

## Electronic Supplementary Information

### Design, synthesis and anticancer activities of novel otobain derivatives

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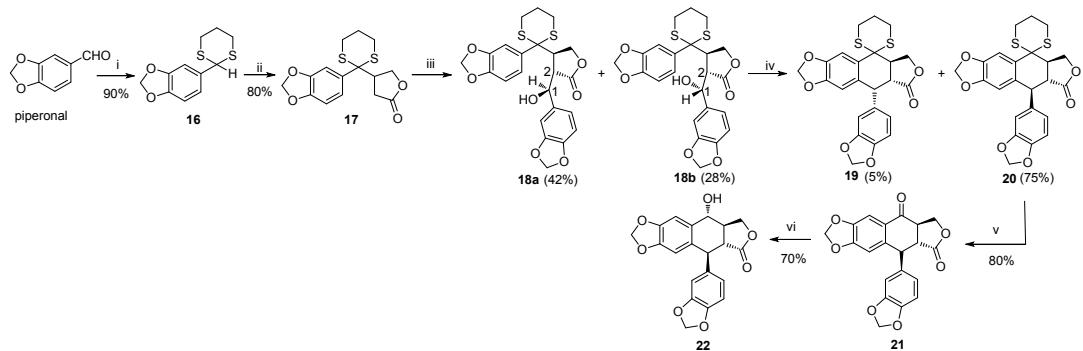
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## 1. General information.

All reagents and chemicals were purchased from commercial suppliers and used without further purification unless otherwise stated.  $^1\text{H}$  (500 MHz) and  $^{13}\text{C}$  (125 MHz) NMR spectra were recorded on a Varian INOVA spectrometer with  $\text{CDCl}_3$  or  $\text{DMSO-d}_6$  and tetramethylsilane (TMS) as the internal standard. All chemical shift values were reported in units of  $\delta$  (ppm). The following abbreviations were used to indicate the peak multiplicity: s = singlet; d = doublet; t = triplet; m = multiplet; br = broad. Analytical thin-layer chromatography (TLC) was carried out on precoated plates (silica gel 60 F254), and spots were visualized with ultraviolet (UV) light. Flash column chromatography was carried out with silica gel (300–350 mesh). Melting points were determined on Yanano MP 500. High-resolution mass data were obtained on a Bruker microOTOFQ II spectrometer.

## 2. General procedures and characterizations of intermediate compounds 16–22, 24 and 25.



**Scheme 1** Synthesis of 22. Reagents and conditions: (i)  $\text{SH}(\text{CH}_2)_3\text{SH}$ ,  $p\text{-TsOH}$ ,  $\text{DCM}$ , rt; (ii)  $n\text{-BuLi}$ , furan-2(5H)-one,  $\text{THF}$ ,  $-78^\circ\text{C}$ ; (iii)  $\text{LDA}$ , piperonal,  $\text{THF}$ ,  $-78^\circ\text{C}$ ; (iv)  $\text{TFA}$ ,  $\text{DCM}$ , rt; (v)  $\text{HgO}$ ,  $\text{BF}_3\cdot\text{Et}_2\text{O}$ ,  $\text{THF}/\text{H}_2\text{O}$  (85/15),  $0^\circ\text{C}$ -rt; (vi)  $\text{NaBH}_4$ ,  $\text{THF}$ ,  $0^\circ\text{C}$ .

### Synthesis of 5-(1,3-dithian-2-yl)benzo[d][1,3]dioxole (16)

To a solution of piperonal (15.00 g, 100 mmol) in dichloromethane (150 mL), propanedithiol (11.02 mL, 110 mmol) and monohydrated p-toluenesulfonic acid (1.90 g, 10 mmol) were added. The mixture was stirred for 24 h at room temperature. Upon completion, saturated sodium carbonate (100 mL) was added and stirred for 2 h. The organic layer was washed with saturated brine ( $3\times 50$  mL), dried over anhydrous  $\text{Na}_2\text{SO}_4$ , filtered, concentrated in vacuo, and recrystallized from ethanol to give white solid of 5-(1,3-dithian-2-yl)benzo[d][1,3]dioxole (**16**) (21.50 g, 90%): m.p.  $97^\circ\text{C}$  (lit.<sup>1</sup> 93–94 °C);  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ):  $\delta$  1.87–1.94(m, 1H), 2.12–2.17(m, 1H), 2.89(d,  $J$  = 14.2 Hz, 2H), 3.01–3.07(m, 2H), 5.09(s, 1H), 5.95(s, 2H), 6.75(d,  $J$  = 8.0 Hz, 1H), 6.93(d,  $J$  = 8.0 Hz, 1H), 6.98(s, 1H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ ):  $\delta$  25.05, 32.19, 51.19, 101.23, 108.39, 121.31, 132.95, 147.64, 147.76.

### Synthesis of 4-(2-(benzo[d][1,3]dioxol-5-yl)-1,3-dithian-2-yl)dihydrofuran-2(3H)-one (17)

To a solution of **16** (4.00 g, 16.64 mmol) in dry  $\text{THF}$  (60 mL) at  $-78^\circ\text{C}$  was added  $n\text{-BuLi}$  (6.96 mL, 2.4 M, in hexane, 16.64 mmol) dropwise within a period of 30 min. Stirring was continued for 30 min, and then a solution of furan-2(5H)-one (1.44 g, 16.64 mmol) in dry  $\text{THF}$  (10 mL) was added dropwise over a period of 20 min at  $-78^\circ\text{C}$ . The reaction was kept in the same conditions for two more hours. After the addition of concentrated acetic acid (3 mL), the mixture was allowed to reach the room temperature. The crude of the reaction was extracted with ethyl acetate and the organic phase was dried over anhydrous  $\text{Na}_2\text{SO}_4$ . Removal of the solvent under reduced pressure yielded yellowish colored oil, which was recrystallized from ethyl acetate to provide white solid of 4-(2-(benzo[d][1,3]dioxol-5-yl)-1,3-dithian-2-yl)dihydrofuran-2(3H)-one (**17**) (4.36 g, 80%): m.p.  $162\text{--}163^\circ\text{C}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ):  $\delta$  1.85–1.99(m, 2H), 2.40–2.46(m, 1H), 2.66–2.76(m, 4H), 2.82–2.87(m, 1H), 2.98–3.04(m, 1H), 4.18–4.22(m, 1H), 4.39–4.43(m, 1H), 6.01(s, 2H), 6.84(d,  $J$  = 8.5 Hz, 1H), 7.46(d,  $J$  = 5.6 Hz, 2H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ ):  $\delta$  24.75, 27.20, 30.19, 48.20, 60.67, 68.54, 101.57, 108.33, 109.37, 123.16, 132.95, 147.22, 148.67, 175.76; HRMS (ESI): calc. for  $\text{C}_{15}\text{H}_{16}\text{O}_4\text{S}_2\text{Na}$  ( $\text{M}+\text{Na}$ )<sup>+</sup> 347.0382, found: 347.0379.

### Synthesis of 3-(benzo[d][1,3]dioxol-5-yl(hydroxymethyl)-4-(2-(benzo[d][1,3]dioxol-5-yl)-1,3-dithian-2-yl)dihydrofuran-2(3H)-one (18)

To a solution of diisopropylamine (1.2 mL, 7.40 mmol) in dry  $\text{THF}$  (5 mL) at  $-78^\circ\text{C}$  was added  $n\text{-BuLi}$  (3.20 mL, 2.4 M, in hexane, 7.40 mmol) dropwise within a period of 30 min. Stirring was continued for 30 min, and then a solution of (**17**) (2.00 g, 6.17 mmol) in dry  $\text{THF}$  (15 mL) was added dropwise over a period of 30 min at  $-78^\circ\text{C}$ . The reaction was stirred at  $-78^\circ\text{C}$  for one more hour. At this point, a solution of piperonal (1.11 g, 7.40 mmol) in dry  $\text{THF}$  (5 mL) was added. After 1 h, acetic acid (1.60 mL) was added and allowed the mixture to rise to the room

temperature. The mixture of the reaction was extracted with ethyl acetate and the product was purified by silica column chromatography (ethyl acetate/petroleum ether, 3:7) to give **18a** (1.23 g, 42%) and **18b** (0.82 g, 28%).

**3-(benzo[d][1,3]dioxol-5-yl(hydroxy)methyl)-4-(2-(benzo[d][1,3]dioxol-5-yl)-1,3-dithian-2-yl)dihydrofuran-2(3H)-one (**18a**):** white solid, m.p. 181-184 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 1.76-1.90(m, 2H), 2.50-2.54(m, 3H), 2.56-2.67(m, 2H), 2.76 (dt, J<sub>1</sub> = 7.9, J<sub>2</sub> = 2.3 Hz, 1H), 2.90(s, 1H), 4.33(dd, J<sub>1</sub> = 9.2, J<sub>2</sub> = 8.2 Hz, 1H), 4.88(dd, J<sub>1</sub> = 9.3, J<sub>2</sub> = 1.9 Hz, 1H), 5.08(t, J = 3.6 Hz, 1H), 5.94(t, J = 1.8 Hz, 2H), 5.99(d, J = 1.6 Hz, 1H), 6.02(d, J = 1.6 Hz, 1H), 6.52(d, J = 8.0 Hz, 2H), 6.60(d, J = 8.2 Hz, 1H), 6.63(d, J = 8.2 Hz, 1H), 7.08(s, 1H), 7.19(dd, J<sub>1</sub> = 8.2, J<sub>2</sub> = 1.9 Hz, 1H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 24.45, 26.70, 27.21, 47.83, 49.49, 62.96, 70.31, 73.62, 101.19, 101.55, 105.89, 107.54, 107.94, 109.27, 118.60, 123.31, 132.55, 134.28, 146.78, 146.97, 147.63, 148.10, 178.31; HRMS (ESI): calc. for C<sub>23</sub>H<sub>22</sub>O<sub>7</sub>S<sub>2</sub>Na (M+Na)<sup>+</sup> 497.0699, found: 497.0702.

**3-(benzo[d][1,3]dioxol-5-yl(hydroxy)methyl)-4-(2-(benzo[d][1,3]dioxol-5-yl)-1,3-dithian-2-yl)dihydrofuran-2(3H)-one (**18b**):** white solid, m.p. 169-170 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 1.81-1.95(m, 2H), 2.59-2.75(m, 6H), 3.15(d, J = 5.9 Hz, 1H), 3.92(t, J = 9.0 Hz, 1H), 4.67(d, J = 9.9 Hz, 1H), 4.72(d, J = 5.9 Hz, 1H), 5.94(s, 1H), 5.97(s, 1H), 5.99(s, 1H), 6.03(s, 1H), 6.65(d, J = 7.9 Hz, 1H), 6.70(d, J = 9.5 Hz, 2H), 6.75(d, J = 8.1 Hz, 1H), 7.32(s, 1H), 7.35(d, J = 8.2 Hz, 1H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 24.43, 26.86, 27.12, 49.48, 50.34, 62.35, 68.35, 73.56, 101.17, 101.60, 106.62, 107.95, 108.05, 109.49, 119.85, 123.40, 132.60, 133.46, 147.15, 147.48, 147.78, 148.50, 176.58; HRMS (ESI): calc. for C<sub>23</sub>H<sub>22</sub>O<sub>7</sub>S<sub>2</sub>Na (M+Na)<sup>+</sup> 497.0708, found: 497.0699.

### Synthesis of lactones (**19**) and (**20**)

TFA (4.27 mL) was added dropwise to a solution of **18** (1.78 g, 3.59 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (25 mL). The reaction mixture was stirred for 24 h at room temperature, and then the reaction was quenched with NaHCO<sub>3</sub> (concentrated solution). The organic layer was washed with saturated brine (3×20 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered, concentrated in vacuo to give the crude product, which was purified by silica column chromatography (ethyl acetate/petroleum ether, 3:7) to give **19** (86 mg, 5%) and **20** (1.29 g, 75%).

**9'-(benzo[d][1,3]dioxol-5-yl)-8a',9'-dihydro-5a'H-spiro[[1,3]dioxol-5-yl]-8'(6'H)-one (**19**):** white solid, m.p. 234-236 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 2.05-2.13(m, 1H), 2.20-2.28(m, 1H), 2.80-2.96(m, 4H), 3.29(dd, J<sub>1</sub> = 13.8 Hz, J<sub>2</sub> = 5.1 Hz, 1H), 3.53-3.59(m, 1H), 4.49 (d, J = 5.1 Hz, 1H), 4.55(dd, J<sub>1</sub> = 10.8 Hz, J<sub>2</sub> = 7.7 Hz, 1H), 4.65-4.68(m, 1H), 5.89-5.95(m, 4H), 6.36(s, 1H), 6.58(dd, J<sub>1</sub> = 8.1 Hz, J<sub>2</sub> = 1.6 Hz, 1H), 6.65(d, J = 1.6 Hz, 1H), 6.69(d, J = 8.1 Hz, 1H), 7.70(s, 1H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 23.68, 29.46, 31.21, 43.52, 43.74, 50.84, 53.49, 70.11, 101.01, 101.63, 107.78, 109.08, 110.79, 111.20, 124.27, 132.25, 133.15, 133.26, 146.71, 147.18, 147.36, 148.46, 173.80; HRMS (ESI): calc. for C<sub>23</sub>H<sub>20</sub>O<sub>6</sub>S<sub>2</sub>Na (M+Na)<sup>+</sup> 479.0594, found: 479.0595.

**9'-(benzo[d][1,3]dioxol-5-yl)-8a',9'-dihydro-5a'H-spiro[[1,3]dioxol-5-yl]-8'(6'H)-one (**20**):** white solid, m.p. 281-284 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 2.05-2.14(m, 1H), 2.23-2.27(m, 1H), 2.82-2.92(m, 2H), 2.96-3.04(m, 2H), 3.21-3.27(m, 1H), 3.33(dd, J<sub>1</sub> = 13.7 Hz, J<sub>2</sub> = 11.1 Hz, 1H), 3.99(d, J = 11.0 Hz, 1H), 4.53(dd, J<sub>1</sub> = 10.6 Hz, J<sub>2</sub> = 8.0 Hz, 1H), 4.71(t, J = 7.2 Hz, 1H), 5.90(d, J = 3.6 Hz, 2H), 5.93(s, 2H), 6.23(s, 1H), 6.60(s, 1H), 6.77(s, 2H), 7.64(s, 1H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 23.61, 29.31, 30.54, 44.27, 46.39, 52.62, 56.53, 68.80, 101.03, 101.52, 108.20, 108.95, 109.14, 109.27, 123.09, 132.81, 134.08, 136.86, 146.60, 146.64, 147.91, 148.01, 174.95; HRMS (ESI): calc. for C<sub>23</sub>H<sub>20</sub>O<sub>6</sub>S<sub>2</sub>Na (M+Na)<sup>+</sup> 479.0595, found: 479.0594.

### Synthesis of 9'-(benzo[d][1,3]dioxol-5-yl)-8a',9'-dihydro-5a'H-spiro[[1,3]dithiane-2,5'-furo[3',4':6,7]naphtho[2,3-d][1,3]dioxol-8'(6'H)-one (**21**)

A solution of **20** (1.50 g, 3.29 mmol) in 60 mL of 85:15 THF/H<sub>2</sub>O was added to a suspension of HgO (1.55 g, 7.51 mmol) in 85:15 THF/H<sub>2</sub>O at 0 °C under argon followed by BF<sub>3</sub>Et<sub>2</sub>O (0.52 mL, 7.51 mmol). The reaction mixture was stirred for 1 h at room temperature, and then CH<sub>2</sub>Cl<sub>2</sub> (30 mL) was added and the precipitate filtered, concentrated in vacuo to give the crude product, which was isolated by silica column chromatography (ethyl acetate/petroleum ether, 3:7) to give **21** (0.96 g, 80%): white solid, m.p. 216-218 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 3.01(dd, J<sub>1</sub> = 15.5 Hz, J<sub>2</sub> = 11.4 Hz, 1H), 3.39(ddd, J<sub>1</sub> = 15.5 Hz, J<sub>2</sub> = 10.6 Hz, J<sub>3</sub> = 7.0 Hz, 1H), 4.23(d, J = 11.4 Hz, 1H), 4.42(dd, J<sub>1</sub> = 10.5 Hz, J<sub>2</sub> = 9.5 Hz, 1H), 4.63(dd, J<sub>1</sub> = 9.3 Hz, J<sub>2</sub> = 7.0 Hz, 1H), 5.96-6.01(m, 4H), 6.39(s, 1H), 6.57(s, 1H), 6.75(d, J = 7.2 Hz, 1H), 6.82(d, J = 7.9 Hz, 1H), 7.44(s, 1H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 47.43, 47.52, 49.67, 66.22, 101.26, 102.25, 105.52, 108.49, 109.63, 127.95, 134.00, 143.72, 147.20, 147.61, 148.18, 152.99, 172.79, 192.15; HRMS (ESI): calc. for C<sub>20</sub>H<sub>14</sub>O<sub>7</sub>Na (M+Na)<sup>+</sup> 389.0632, found: 389.0632.

### Synthesis of 5-(benzo[d][1,3]dioxol-5-yl)-9-hydroxy-5,5a,8a,9-tetrahydrofuro[3',4':6,7]naphtho[2,3-d][1,3]dioxol-6(8H)-one (**22**)

To a mixture of **21** (500 mg, 1.37 mmol) in THF (20 mL) NaBH<sub>4</sub> (150 mg, 4.01 mmol) was added and allowed to react at room temperature, for 1 h. The reaction was quenched with a saturated NH<sub>4</sub>Cl solution. The aqueous phase was extracted with Et<sub>2</sub>O (3×20 mL). The collected organic phases were dried, concentrated in vacuum, and purified by flash chromatography (dichloromethane/acetone, 20:1) to give **22** (353 mg, 70%): white solid, m.p.

261-263°C ;  $^1\text{H}$  NMR (500 MHz, *d*-DMSO):  $\delta$  2.53-2.56(m, 1H), 3.07(dd,  $J_1 = 13.8$  Hz,  $J_2 = 11.5$  Hz, 1H), 4.06(d,  $J = 11.4$  Hz, 1H), 4.13(dd,  $J_1 = 10.6$  Hz,  $J_2 = 8.6$  Hz, 1H), 4.51(dd,  $J_1 = 8.2$  Hz,  $J_2 = 7.1$  Hz, 1H), 4.82(d,  $J = 10.0$  Hz, 1H), 5.91(s, 1H), 5.94(s, 1H), 5.99(d,  $J = 3.6$  Hz, 1H), 6.09(s, 1H), 6.70-6.72(m, 2H), 6.86(d,  $J = 7.8$  Hz, 1H). 7.05(s, 1H);  $^{13}\text{C}$  NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  45.18, 45.81, 46.78, 70.22, 70.67, 101.29, 101.41, 106.23, 108.32, 108.70, 109.57, 123.22, 133.45, 135.37, 137.94, 146.25, 146.39, 146.65, 147.65, 175.85; HRMS (ESI): calc. for C<sub>20</sub>H<sub>16</sub>O<sub>7</sub>S<sub>2</sub>Na (M+Na)<sup>+</sup> 391.0788, found: 391.0783.

**Synthesis of 9-azido-5-(benzo[d][1,3]dioxol-5-yl)-5,5a,8a,9-tetrahydrofuro[3',4':6,7]naphtho[2,3-d][1,3]dioxol-6(8H)-one (24)**

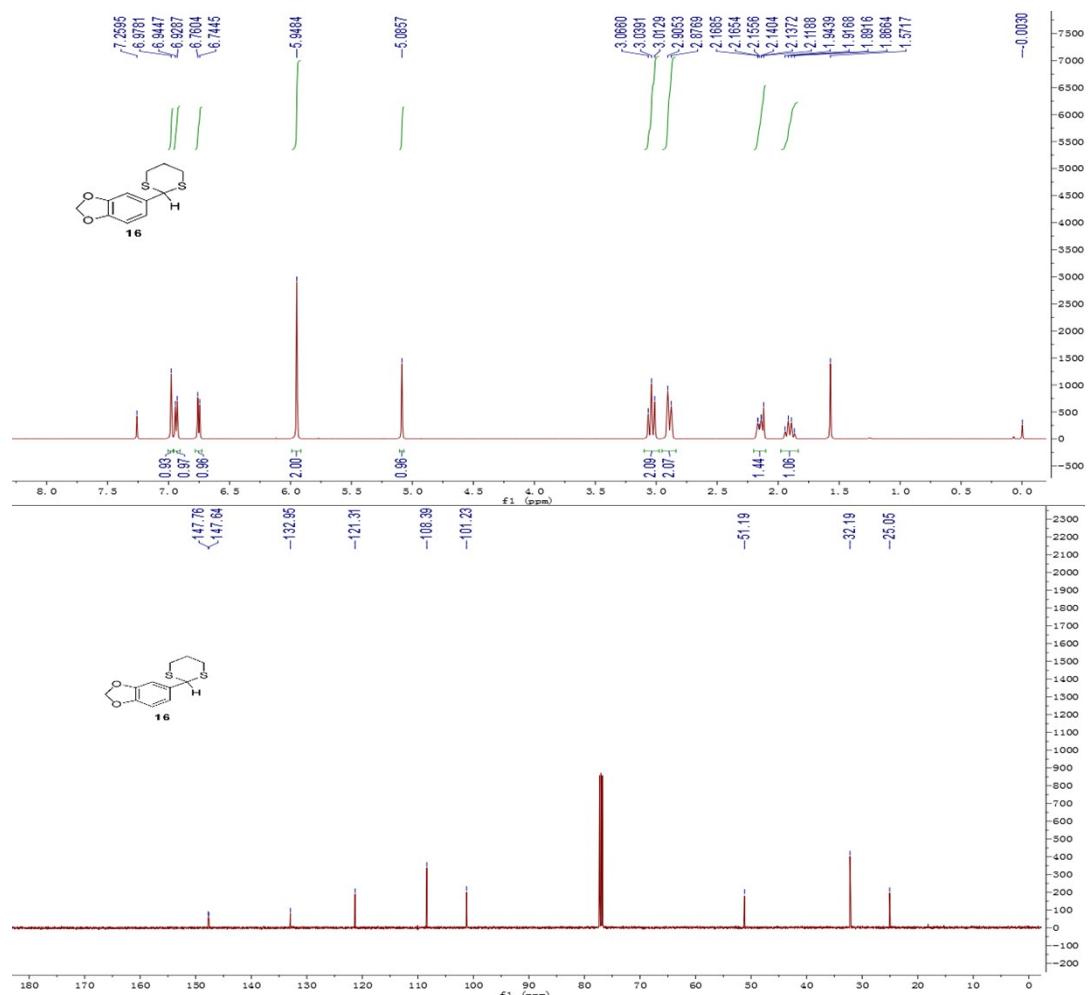
To a solution of **22** (294.4 mg, 0.80 mmol) and sodium azide (264.0 mg, 4.00 mmol) in CHCl<sub>3</sub> (20 mL), was added trifluoroacetic acid (1.6 mL, 2.07 mmol) dropwise in an ice bath. The reaction mixture was brought up to room temperature stirred for 5 h. Saturated aqueous sodium bicarbonate solution was added (2×20 mL). The organic layer was washed with water (1×20 mL), brine (1×20 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. After the solvent was removed, the crude product was purified by column chromatography (ethyl acetate/petroleum ether, 1:3) to afford compound **24** (267 mg, 85%): white solid, m.p. 174-176 °C  $^1\text{H}$  NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  2.71-2.78(m, 1H), 2.99(dd,  $J_1 = 13.9$ ,  $J_2 = 11.4$  Hz, 1H), 3.99(d,  $J = 11.4$  Hz, 1H), 4.25(dd,  $J_1 = 10.6$  Hz,  $J_2 = 8.7$  Hz, 1H), 4.36-4.40(m, 1H), 4.71(d,  $J = 3.1$  Hz, 1H), 5.95(d,  $J = 6.3$  Hz, 4H), 6.42(s, 1H), 6.61(s, 1H), 6.71(s, 1H), 6.76(d,  $J = 8.2$  Hz, 1H), 6.79(d,  $J = 8.2$  Hz, 1H);  $^{13}\text{C}$  NMR(125 MHz, CDCl<sub>3</sub>): 42.20, 43.35, 45.84, 59.37, 67.09, 101.10, 101.70, 108.31, 108.83, 109.08, 110.62, 123.01, 126.00, 134.07, 136.11, 146.76, 147.77, 147.95, 148.96, 174.83; HRMS (ESI): calc. for C<sub>20</sub>H<sub>15</sub>N<sub>3</sub>O<sub>6</sub>Na (M+Na)<sup>+</sup> 416.0853, found: 416.0863.

**Synthesis of 9-amino-5-(benzo[d][1,3]dioxol-5-yl)-5,5a,8a,9-tetrahydrofuro[3',4':6,7]naphtho[2,3-d][1,3]dioxol-6(8H)-one (25)**

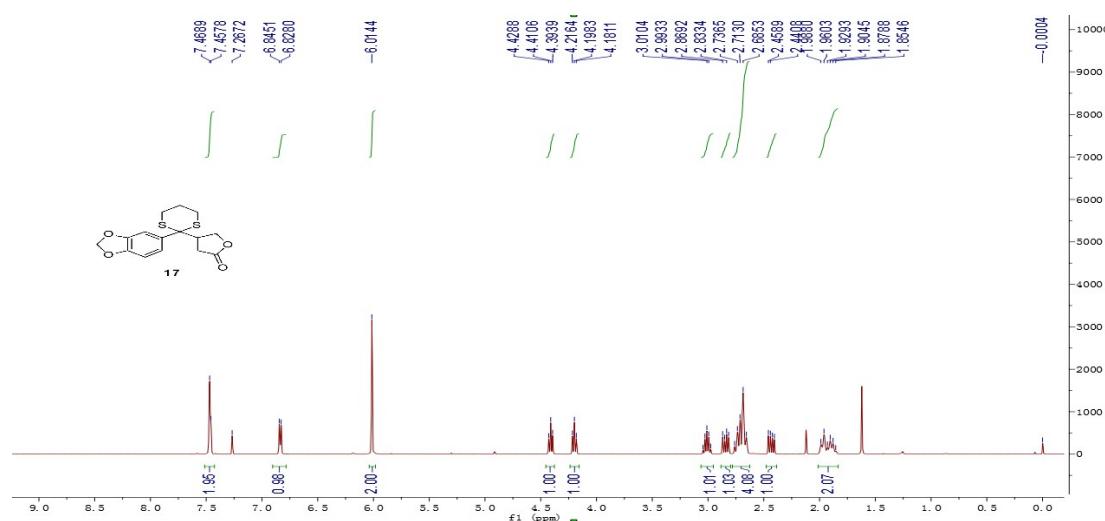
To a solution of **24** (100 mg, 0.25 mmol) in ethyl acetate (10 mL) was added 5% palladium on activated carbon (45 mg). The mixture was stirred overnight under hydrogen. The reaction mixture was filtered, and the filtrate was evaporated. The crude product was purified by column chromatography (dichloromethane/methanol, 120:1) to afford compound **25** (74 mg, 80%): white solid, m.p. 246-248 °C;  $^1\text{H}$  NMR (500 MHz, *d*-DMSO)  $\delta$  2.07(s, 2H), 2.51-2.68(m, 1H), 3.19(dd,  $J_1 = 13.8$  Hz,  $J_2 = 11.6$  Hz, 1H), 3.91(d,  $J = 11.4$  Hz, 1H), 3.99(d,  $J = 3.8$  Hz, 1H), 4.25(dd,  $J_1 = 10.8$  Hz,  $J_2 = 8.5$  Hz, 1H), 4.34(t,  $J = 7.8$  Hz, 1H), 5.90(s, 1H), 5.92(s, 1H), 5.99(d,  $J = 2.6$  Hz, 2H), 6.08(s, 1H), 6.76-6.78(m, 2H), 6.86(d,  $J = 7.8$  Hz, 1H), 6.88(s, 1H);  $^{13}\text{C}$  NMR(125 MHz, *d*-DMSO): 40.39, 44.36, 45.97, 49.20, 68.39, 101.26, 101.38, 108.24, 108.81, 109.66, 109.78, 123.40, 133.09, 135.48, 137.93, 146.15, 146.21, 146.81, 147.67, 176.57; HRMS (ESI): calc. for C<sub>20</sub>H<sub>17</sub>NO<sub>6</sub>Na (M+Na)<sup>+</sup> 390.0948, found: 390.0963.

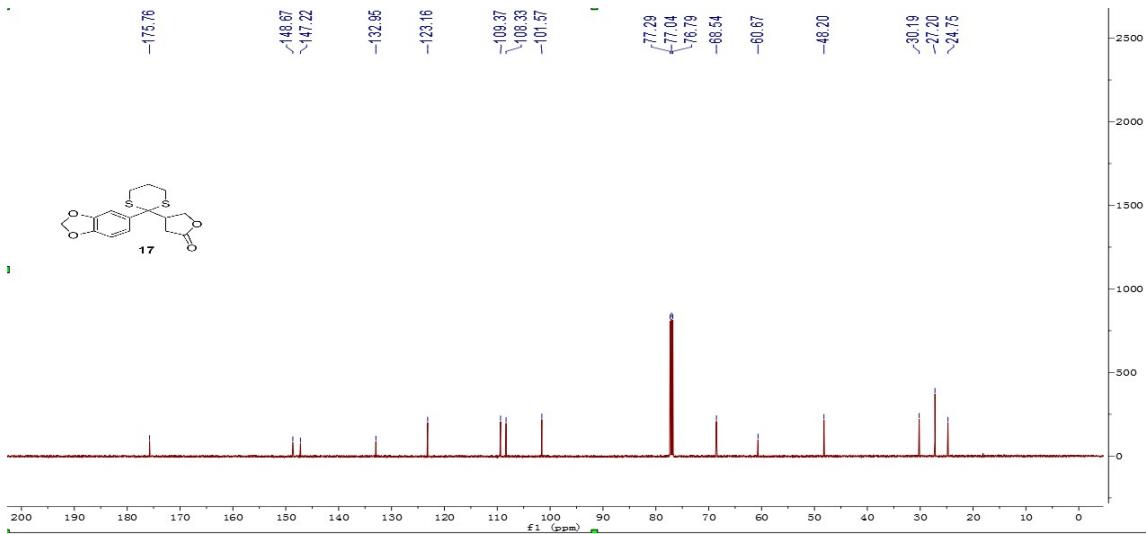
### 3. Copies of NMR Data for All Compounds.

$^1\text{H}$  NMR (500 MHz) and  $^{13}\text{C}$  NMR (125 MHz) spectra of compound **16**.

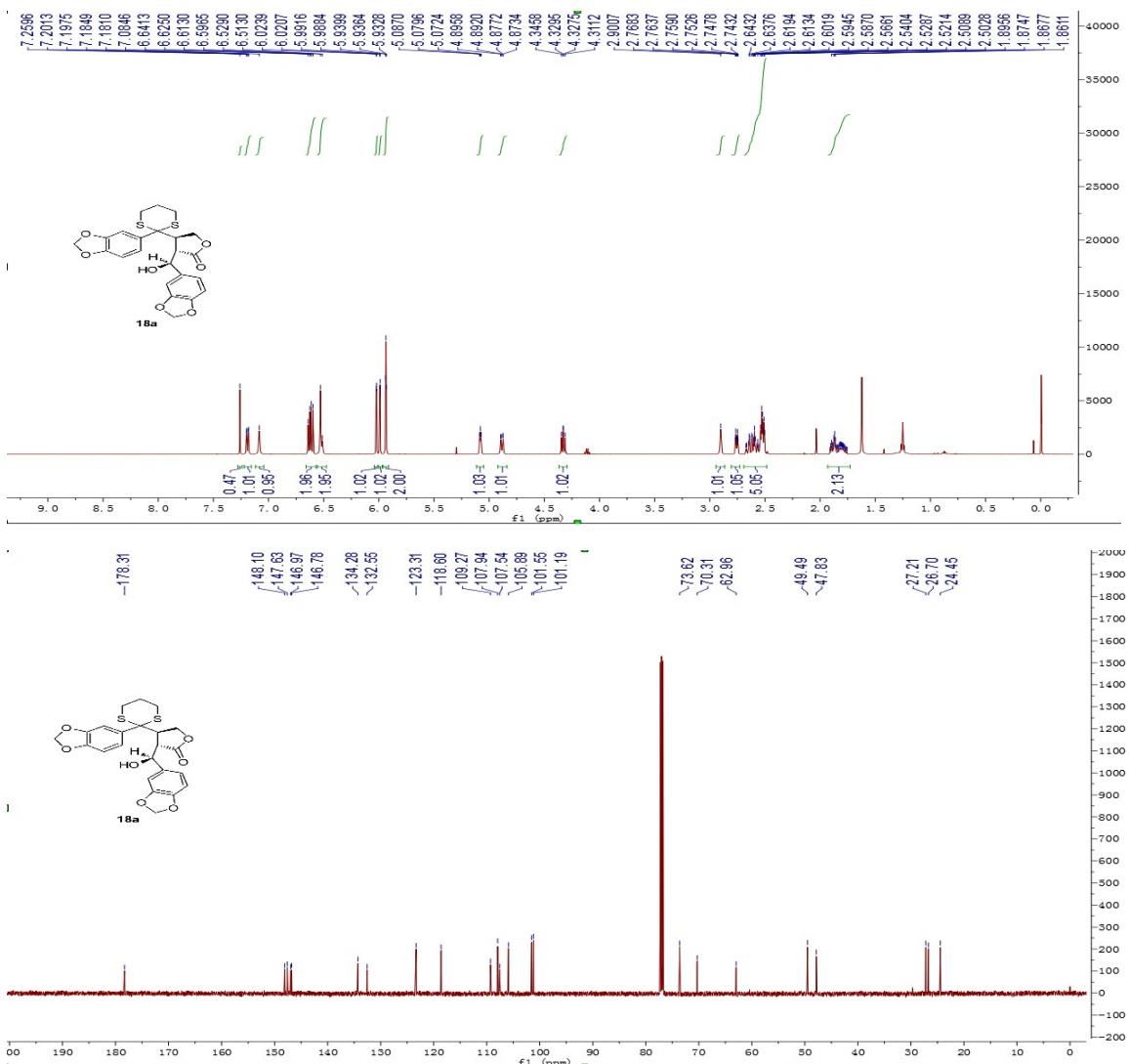


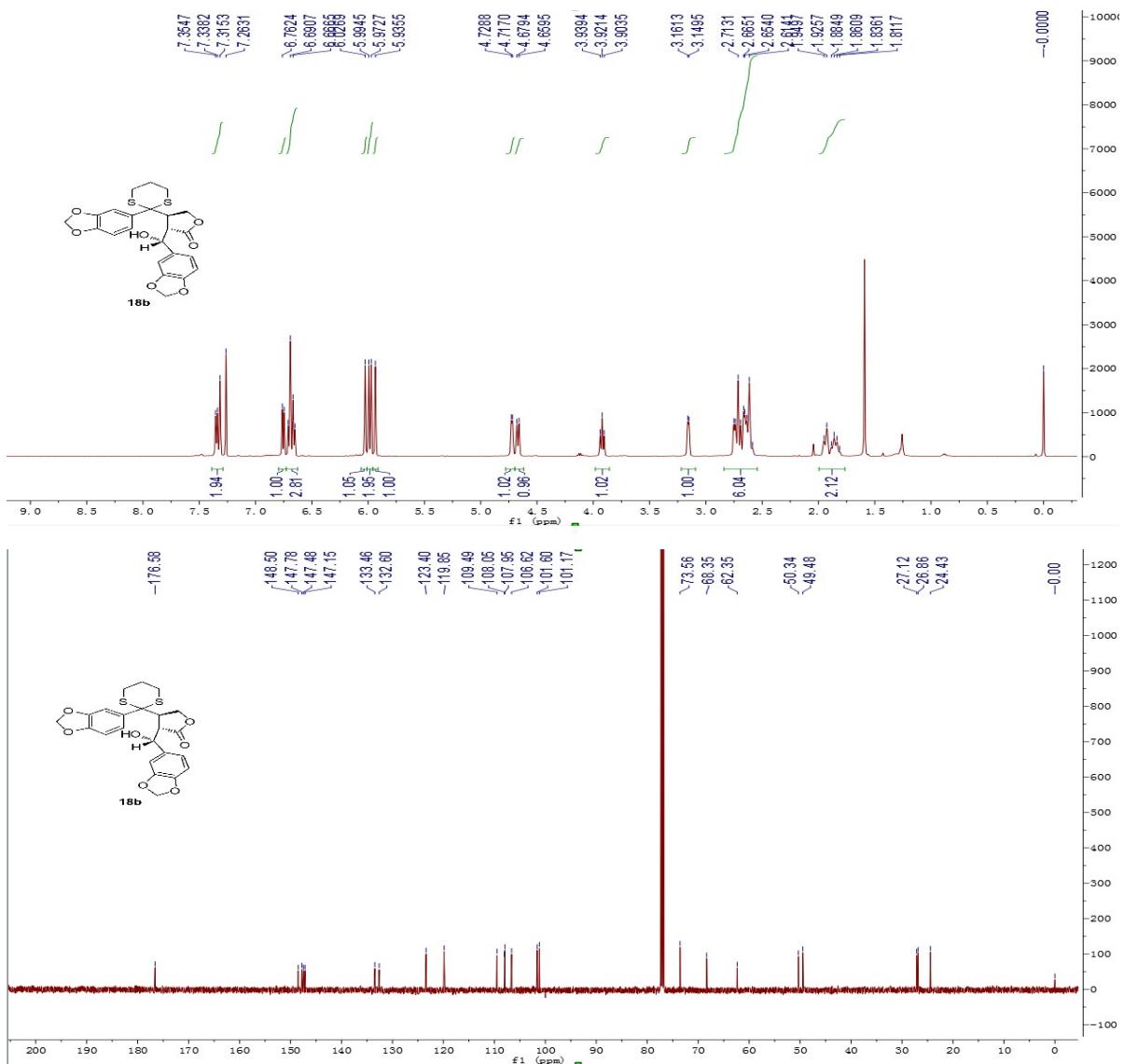
$^1\text{H}$  NMR (500 MHz) and  $^{13}\text{C}$  NMR (125 MHz) spectra of compound **17**.



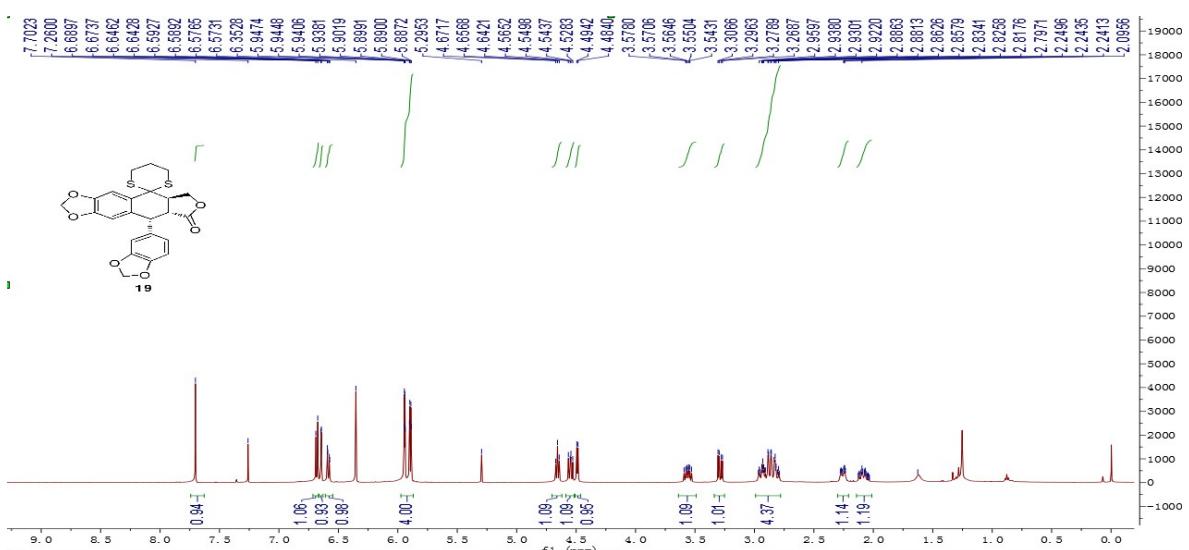


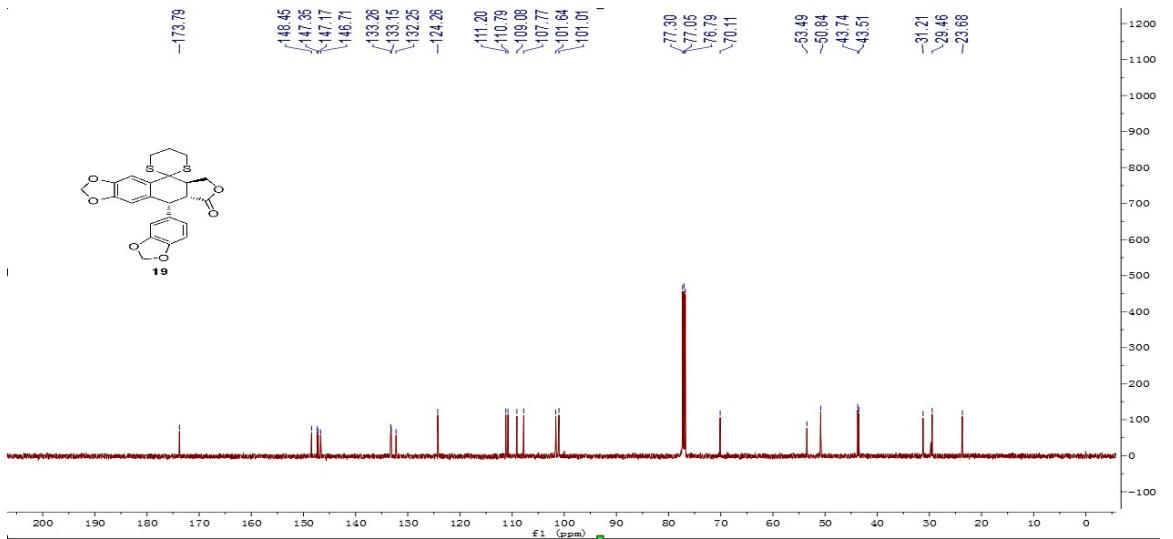
<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound **18a** and **18b**.



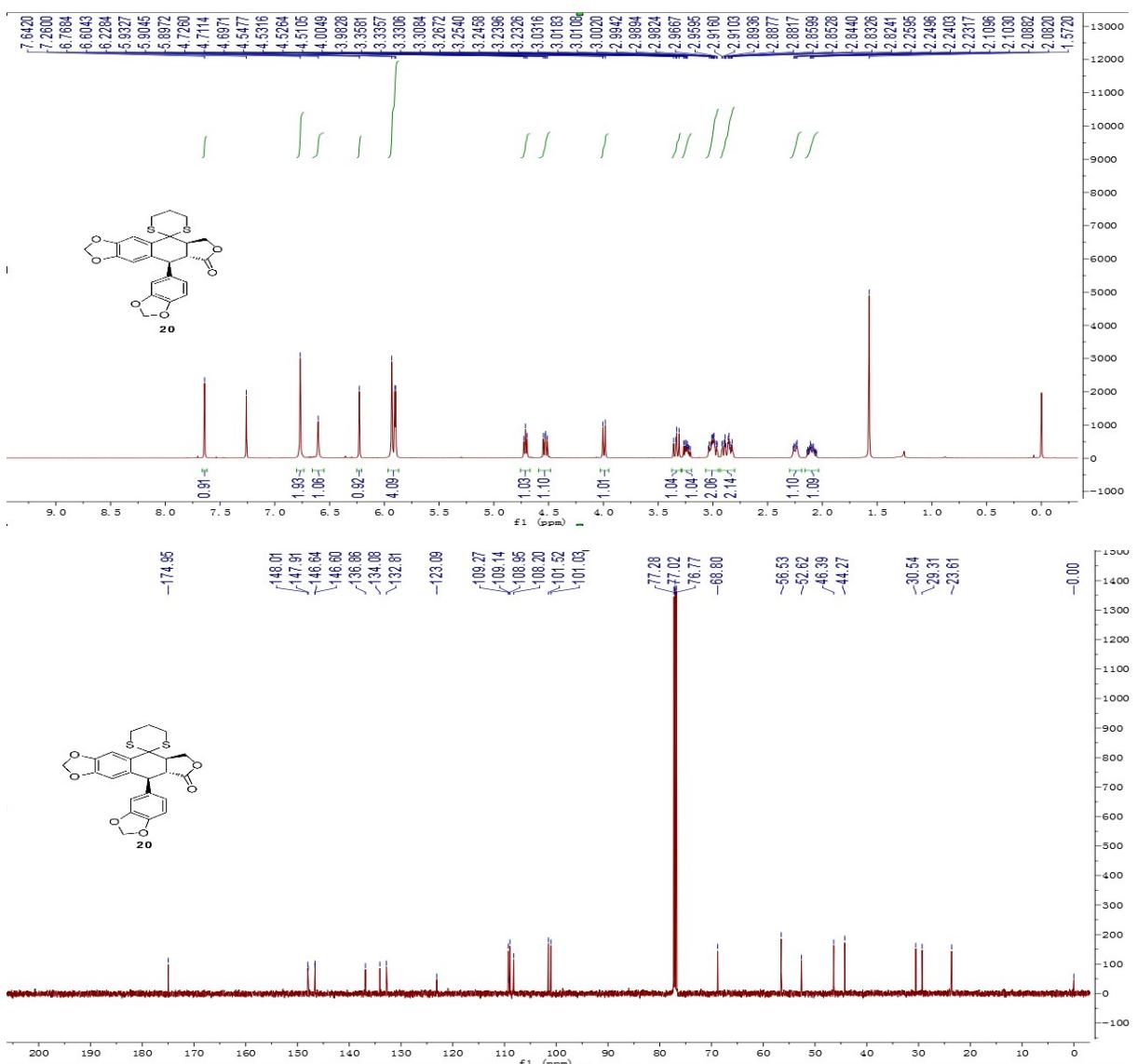


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound **19**.

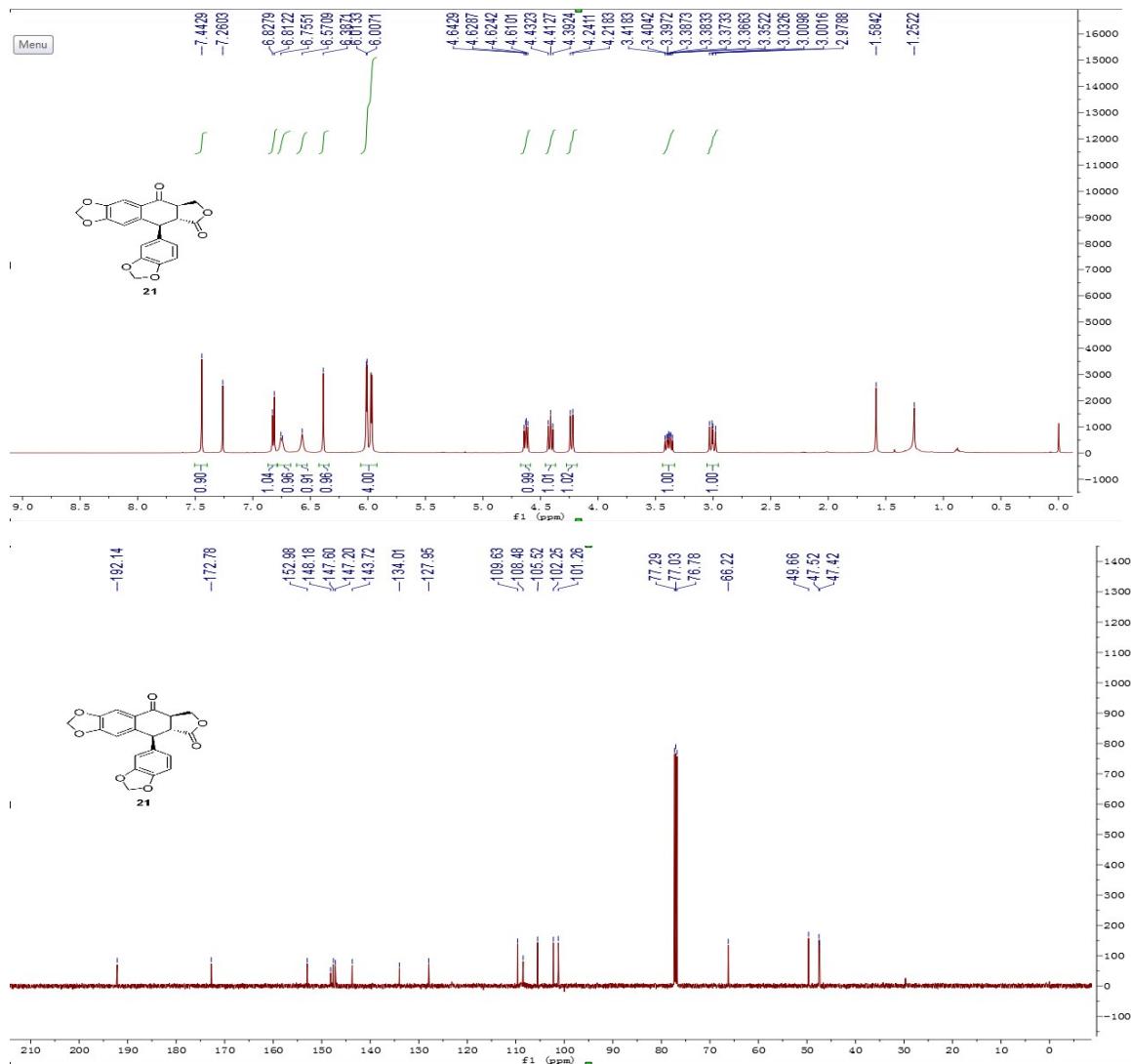




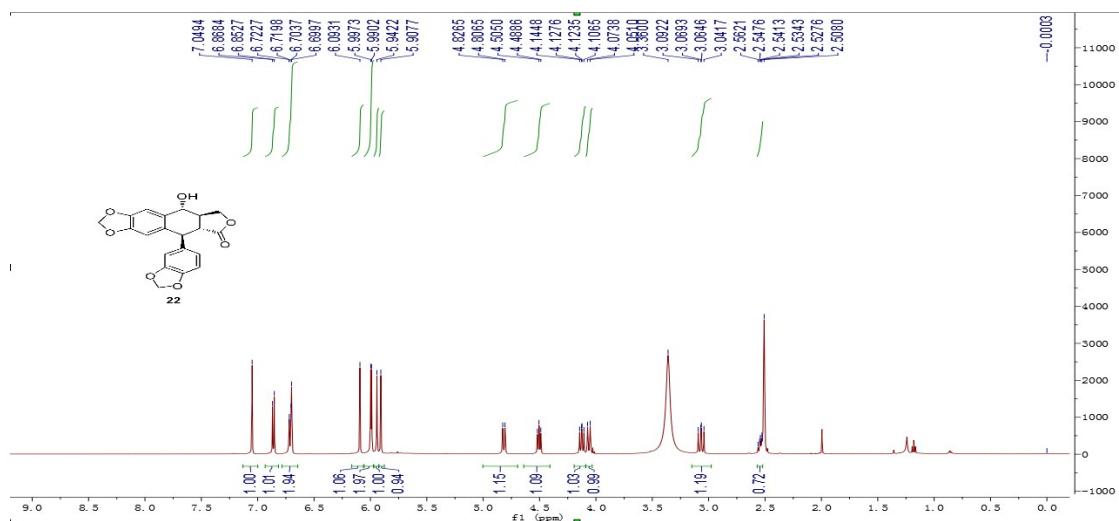
<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound **20**.

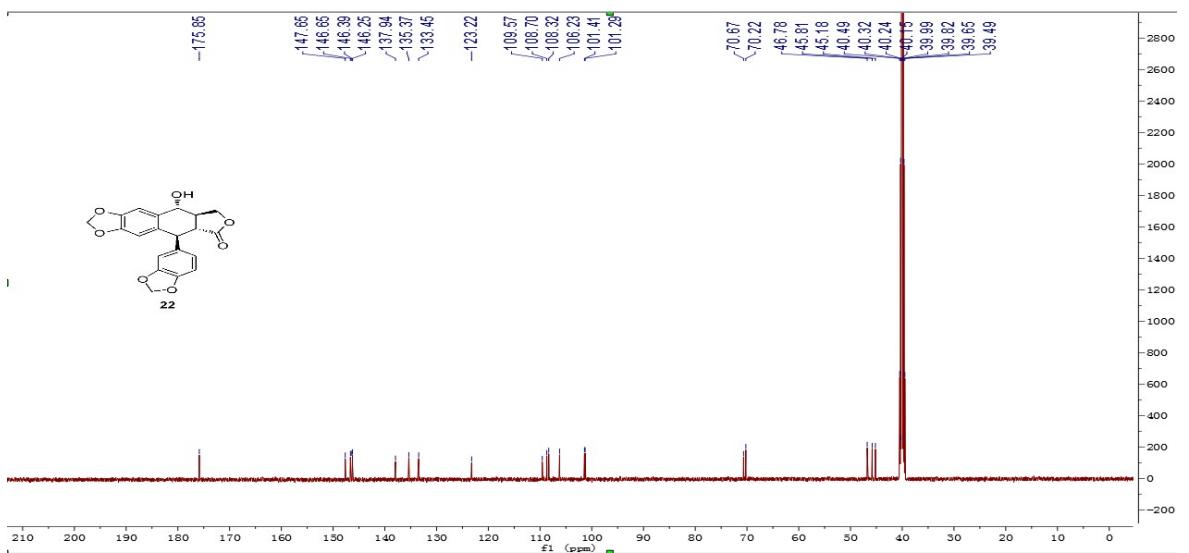


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 21.

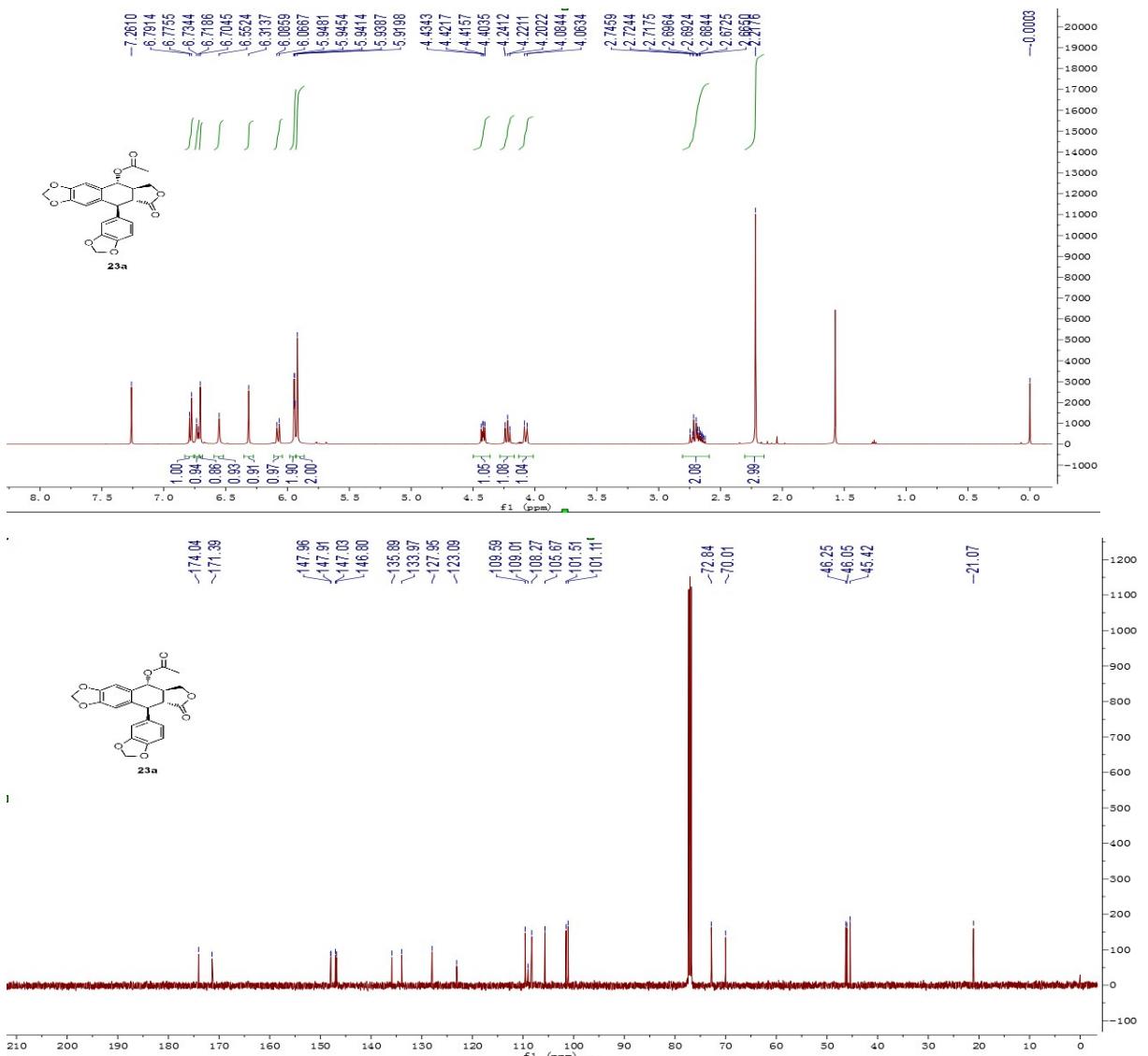


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 22.

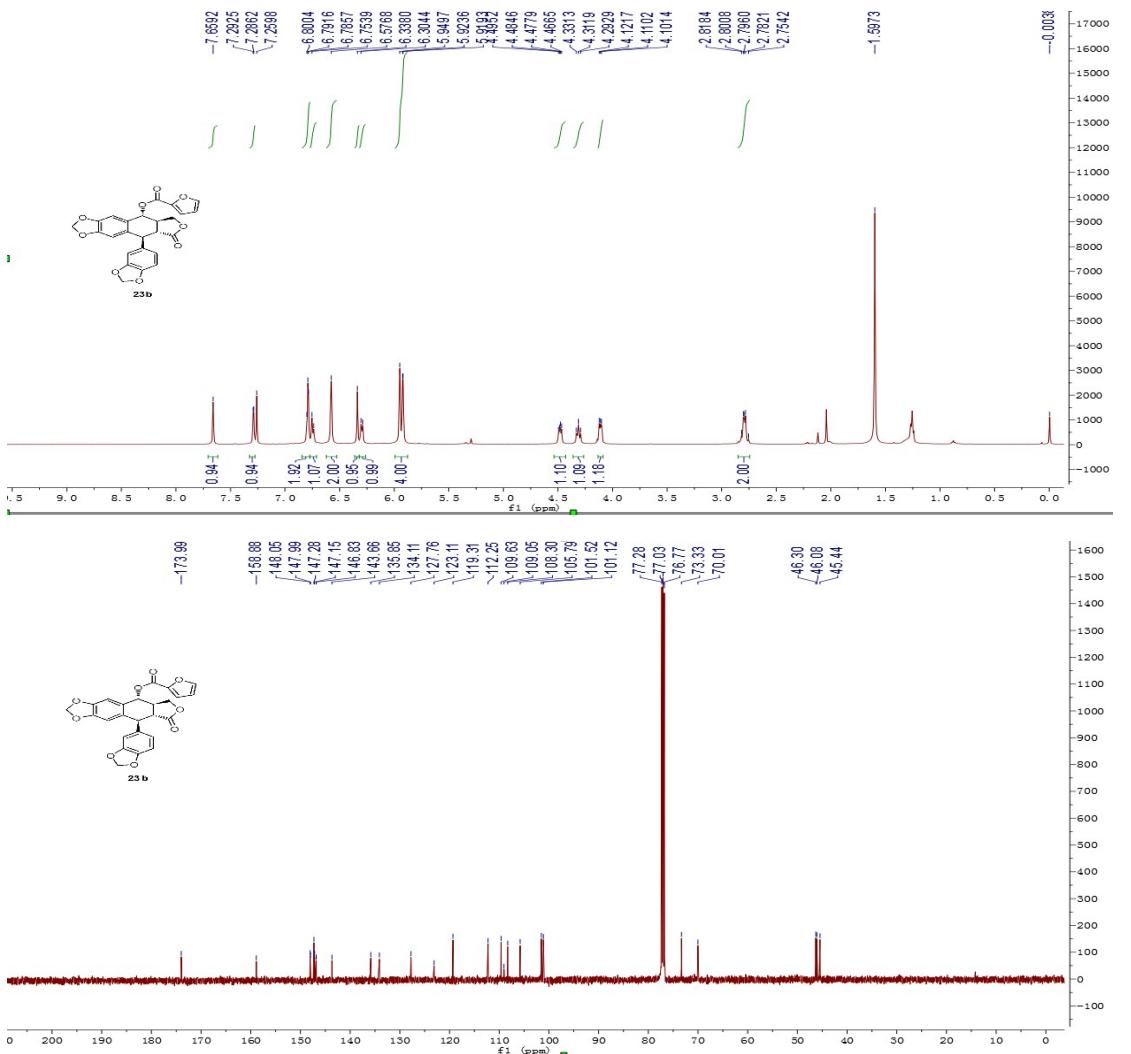




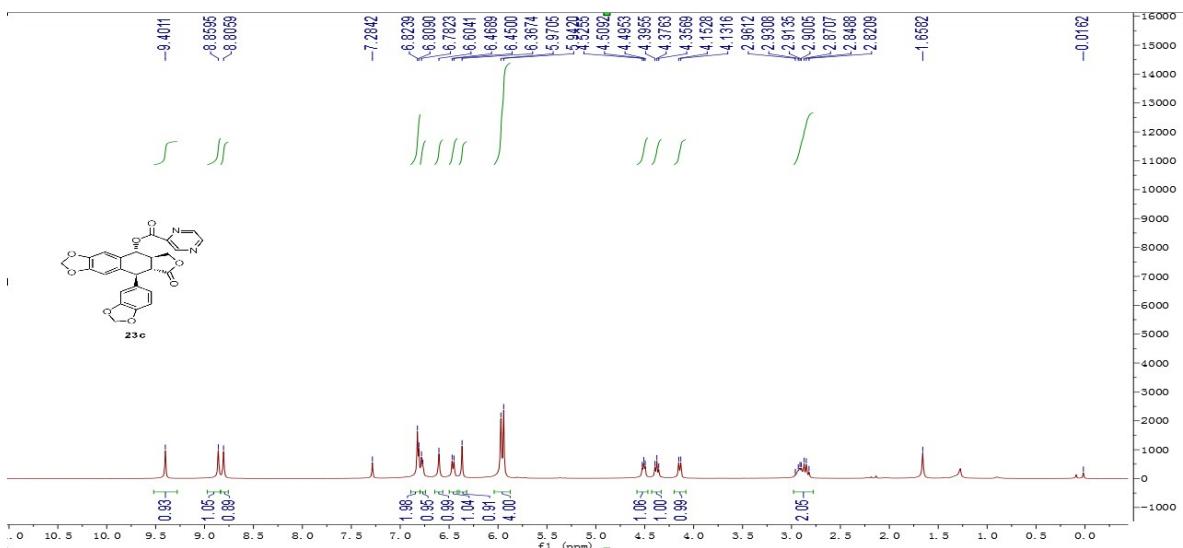
$^1\text{H}$  NMR (500 MHz) and  $^{13}\text{C}$  NMR (125 MHz) spectra of compound **23a**.

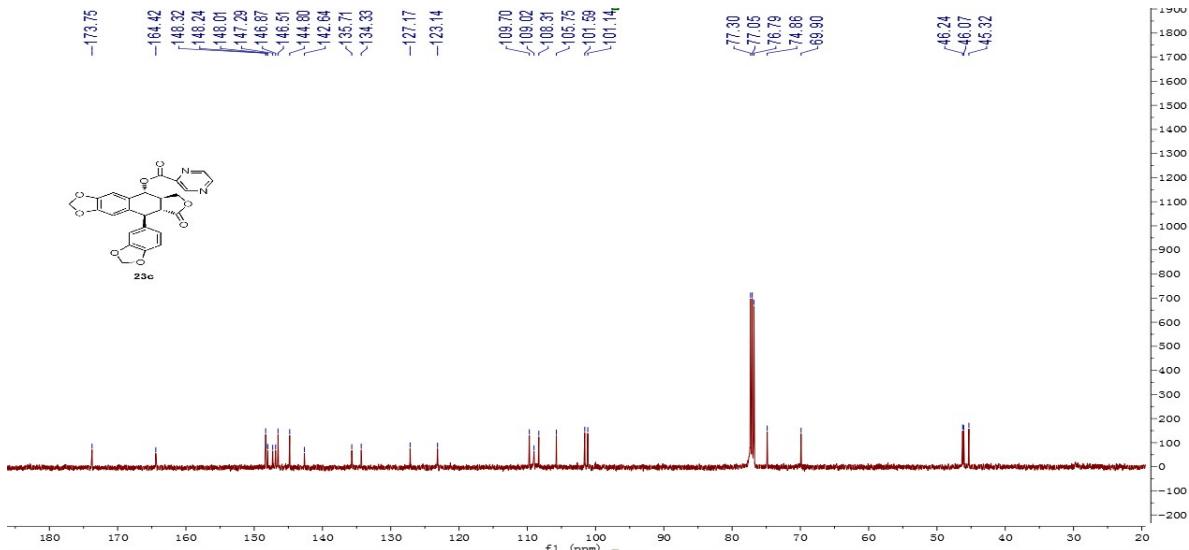


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 23b.

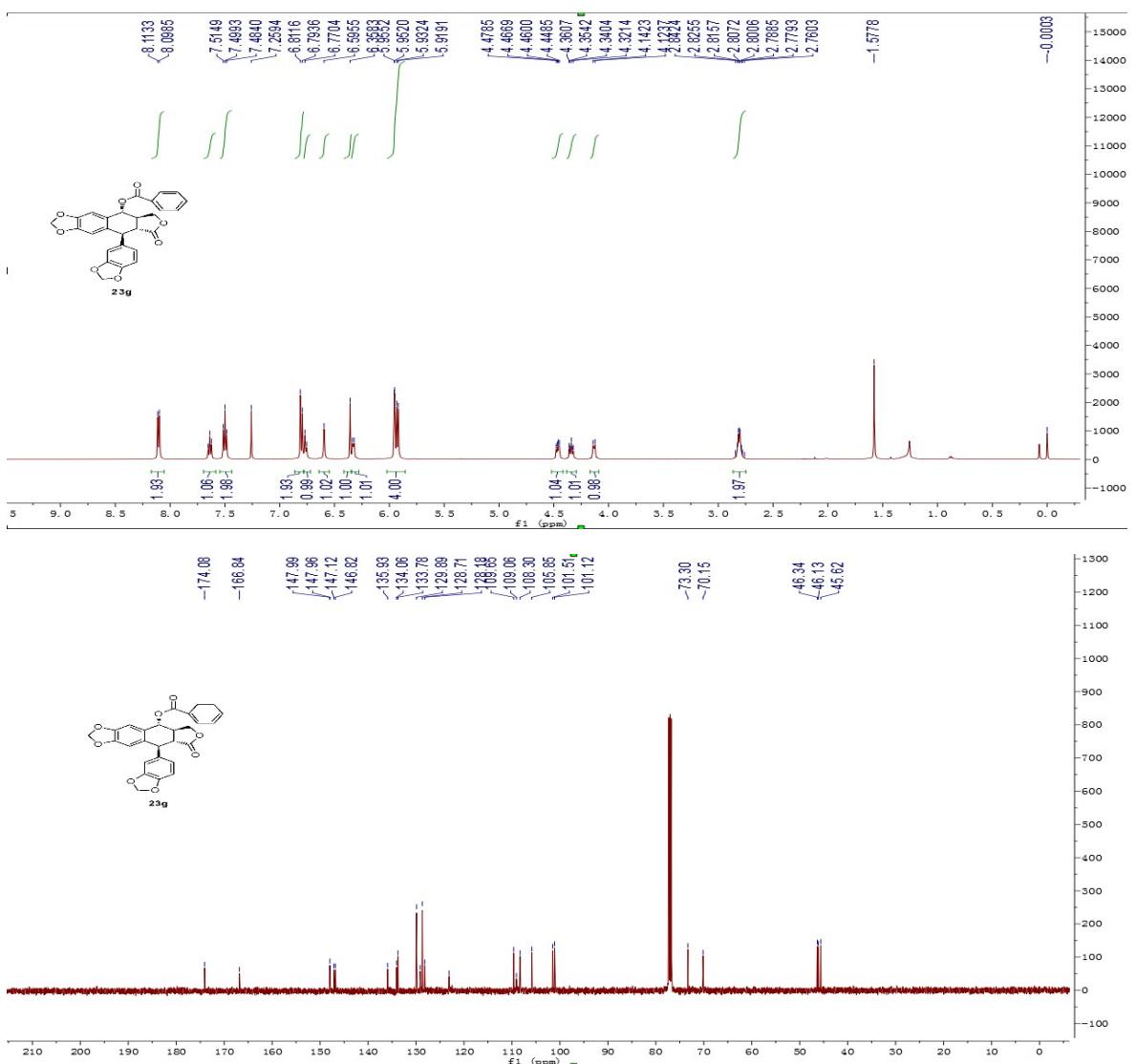


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 23c.

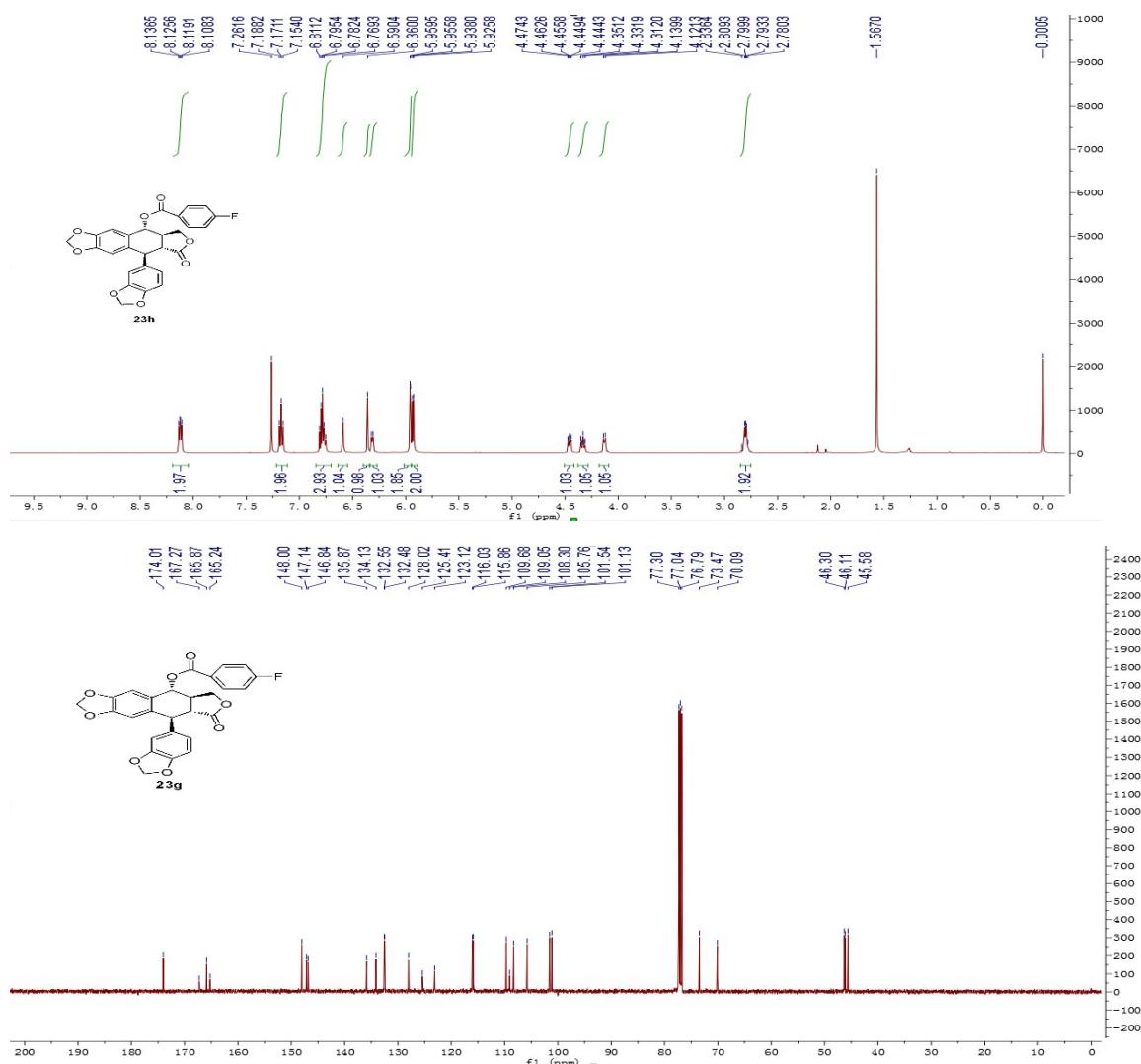




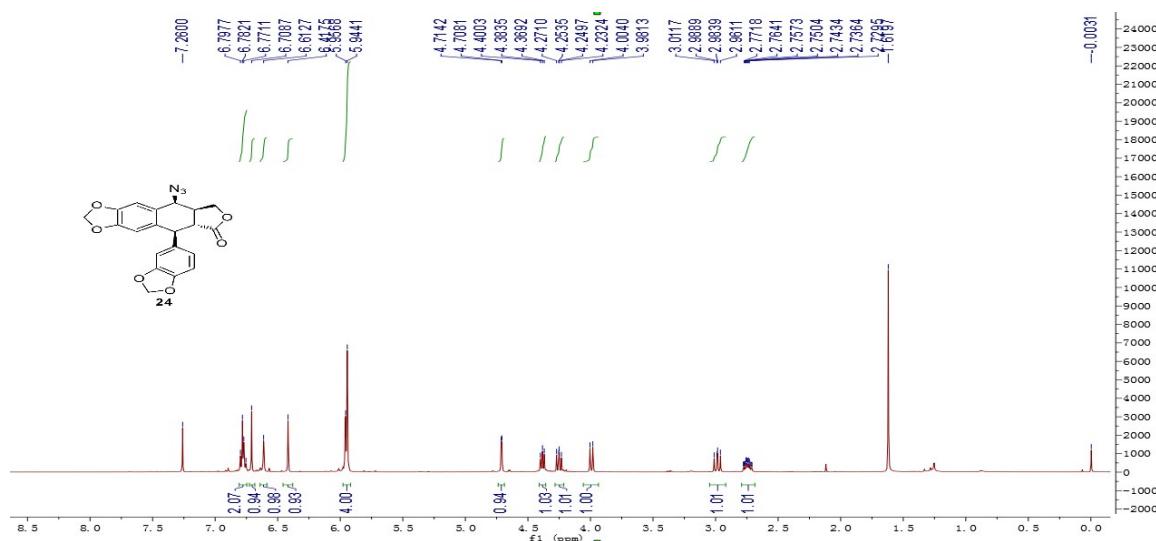
<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 23g.

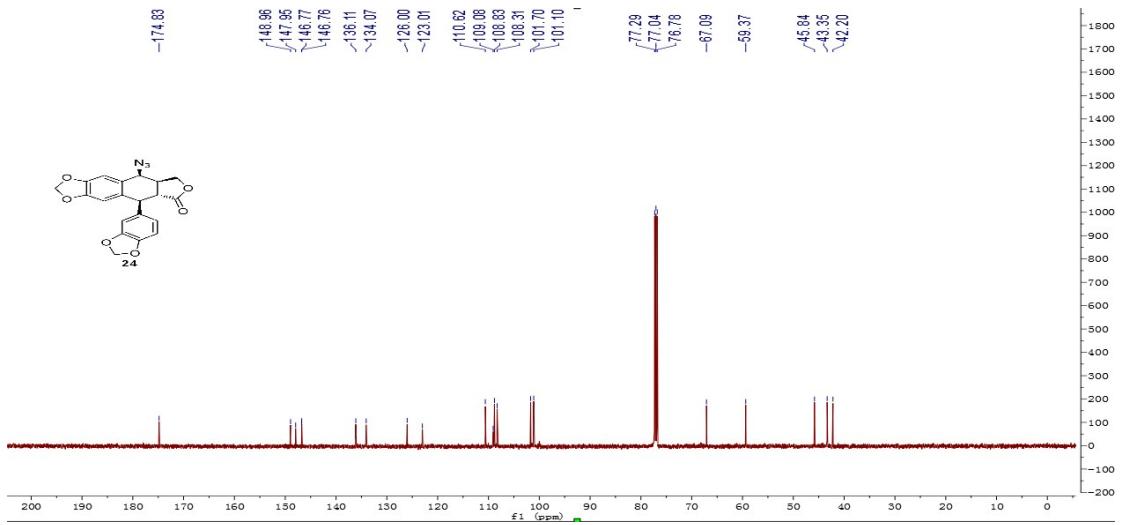


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound **23h**.

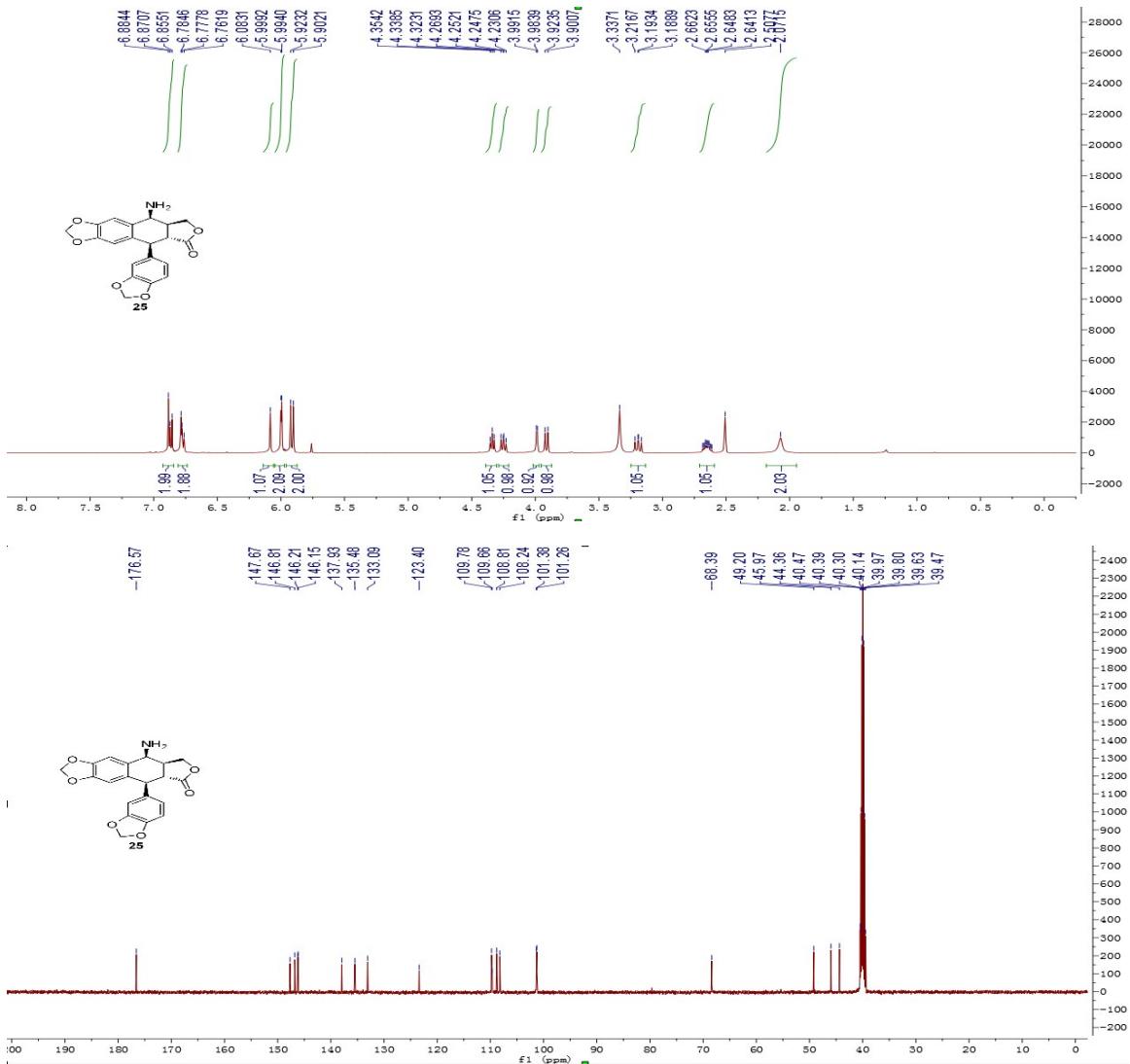


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 24.

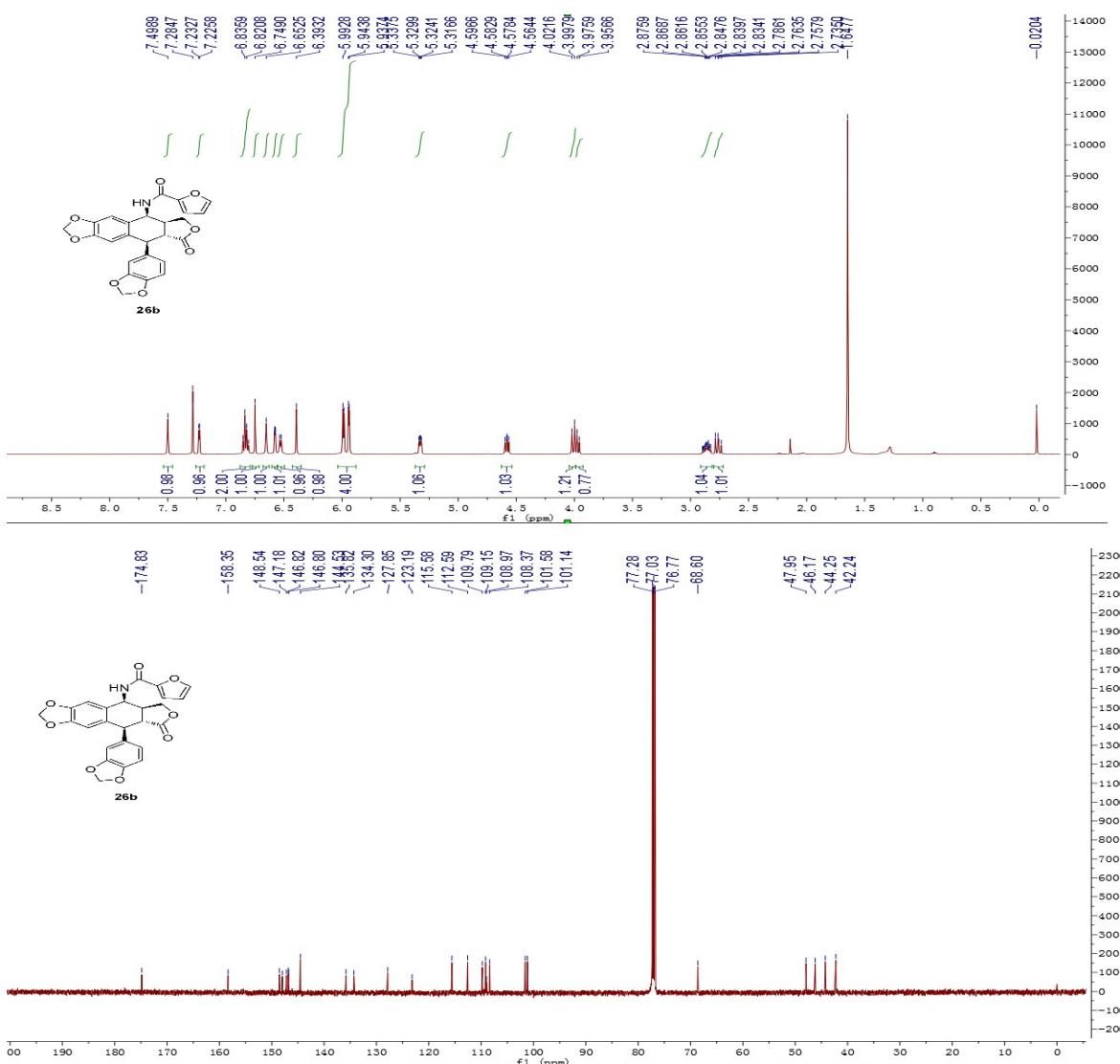




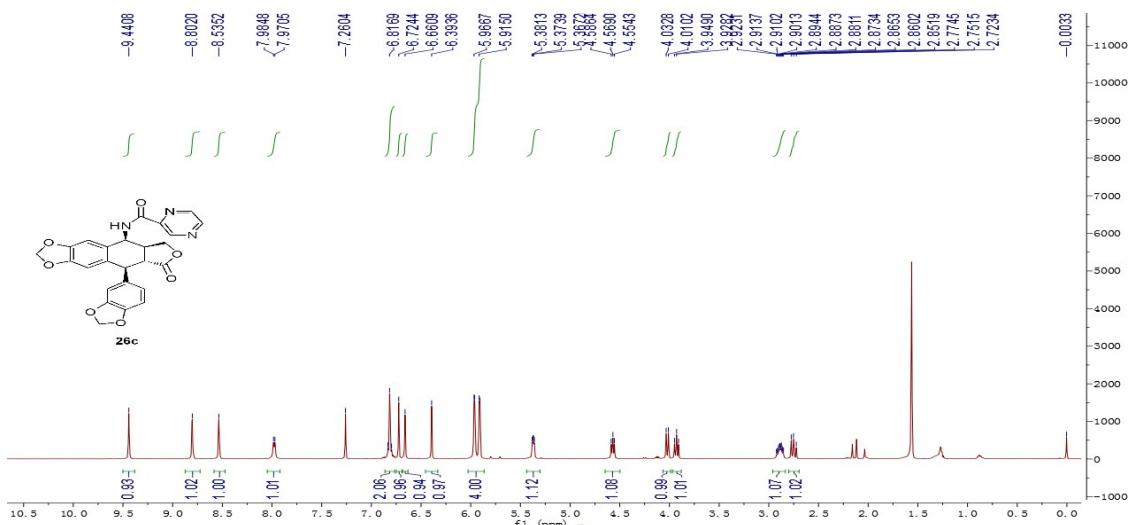
<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 25.

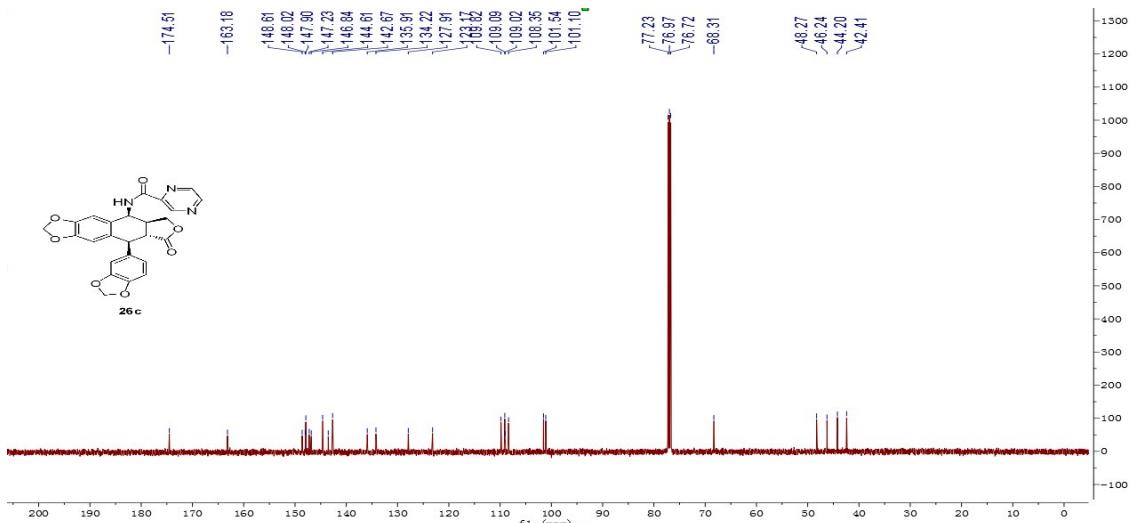


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound **26b**.

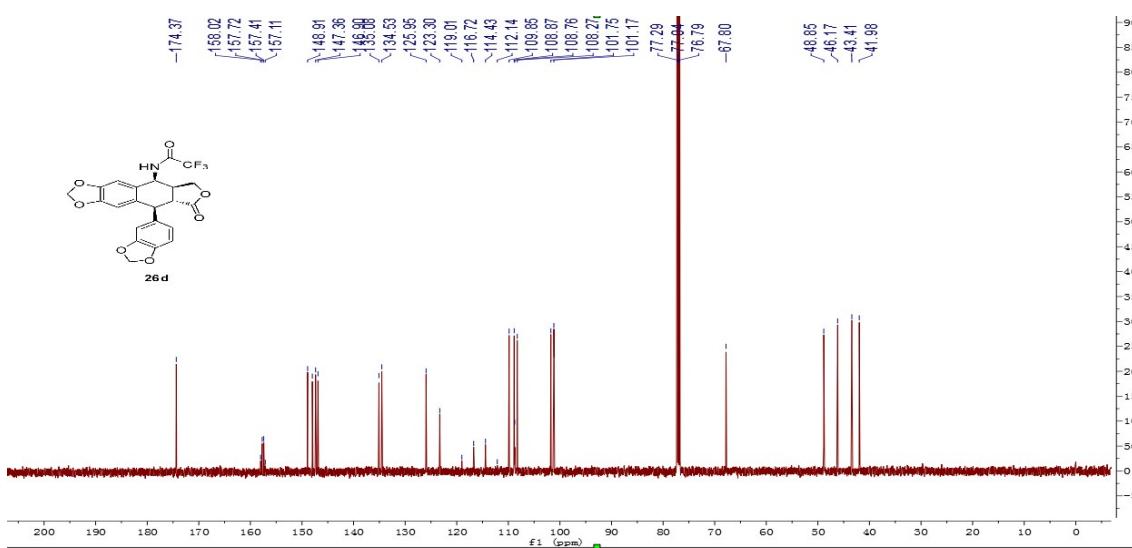
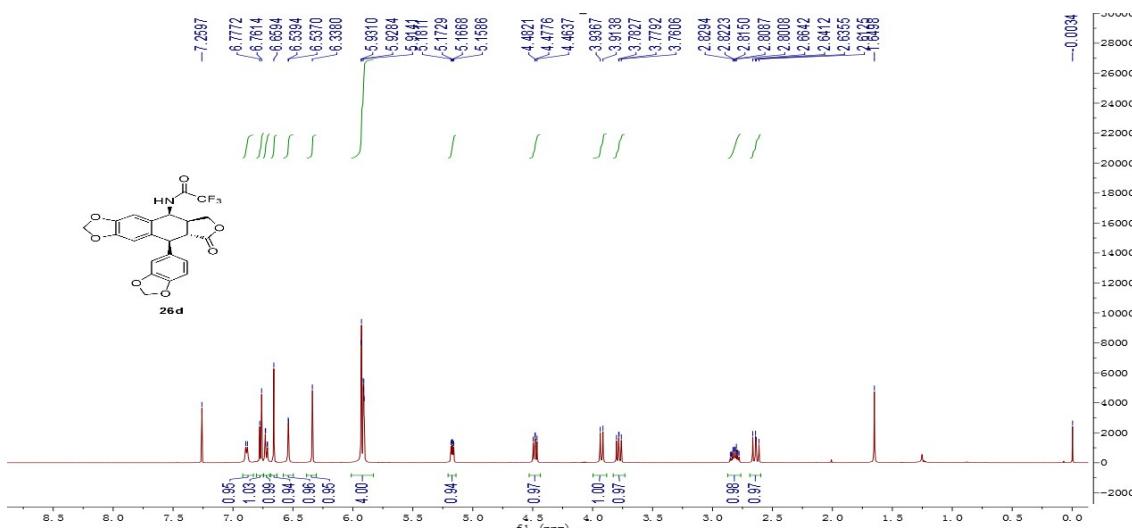


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound **26c**.

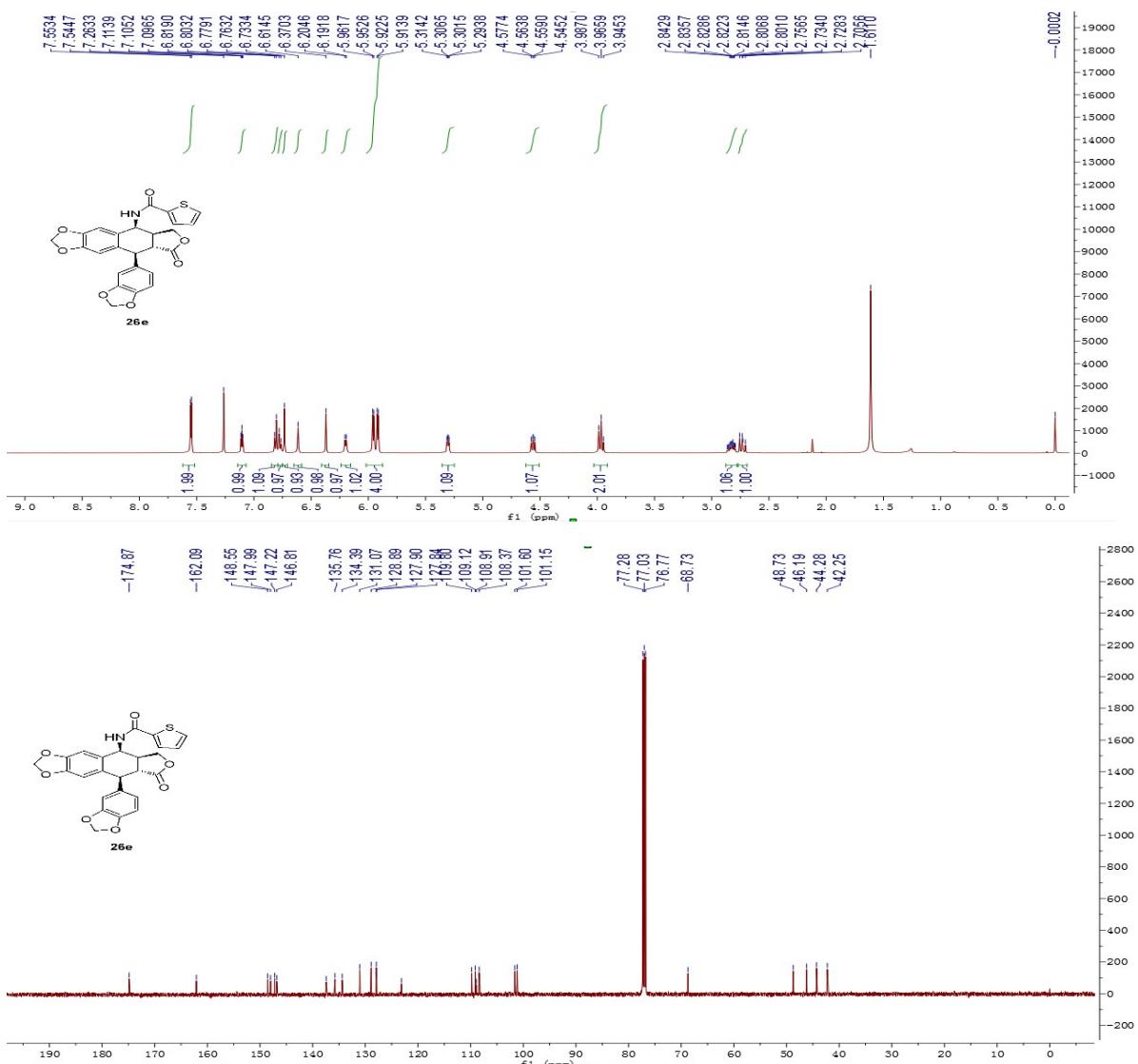




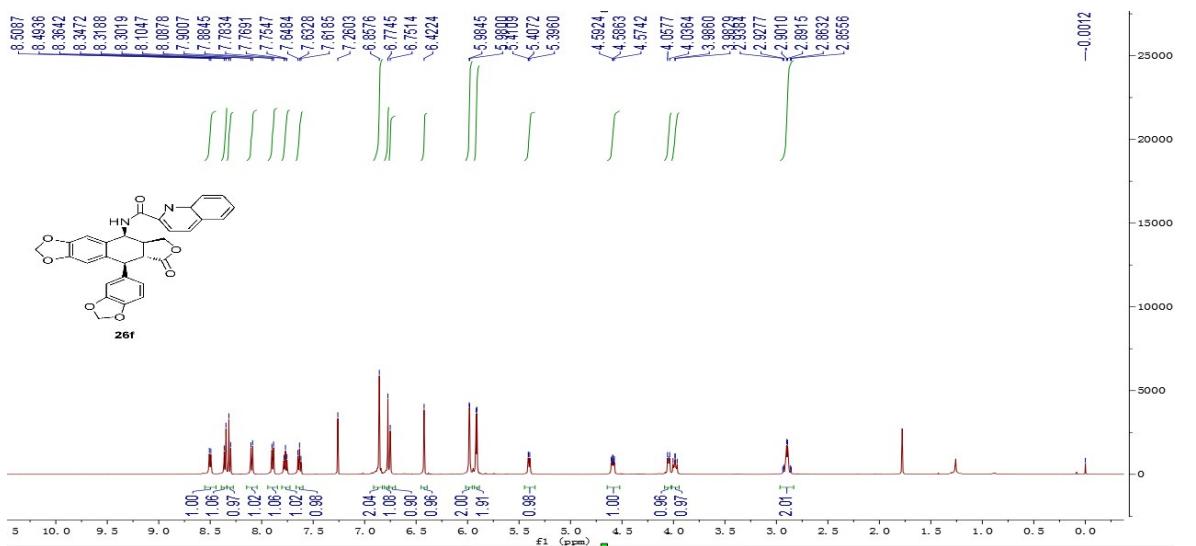
<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound **26d**.

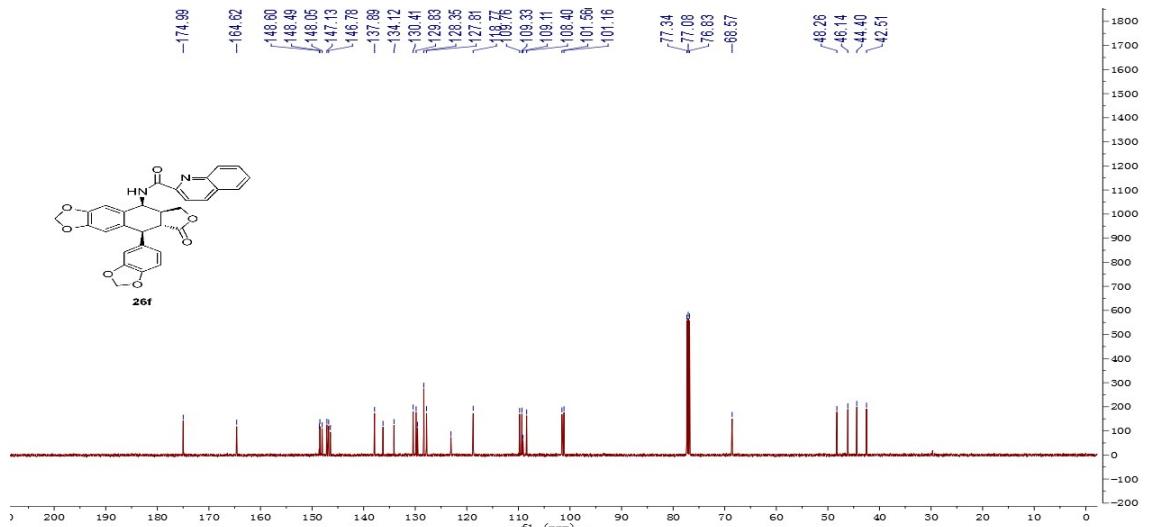


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 26e.

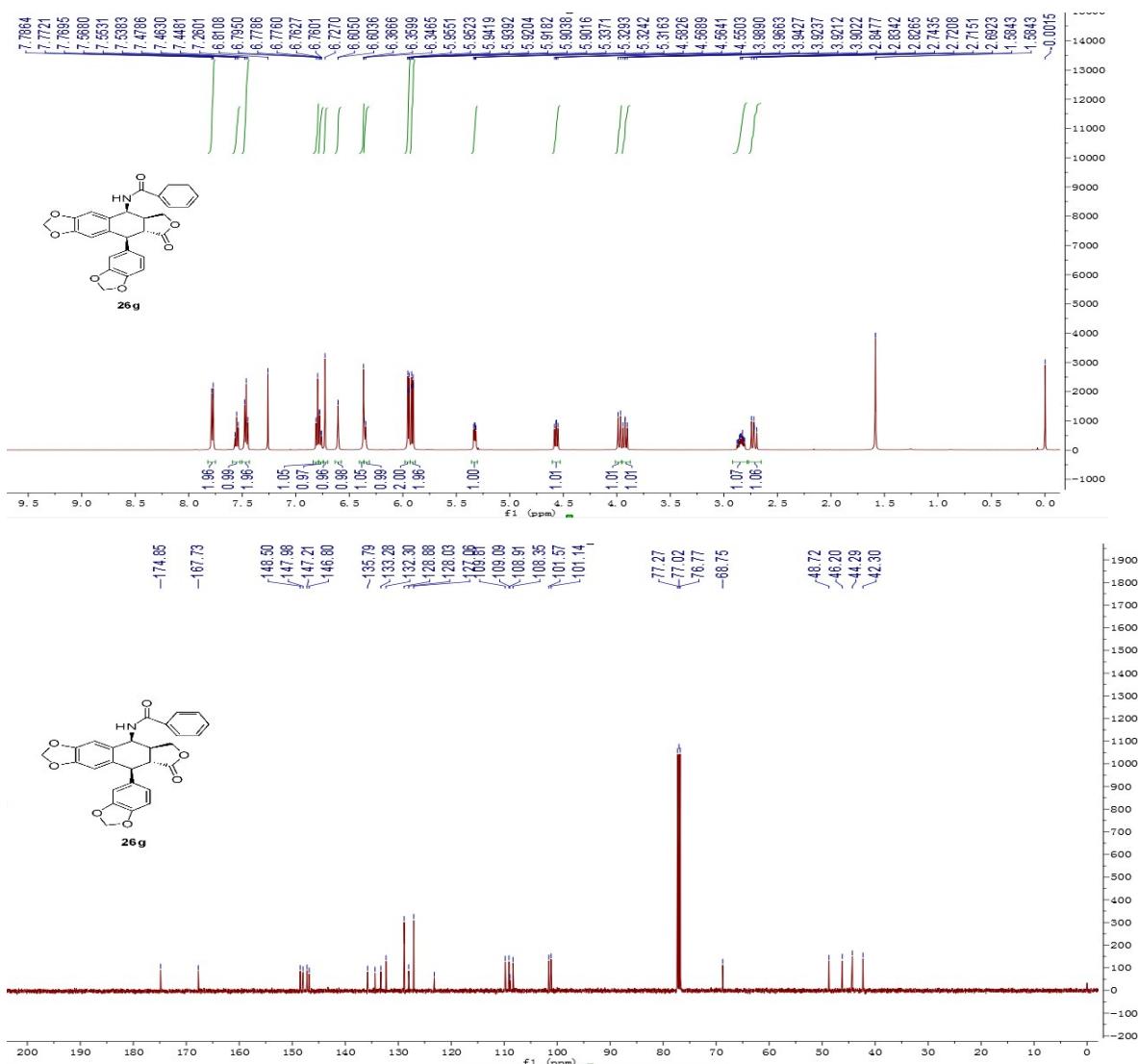


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 26f.

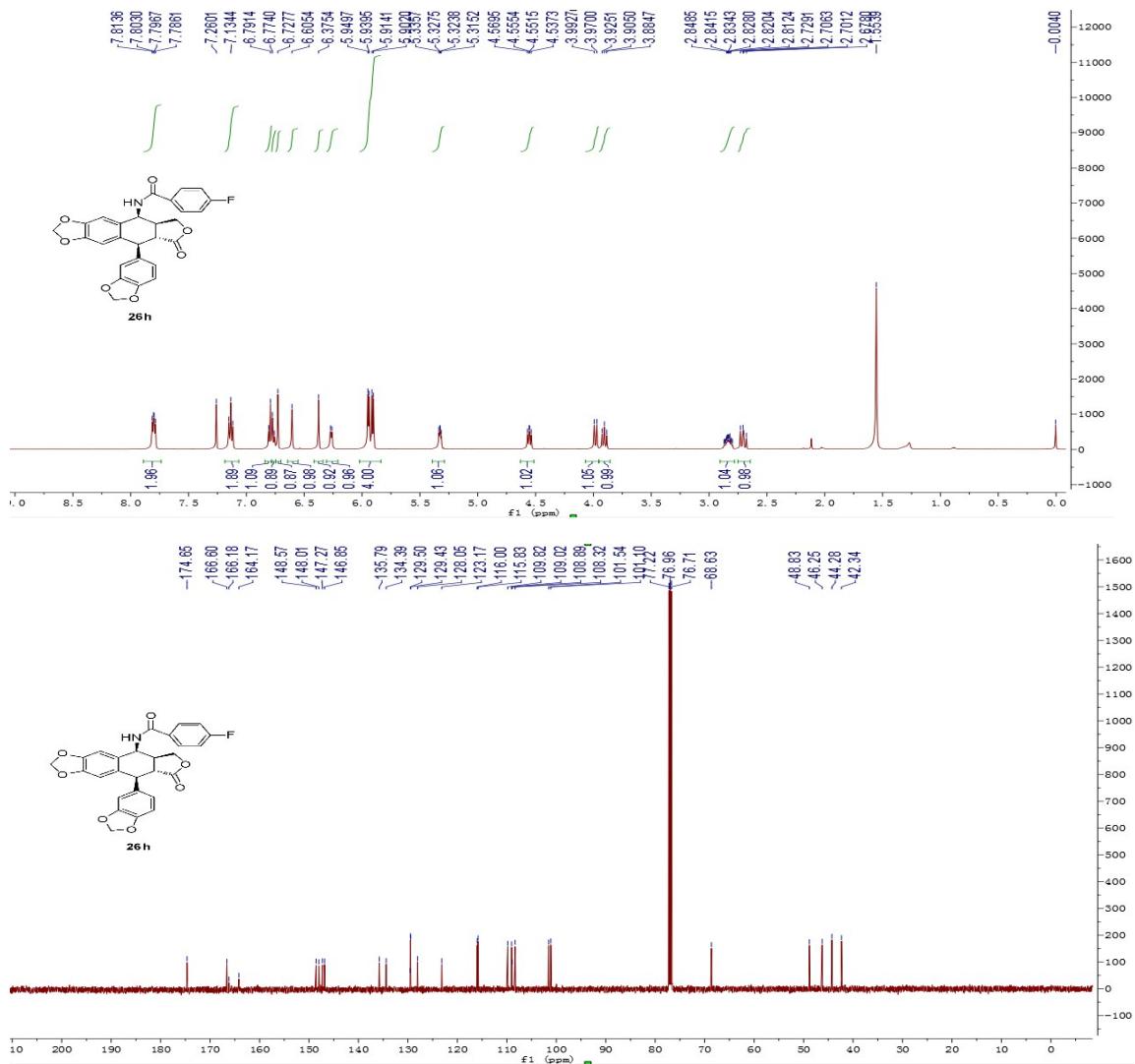




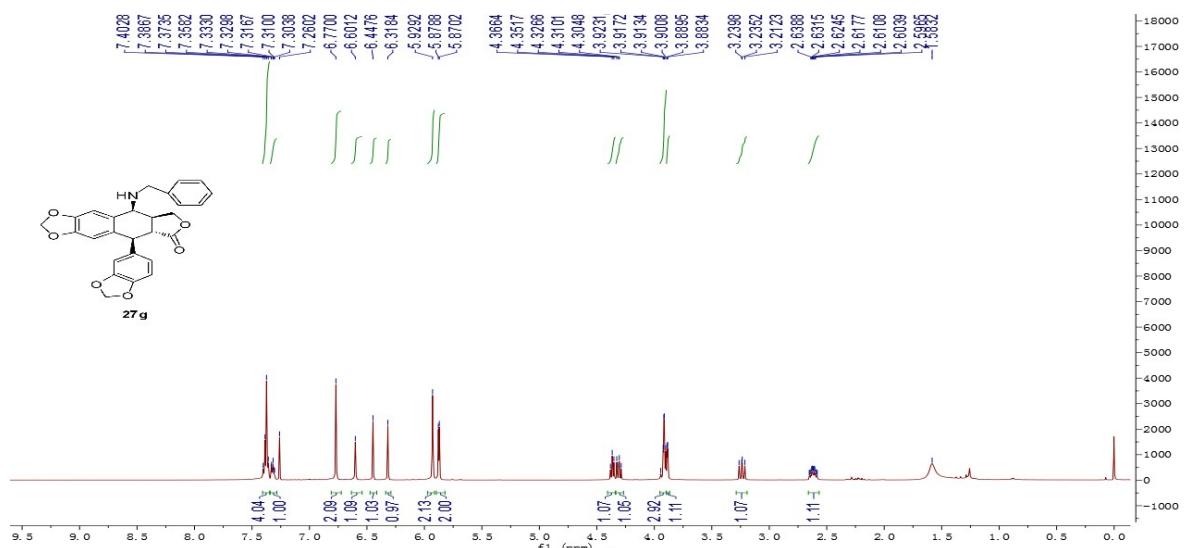
<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 26g.

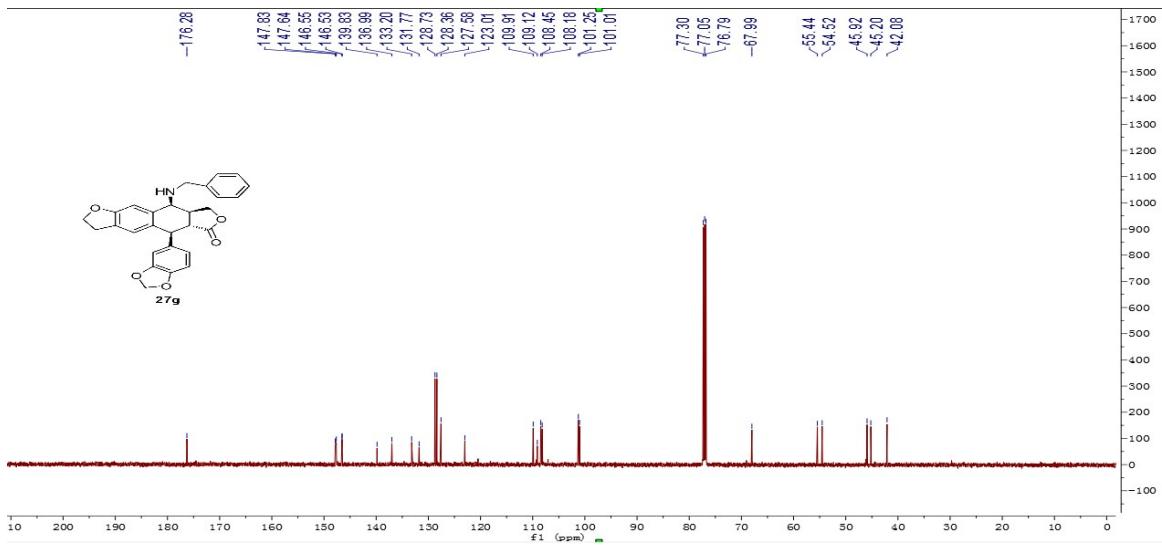


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound **26h**.

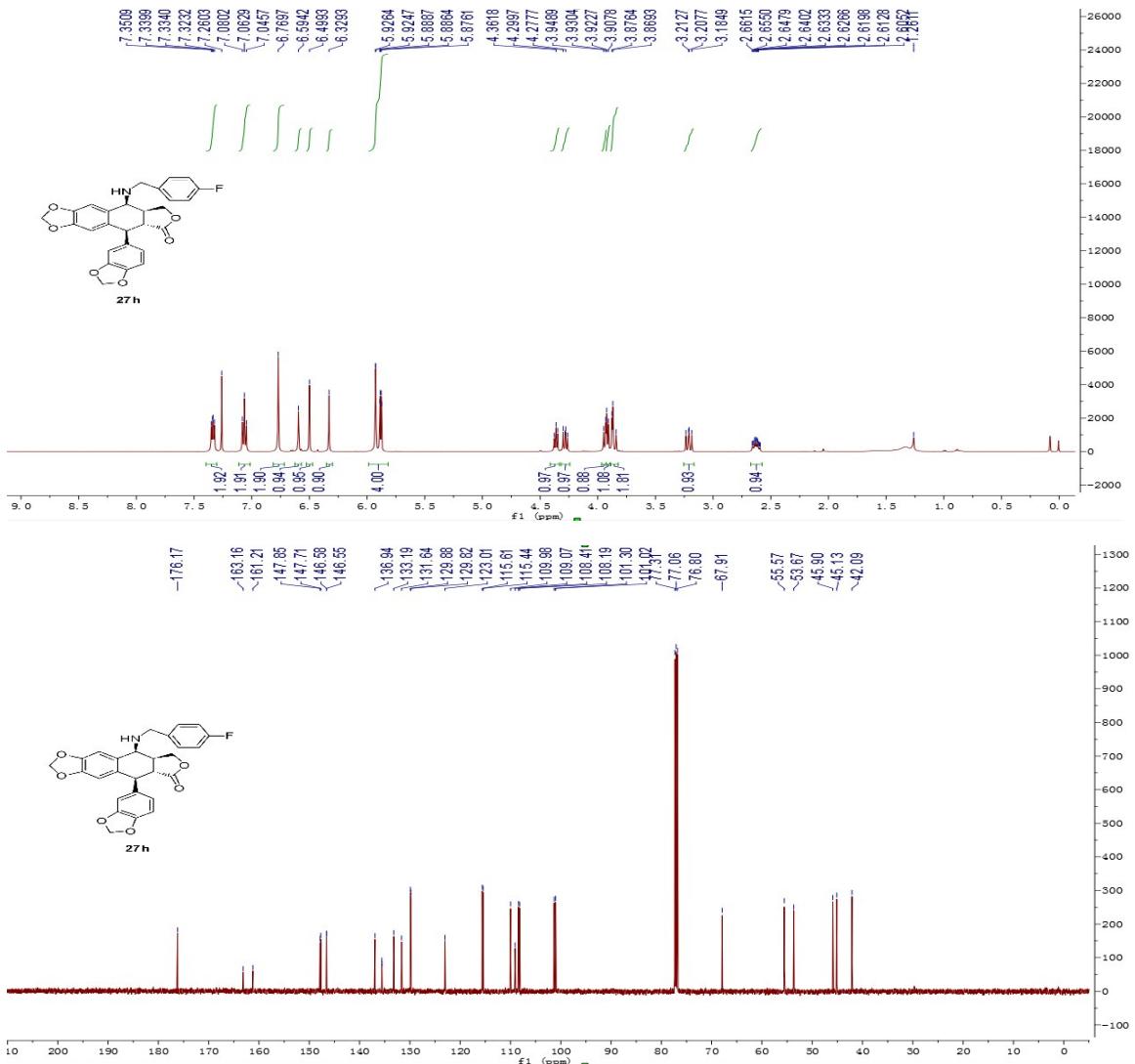


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 27g.

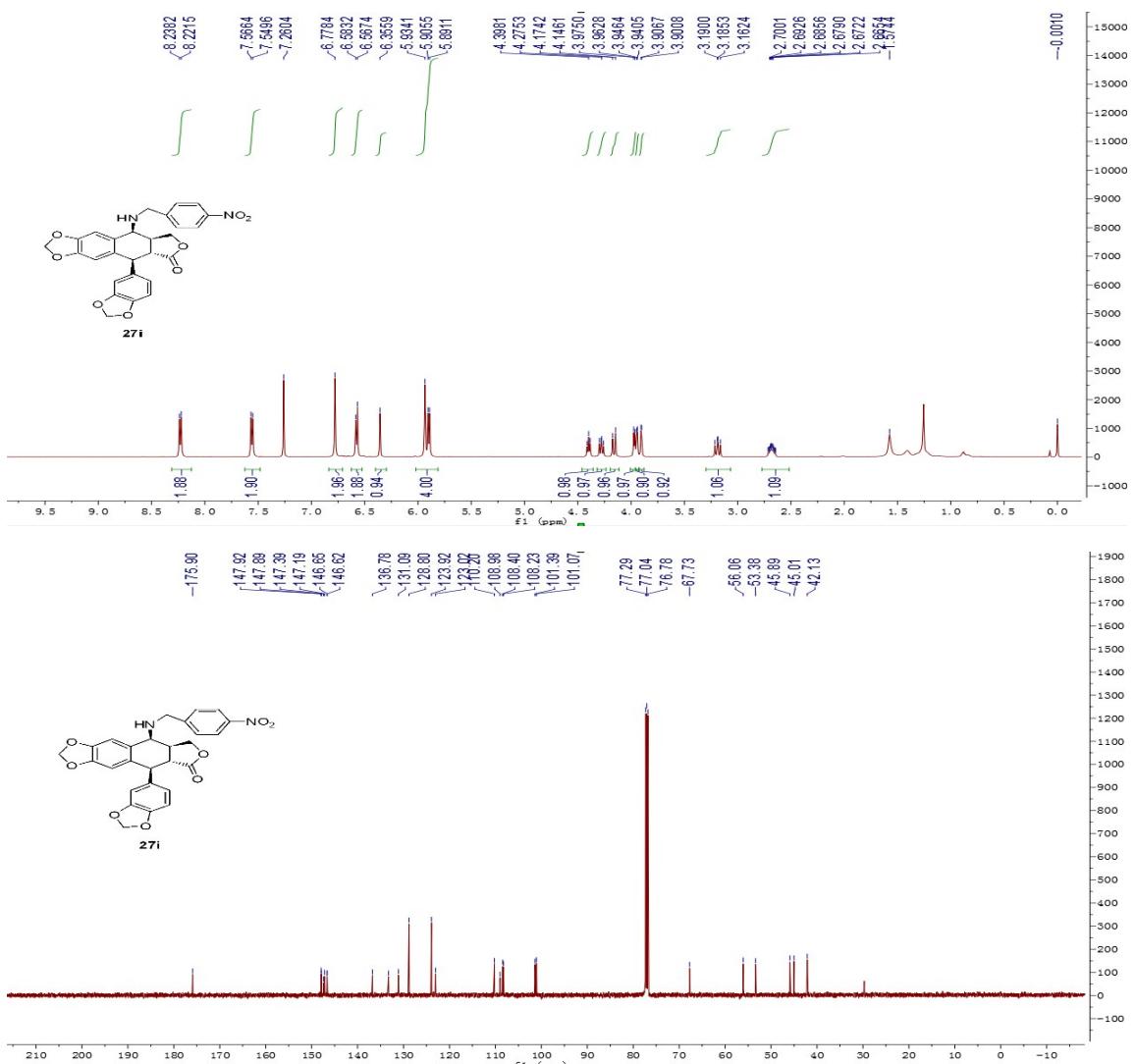




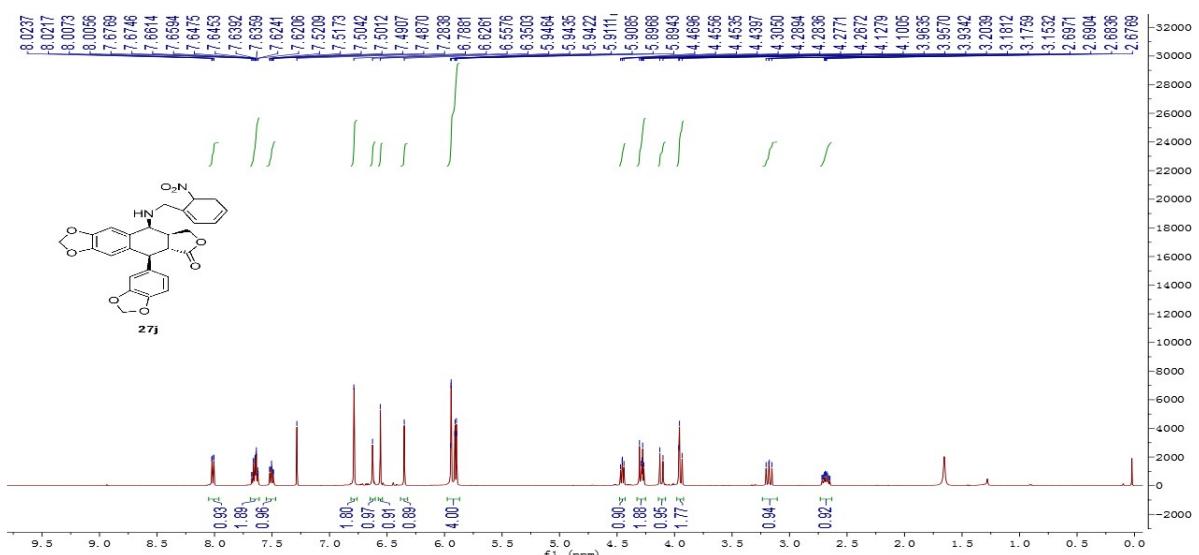
$^1\text{H}$  NMR (500 MHz) and  $^{13}\text{C}$  NMR (125 MHz) spectra of compound **27h**.

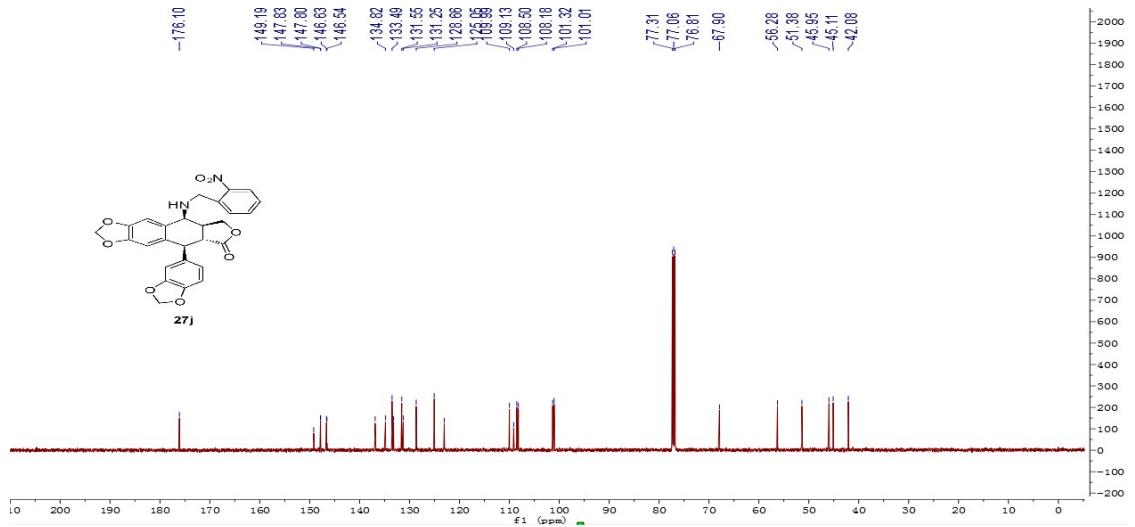


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 27i.

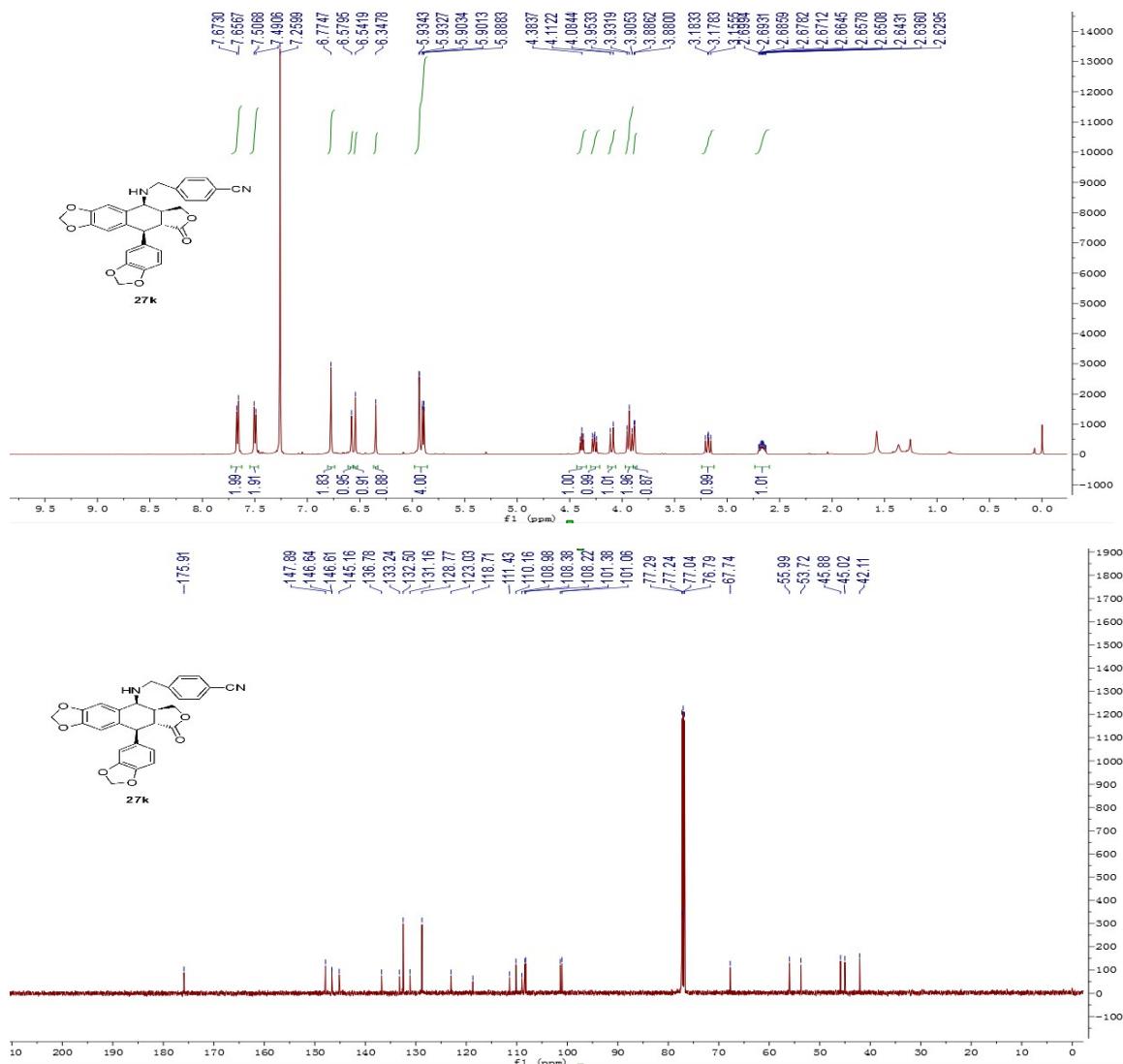


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 27j.

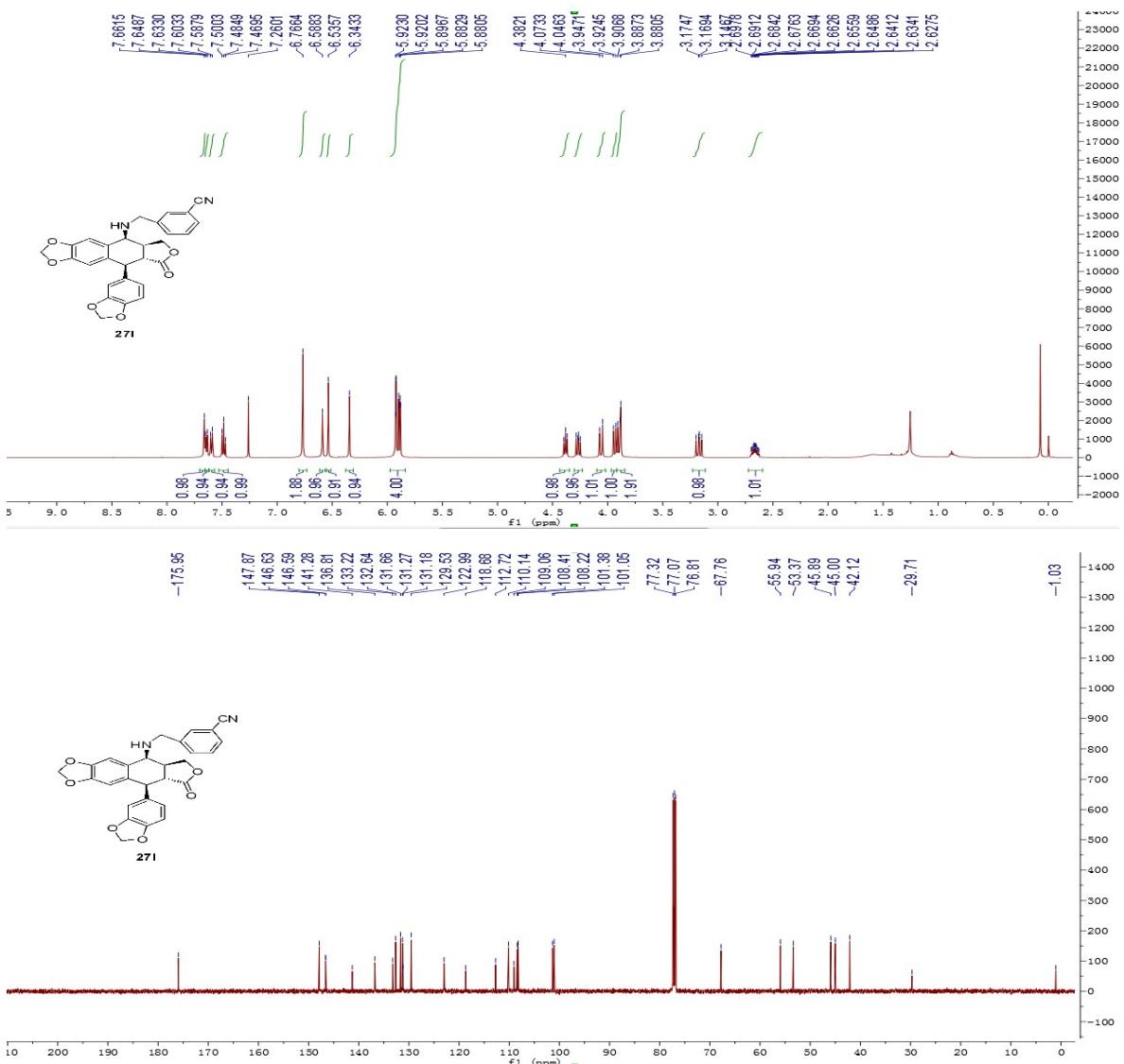




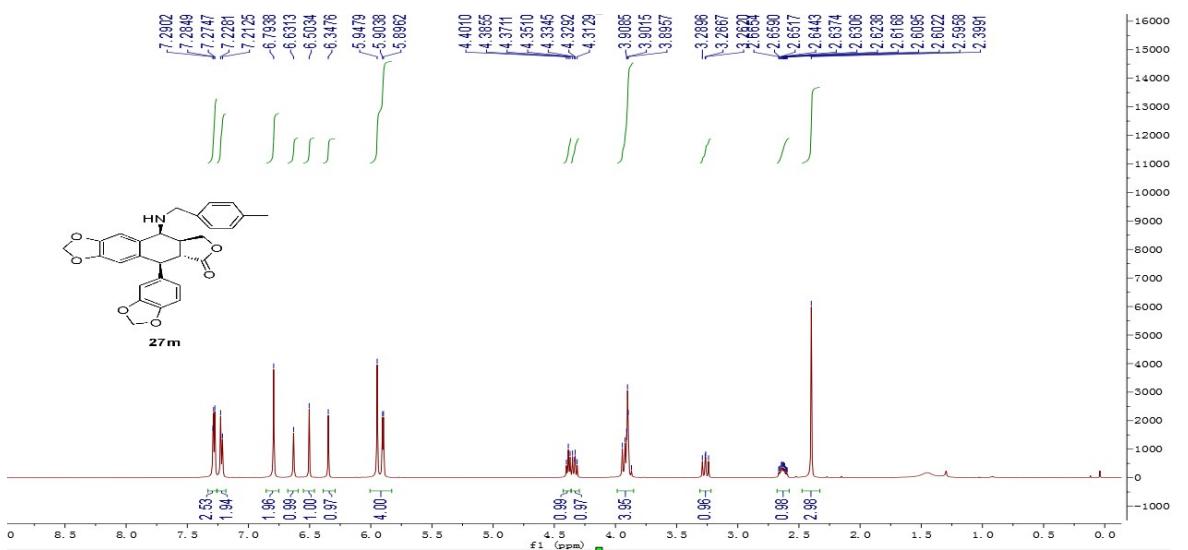
<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 27k.

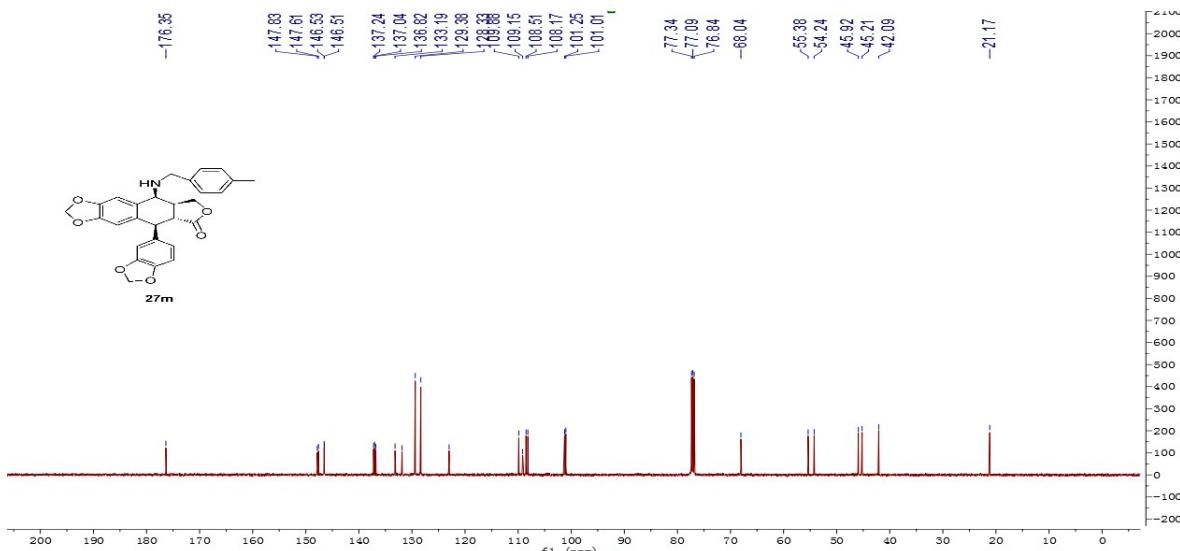


<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 27l.



<sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra of compound 27m.





#### 4. X-ray Data of Compound 23a and 24

##### 4.1. Single Crystal X-ray Crystallography of compound 23a (CCDC 1430247)

Data intensity of **23a** was collected using a Bruker SMART APEX II (Mo radiation). The X-ray condition of was  $50\text{ kV} \times 30\text{ mA}$ . Data collection and reduction were done by using the Bruker ApexII software package. The structure was solved by direct methods and refined by full-matrix least-squares on  $F^2$  with anisotropic displacement parameters for non-H atoms using SHELX-97. Hydrogen atoms were added at their geometrically ideal positions and refined isotropically. Crystal data for **23a**:  $\text{C}_{22}\text{H}_{18}\text{O}_8$ ,  $M = 410.36$ ,  $T = 173(2)\text{ K}$ ,  $\lambda = 0.71073\text{ \AA}$ , triclinic, space group P-1,  $a = 9.4301(6)\text{ \AA}$ ,  $b = 9.8683(6)\text{ \AA}$ ,  $c = 11.5227(8)\text{ \AA}$ ,  $V = 923.30(10)\text{ \AA}^3$ ,  $z = 2$ ,  $d_{\text{calc}} = 1.476\text{ mg/m}^3$ , 10831 reflections measured, 3228 unique [ $R_{\text{int}} = 0.0296$ ].  $R_1 = 0.0403$ ,  $wR_2 = 0.1065$  ( $I > 2\sigma(I)$ , final).  $R_1 = 0.0527$ ,  $wR_2 = 0.1185$  (all data). GOF = 1.080, and 271 parameters.

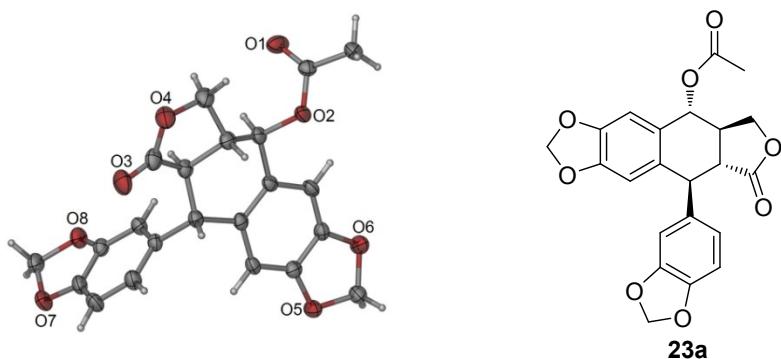


Table 1. Crystal data and structure refinement for z.

Identification code	z
Empirical formula	$\text{C}_{22}\text{H}_{18}\text{O}_8$
Formula weight	410.36
Temperature	173(2) K

Wavelength  $0.71073\text{ \AA}$

Crystal system, space group      Triclinic, P-1  
 Unit cell dimensions       $a = 9.4301(6)$  Å    $\alpha = 81.308(2)$  deg.  
                                      $b = 9.8683(6)$  Å    $\beta = 68.130(2)$  deg.  
                                      $c = 11.5227(8)$  Å    $\gamma = 68.113(2)$  deg.  
 Volume       $923.30(10)$  Å<sup>3</sup>  
 Z, Calculated density      2, 1.476 Mg/m<sup>3</sup>  
 Absorption coefficient      0.114 mm<sup>-1</sup>  
 F(000)      428  
 Crystal size      0.48 x 0.29 x 0.13 mm  
 Theta range for data collection    1.90 to 25.01 deg.  
 Limiting indices      -11≤h≤11, -11≤k≤11, -13≤l≤13  
 Reflections collected / unique    10831 / 3228 [R(int) = 0.0296]  
 Completeness to theta = 25.01    99.1 %  
 Absorption correction      Semi-empirical from equivalents  
 Max. and min. transmission    0.9854 and 0.9475  
 Refinement method      Full-matrix least-squares on F<sup>2</sup>  
 Data / restraints / parameters    3228 / 0 / 271  
 Goodness-of-fit on F<sup>2</sup>      1.080  
 Final R indices [I>2sigma(I)]    R1 = 0.0403, wR2 = 0.1065  
 R indices (all data)      R1 = 0.0527, wR2 = 0.1185  
 Largest diff. peak and hole    0.226 and -0.223 e.Å<sup>-3</sup>

Table 2. Atomic coordinates (x 10<sup>4</sup>) and equivalent isotropic  
 displacement parameters (Å<sup>2</sup> x 10<sup>3</sup>) for z.  
 U(eq) is defined as one third of the trace of the orthogonalized  
 Uij tensor.

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x	y	z	U(eq)
O(1)	8533(2)	5524(2)	1594(2)    42(1)

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O(2)	8438(2)	7821(2)	1005(1)	31(1)
O(3)	15354(2)	6848(2)	35(2)	45(1)
O(4)	13304(2)	6285(2)	1501(1)	42(1)
O(5)	9704(2)	11104(2)	-4049(1)	34(1)
O(6)	7476(2)	10409(2)	-2907(2)	38(1)
O(7)	18654(2)	5857(2)	-5663(1)	39(1)
O(8)	16764(2)	4701(2)	-5012(1)	34(1)
C(1)	8005(3)	11425(2)	-3828(2)	34(1)
C(2)	10028(2)	10142(2)	-3100(2)	27(1)
C(3)	11431(2)	9605(2)	-2828(2)	26(1)
C(4)	11469(2)	8650(2)	-1782(2)	25(1)
C(5)	13038(2)	8123(2)	-1471(2)	26(1)
C(6)	12951(2)	6989(2)	-428(2)	28(1)
C(7)	14045(3)	6727(2)	331(2)	36(1)
C(8)	11719(3)	6260(3)	1607(2)	38(1)
C(9)	11294(2)	7346(2)	596(2)	29(1)
C(10)	10071(2)	7287(2)	79(2)	28(1)
C(11)	10100(2)	8274(2)	-1074(2)	26(1)
C(12)	8694(2)	8808(2)	-1412(2)	29(1)
C(13)	8696(2)	9735(2)	-2414(2)	28(1)
C(14)	7817(3)	6818(2)	1703(2)	29(1)
C(15)	6158(3)	7514(3)	2597(2)	39(1)
C(16)	14558(2)	7564(2)	-2611(2)	26(1)
C(17)	15724(3)	8235(2)	-3003(2)	33(1)
C(18)	17157(3)	7745(2)	-4028(2)	37(1)
C(19)	17375(2)	6545(2)	-4618(2)	32(1)
C(20)	16234(2)	5862(2)	-4232(2)	27(1)
C(21)	14800(2)	6356(2)	-3255(2)	27(1)
C(22)	18460(3)	4491(2)	-5662(2)	36(1)

Table 3. Bond lengths [Å] and angles [deg] for z.

O(1)-C(14)	1.203(2)
O(2)-C(14)	1.346(2)
O(2)-C(10)	1.459(2)
O(3)-C(7)	1.200(3)
O(4)-C(7)	1.358(3)
O(4)-C(8)	1.463(3)
O(5)-C(2)	1.381(2)
O(5)-C(1)	1.440(2)
O(6)-C(13)	1.374(2)
O(6)-C(1)	1.427(3)
O(7)-C(19)	1.386(3)
O(7)-C(22)	1.425(3)
O(8)-C(20)	1.385(2)
O(8)-C(22)	1.439(2)
C(1)-H(1A)	0.9900
C(1)-H(1B)	0.9900
C(2)-C(3)	1.367(3)
C(2)-C(13)	1.380(3)
C(3)-C(4)	1.416(3)
C(3)-H(3A)	0.9500
C(4)-C(11)	1.397(3)
C(4)-C(5)	1.534(3)
C(5)-C(6)	1.514(3)
C(5)-C(16)	1.520(3)
C(5)-H(5A)	1.0000
C(6)-C(9)	1.516(3)
C(6)-C(7)	1.517(3)
C(6)-H(6A)	1.0000
C(8)-C(9)	1.523(3)
C(8)-H(8A)	0.9900
C(8)-H(8B)	0.9900
C(9)-C(10)	1.503(3)

C(9)-H(9A)	1.0000
C(10)-C(11)	1.522(3)
C(10)-H(10A)	1.0000
C(11)-C(12)	1.410(3)
C(12)-C(13)	1.360(3)
C(12)-H(12A)	0.9500
C(14)-C(15)	1.480(3)
C(15)-H(15A)	0.9800
C(15)-H(15B)	0.9800
C(15)-H(15C)	0.9800
C(16)-C(17)	1.390(3)
C(16)-C(21)	1.401(3)
C(17)-C(18)	1.397(3)
C(17)-H(17A)	0.9500
C(18)-C(19)	1.366(3)
C(18)-H(18A)	0.9500
C(19)-C(20)	1.380(3)
C(20)-C(21)	1.369(3)
C(21)-H(21A)	0.9500
C(22)-H(22A)	0.9900
C(22)-H(22B)	0.9900
C(14)-O(2)-C(10)	117.34(15)
C(7)-O(4)-C(8)	109.86(16)
C(2)-O(5)-C(1)	105.06(15)
C(13)-O(6)-C(1)	106.22(15)
C(19)-O(7)-C(22)	103.50(16)
C(20)-O(8)-C(22)	103.65(15)
O(6)-C(1)-O(5)	107.63(16)
O(6)-C(1)-H(1A)	110.2
O(5)-C(1)-H(1A)	110.2
O(6)-C(1)-H(1B)	110.2

O(5)-C(1)-H(1B)	110.2
H(1A)-C(1)-H(1B)	108.5
C(3)-C(2)-C(13)	121.67(19)
C(3)-C(2)-O(5)	128.11(19)
C(13)-C(2)-O(5)	110.22(17)
C(2)-C(3)-C(4)	118.08(19)
C(2)-C(3)-H(3A)	121.0
C(4)-C(3)-H(3A)	121.0
C(11)-C(4)-C(3)	119.77(18)
C(11)-C(4)-C(5)	123.63(18)
C(3)-C(4)-C(5)	116.57(17)
C(6)-C(5)-C(16)	111.84(16)
C(6)-C(5)-C(4)	109.81(16)
C(16)-C(5)-C(4)	112.60(16)
C(6)-C(5)-H(5A)	107.4
C(16)-C(5)-H(5A)	107.4
C(4)-C(5)-H(5A)	107.4
C(5)-C(6)-C(9)	113.65(16)
C(5)-C(6)-C(7)	117.85(18)
C(9)-C(6)-C(7)	101.09(16)
C(5)-C(6)-H(6A)	107.9
C(9)-C(6)-H(6A)	107.9
C(7)-C(6)-H(6A)	107.9
O(3)-C(7)-O(4)	121.5(2)
O(3)-C(7)-C(6)	130.1(2)
O(4)-C(7)-C(6)	108.39(18)
O(4)-C(8)-C(9)	102.77(17)
O(4)-C(8)-H(8A)	111.2
C(9)-C(8)-H(8A)	111.2
O(4)-C(8)-H(8B)	111.2
C(9)-C(8)-H(8B)	111.2
H(8A)-C(8)-H(8B)	109.1

C(10)-C(9)-C(6)	109.46(16)
C(10)-C(9)-C(8)	120.08(18)
C(6)-C(9)-C(8)	100.37(16)
C(10)-C(9)-H(9A)	108.8
C(6)-C(9)-H(9A)	108.8
C(8)-C(9)-H(9A)	108.8
O(2)-C(10)-C(9)	109.76(16)
O(2)-C(10)-C(11)	107.72(15)
C(9)-C(10)-C(11)	109.76(17)
O(2)-C(10)-H(10A)	109.9
C(9)-C(10)-H(10A)	109.9
C(11)-C(10)-H(10A)	109.9
C(4)-C(11)-C(12)	120.53(18)
C(4)-C(11)-C(10)	121.48(17)
C(12)-C(11)-C(10)	117.99(17)
C(13)-C(12)-C(11)	118.00(18)
C(13)-C(12)-H(12A)	121.0
C(11)-C(12)-H(12A)	121.0
C(12)-C(13)-O(6)	128.74(19)
C(12)-C(13)-C(2)	121.88(18)
O(6)-C(13)-C(2)	109.38(17)
O(1)-C(14)-O(2)	123.40(19)
O(1)-C(14)-C(15)	125.1(2)
O(2)-C(14)-C(15)	111.45(18)
C(14)-C(15)-H(15A)	109.5
C(14)-C(15)-H(15B)	109.5
H(15A)-C(15)-H(15B)	109.5
C(14)-C(15)-H(15C)	109.5
H(15A)-C(15)-H(15C)	109.5
H(15B)-C(15)-H(15C)	109.5
C(17)-C(16)-C(21)	119.87(19)
C(17)-C(16)-C(5)	120.10(18)

C(21)-C(16)-C(5)	120.03(18)
C(16)-C(17)-C(18)	122.3(2)
C(16)-C(17)-H(17A)	118.8
C(18)-C(17)-H(17A)	118.8
C(19)-C(18)-C(17)	116.4(2)
C(19)-C(18)-H(18A)	121.8
C(17)-C(18)-H(18A)	121.8
C(18)-C(19)-C(20)	121.9(2)
C(18)-C(19)-O(7)	128.3(2)
C(20)-C(19)-O(7)	109.74(19)
C(21)-C(20)-C(19)	122.43(19)
C(21)-C(20)-O(8)	128.36(19)
C(19)-C(20)-O(8)	109.16(18)
C(20)-C(21)-C(16)	117.05(19)
C(20)-C(21)-H(21A)	121.5
C(16)-C(21)-H(21A)	121.5
O(7)-C(22)-O(8)	107.02(16)
O(7)-C(22)-H(22A)	110.3
O(8)-C(22)-H(22A)	110.3
O(7)-C(22)-H(22B)	110.3
O(8)-C(22)-H(22B)	110.3
H(22A)-C(22)-H(22B)	108.6

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Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ( $\text{Å}^2 \times 10^3$ ) for z.

The anisotropic displacement factor exponent takes the form:

$$-2 \pi^2 [ h^2 a^*{}^2 U_{11} + \dots + 2 h k a^* b^* U_{12} ]$$


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U11	U22	U33	U23	U13	U12
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O(1)	50(1)	28(1)	35(1)	-2(1)	-3(1)	-12(1)
O(2)	26(1)	27(1)	32(1)	-2(1)	0(1)	-8(1)
O(3)	30(1)	63(1)	42(1)	-3(1)	-16(1)	-11(1)
O(4)	37(1)	54(1)	30(1)	2(1)	-13(1)	-10(1)
O(5)	30(1)	39(1)	32(1)	7(1)	-12(1)	-13(1)
O(6)	31(1)	40(1)	51(1)	10(1)	-21(1)	-17(1)
O(7)	28(1)	40(1)	34(1)	-5(1)	2(1)	-8(1)
O(8)	29(1)	35(1)	30(1)	-9(1)	-4(1)	-6(1)
C(1)	29(1)	38(1)	32(1)	4(1)	-10(1)	-12(1)
C(2)	28(1)	26(1)	24(1)	-4(1)	-7(1)	-7(1)
C(3)	21(1)	27(1)	29(1)	-6(1)	-4(1)	-8(1)
C(4)	22(1)	24(1)	25(1)	-6(1)	-5(1)	-5(1)
C(5)	22(1)	28(1)	26(1)	-5(1)	-5(1)	-7(1)
C(6)	25(1)	28(1)	25(1)	-6(1)	-6(1)	-5(1)
C(7)	31(1)	38(1)	31(1)	-7(1)	-9(1)	-2(1)
C(8)	34(1)	43(1)	30(1)	1(1)	-9(1)	-9(1)
C(9)	28(1)	28(1)	25(1)	-5(1)	-4(1)	-6(1)
C(10)	23(1)	26(1)	26(1)	-4(1)	-2(1)	-5(1)
C(11)	24(1)	23(1)	25(1)	-4(1)	-3(1)	-7(1)
C(12)	24(1)	29(1)	33(1)	-2(1)	-4(1)	-13(1)
C(13)	23(1)	28(1)	32(1)	-3(1)	-10(1)	-8(1)
C(14)	32(1)	30(1)	24(1)	2(1)	-9(1)	-12(1)
C(15)	30(1)	41(1)	38(1)	5(1)	-3(1)	-13(1)
C(16)	21(1)	28(1)	27(1)	-1(1)	-8(1)	-6(1)
C(17)	27(1)	32(1)	38(1)	-5(1)	-8(1)	-10(1)
C(18)	26(1)	38(1)	43(1)	-1(1)	-4(1)	-14(1)
C(19)	22(1)	36(1)	31(1)	-2(1)	-4(1)	-6(1)
C(20)	26(1)	28(1)	25(1)	-3(1)	-8(1)	-5(1)
C(21)	23(1)	32(1)	28(1)	-2(1)	-8(1)	-11(1)
C(22)	28(1)	35(1)	34(1)	-3(1)	-4(1)	-5(1)

Table 5. Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for z.

	x	y	z	U(eq)
H(1A)	7366	12436	-3527	40
H(1B)	7850	11329	-4611	40
H(3A)	12355	9865	-3325	32
H(5A)	13099	8980	-1152	32
H(6A)	13203	6038	-800	33
H(8A)	10903	6577	2445	45
H(8B)	11796	5273	1448	45
H(9A)	10954	8358	892	35
H(10A)	10331	6260	-145	34
H(12A)	7773	8528	-954	35
H(15A)	5741	6757	3090	59
H(15B)	5439	8100	2135	59
H(15C)	6196	8146	3156	59
H(17A)	15539	9056	-2558	40
H(18A)	17937	8222	-4300	44
H(21A)	14005	5900	-3024	33
H(22A)	19136	3732	-5232	43
H(22B)	18797	4178	-6530	43

Table 6. Torsion angles [deg] for z.

C(13)-O(6)-C(1)-O(5)	12.1(2)
C(2)-O(5)-C(1)-O(6)	-11.6(2)
C(1)-O(5)-C(2)-C(3)	-173.9(2)
C(1)-O(5)-C(2)-C(13)	6.9(2)

C(13)-C(2)-C(3)-C(4)	-2.4(3)
O(5)-C(2)-C(3)-C(4)	178.49(18)
C(2)-C(3)-C(4)-C(11)	0.7(3)
C(2)-C(3)-C(4)-C(5)	-177.41(17)
C(11)-C(4)-C(5)-C(6)	8.3(3)
C(3)-C(4)-C(5)-C(6)	-173.73(16)
C(11)-C(4)-C(5)-C(16)	133.60(19)
C(3)-C(4)-C(5)-C(16)	-48.4(2)
C(16)-C(5)-C(6)-C(9)	-166.27(16)
C(4)-C(5)-C(6)-C(9)	-40.5(2)
C(16)-C(5)-C(6)-C(7)	75.8(2)
C(4)-C(5)-C(6)-C(7)	-158.47(17)
C(8)-O(4)-C(7)-O(3)	179.0(2)
C(8)-O(4)-C(7)-C(6)	-1.4(2)
C(5)-C(6)-C(7)-O(3)	-29.6(3)
C(9)-C(6)-C(7)-O(3)	-154.0(2)
C(5)-C(6)-C(7)-O(4)	150.85(18)
C(9)-C(6)-C(7)-O(4)	26.4(2)
C(7)-O(4)-C(8)-C(9)	-24.1(2)
C(5)-C(6)-C(9)-C(10)	66.4(2)
C(7)-C(6)-C(9)-C(10)	-166.30(17)
C(5)-C(6)-C(9)-C(8)	-166.34(17)
C(7)-C(6)-C(9)-C(8)	-39.1(2)
O(4)-C(8)-C(9)-C(10)	158.86(17)
O(4)-C(8)-C(9)-C(6)	39.0(2)
C(14)-O(2)-C(10)-C(9)	-98.3(2)
C(14)-O(2)-C(10)-C(11)	142.21(17)
C(6)-C(9)-C(10)-O(2)	-172.29(15)
C(8)-C(9)-C(10)-O(2)	72.5(2)
C(6)-C(9)-C(10)-C(11)	-54.1(2)
C(8)-C(9)-C(10)-C(11)	-169.26(17)
C(3)-C(4)-C(11)-C(12)	1.7(3)

C(5)-C(4)-C(11)-C(12)	179.61(18)
C(3)-C(4)-C(11)-C(10)	-178.19(17)
C(5)-C(4)-C(11)-C(10)	-0.2(3)
O(2)-C(10)-C(11)-C(4)	142.99(18)
C(9)-C(10)-C(11)-C(4)	23.5(2)
O(2)-C(10)-C(11)-C(12)	-36.9(2)
C(9)-C(10)-C(11)-C(12)	-156.33(17)
C(4)-C(11)-C(12)-C(13)	-2.3(3)
C(10)-C(11)-C(12)-C(13)	177.53(18)
C(11)-C(12)-C(13)-O(6)	-178.86(19)
C(11)-C(12)-C(13)-C(2)	0.7(3)
C(1)-O(6)-C(13)-C(12)	171.6(2)
C(1)-O(6)-C(13)-C(2)	-8.0(2)
C(3)-C(2)-C(13)-C(12)	1.7(3)
O(5)-C(2)-C(13)-C(12)	-179.00(18)
C(3)-C(2)-C(13)-O(6)	-178.67(18)
O(5)-C(2)-C(13)-O(6)	0.6(2)
C(10)-O(2)-C(14)-O(1)	-0.4(3)
C(10)-O(2)-C(14)-C(15)	-179.82(17)
C(6)-C(5)-C(16)-C(17)	-116.6(2)
C(4)-C(5)-C(16)-C(17)	119.1(2)
C(6)-C(5)-C(16)-C(21)	62.3(2)
C(4)-C(5)-C(16)-C(21)	-61.9(2)
C(21)-C(16)-C(17)-C(18)	-0.3(3)
C(5)-C(16)-C(17)-C(18)	178.65(19)
C(16)-C(17)-C(18)-C(19)	-1.3(3)
C(17)-C(18)-C(19)-C(20)	0.8(3)
C(17)-C(18)-C(19)-O(7)	178.4(2)
C(22)-O(7)-C(19)-C(18)	165.6(2)
C(22)-O(7)-C(19)-C(20)	-16.5(2)
C(18)-C(19)-C(20)-C(21)	1.4(3)
O(7)-C(19)-C(20)-C(21)	-176.71(18)

C(18)-C(19)-C(20)-O(8)	178.94(19)
O(7)-C(19)-C(20)-O(8)	0.9(2)
C(22)-O(8)-C(20)-C(21)	-167.6(2)
C(22)-O(8)-C(20)-C(19)	15.0(2)
C(19)-C(20)-C(21)-C(16)	-2.9(3)
O(8)-C(20)-C(21)-C(16)	-179.95(18)
C(17)-C(16)-C(21)-C(20)	2.3(3)
C(5)-C(16)-C(21)-C(20)	-176.63(17)
C(19)-O(7)-C(22)-O(8)	25.7(2)
C(20)-O(8)-C(22)-O(7)	-25.3(2)

Symmetry transformations used to generate equivalent atoms:

Table 7. Hydrogen bonds for z [Å and deg.].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
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#### 4.2. Single Crystal X-ray Crystallography of compound **24** (CCDC 1430248)

Data intensity of **24** was collected using a Bruker SMART APEX II (Mo radiation). The X-ray condition of was 50 kV × 30 mA. Data collection and reduction were done by using the Bruker ApexII software package. The structure was solved by direct methods and refined by full-matrix least-squares on  $F^2$  with anisotropic displacement parameters for non-H atoms using SHELX-97. Hydrogen atoms were added at their geometrically ideal positions and refined isotropically. Crystal data for compound **24**:  $C_{20}H_{15}N_3O_6$ ,  $M = 393.35$ ,  $T = 173(2)$  K,  $\lambda = 0.71073$  Å, monoclinic, space group P2(1)/c,  $a = 8.8536(5)$  Å,  $b = 11.6840(7)$  Å,  $c = 16.8730(10)$  Å,  $V = 1724.97(17)$  Å<sup>3</sup>,  $z = 4$ ,  $d_{\text{calc}} = 1.515$  mg/m<sup>3</sup>, 19705 reflections measured, 3046 unique [ $R_{\text{int}} = 0.0523$ ].  $R_1 = 0.0354$ ,  $wR_2 = 0.0758$  ( $I > 2\sigma(I)$ , final).  $R_1 = 0.0528$ ,  $wR_2 = 0.0854$  (all data). GOF = 1.037, and 262 parameters.

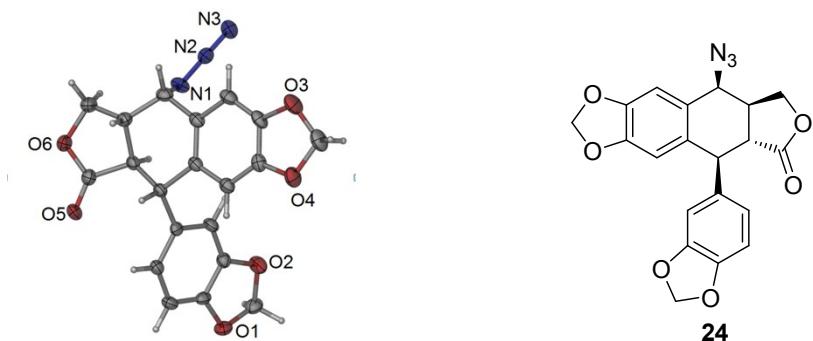


Table 1. Crystal data and structure refinement for z.

Identification code	z
Empirical formula	$C_{20}H_{15}N_3O_6$

Formula weight                    393.35  
 Temperature                      173(2) K  
 Wavelength                      0.71073 Å  
 Crystal system, space group    Monoclinic, P2(1)/c  
 Unit cell dimensions          a = 8.8536(5) Å alpha = 90 deg.  
                                      b = 11.6840(7) Å beta = 98.782(2) deg.  
                                      c = 16.8730(10) Å gamma = 90 deg.  
 Volume                            1724.97(17) Å<sup>3</sup>  
 Z, Calculated density          4, 1.515 Mg/m<sup>3</sup>  
 Absorption coefficient        0.114 mm<sup>-1</sup>  
 F(000)                         816  
 Crystal size                    0.27 x 0.26 x 0.15 mm  
 Theta range for data collection 2.13 to 25.00 deg.  
 Limiting indices               -10<=h<=10, -13<=k<=13, -20<=l<=20  
 Reflections collected / unique 19705 / 3046 [R(int) = 0.0523]  
 Completeness to theta = 25.00 100.0 %  
 Absorption correction        Semi-empirical from equivalents  
 Max. and min. transmission   0.9831 and 0.9698  
 Refinement method            Full-matrix least-squares on F<sup>2</sup>  
 Data / restraints / parameters 3046 / 1 / 262  
 Goodness-of-fit on F<sup>2</sup>       1.037  
 Final R indices [I>2sigma(I)] R1 = 0.0354, wR2 = 0.0758  
 R indices (all data)        R1 = 0.0528, wR2 = 0.0854  
 Largest diff. peak and hole   0.190 and -0.177 e.Å<sup>-3</sup>

Table 2. Atomic coordinates (x 10<sup>4</sup>) and equivalent isotropic displacement parameters (Å<sup>2</sup> x 10<sup>3</sup>) for z.  
 U(eq) is defined as one third of the trace of the orthogonalized Uij tensor.

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	x	y	z	U(eq)
O(1)	9825(1)	4690(1)	1354(1)	34(1)

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O(2)	11114(1)	6285(1)	1007(1)	40(1)
O(3)	7886(2)	12444(1)	689(1)	43(1)
O(4)	7531(2)	10987(1)	1573(1)	38(1)
O(5)	5302(2)	6030(1)	-1696(1)	37(1)
O(6)	5479(2)	7236(1)	-2695(1)	41(1)
N(1)	8682(2)	9696(1)	-1893(1)	35(1)
N(2)	9460(2)	10572(1)	-1914(1)	31(1)
N(3)	10268(2)	11316(1)	-1950(1)	41(1)
C(1)	6054(2)	8397(2)	-2790(1)	40(1)
C(2)	6121(2)	8903(2)	-1957(1)	30(1)
C(3)	7106(2)	9930(2)	-1702(1)	30(1)
C(4)	7148(2)	10126(1)	-810(1)	26(1)
C(5)	7478(2)	11244(1)	-529(1)	32(1)
C(6)	7604(2)	11426(1)	276(1)	30(1)
C(7)	8165(2)	12119(2)	1513(1)	43(1)
C(8)	7371(2)	10563(2)	799(1)	29(1)
C(9)	6979(2)	9477(1)	539(1)	27(1)
C(10)	6881(2)	9244(1)	-284(1)	24(1)
C(11)	6435(2)	8024(1)	-562(1)	24(1)
C(12)	6637(2)	7870(1)	-1437(1)	25(1)
C(13)	5757(2)	6925(2)	-1913(1)	30(1)
C(14)	7288(2)	7110(1)	-30(1)	22(1)
C(15)	8873(2)	7210(1)	206(1)	27(1)
C(16)	9598(2)	6347(1)	664(1)	26(1)
C(17)	8827(2)	5400(1)	878(1)	25(1)
C(18)	7291(2)	5281(2)	662(1)	30(1)
C(19)	6527(2)	6165(1)	205(1)	26(1)
C(20)	11307(2)	5166(2)	1338(1)	33(1)

Table 3. Bond lengths [Å] and angles [deg] for z.

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O(1)-C(17)            1.377(2)

O(1)-C(20)	1.429(2)
O(2)-C(16)	1.379(2)
O(2)-C(20)	1.423(2)
O(3)-C(6)	1.381(2)
O(3)-C(7)	1.426(2)
O(4)-C(8)	1.384(2)
O(4)-C(7)	1.446(2)
O(5)-C(13)	1.198(2)
O(6)-C(13)	1.355(2)
O(6)-C(1)	1.466(2)
N(1)-N(2)	1.237(2)
N(1)-C(3)	1.504(2)
N(2)-N(3)	1.134(2)
C(1)-C(2)	1.516(3)
C(1)-H(1A)	0.9900
C(1)-H(1B)	0.9900
C(2)-C(3)	1.507(2)
C(2)-C(12)	1.521(2)
C(2)-H(2A)	1.0000
C(3)-C(4)	1.516(2)
C(3)-H(3A)	1.0000
C(4)-C(10)	1.404(2)
C(4)-C(5)	1.405(2)
C(5)-C(6)	1.364(3)
C(5)-H(5A)	0.9500
C(6)-C(8)	1.376(2)
C(7)-H(7A)	0.9900
C(7)-H(7B)	0.9900
C(8)-C(9)	1.369(2)
C(9)-C(10)	1.405(2)
C(9)-H(9A)	0.9500
C(10)-C(11)	1.533(2)

C(11)-C(14)	1.520(2)
C(11)-C(12)	1.525(2)
C(11)-H(11A)	1.0000
C(12)-C(13)	1.509(2)
C(12)-H(12A)	1.0000
C(14)-C(19)	1.383(2)
C(14)-C(15)	1.404(2)
C(15)-C(16)	1.369(2)
C(15)-H(15A)	0.9500
C(16)-C(17)	1.377(2)
C(17)-C(18)	1.360(2)
C(18)-C(19)	1.400(2)
C(18)-H(18A)	0.9500
C(19)-H(19A)	0.9500
C(20)-H(20A)	0.9900
C(20)-H(20B)	0.9900
C(17)-O(1)-C(20)	105.29(13)
C(16)-O(2)-C(20)	105.32(13)
C(6)-O(3)-C(7)	104.70(14)
C(8)-O(4)-C(7)	104.32(14)
C(13)-O(6)-C(1)	109.85(13)
N(2)-N(1)-C(3)	113.22(14)
N(3)-N(2)-N(1)	174.14(19)
O(6)-C(1)-C(2)	103.00(14)
O(6)-C(1)-H(1A)	111.2
C(2)-C(1)-H(1A)	111.2
O(6)-C(1)-H(1B)	111.2
C(2)-C(1)-H(1B)	111.2
H(1A)-C(1)-H(1B)	109.1
C(3)-C(2)-C(1)	120.99(15)
C(3)-C(2)-C(12)	110.90(14)
C(1)-C(2)-C(12)	100.99(14)

C(3)-C(2)-H(2A)	107.7
C(1)-C(2)-H(2A)	107.7
C(12)-C(2)-H(2A)	107.7
N(1)-C(3)-C(2)	108.00(14)
N(1)-C(3)-C(4)	110.92(14)
C(2)-C(3)-C(4)	109.28(14)
N(1)-C(3)-H(3A)	109.5
C(2)-C(3)-H(3A)	109.5
C(4)-C(3)-H(3A)	109.5
C(10)-C(4)-C(5)	121.07(16)
C(10)-C(4)-C(3)	122.32(15)
C(5)-C(4)-C(3)	116.61(15)
C(6)-C(5)-C(4)	117.55(16)
C(6)-C(5)-H(5A)	121.2
C(4)-C(5)-H(5A)	121.2
C(5)-C(6)-C(8)	121.78(16)
C(5)-C(6)-O(3)	128.18(16)
C(8)-C(6)-O(3)	109.96(16)
O(3)-C(7)-O(4)	107.44(15)
O(3)-C(7)-H(7A)	110.2
O(4)-C(7)-H(7A)	110.2
O(3)-C(7)-H(7B)	110.2
O(4)-C(7)-H(7B)	110.2
H(7A)-C(7)-H(7B)	108.5
C(9)-C(8)-C(6)	122.00(17)
C(9)-C(8)-O(4)	128.20(16)
C(6)-C(8)-O(4)	109.74(15)
C(8)-C(9)-C(10)	118.07(16)
C(8)-C(9)-H(9A)	121.0
C(10)-C(9)-H(9A)	121.0
C(4)-C(10)-C(9)	119.43(15)
C(4)-C(10)-C(11)	123.39(15)

C(9)-C(10)-C(11)	117.16(14)
C(14)-C(11)-C(12)	111.59(13)
C(14)-C(11)-C(10)	113.04(13)
C(12)-C(11)-C(10)	110.04(13)
C(14)-C(11)-H(11A)	107.3
C(12)-C(11)-H(11A)	107.3
C(10)-C(11)-H(11A)	107.3
C(13)-C(12)-C(2)	100.88(13)
C(13)-C(12)-C(11)	118.28(14)
C(2)-C(12)-C(11)	113.32(14)
C(13)-C(12)-H(12A)	107.9
C(2)-C(12)-H(12A)	107.9
C(11)-C(12)-H(12A)	107.9
O(5)-C(13)-O(6)	120.76(16)
O(5)-C(13)-C(12)	130.32(16)
O(6)-C(13)-C(12)	108.91(14)
C(19)-C(14)-C(15)	119.68(15)
C(19)-C(14)-C(11)	120.71(15)
C(15)-C(14)-C(11)	119.59(14)
C(16)-C(15)-C(14)	117.47(15)
C(16)-C(15)-H(15A)	121.3
C(14)-C(15)-H(15A)	121.3
C(15)-C(16)-C(17)	122.09(16)
C(15)-C(16)-O(2)	128.16(15)
C(17)-C(16)-O(2)	109.69(14)
C(18)-C(17)-O(1)	128.37(15)
C(18)-C(17)-C(16)	121.80(16)
O(1)-C(17)-C(16)	109.74(15)
C(17)-C(18)-C(19)	116.86(16)
C(17)-C(18)-H(18A)	121.6
C(19)-C(18)-H(18A)	121.6
C(14)-C(19)-C(18)	122.08(16)

C(14)-C(19)-H(19A)	119.0
C(18)-C(19)-H(19A)	119.0
O(2)-C(20)-O(1)	107.99(14)
O(2)-C(20)-H(20A)	110.1
O(1)-C(20)-H(20A)	110.1
O(2)-C(20)-H(20B)	110.1
O(1)-C(20)-H(20B)	110.1
H(20A)-C(20)-H(20B)	108.4

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for z.

The anisotropic displacement factor exponent takes the form:

$$-2 \pi^2 [ h^2 a^*{}^2 U_{11} + \dots + 2 h k a^* b^* U_{12} ]$$

	U11	U22	U33	U23	U13	U12
O(1)	39(1)	27(1)	35(1)	10(1)	0(1)	3(1)
O(2)	27(1)	30(1)	58(1)	10(1)	-6(1)	1(1)
O(3)	50(1)	26(1)	54(1)	-11(1)	10(1)	-6(1)
O(4)	41(1)	35(1)	39(1)	-12(1)	8(1)	-4(1)
O(5)	44(1)	29(1)	35(1)	0(1)	0(1)	-13(1)
O(6)	56(1)	37(1)	27(1)	1(1)	-5(1)	-14(1)
N(1)	38(1)	29(1)	40(1)	3(1)	10(1)	-4(1)
N(2)	36(1)	32(1)	26(1)	0(1)	3(1)	-4(1)
N(3)	44(1)	39(1)	39(1)	-1(1)	3(1)	-11(1)
C(1)	50(1)	34(1)	33(1)	6(1)	-2(1)	-10(1)
C(2)	30(1)	29(1)	29(1)	7(1)	-1(1)	-1(1)
C(3)	30(1)	24(1)	35(1)	8(1)	2(1)	2(1)
C(4)	21(1)	21(1)	36(1)	2(1)	3(1)	2(1)
C(5)	28(1)	21(1)	46(1)	4(1)	6(1)	0(1)
C(6)	26(1)	19(1)	47(1)	-6(1)	7(1)	0(1)

C(7)	41(1)	31(1)	56(1)	-14(1)	6(1)	-3(1)
C(8)	22(1)	29(1)	35(1)	-6(1)	3(1)	3(1)
C(9)	22(1)	24(1)	34(1)	1(1)	3(1)	0(1)
C(10)	19(1)	20(1)	33(1)	1(1)	2(1)	1(1)
C(11)	21(1)	22(1)	28(1)	2(1)	1(1)	-2(1)
C(12)	22(1)	23(1)	29(1)	2(1)	-1(1)	-3(1)
C(13)	29(1)	32(1)	28(1)	1(1)	-1(1)	-2(1)
C(14)	26(1)	19(1)	21(1)	-3(1)	2(1)	0(1)
C(15)	26(1)	20(1)	34(1)	3(1)	3(1)	-4(1)
C(16)	24(1)	24(1)	28(1)	-1(1)	1(1)	1(1)
C(17)	36(1)	19(1)	20(1)	2(1)	4(1)	3(1)
C(18)	36(1)	22(1)	31(1)	3(1)	6(1)	-6(1)
C(19)	26(1)	25(1)	26(1)	-1(1)	3(1)	-4(1)
C(20)	38(1)	28(1)	33(1)	5(1)	0(1)	7(1)

Table 5. Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for z.

	x	y	z	U(eq)
H(1A)	5350	8837	-3190	48
H(1B)	7080	8380	-2955	48
H(2A)	5053	9092	-1879	36
H(3A)	6671	10620	-2005	36
H(5A)	7609	11851	-887	38
H(7A)	7672	12668	1840	51
H(7B)	9277	12112	1711	51
H(9A)	6779	8899	905	32
H(11A)	5323	7930	-532	29
H(12A)	7749	7746	-1455	30
H(15A)	9421	7852	53	32
H(18A)	6762	4629	814	35

H(19A)	5451	6113	51	31
H(20A)	11886	5201	1888	40
H(20B)	11884	4682	1007	40

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Table 6. Torsion angles [deg] for z.

C(3)-N(1)-N(2)-N(3)	-179(100)
C(13)-O(6)-C(1)-C(2)	-22.0(2)
O(6)-C(1)-C(2)-C(3)	159.99(15)
O(6)-C(1)-C(2)-C(12)	37.27(18)
N(2)-N(1)-C(3)-C(2)	161.48(15)
N(2)-N(1)-C(3)-C(4)	-78.79(18)
C(1)-C(2)-C(3)-N(1)	-50.2(2)
C(12)-C(2)-C(3)-N(1)	67.67(18)
C(1)-C(2)-C(3)-C(4)	-170.96(16)
C(12)-C(2)-C(3)-C(4)	-53.09(19)
N(1)-C(3)-C(4)-C(10)	-94.51(18)
C(2)-C(3)-C(4)-C(10)	24.5(2)
N(1)-C(3)-C(4)-C(5)	85.16(18)
C(2)-C(3)-C(4)-C(5)	-155.88(15)
C(10)-C(4)-C(5)-C(6)	3.2(3)
C(3)-C(4)-C(5)-C(6)	-176.49(15)
C(4)-C(5)-C(6)-C(8)	-2.1(3)
C(4)-C(5)-C(6)-O(3)	-178.45(16)
C(7)-O(3)-C(6)-C(5)	-170.22(18)
C(7)-O(3)-C(6)-C(8)	13.04(19)
C(6)-O(3)-C(7)-O(4)	-19.42(18)
C(8)-O(4)-C(7)-O(3)	18.54(18)
C(5)-C(6)-C(8)-C(9)	-0.9(3)
O(3)-C(6)-C(8)-C(9)	176.07(16)
C(5)-C(6)-C(8)-O(4)	-178.46(16)
O(3)-C(6)-C(8)-O(4)	-1.5(2)

C(7)-O(4)-C(8)-C(9)	172.07(18)
C(7)-O(4)-C(8)-C(6)	-10.58(19)
C(6)-C(8)-C(9)-C(10)	2.7(3)
O(4)-C(8)-C(9)-C(10)	179.78(16)
C(5)-C(4)-C(10)-C(9)	-1.4(2)
C(3)-C(4)-C(10)-C(9)	178.24(15)
C(5)-C(4)-C(10)-C(11)	176.73(15)
C(3)-C(4)-C(10)-C(11)	-3.6(3)
C(8)-C(9)-C(10)-C(4)	-1.5(2)
C(8)-C(9)-C(10)-C(11)	-179.78(15)
C(4)-C(10)-C(11)-C(14)	136.57(16)
C(9)-C(10)-C(11)-C(14)	-45.3(2)
C(4)-C(10)-C(11)-C(12)	11.1(2)
C(9)-C(10)-C(11)-C(12)	-170.77(14)
C(3)-C(2)-C(12)-C(13)	-167.67(14)
C(1)-C(2)-C(12)-C(13)	-38.21(17)
C(3)-C(2)-C(12)-C(11)	64.87(19)
C(1)-C(2)-C(12)-C(11)	-165.67(14)
C(14)-C(11)-C(12)-C(13)	75.50(19)
C(10)-C(11)-C(12)-C(13)	-158.17(14)
C(14)-C(11)-C(12)-C(2)	-166.77(14)
C(10)-C(11)-C(12)-C(2)	-40.44(19)
C(1)-O(6)-C(13)-O(5)	175.80(17)
C(1)-O(6)-C(13)-C(12)	-3.1(2)
C(2)-C(12)-C(13)-O(5)	-152.1(2)
C(11)-C(12)-C(13)-O(5)	-27.9(3)
C(2)-C(12)-C(13)-O(6)	26.68(18)
C(11)-C(12)-C(13)-O(6)	150.81(15)
C(12)-C(11)-C(14)-C(19)	-99.29(17)
C(10)-C(11)-C(14)-C(19)	136.04(15)
C(12)-C(11)-C(14)-C(15)	78.76(18)
C(10)-C(11)-C(14)-C(15)	-45.9(2)

C(19)-C(14)-C(15)-C(16)	0.1(2)
C(11)-C(14)-C(15)-C(16)	-177.96(15)
C(14)-C(15)-C(16)-C(17)	1.3(3)
C(14)-C(15)-C(16)-O(2)	-175.57(16)
C(20)-O(2)-C(16)-C(15)	-173.59(18)
C(20)-O(2)-C(16)-C(17)	9.26(18)
C(20)-O(1)-C(17)-C(18)	175.84(17)
C(20)-O(1)-C(17)-C(16)	-7.72(18)
C(15)-C(16)-C(17)-C(18)	-1.6(3)
O(2)-C(16)-C(17)-C(18)	175.75(15)
C(15)-C(16)-C(17)-O(1)	-178.32(15)
O(2)-C(16)-C(17)-O(1)	-0.96(19)
O(1)-C(17)-C(18)-C(19)	176.53(16)
C(16)-C(17)-C(18)-C(19)	0.5(2)
C(15)-C(14)-C(19)-C(18)	-1.2(2)
C(11)-C(14)-C(19)-C(18)	176.83(15)
C(17)-C(18)-C(19)-C(14)	0.9(2)
C(16)-O(2)-C(20)-O(1)	-13.95(18)
C(17)-O(1)-C(20)-O(2)	13.39(18)

Symmetry transformations used to generate equivalent atoms:

Table 7. Hydrogen bonds for z [Å and deg.].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
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## 5. References.

- 1、 R. B. Kothapalli, R. Niddana, R. Balamurugan, *Org. Lett.*, 2014, **16**, 1278-1281.