

Electronic Supplementary Information (ESI) for RSC Advances.

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Enhanced formaldehyde sensing properties of SnO₂ nanorods coupled with Zn₂SnO₄

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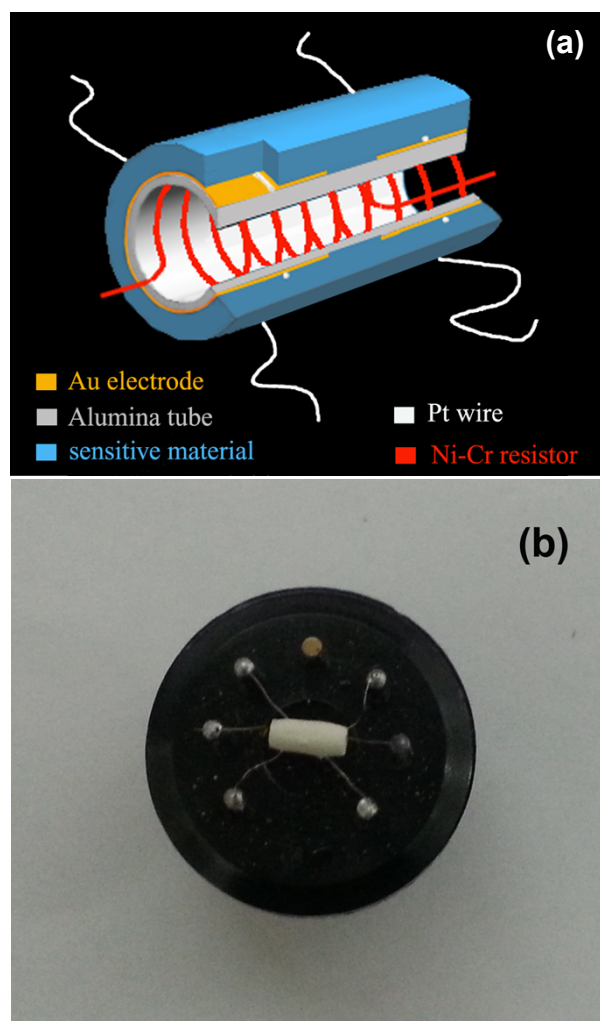


Figure S1.

(a) Sketch of the structure of a typical $\text{SnO}_2\text{-Zn}_2\text{SnO}_4$ nanocomposites gas sensor;

(b) Photograph of the gas sensor.

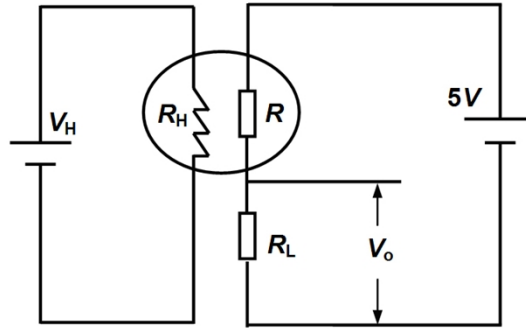


Figure S2. Schematic diagram of testing principle for SnO₂-Zn₂SnO₄ nanocomposites gas sensors. V_H is heating voltage and R_H is heating resistance.

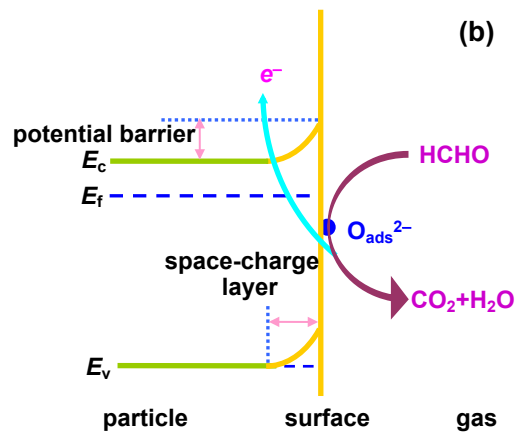
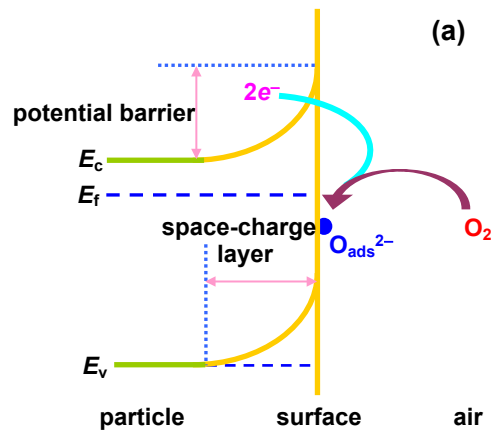
According to [Figure S2](#), the electrical resistance of sensor can be obtained as following:

$$R = \frac{5 - V_o}{V_o} \cdot R_L$$

where R is the resistance of the sensor, R_L is a constant load resistance unchanged with the surrounding gas partial pressure, V_o is the sensor export voltage. The gas response β was defined as the ratio of the electrical resistance in air (R_o) to that in gases (R_g):

$$R_o = \frac{5 - (V_o)_{\text{air}}}{(V_o)_{\text{air}}} \cdot R_L, \quad R_g = \frac{5 - (V_o)_g}{(V_o)_g} \cdot R_L, \quad \beta = \frac{R_o}{R_g}$$

where $(V_o)_{\text{air}}$ is the export voltage in air, and $(V_o)_{\text{gas}}$ is in gases.



Scheme S3.

A schematic diagram of the proposed reaction mechanism of $SnO_2-Zn_2SnO_4$ nanocomposites based sensor to formaldehyde. (a) in air, (b) in formaldehyde.

Table S1 The lattice constants and average grain size of SnO₂ and SnZn_x composites.

Samples	SnO ₂		Zn ₂ SnO ₄	Average grain size [§] (nm)
	<i>a</i> (Å)	<i>c</i> (Å)	<i>a</i> (Å)	
SnO ₂	4.740	3.184	–	36.4
SnZn ₅	4.745	3.186	–	34.2
SnZn ₁₅	4.758	3.191	8.618	16.2
SnZn ₂₅	4.759	3.193	8.610	10.7
SnZn ₃₅	4.765	3.197	8.607	9.2
SnZn ₄₅	4.767	3.201	8.601	11.1

SnO₂ The average lattice constant was calculated from (110), (101), (200), (111), (211), and (220) reflections. JCPDS 41-1445, System: Tetragonal, Lattice: Body-centered, Space group: *P42/mnm*(136), *a* = 4.738 Å, *c* = 3.187 Å.

Zn₂SnO₄ The average lattice constant was calculated from (311) and (440) reflections. JCPDS 24-1470, System: Cubic, Space group: *Fd $\bar{3}m$* (227), *a* = 8.657 Å.

[§] The crystallite size *D* is deduced from the (101) peak width using Scherrer's equation.