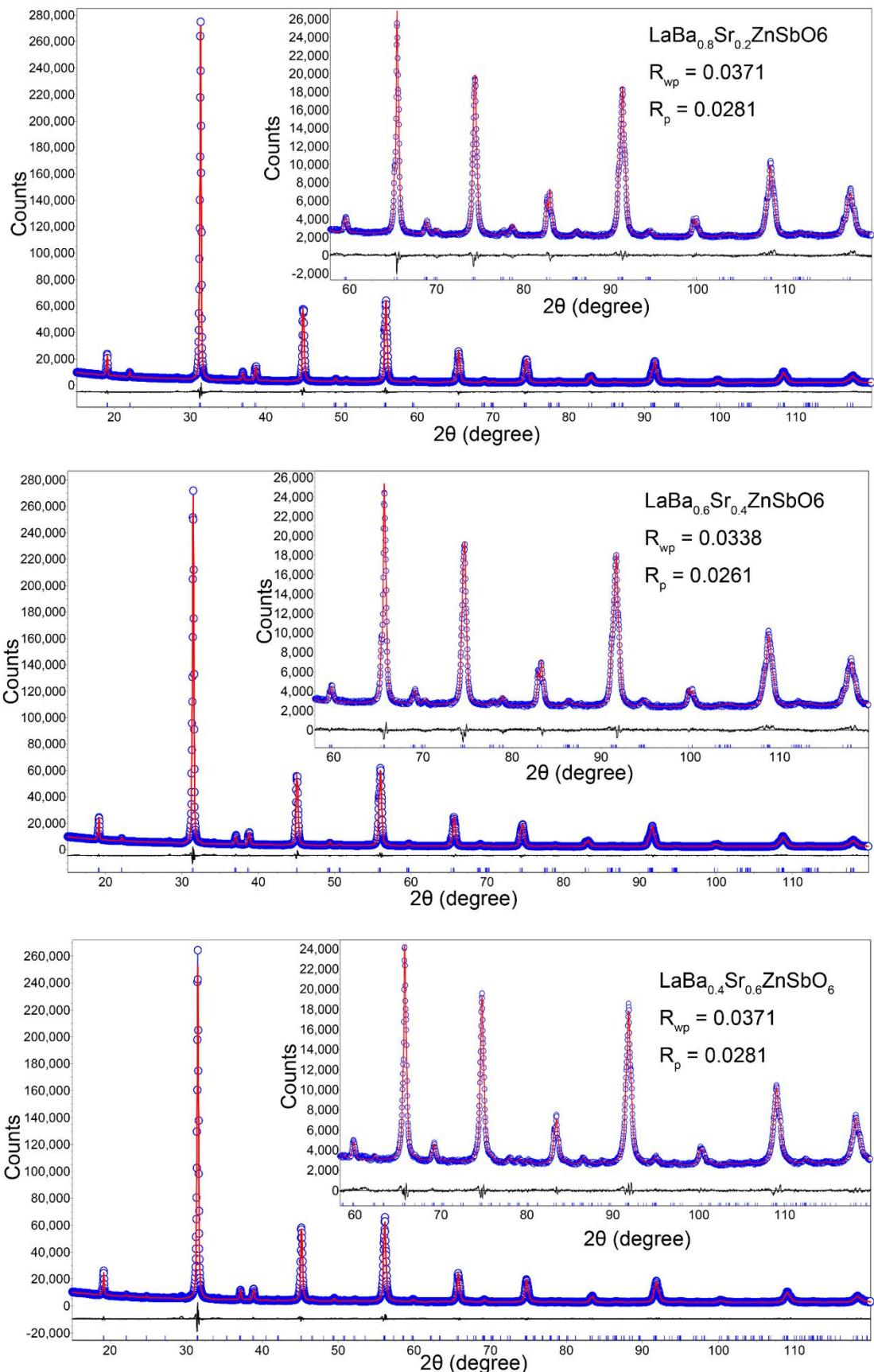


Fig. S2 XRD patterns for  $\text{La}_{0.95}\text{Eu}_{0.05}\text{Ba}_{1-x}\text{Sr}_x\text{ZnSbO}_6$  ( $0 \leq x \leq 1$ ) and  $\text{La}_{1-x}\text{Eu}_x\text{SrZnSbO}_6$  ( $x = 0.025, 0.05, 0.08$ ).



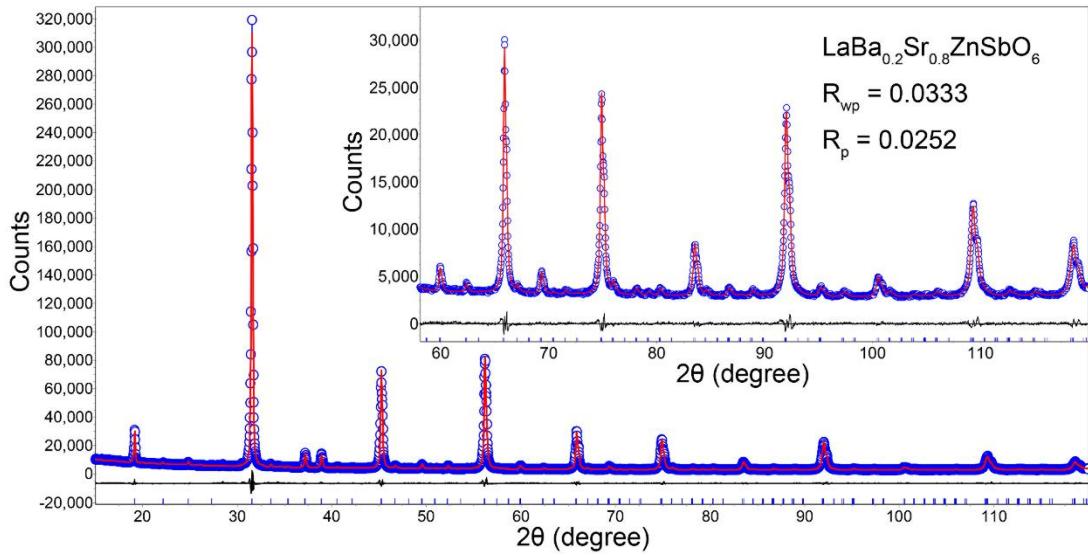


Fig. S3 Final convergence of Rietveld refinement for  $\text{LaBa}_{1-x}\text{Sr}_x\text{ZnSbO}_6$  ( $x = 0.2, 0.4, 0.6, 0.8$ ). The blue symbol  $\circ$  represents observed data and the red solid line is the calculated pattern; the blue marks below the diffraction patterns are the expected reflection positions, and the difference curve is also shown at the bottom. Enlargements of the high angle part are shown in the top-right corner.

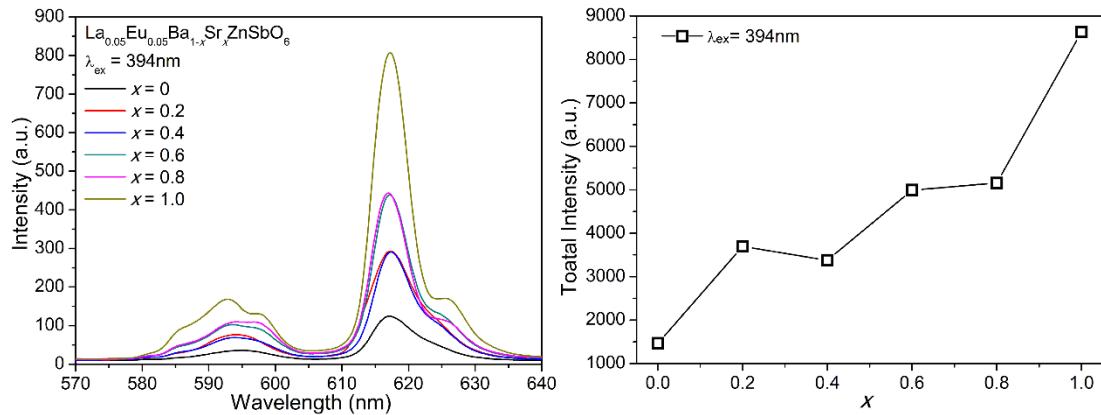


Fig. S4 (a) emission spectra of  $\text{La}_{0.95}\text{Eu}_{0.05}\text{Ba}_{1-x}\text{Sr}_x\text{ZnSbO}_6$  excited by 394 nm; (b) total emission intensities excited under 394 nm along with the content of  $\text{Sr}^{3+}$ .

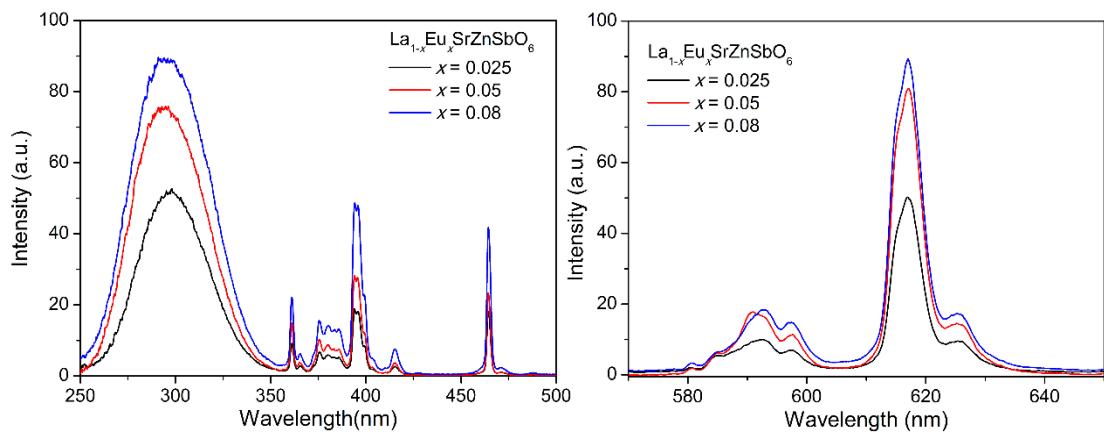


Fig. S5 Excitation (left) and emission (right) spectra for  $\text{La}_{1-x}\text{Eu}_x\text{SrZnSbO}_6$  ( $x = 0.025, 0.05, 0.08$ ).