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Supporting Information for **Oxidative Deamination of Azafulleroids into** C_{60} **by Peracids**

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Fig. S1. ¹H NMR (in DMSO-d6) of oxidized products of azafulleroid 1g. (too low solubility of the oxidized products inhibited ¹³C NMR measurement)



Wavenumber (cm^{-1})

Fig. S2. IR of the insoluble product of **1g** with *m*CPBA.



Scheme S1. Endo/exo definition for substituted azafulleroid.



Fig. S3. HOMO of metastable methyl azafulleroid 1a/5.



Fig. S4. HOMO of metastable TMS-methyl azafulleroid **1b** conformers. The values in parenthesis are relative energy (kJ/mol) to the most stable 6-exo isomer (Fig. 1b).



Fig. S5. HOMO of metastable conformers of benzylic azafulleroid **1e**. The values in parenthesis are relative energy (kJ/mol) to the most stable 6-exo isomer (Fig. 1c).



Fig. S6. HOMO of metastable phenyl azafulleroid 1f/6.



Fig. S7. HOMO and relative energies (kJ/mol) of metastable conformers of tosyl azafulleroid **1g.** The values in parenthesis are relative energy (kJ/mol) to the most stable 5-exo isomer (Fig. 1e)



Fig. S8. (a) Three plausible conformer of *m*CPBA with B3LYP/6-31G(d) energy. (b) LUMO+2 of *m*CPBA (geom-1). (c) LUMO+1 of PAA.



Fig. S9. N-flipping transition states of **1a** (top) and **1f** (bottom) (B3LYP/6-31G(d) with IEFPCM (*o*-DCB)). The values in parentheses are relative energies (kJ/mol).





1a/6: N-attack of *m*CPBA (geom-2) $\Delta E^{\ddagger} = +71.7 \text{ kJ/mol}$

1a/6: C α -attack of *m*CPBA (geom-2) : ΔE^{\ddagger} = +71.4 kJ/mol





1a/6: N-attack of *m*CPBA (geom-3) $\Delta E^{\ddagger} = +44.6 \text{ kJ/mol}$

1a/6: C α -attack of *m*CPBA (geom-2) : ΔE^{\ddagger} = +51.7 kJ/mol

Fig. S10. Representative results of transition states of 1a with geom-2 and geom-3 *m*CPBA.



Fig. S11. Transition state geometry of **1a** with peracetic acid.



Fig. S12. Transition state geometry of **1f** with *m*CPBA.



Scheme S2. Expected intermediates from $C\alpha/C\beta$ attack to azafulleroid, in comparison with those of fulleroid.

NMR Chart

1H and 13C NMR spectra of 1a (N-Methyl azafulleroid)



Calculations: Energy summary

Calculated energy table of transition states for 1a with *m*CPBA (Fig. 2, B3LYP/6-31G(d), with IEFPCM (*o*-DCB)

Initial State /au		Transition State (with vi) /au		$\Delta E^0, \Delta E^{\ddagger}, /kJ/mol$
1a /6	-2380.803978			0
1a /5	-2380.802525			3.81
		5/6-fliping (154.1i)	-2380.791043	34.0
mCPBA (geom-1)	-955.5466729			
		N-attack/6 (374.0i)	-3336.333707	44.5 ^a
		Cα-attack/6 (554.5i)	-3336.331084	51.4 ^a
		Cβ-attack/6 (530.7i)	-3336.328487	58.2 ^a
		Cγ-attack/6 (525.5i)	-3336.325792	65.3 ^a
		N-attack/5 (366.9i)	-3336.331931	49.1 ^a
		Cα-attack/5 (531.5i)	-3336.329781	54.8 ^a
		Cβ-attack/5 (515.6i)	-3336.327064	61.9 ^a
		Cγ-attack/5 (508.2i)	-3336.325568	65.8 ^a
mCPBA (geom-2)				
		N-attack/6 (333.6i)	-3336.323323	71.7 ^ª
		Cα-attack/6 (565.2i)	-3336.323450	71.4 ^ª
mCPBA (geom-3)				
		N-attack/6 (368.3i)	-3336.333644	44.6 ^a
		Cα-attack/6 (555.7i)	-3336.330954	51.7 ^a
<i>m</i> -CBA ^b	-880.4230826			
2	-2455.933987			-16.9 ^c
4	-2455.944849			-45.4 ^c
Nitroso	-169.7979863			
methane				
C60	-2286.174951			-119.1 ^d

^a $\Delta E^{\ddagger} = E(TS) - E(\mathbf{1a}/6) - E(mCPBA, geom-1)$

^b *m*-chlorobenzoic acid

^c $\Delta E^{\circ} = E(\mathbf{2} \text{ or } \mathbf{4}) + E(m\text{-CBA}) - E(\mathbf{1a}/6) - E(m\text{CPBA}, \text{ geom-1})$

^d $\Delta E^{\circ} = E(C_{60}) + E(m\text{-CBA}) + E(\text{nitrosomethane}) - E(\mathbf{1a}/6) - E(m\text{CPBA}, \text{geom-1})$

Initial State /au		Transition State (with vi) /au		$\Delta E^{\ddagger}/\mathrm{kJ/mol}$
PAA	-304.2136119			
		N-attack/6 (368.4i)	-2684.996978	54.1 ^a
		Cα-attack/6 (550.6i)	-2684.996111	56.4ª
		Cβ-attack/6 (526.8i)	-2684.992887	64.8 ^ª
		Cγ-attack/6 (523.6i)	-2684.989543	73.6ª
		N-attack/5 (366.3i)	-2684.994791	59.8 ª
		Cα-attack/5 (532.1i)	-2684.994757	59.9 ^a
		Cβ-attack/5 (510.6i)	-2684.990592	70.9 ^ª
		Cγ-attack/5 (510.4i)	-2684.989855	72.8 ^ª

Calculated energy table of transition states for **1a** with PAA (Fig. 3, B3LYP/6-31G(d), with IEFPCM (o-DCB)

^a $\Delta E^{\ddagger} = E(TS) - E(\mathbf{1a}/6) - E(PAA)$

Calculated energy table of transition states for 1f with *m*CPBA (Fig. 3, B3LYP/6-31G(d), with IEFPCM (o-DCB)

Initial State /au		Transition State (with vi) /au		$\Delta E^{0}, \Delta E^{\ddagger}, /kJ/mol$
1f /6	-2572.545125			0.61
1 f /5	-2572.545358			0
		5/6-fliping (32.8i)	-2572.541745	8.87
mCPBA (geom-1)	-955.5466729			
		N-attack/6 (359.7i)	-3528.068000	63.1 ^a
		Cα-attack/6 (549.2i)	-3528.062003	78.8 ª
		Cβ-attack/6 (552.3i)	-3528.064189	73.1 ª
		Cγ-attack/6 (541.5i)	-3528.064241	72.9 ª
		N-attack/5 (371.7i)	-3528.067831	63.5 ^a
		Cα-attack/5 (509.7i)	-3528.067755	63.7 ^ª
		Cβ-attack/5 (519.7i)	-3528.068467	61.9 ^ª
		Cγ-attack/5 (516.7i)	-3528.067347	64.8 ^a

^a $\Delta E^{\ddagger} = E(TS) - E(\mathbf{1f}/5) - E(mCPBA, \text{geom-1})$

Reference in ESI and full citation of software.

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