Rhodamine Embedded bio-compatible Smart Molecule Mimicking Combinatorial Logic Circuit and 'Key-pad Lock' memory device for defending information risk.

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Supporting Information for Publication



Fig S1. ¹H NMR spectrum of L^Bin CDCl₃.











Fig S4(B). ¹H NMR spectrum of L^c in CD₃OD (Illustrating mainly aliphatic proton)



Fig. S4(C). ¹H NMR spectrum of L^c in CD₃OD (Illustrating mainly aromatic proton).



Fig. S5. ¹³C-NMR spectra of L^c in DMSO (d₆)



Fig S6. Mass Spectra of L^{c} (C₃₆H₃₈N₄O₅+H⁺).



Fig. S7. IR spectrum of L^c.



Fig. S8. UV-Vis spectral studies of selective binding of L^{c} (20 μ M) toward Al³⁺ over other metal ions in HEPES buffer at pH 7.0 in H₂O–MeOH = 3:7 (v/v) at 25 °C



Fig. S9. Fluorescence studies of selective binding of L^c (10 μ M) toward Al³⁺ over other metal ions in HEPES buffer at pH 7.0 in H₂O–MeOH = 3:7 (v/v) at 25 °C.



Fig. S10. Mass Spectra of Al-Complex of Receptor (L^{c}) ($C_{36}H_{38}N_4O_5+AI^{3+}$)



Fig. S11. Mass Spectra of Cu-Complex of Receptor (L^C).



Fig. S12 (A). Change in absorption upon addition of EDTA to L^{c} - Al³⁺ complex in HEPES buffer at pH 7.0 in H₂O–MeOH = 3:7 (v/v) at 25 °C.



Fig. S12 (B). Change in emission upon addition of EDTA to L^{c} -Al³⁺ complex in HEPES buffer at pH 7.0 in H₂O–MeOH = 3:7 (v/v) at 25 °C.



Fig. S13. Change in absorption upon addition of EDTA to $L^{c} - Cu^{2+}$ complex in HEPES buffer at pH 7.0 in H₂O-MeOH = 3:7 (v/v) at 25 °C.



Fig. S14. Comparative IR spectra of L^C and complex (L^C+Al³⁺).



Fig. S15 (A). ¹H NMR Studies of the L^C-Al³⁺ complex in CD₃OD.



Fig. S15 (B). ¹H NMR Studies of the L^C-Al³⁺ in CD₃OD (illustrating mainly aromatic protons).



Fig. S16. ¹³C-NMR spectra of complex ($L^{C}-AI^{3+}$) in DMSO (d_{6}).



Fig. S17. (A) Depicts % cell viability of HepG2 and HCT116 cells treated with different concentrations (1 μ M-100 μ M) of L^c for 12 hours determined by MTT assay.