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Supporting information for

Three Dimensional Carbon Nanotube/Nickel Hydroxides Gels for Advanced Supercapacitors

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**Fig. S1** FESEM images of gel composites in smaller magnification: (a) Ni(OH)<sub>2</sub> at nickel foam (b) CNTs hydrogel (c) CNTs xerogel (d) NiC hydrogel (NiC-hydro) composite and (e) NiC xerogel (NiC-xero) composite



Fig. S2 Cross sectional area image of (a) NiC-hydro and (b) NiC-xero .



**Fig. S3** FESEM images of the Ni(OH)<sub>2</sub>/SWNT xerogel (NiC-xero) composites having Ni(OH)<sub>2</sub> deposited at difference charges: (a) 1.11C (b) 3.33C (c) 5.55C (d) 7.77C.



**Fig. S4** TEM images of (a) NiC-hydro (b) NiC- xero, SAED pattern of the (c) NiC-hydro and (d) NiC-xero



**Fig. S5** Electrochemical performance of xerogel electrodes having different Ni(OH)<sub>2</sub> loadings: (a) CV curves of difference Ni(OH)<sub>2</sub>/SNWT xerogel electrodes (b) specific capacitances of the Ni(OH)<sub>2</sub>/SNWT xerogel electrodes calculated from different scan rates (c) discharge profiles of the Ni(OH)<sub>2</sub>/SNWT xerogel electrodes and (d) specific capacitances of the Ni(OH)<sub>2</sub>/SNWT xerogel electrodes calculated from different galvanic discharge rates.



**Fig. S6** Comparison o areal capacitance of (a) different Ni(OH)<sub>2</sub>/SWNT gel composites and (b) Ni(OH)<sub>2</sub>/SWNTs xerogel composites synthesize through different accumulated during synthesis.



**Fig. S7** Electrochemical impedance spectroscopy: (a) fitting results of NiC-xero and NiChydro and (b) equivalent circuit model.