Supporting Information

Amine-imine palladium catalysts for living polymerization of ethylene and copolymerization of ethylene with methyl acrylate: Incorporation of acrylate units into main chain and branch end

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| | 1 | C1.0.5H2O | | |
|---|---|---|--|--|
| Empirical formula | C ₃₀ H ₄₇ ClN ₂ Pd | C ₆₄ H ₆₂ BF ₂₄ N ₃ O _{0.5} Pd | | |
| Formula weight | 577.55 | 1454.38 | | |
| Crystal system | Monoclinic | Triclinic | | |
| Space group | $P2_1/n$ | <i>P</i> -1 | | |
| a (Å) | 15.919(2) | 12.1356(2) | | |
| b (Å) | 11.3592(16) | 13.0597(2) | | |
| c (Å) | 17.894(3) | 22.3007(3) | | |
| α (deg) | 90 | 103.2350(10) | | |
| β (deg) | 114.701(2) | 98.4960(10) | | |
| γ (deg) | 90 | 101.5750(10) | | |
| Volume (Å ³) | 2939.7(7) | 3300.29(9) | | |
| Z | 4 | 2 | | |
| $D(calc.) (g/cm^3)$ | 1.305 | 1.464 | | |
| F(000) | 1216 | 1476 | | |
| θ range (deg) | 2.19 to 27.05 | 1.65 to 24.67 | | |
| Index ranges | $-20 \le h \le 16$ | $-14 \le h \le 14$ | | |
| | $-14 \le k \le 12$ | $-15 \le k \le 15$ | | |
| | $-22 \le l \le 15$ | $-26 \le l \le 26$ | | |
| Reflections collected/unique | 14657/6309 | 48203/10990 | | |
| | $(R_{int} = 0.0261)$ | $(R_{int} = 0.0572)$ | | |
| Data completeness | 97.6 % | 97.8% | | |
| Data/restraints/parameters | 6309/0/339 | 10990/35/909 | | |
| Goodness-of-fit on F^2 | 1.021 | 1.069 | | |
| Final R indices $[I \ge 2\sigma(I)]$ | $R_1 = 0.0308$ | $R_1 = 0.0706$ | | |
| | $wR_2 = 0.0716$ | $wR_2 = 0.1780$ | | |
| R indices (all data) | $R_1 = 0.0506$ | $R_1 = 0.0722$ | | |
| | $wR_2 = 0.0807$ | $wR_2 = 0.1794$ | | |
| Largest diff. peak and hole (eA ⁻³) | 0.498/-0.425 | 2.202/-1.329 | | |

Table S1. Crystal data and structure refinement for palladium complexes



Figure S1. ¹H NMR spectrum of palladium complex 1



Figure S2. ¹H NMR spectrum of palladium complex 2



Figure S3. ¹H NMR spectrum of palladium complex C1



Figure S4. ¹H NMR spectrum of palladium complex C2



Figure S5. Influence of reaction temperature on ethylene polymerization activity with α -diimine palladium C3. (Conditions: 10 µmol Pd catalyst, 20 atm ethylene pressure 58 mL toluene and 2 mL CH₂Cl₂, 4 h.)



Figure S6. ¹³C NMR spectrum of the PE obtained by C1 (entry 2 in Table 1)



Figure S7. ¹³C NMR spectrum of the PE obtained by C2 (entry 5 in Table 1)



Figure S8. ¹³C NMR spectrum of the PE obtained by C3 (entry 12 in Table 1)

 Table S2. Branching distribution of PE samples obtained by different palladium

 catalysts

| Entry | Cat | Branching distribution (/1000C) | | | | | BD | | |
|-------|-----------|---------------------------------|------|-----|------|-----|------|-----------------|----------|
| | | Me | Et | Pr | Bu | Am | Lg | ^s Bu | (/1000C) |
| 2 | C1 | 53.9 | 14.6 | 3.7 | 8.8 | 2.4 | 16.1 | 2.7 | 102.2 |
| 5 | C2 | 53.1 | 17.8 | 4.2 | 12.0 | 6.6 | 20.1 | 8.2 | 122.0 |
| 12 | C3 | 54.2 | 13.8 | 4.0 | 9.7 | 5.0 | 18.4 | 5.5 | 110.6 |

Branching density is determined by ¹³C NMR spectroscopy in number of branches per 1000C according to previous reports (*Macromolecules*, **1999**, 32, 1620, *Science* **1999**, 283, 2059, *ACS Catal.* **2015**, *5*, 456).



Synthesis of cationic MA chelate palladium complexes

To a solution of complex 1 (0.44 g, 0.76 mmol) in dry CH_2Cl_2 (20 mL) was added NaBArF (0.70 g, 0.79 mmol) and MA (2 mL) under nitrogen atmosphere, and the reaction mixture was stirred for 24 h at room temperature. Sodium chloride was removed from the reaction mixture via filtration, yielding a clear solution. The CH_2Cl_2 was removed in vacuum, and the product was precipitated by 30 mL hexane. The resulting solid was filtered through Celite, washed with 3 × 10 mL hexane and dried in vacuum.



Figure S10. ¹H NMR spectrum of MA chelate amine-imine palladium complexes

 $(N \cap NH \text{ is amine-imine ligand}).$