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Supporting Information

Sulfuration of C(sp2)-H Bond of Enaminones: A Protocol for The Synthesis of Thioether Using Elemental Sulfur as

Sulfurating Reagent †

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Electronic (UV-Visible) and	vibrational (Raman) experiments	.S2
NMR spectra of compounds	3	S3

Electronic (UV-Visible) and vibrational (Raman) experiments



Figure S1. (A) UV-Visible absorption spectra and (B) Raman spectra experiments

Notes: Beside electron spin resonance (ESR) spectroscopy, Chivers and co-workers reported that electronic (UV-Visible) and vibrational (IR and Raman) are also the useful spectroscopic techniques for the characterization of the trisulfur radical anion.¹ They observed that the bright blue S_3^- is associated with a visible absorption band in the range 595–620 nm in UV-visible absorption spectra ². We conducted the following experiments (Figure 1). The reaction mixture excluding enaminone was detected with the UV-vis. spectroscopy, but no absorption peak in the range of 595–620 nm was observed. Further more, in Raman spectroscopy, the characteristic peak at 535 cm⁻¹ associated with trisulfur radical anion^{1, 2b, 3} was not observed, either. These results from UV-visible absorption spectra and Raman spectra suggest that the S_3^- did not exist in our reaction system.

1 T. Chivers and P. J. W. Elder, Chem. Soc. Rev. 2013, 42, 5996.

^{2 (}a) T. Chivers, in *Homoatomic Rings, Chains and Macromolecules of Main-Group Elements*, ed. A. L. Rheingold, Elsevier, Amsterdam, 1977, ch. 22, pp. 499–537; (b) T. Chivers, *Nature* 1974, 252, 32.

^{3. (}a) J.-T. Yeon, J.-Y. Jang, J.-G. Han, J. Cho, K. T. Lee, N.-S. Choi, J. Electrochem. Soc. 2012, 159, 1308; (b) R. J. H. Clark, D.

G. Cobbold, Inorg. Chem. 1978, 17, 3169; (c) T. Chivers, C. Lau, Inorg. Chem. 1982, 21, 453;

NMR spectrum of compounds 3



¹³C NMR spectrum of compound 3aa



¹³C NMR spectrum of compound 3ba



¹³C NMR spectrum of compound 3ca



¹³C NMR spectrum of compound 3da



¹H NMR spectrum of compound 3ea



¹³C NMR spectrum of compound 3ea



¹³C NMR spectrum of compound 3fa





¹³C NMR spectrum of compound 3ga



C INFIR Spectrum of compound Si



¹³C NMR spectrum of compound 3ia



11.5 11.0 10.5 10.0 9.5 9.0 8.5 8.0 7.5 7.0 6.5 6.0 5.5 5.0 4.5 4.0 3.5 3.0 2.5 2.0 1.5 1.0 0.5 fl (ppm)

¹H NMR spectrum of compound 3ja



¹³C NMR spectrum of compound 3ja

¹³C NMR spectrum of compound 3la

¹H NMR spectrum of compound 3ma

¹³C NMR spectrum of compound 3ma

¹³C NMR spectrum of compound 3na

¹⁹F NMR spectrum of compound 3na

13.513.012.512.011.511.010.510.09.5 9.0 8.5 8.0 7.5 7.0 6.5 6.0 5.5 5.0 4.5 4.0 3.5 3.0 2.5 2.0 1.5 1.0 0.5 f1 (ppm)

¹H NMR spectrum of compound 3oa

¹³C NMR spectrum of compound 3oa

¹³C NMR spectrum of compound 3pa

¹³C NMR spectrum of compound 3qa

¹³C NMR spectrum of compound 3ra

¹H NMR spectrum of compound 3sa

¹³C NMR spectrum of compound 3sa

¹³C NMR spectrum of compound 3ta

¹³C NMR spectrum of compound 3ua

¹³C NMR spectrum of compound 3ab

¹H NMR spectrum of compound 3bb

¹³C NMR spectrum of compound 3bb

¹³C NMR spectrum of compound 3eb