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## **Supporting Information**

## Cadmium metal-organic frameworks: Ln<sup>3+</sup> ions functionalized

## assembly, fluorescence tuning and polymer film preparation

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Scheme S1 The coordination environment of central atom Cd and the structure of Cd-MOF



Fig. S1 FTIR spectra of as-synthesized Cd-MOF, Eu<sup>3+</sup>@Cd-MOF and Sm<sup>3+</sup><sub>0.98</sub>/ Tb<sup>3+</sup><sub>0.02</sub>@Cd-MOF.



Fig. S2 Thermogravimetric analysis of Cd-MOF and Eu<sup>3+</sup>@Cd-MOF.



Fig. S3 CIE diagrams of Ln<sup>3+</sup>@Cd-MOF: (a) Ln=Eu; (b) Ln=Tb; (c) Ln=Sm; (d) Ln=Dy.



Fig. S4 Excitation and emission spectra of pure ligand 5-tbip.



Fig. S5 CIE diagram of  $Eu^{3+}$  and  $Tb^{3+}$  co-actived MOF with a series of ratio: (a)  $Eu^{3+}:Tb^{3+}=1:9$ ; (b)  $Eu^{3+}:Tb^{3+}=2:8$ ; (c)  $Eu^{3+}:Tb^{3+}=3:7$ ; (d)  $Eu^{3+}:Tb^{3+}=4:5$ ; (e)  $Eu^{3+}:Tb^{3+}=5:5$ .



Fig. S6 PXRD patterns of  $Ln^{3+}$ @Cd-MOF-2 (Ln = Eu, Tb, Sm, Dy).





Fig. S7 Emission spectra of  $Ln^{3+}$ @Cd-MOF-2when excited at 293 nm. (a) Ln = Eu; (b) Ln = Tb; (c) Ln = Sm; (d) Ln=Dy.





**Fig. S8** Emission spectra of  $Ln^{3+}$ @Cd-MOF-2 polymer film when excited at 293 nm. (a) Ln = Eu; (b) Ln = Tb; (c) Ln = Sm; (d) Ln=Dy.

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Samples	Cd <sup>2+</sup> (mg / L)	Ln <sup>3+</sup> (mg / L)	atomic ratio of Cd <sup>2+</sup> :Ln <sup>3+</sup>
Eu³+@Cd-MOF	253.652	7.458	1:0.029
Tb³⁺@Cd-MOF	214.375	7.197	1:0.034
Sm <sup>3+</sup> @Cd-MOF	220.854	6.257	1:0.028
Dy³⁺@Cd-MOF	287.472	7.240	1:0.025
Eu³⁺@Cd-MOF-2	276.430	5.878	1:0.021
Tb <sup>3+</sup> @Cd-MOF-2	201.641	4.920	1:0.024
Sm <sup>3+</sup> @Cd-MOF-2	187.785	3.674	1:0.020
Dy³⁺@Cd-MOF-2	260.542	4.751	1:0.018

Table S1 The detailed ICP-OES studies of  $Ln^{3+}$ @Cd-MOF and  $Ln^{3+}$ @Cd-MOF-2