Electronic Supplementary Information (ESI)

Luminescent monometallic Cu(I) triphenylphosphine complexes based on the methylated 5-trifluoromethyl-3-(2'-pyridyl)-1,2,4-triazole ligands

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Figure S1 ¹H NMR spectra of 1 in CD_2Cl_2 .



Figure S2 ¹H NMR spectra of 2 in CD_2Cl_2 .



Figure S3 ¹H NMR spectra of 3 in CD_2Cl_2 .















Figure S9 31 P NMR spectrum of 4 in CD₂Cl₂.



Figure S10 ³¹P NMR spectrum of 5 in CD₂Cl₂.

orbital		nergy (eV)	MO contribution (%)			
			Cu (s/p/d)	PPh ₃	<i>p</i> -mpftz	
LUMO+8		-0.28	8.56 (2/97/1) 89.85		1.59	
LUN	1O+5	-0.53	9.69 (6/91/3)	41.71	48.59	
LUN	1O+3	-0.72	4.86 (15/74/11)	91.07	4.08	
LUMO+2		-0.76	6.81 (44/53/4)	91.74	1.46	
LU	MO	-1.16	6.29 (11/76/13)	11.13	82.58	
НО	MO	-5.85	29.44 (0/22/78)	59.05	11.51	
HOM	AO-1	-6.34	38.92 (7/16/77)	35.12	25.95	
HOM	МО-2	-6.53	33.18 (7/10/83)	14.95	51.87	
HON	мо-з	-6.59	21.69 (3/19/78)	13.04	65.26	
state	E/nm (eV)	O.S.	transition		assignment	
S_1	333 (3.72)	0.1159	HOMO→LUMO	(90%)	¹ LLCT/ ¹ MLCT/ ¹ IL	
S_2	309 (4.01)	0.0277	HOMO-1→LUM	O (53%)	¹ IL/ ¹ MLCT/ ¹ LLCT	
			HOMO-2→LUM	O (26%)		
S_6	294 (4.22)	0.0444	HOMO-2→LUM	O (37%)	¹ IL/ ¹ MLCT/ ¹ LLCT	
			HOMO-1→LUMO (31%)			
			HOMO-3→LUMO (10%)			
S ₁₀ 278 (4.47) 0.1184		0.1184	HOMO-3→LUMO (47%)		¹ IL/ ¹ MLCT/ ¹ LLCT	
			HOMO→LUMO	+5 (24%)		
			HOMO-2→LUMO (15%)			
S_{15}	264 (4.70)	0.1119	HOMO→LUMO	+8 (64%)	¹ IL/ ¹ MLCT/ ¹ LLCT	
		HOMO-1→LUM	O+2 (21%)			

Table S1 Partial molecular orbital compositions (%) by SCPA approach (C-squared population analysis proposed by Ros and Schuit) and the absorption transitions for complex 1 in CH_2Cl_2 media calculated by TDDFT method at the M06 level.





Figure S11 Plots of the frontier molecular orbitals involved in the absorption transitions for complex 1 in CH_2Cl_2 media by TDDFT method at the M06 level (isovalue = 0.025). The red and green parts represent different phases, respectively. For clarity, the hydrogen atoms are omitted.

ort	oital e	nergy (eV)		MO contri	bution (%)
			Cu (s/p/d)	PPh ₃	<i>m</i> -mpftz
LUN	AO+5	-0.55	15.85 (4/95/1)	51.95	32.19
LUN	AO+4	-0.66	3.93 (31/68/1)	73.95	22.11
LUN	AO+2	-0.78	5.74 (35/58/7)	81.52	12.73
LU	MO	-1.20	6.43 (3/85/12)	8.66	84.91
НО	OMO	-6.03	26.18 (0/19/81)	60.88	12.95
HON	мо-2	-6.48	20.08 (6/15/79)	12.42	67.50
НОМО-3		-6.52	35.63 (3/10/87) 9.71		54.66
state	E/nm (eV)	O.S.	transition		assignment
S ₁	320 (3.87)	0.0853	HOMO→LUMO (869	%)	¹ LLCT/ ¹ MLCT/ ¹ IL
S_6	286 (4.34)	0.2226	HOMO-2→LUMO (4	2%)	¹ IL/ ¹ MLCT
			HOMO-3→LUMO (2	20%)	
			HOMO→LUMO+2 (20%)		
S ₂₂	254 (4.88)	0.1349	HOMO-3→LUMO+2	2 (23%)	¹ LLCT/ ¹ MLCT/ ¹ IL
			HOMO-2→LUMO+4	(19%)	
			HOMO-2→LUMO+2	(10%)	

Table S2 Partial molecular orbital compositions (%) by SCPA approach (C-squared population analysis proposed by Ros and Schuit) and the absorption transitions for complex **2** in CH_2Cl_2 media calculated by TDDFT method at the M06 level.





Figure S12 Plots of the frontier molecular orbitals involved in the absorption transitions for complex 2 in CH_2Cl_2 media by TDDFT method at the M06 level (isovalue = 0.025). The red and green parts represent different phases, respectively. For clarity, the hydrogen atoms are omitted.

orbital energy (eV) MO contribution (%) Cu(s/p/d)PPh₃ *o*-mpftz LUMO+3 -0.59 4.63 (5/91/4) 46.04 49.33 LUMO -1.18 6.00 (2/83/15) 14.58 79.42 HOMO -5.79 27.37 (0/23/77) 64.96 7.67 HOMO-1 -6.28 42.73 (5/16/79) 27.22 30.04 HOMO-2 -6.43 10.96 (11/15/74) 10.00 79.04 HOMO-3 -6.61 39.14 (5/14/81) 25.34 35.52 state E/nm (eV) O.S. transition assignment S_1 ¹LLCT/¹MLCT/¹IL 339 (3.65) 0.1249 HOMO→LUMO (96%) ¹IL/¹MLCT/¹LLCT 296 (4.19) 0.0737 HOMO-2→LUMO (62%) S_5 HOMO-1 \rightarrow LUMO (10%) ¹IL/¹MLCT/¹LLCT S_8 286 (4.34) 0.1118 HOMO-3→LUMO (47%) HOMO-2→LUMO (22%) HOMO→LUMO+3 (13%)

Table S3 Partial molecular orbital compositions (%) by SCPA approach (C-squared population analysis proposed by Ros and Schuit) and the absorption transitions for complex **3** in CH_2Cl_2 media calculated by TDDFT method at the M06 level.



Figure S13 Plots of the frontier molecular orbitals involved in the absorption transitions for complex 3 in CH_2Cl_2 media by TDDFT method at the M06 level (isovalue = 0.025). The red and green parts represent different phases, respectively. For clarity, the hydrogen atoms are omitted.

orbital	ene	rgy (eV)	MO contribution (%)		
			Cu (s/p/d)	PPh ₃	<i>p</i> -mpftzH
LUN	1O+6	-0.86	18.58 (57/42/1)	78.21	3.21
LUN	4O+5	-0.91	7.65 (7/91/2)	87.94	4.42
LUN	4O+3	-1.05	5.04 (3/84/12)	91.36	3.60
LUN	4O+2	-1.13	9.73 (46/50/4)	83.17	7.10
LUN	1O+1	-1.45	7.31 (8/88/4)	15.90	76.79
LU	MO	-2.37	1.25 (22/38/40)	4.59	94.17
НО	MO	-6.52	25.35 (0/46/54)	71.97	2.68
HON	MO-1	-6.80	34.87 (30/12/58)	55.97	9.16
state	E/nm (eV)	O.S.	transition	assign	nment
S ₁	379 (3.27)	0.0320	HOMO→LUMO ((98%) ¹ LLC	T/ ¹ MLCT
S_2	357 (3.47)	0.0139	HOMO-1→LUM0	D(94%) ¹ LLC	T/ ¹ MLCT
S_4	294 (4.22)	0.0220	HOMO→LUMO+	-1 (94%) ¹ LLC	T/ ¹ MLCT/ ¹ IL
S ₉	276 (4.50)	0.1124	HOMO→LUMO+	-3 (60%) ¹ IL/ ¹ M	ALCT
S_{15}	266 (4.65)	0.1342	HOMO→LUMO+	-5 (65%) ¹ IL/ ¹ M	ALCT
			HOMO-1→LUM0	D+3 (19%)	
\mathbf{S}_{18}	264 (4.69)	0.1346	HOMO→LUMO+	-6 (59%) ¹ IL/ ¹ M	ИLСТ
			HOMO-1→LUM0	D+2 (19%)	

Table S4 Partial molecular orbital compositions (%) by SCPA approach (C-squared population analysis proposed by Ros and Schuit) and the absorption transitions for complex 4 in CH_2Cl_2 media calculated by TDDFT method at the M06 level.





Figure S14 Plots of the frontier molecular orbitals involved in the absorption transitions for complex 4 in CH_2Cl_2 media by TDDFT method at the M06 level (isovalue = 0.025). The red and green parts represent different phases, respectively. For clarity, the hydrogen atoms except for the ones bonded to N atoms are omitted.

orbital	ener	rgy (eV)	MO contribution (%	o)	
			Cu (s/p/d)	PPh ₃	<i>m</i> -mpftzH
LUMC	0+5 -0.9	0	1.73 (2/78/19)	93.52	4.75
LUMC)+3 -1.1	0	9.16 (4/93/3)	87.67	3.17
LUMC)+2 -1.1	4	5.66 (21/70/9)	92.67	1.67
LUMC)+1 -1.4	4	7.14 (17/79/4)	9.38	83.48
LUMC	-2.2	6	1.37 (3/57/39)	4.00	94.62
HOMO	-6.5	3	23.79 (1/38/61)	73.76	2.45
HOMO	D-1 -6.8	4	37.00 (27/15/58)	52.85	10.16
state	E/nm (eV)	O.S.	transition		assignment
S ₁	366 (3.39)	0.0424	HOMO→LUMO (9	98%)	¹ LLCT/ ¹ MLCT
S_2	348 (3.57)	0.0064	HOMO-1→LUMO	0 (90%)	¹ LLCT/ ¹ MLCT/ ¹ IL
S_5	281 (4.41)	0.0878	HOMO→LUMO+2	2 (61%)	¹ IL/ ¹ MLCT/ ¹ LLCT
			HOMO-1→LUMO	+1 (24%)	
S_6	279 (4.44)	0.0656	HOMO-1→LUMO	0+1 (54%)	¹ LLCT/ ¹ MLCT/ ¹ IL
			HOMO→LUMO+2	2 (16%)	
			HOMO→LUMO+2	3 (14%)	
S_{13}	266 (4.66)	0.2406	HOMO→LUMO+:	5 (63%)	¹ IL/ ¹ MLCT
			HOMO-1→LUMO	0+2 (12%)	

Table S5 Partial molecular orbital compositions (%) by SCPA approach (C-squared population analysis proposed by Ros and Schuit) and the absorption transitions for complex **5** in CH₂Cl₂ media calculated by TDDFT method at the M06 level.





Figure S15 Plots of the frontier molecular orbitals involved in the absorption transitions for complex 5 in CH_2Cl_2 media by TDDFT method at the M06 level (isovalue = 0.025). The red and green parts represent different phases, respectively. For clarity, the hydrogen atoms except for the ones bonded to N atoms are omitted.

orbital	energy (eV)		MO contribu	tion (%)
	1	Cu (s/p/d)	PPh ₃	<i>p</i> -mpftz
LUMO	-1.55	4.59 (5/72/23)	7.27	88.14
НОМО	-5.83	26.47 (0/21/79)	54.62	18.90
HOMO-1	-6.15	7.69 (4/35/61)	10.71	81.59
	2	Cu (s/p/d)	PPh ₃	<i>m</i> -mpftz
LUMO	-1.55	4.96 (9/71/20)	7.09	87.96
НОМО	-5.92	13.99 (1/11/88)	30.39	55.63
HOMO-1	-6.10	17.44 (1/33/66)	42.34	40.22
НОМО-3	-6.47	47.68 (3/10/88)	10.68	41.65
	3	Cu (s/p/d)	PPh ₃	o-mpftz
LUMO	-1.56	4.87 (8/69/23)	8.68	86.46
НОМО	-5.86	24.73 (0/22/77)	58.59	16.68
	4	Cu (s/p/d)	PPh ₃	<i>p</i> -mpftzH
LUMO	-2.31	2.25 (5/33/62)	4.85	92.89
НОМО	-5.78	22.21 (0/26/74)	70.39	7.40
HOMO-1	-6.76	28.56 (29/22/50)	62.73	8.71
НОМО-2	-7.20	57.35 (1/7/93)	16.10	26.56
HOMO-14	-7.97	34.65 (1/0/99)	39.20	26.14
	5	Cu (s/p/d)	PPh ₃	<i>m</i> -mpftzH
LUMO	-2.44	3.06 (3/45/52)	4.37	92.56
НОМО	-5.89	27.90 (0/16/84)	59.53	12.57

Table S6 Partial molecular orbital compositions (%) by SCPA approach (C-squared population analysis proposed by Ros and Schuit) in the lowest triplet state for complexes 1-5, respectively, in CH₂Cl₂ media calculated by TDDFT method at the M06 level.

complex	state	E/nm (eV)	O.S.	transition	assignment
1	T ₁	515 (2.41)	0.0000	HOMO→LUMO (74%)	³ LLCT/ ³ MLCT/ ³ IL
				HOMO-1→LUMO (13%)	
2	T ₁	535 (2.32)	0.0000	HOMO-1→LUMO (39%)	³ IL/ ³ LLCT/ ³ MLCT
				HOMO→LUMO (36%)	
				HOMO-3→LUMO (14%)	
3	T ₁	518 (2.40)	0.0000	HOMO→LUMO (81%)	³ LLCT/ ³ MLCT/ ³ IL
1	T ₁	567 (2.19)	0.0000	HOMO-1→LUMO (52%)	³ LLCT/ ³ MLCT/ ³ IL
				HOMO-2→LUMO (18%)	
				HOMO-14→LUMO (14%)	
5	T_1	576 (2.15)	0.0000	HOMO→LUMO (94%)	³ LLCT/ ³ MLCT/ ³ IL

Table S7 The emission transitions for complexes 1-5 in CH₂Cl₂ media calculated by TDDFT method at the M06 level.



Figure S16 The optimized structure, plots of frontier molecular orbitals (isovalue = 0.025) involved in the emission transitions, and the charge density difference (CDD) map (isovalue = 0.0006) between lowest-energy triplet state T₁ and the ground state S₀ for complex **1** in CH₂Cl₂ media calculated by TDDFT method at the M06 level. The red and green parts represent different phases. And the purple and blue parts display the electron accumulation and depletion region, respectively. For clarity, the hydrogen atoms are omitted.



Figure S17 The optimized structure, plots of frontier molecular orbitals (isovalue = 0.025) involved in the emission transitions, and the charge density difference (CDD) map (isovalue = 0.0006) between lowest-energy triplet state T₁ and the ground state S₀ for complex **2** in CH₂Cl₂ media calculated by TDDFT method at the M06 level. The red and green parts represent different phases. And the purple and blue parts display the electron accumulation and depletion region, respectively. For clarity, the hydrogen atoms are omitted.



Figure S18 The optimized structure, plots of frontier molecular orbitals (isovalue = 0.025) involved in the emission transitions, and the charge density difference (CDD) map (isovalue = 0.0006) between lowest-energy triplet state T₁ and the ground state S₀ for complex **3** in CH₂Cl₂ media calculated by TDDFT method at the M06 level. The red and green parts represent different phases. And the purple and blue parts display the electron accumulation and depletion region, respectively. For clarity, the hydrogen atoms are omitted.



Figure S19 The optimized structure, plots of frontier molecular orbitals (isovalue = 0.025) involved in the emission transitions, and the charge density difference (CDD) map (isovalue = 0.0006) between lowest-energy triplet state T₁ and the ground state S₀ for complex **4** in CH₂Cl₂ media calculated by TDDFT method at the M06 level. The red and green parts represent different phases. And the purple and blue parts display the electron accumulation and depletion region, respectively. For clarity, the hydrogen atoms except for the ones bonded to N atoms are omitted.



Figure S20 The optimized structure, plots of frontier molecular orbitals (isovalue = 0.025) involved in the emission transitions, and the charge density difference (CDD) map (isovalue = 0.0006) between lowest-energy triplet state T_1 and the ground state S_0 for complex **5** in CH₂Cl₂ media calculated by TDDFT method at the M06 level. The red and green parts represent different phases. And the purple and blue parts display the electron accumulation and depletion region, respectively. For clarity, the hydrogen atoms except for the ones bonded to N atoms are omitted.







Figure S22 The ESI mass spectra of 4–5.



Figure S23 The PXRD patterns of 1–5.