

Supporting Information

Natural kaolin derived stable SBA-15 as support for Fe/BiOCl: A novel
and efficient fenton-like catalyst for degradation of 2-nitrophenol

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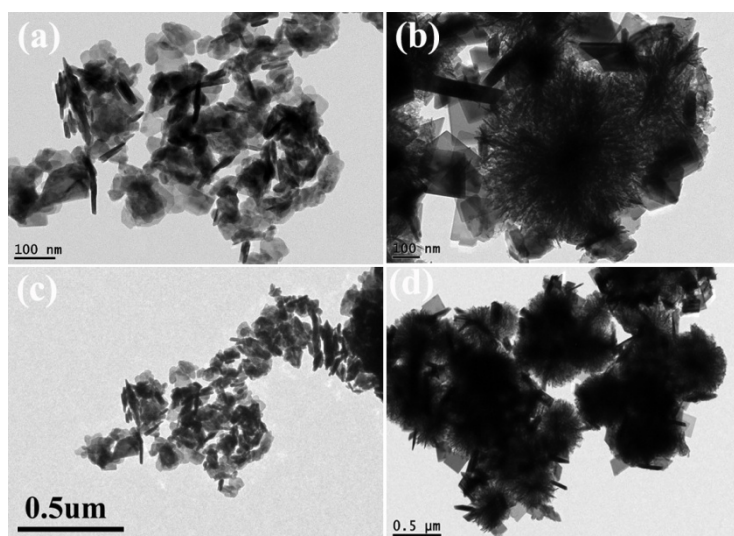


Figure S1 TEM images of pure BiOCl (a and c) and Fe/BiOCl (b and d)

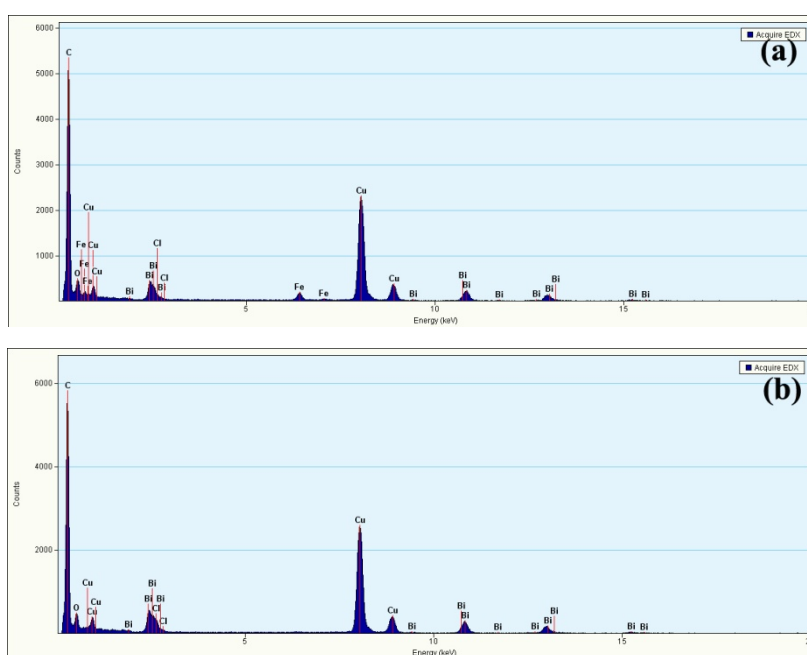


Figure S2 EDS spectrum of flower-like Fe/BiOCl (a) and nanosheet Fe/BiOCl (b) in the catalyst

0.5Fe/BiOCl-FCSBA-15(0.6)

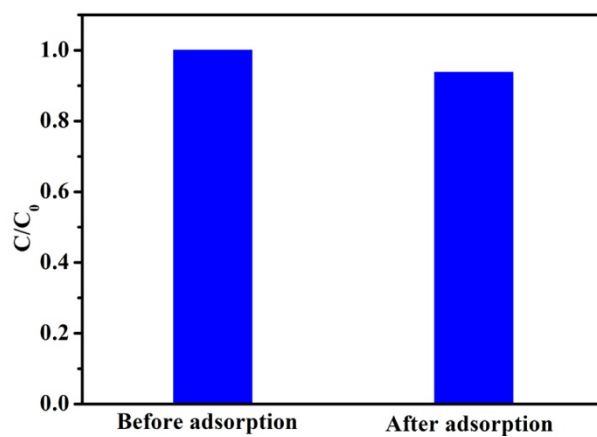


Figure S3 The adsorption behavior of catalyst 0.5Fe/BiOCl-FCSBA-15(0.6) on 2-nitrophenol

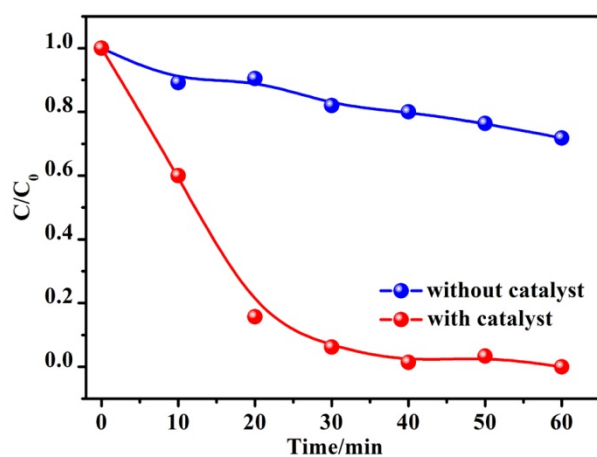


Figure S4 The degradation efficiency of 2-nitrophenol with or without catalyst ($T=60^{\circ}\text{C}$, $\text{pH}=4$, H_2O_2 dosage=800 mg/L, Fe/Bi molar ratio=0.5, Bi/Si molar ratio=0.6)

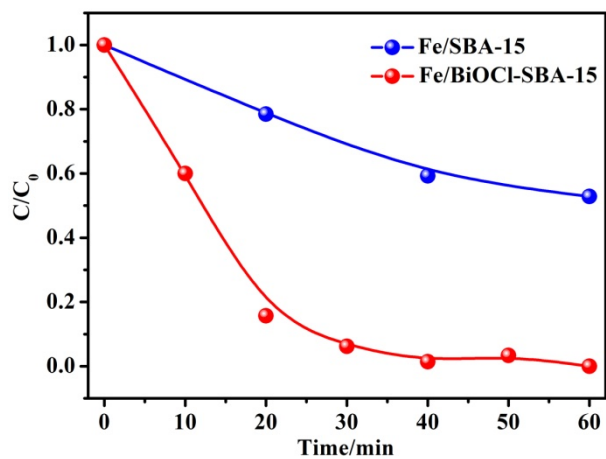


Figure S5 The degradation efficiency of 2-nitrophenol over catalysts Fe/FCSBA-15 and Fe/BiOCl-FCSBA-15 ($T=60^{\circ}\text{C}$, $\text{pH}=4$, H_2O_2 dosage=800 mg/L)

The synthesis procedure of Fe/FCSBA-15 was as follows: 2.02 g $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ was dissolved in 20 mL deionized water, subsequently a certain amount of FCSBA-15 was added with stirring for 1 h. The mixture was then transferred into a 100 mL Teflon-lined stainless steel autoclave, heated up to 110°C and kept at this temperature for 8 h. The resulting precipitate was collected by centrifugation, then washed with ethanol and deionized water for several times, then dried at 60°C in vacuum and treated at 400°C for 3 h.

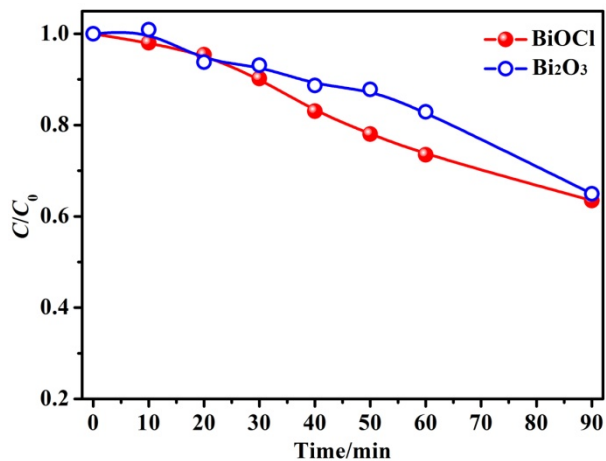


Figure S6 The degradation efficiency of 2-nitrophenol over BiOCl and Bi₂O₃ ($T=60^{\circ}\text{C}$, $\text{pH}=4$, H_2O_2 dosage=800 mg/L)

The synthesis procedure of BiOCl was as follows: 4 mmol Bi(NO₃)₃·5H₂O was dissolved in 190 mL of glycerol, denoted as solution A. Then, 4 mmol KCl was dissolved in 10 mL of deionized water (solution B), which was subsequently poured into solution A. The mixture was transferred into a 200 mL Teflon-lined stainless steel autoclave, heated to 150 °C and kept at this temperature for 8 h. The resulting precipitate was collected by centrifugation, then washed with ethanol and deionized water for several times, and dried at 60 °C in vacuum to obtain BiOCl powder.

In a typical synthesis procedure of Bi₂O₃, 0.364 g Bi(NO₃)₃·5H₂O was dissolved in 10 mL of 1M dilute HNO₃. Then 0.16 g urea, a certain amount of ethylene glycol and subsequently 0.6 g PVP were added with stirring. The obtained mixture was transferred into a Teflon-lined stainless steel autoclave and kept at 150 °C for 3 h. The resulting precipitate was collected by centrifugation, then washed with ethanol and deionized water for several times, and dried at 60 °C for 12 h. After treatment at 600 °C for 3 h, Bi₂O₃ powder was obtained.

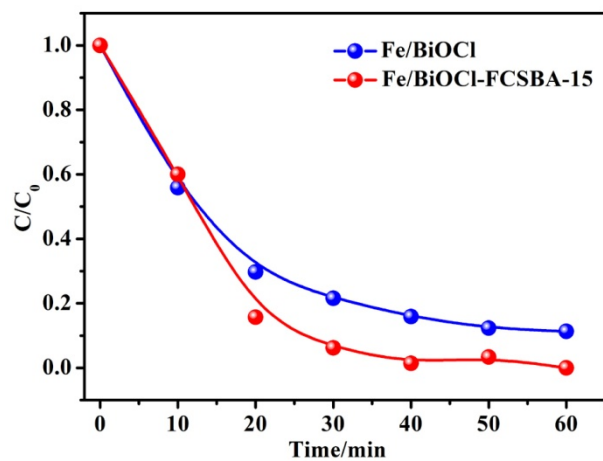


Figure S7 The degradation efficiency of 2-nitrophenol over catalysts Fe/BiOCl with or without matrix FCSBA-15 ($T=60^{\circ}\text{C}$, $\text{pH}=4$, H_2O_2 dosage=800 mg/L, Fe/Bi molar ratio=0.5, Bi/Si molar ratio=0.6)

The synthesis process of Fe/BiOCl was the same as that of Fe/BiOCl-FCSBA-15, except the addition of the matrix FCSBA-15.

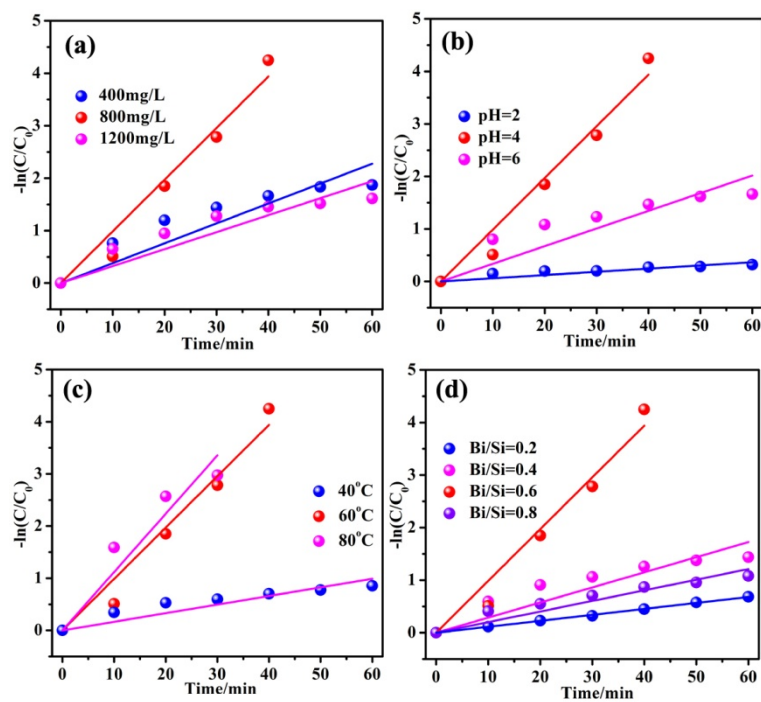


Figure S8 Effect of various variables on reaction rate constants in the degradation of 2-nitrophenol over catalyst Fe/BiOCl-FCSBA-15. (a) Effect of H_2O_2 dosage ($T=60^\circ C$, $pH=4$, Fe/Bi molar ratio=0.5, Bi/Si molar ratio=0.6); (b) Effect of pH value ($T=60^\circ C$, H_2O_2 dosage=800 mg/L, Fe/Bi molar ratio=0.5, Bi/Si molar ratio=0.6); (c) Effect of oxidation temperature ($pH=4$, H_2O_2 dosage=800 mg, Fe/Bi molar ratio=0.5, Bi/Si molar ratio=0.6); (d) Effect of Bi/Si molar ratio ($T=60^\circ C$, $pH=4$, H_2O_2 dosage=800 mg/L, Fe/Bi molar ratio=0.5)

Table S1

Reaction rate constants in the degradation of 2-nitrophenol over catalyst Fe/BiOCl-FCSBA-15

Parameters		Reaction rate constants (10^{-2} min^{-1})
H ₂ O ₂ dosage (mg/L)	400	3.80
	800	9.85
	1200	3.24
pH values	2	0.61
	4	9.85
	6	3.36
Temperature (°C)	40	1.65
	60	9.85
	80	11.18
Fe/Bi molar ratio	0.2	2.74
	0.5	9.85
	1.0	4.30
Bi/Si molar ratio	0.2	1.13
	0.4	2.02
	0.6	9.85
	0.8	2.87

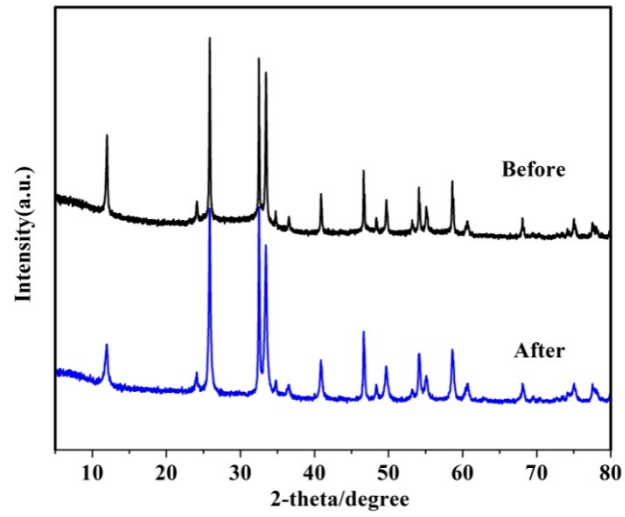


Figure S9 XRD patterns of catalyst 0.5Fe/BiOCl-FCSBA-15(0.6) before and after Fenton-like reaction ($T=60^{\circ}\text{C}$, $\text{pH}=4$, H_2O_2 dosage=800 mg/L)

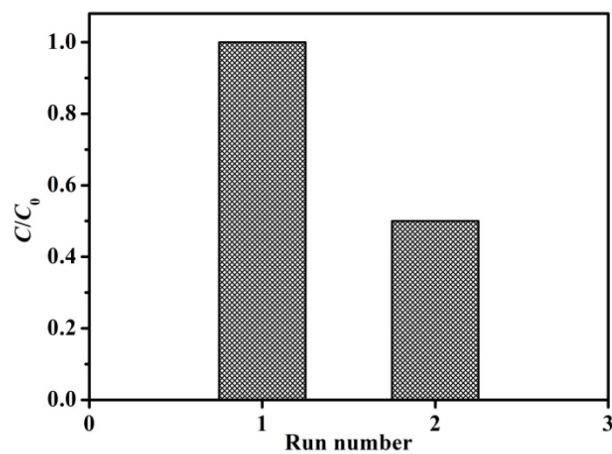


Figure S10 The reusability of catalyst 0.5Fe/BiOCl-FCSBA-15(0.6) ($T=60^{\circ}\text{C}$, $\text{pH}=4$, H_2O_2 dosage=800 mg/L)