## Mesoporous $In_2O_3$ nanocrystals: synthesis, characterization and $NO_x$ gas sensor at room temperature

Jun Gao<sup>‡a</sup>, Hongyuan Wu<sup>‡a,c,</sup>, Jiao Zhou<sup>a</sup>, Liyuan Yao<sup>a</sup>, Guo Zhang<sup>a</sup>, Shuang Xu<sup>a</sup>, Yu Xie<sup>d</sup>, Li Li<sup>\*a,</sup>

<sup>b</sup>, Keying Shi\*a

<sup>a</sup> Key Laboratory of Functional Inorganic Material Chemistry, Ministry of Education. Key

Laboratory of Physical Chemistry, School of Chemistry and Material Science, Heilongjiang

University, Harbin, 150080, P. R. China.

<sup>b</sup> Key Laboratory of Chemical Engineering Process & Technology for High-efficiency

Conversion, School of Chemistry and Material Science, Heilongjiang University, Harbin 150080,

P. R. China.

<sup>c</sup> College of Chemistry and Chemical Engineering, Qiqihar University, Qiqihar 161006, P.R. China

<sup>d</sup> Department of Materials Chemistry, Nanchang Hangkong University, Nanchang 330063, P.R. China



Fig. S1 Low-angle XRD X-ray diffraction diagrams of SBA-16 (1) and mesoporous In<sub>2</sub>O<sub>3</sub> materials replicated from mesoporous silica of INS-1, INS-2, INS-3, INS-4. respectively.



Fig. S2 TEM image of the INS-2.

Fig.S2 exhibited INS-2 sample was composed of  $In_2O_3$  nanoparticles (5~10 nm) with some mesopores.



Fig. S3 TEM/HRTEM images of the INS-3. (a) TEM image, (b-d) HRTEM images of (a) INS-3.

Volume concentration /ppm		97.0	48.5	29.1	9.70	4.85	2.91	0.97
INS-1	Response	9.74	3.81	3.42	2.7	0.53		-
	Response time/s	105	131	56	167	35		-
INS-2	Response	41.2	17.7	13.5	5.36	1.58	1.19	1.2
	Response time/s	196	276	409	450	409	323	165
INS-3	Response	158.7	50.7	36.8	19.7	12.6	7.2	1.9
	Response time/s	87	65	206	131	48	49	45
INS-4	Response	7.39	6.9	5.84	4.4	0.76	-	-
	Response time/s	267	200	257	333	89.3	-	-

Table S1 Comparison of the response-recovery results of response and response time(s) with different samples to NO<sub>x</sub> (RH: 42%)



Fig. S4 (a) The Mott-Schottky curves of INS-1, INS-2, INS-3 and INS-4 samples, (b) The EIS curves of INS-1, INS-2 and INS-3 samples. The right inset is the corresponding equivalent circuit model, the inset in left shows the logarithmic plot of the imaginary part of the impedance.

Here  $R_{\Omega}$  indicates the uncompensated bulk resistance of the electrolyte, separator and electrode. The Q1R1 parallel element corresponds to the electrode film interface capacitance (Q1) and the surface pore resistance (R1).<sup>1</sup> The Q2R2 parallel element might be attributed to the possible break down of the electrolyte and the electrode material's internal microstructures.  $R_{ct}$  is attributed to the charge-transfer resistance at the active material interface and C is the constant phase angle element, involving double layer capacitance.

Table S2 Carrier concentrations and fitted impedance parameters of the INS-1, INS-2 and INS-3.

Samples	$R_{\Omega}(\Omega)$	$R_1(\Omega)$	$R_2(\Omega)$	$R_{ct}(\Omega)$	Carrier concentration (cm <sup>-3</sup> )
INS-1	972.8	5.71×10 <sup>5</sup>	1987	5.67×10 <sup>4</sup>	3.7×10 <sup>17</sup>
INS-2	846.6	3.34×10 <sup>5</sup>	2509	4.89×10 <sup>4</sup>	2.7×10 <sup>17</sup>
INS-3	562.4	6639	1314	125.2	9.3 ×10 <sup>17</sup>

## Notes and references

 R. Sakamoto, Sh. Katagiri, H. Maeda, H. Nishihara, *Coordination Chemistry Reviews*, 2013, 257, 1493-1506.