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Supplementary Material

A novel example of double reactivity by either photochemical [2+2] or thermal additions of an ionic organic supramolecular assembly

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Figure S1. Figure S1. (a) Monitoring the photoreaction of **1** by FT-IR at different UVirradiation time periods. (b) Comparison of the FT-IR spectra of **1** before (blue) and after the photoreaction for 24 h (red), respectively.



(b)

Figure S2. Comparison of the ¹H NMR spectra obtained from the photoreaction of compound **1** after irradiation for 48 h and the mixture after a second grinding-irradiation step (additional UV-irradiation for 1 day).



Figure S3. (a) Monitoring structural changes of the photoreaction of 1 by X-Ray Powder Diffraction at different UV-irradiation time periods.



Figure S4. Thermal analyses TGA/DSC of **1** heated at 190 °C for 30 min and cool down at 100 °C and heated until 600 °C.



Figure S5. FT-IR spectrum of **1** fresh heated in the solid state upon nitrogen atmosphere at 190 °C for 30 min



Figure S6. (a) Comparison of the ¹H NMR spectra of compound **1** in DMSO after UV irradiation for 2 days at 254 nm and after heating under hydrothermal conditions at 140 and 190 °C for 2 days, showing the isomerisation of *rctt* to *rtct*-isomer. (b)



Figure S7. The ¹H NMR spectrum of compound **1** in DMSO-D₆ after UV irradiation for 2 days at 254 nm and heated under hydrothermal conditions at 120 °C for 2 days, showing the isomerisation of *rctt* to *rcct* and *rtct*-isomers together with the products of the thermal ring cleavage (Maleic and Fumaric acid) and traces of hydroamination product.

