

Supporting Information

Controllable synthesis of InTaO₄ catalysts with different morphologies using a versatile sol precursor for photocatalytic evolution of H₂

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Table S1. Brunauer-Emmett-Teller (BET) surface areas of the as-synthesized catalysts.

Samples	BET (m ² /g)
InTaO ₄ nanoparticles	3.1405
SiO ₂ @InTaO ₄ core-shell nanospheres	2.9093
InTaO ₄ nanofibers	4.1579

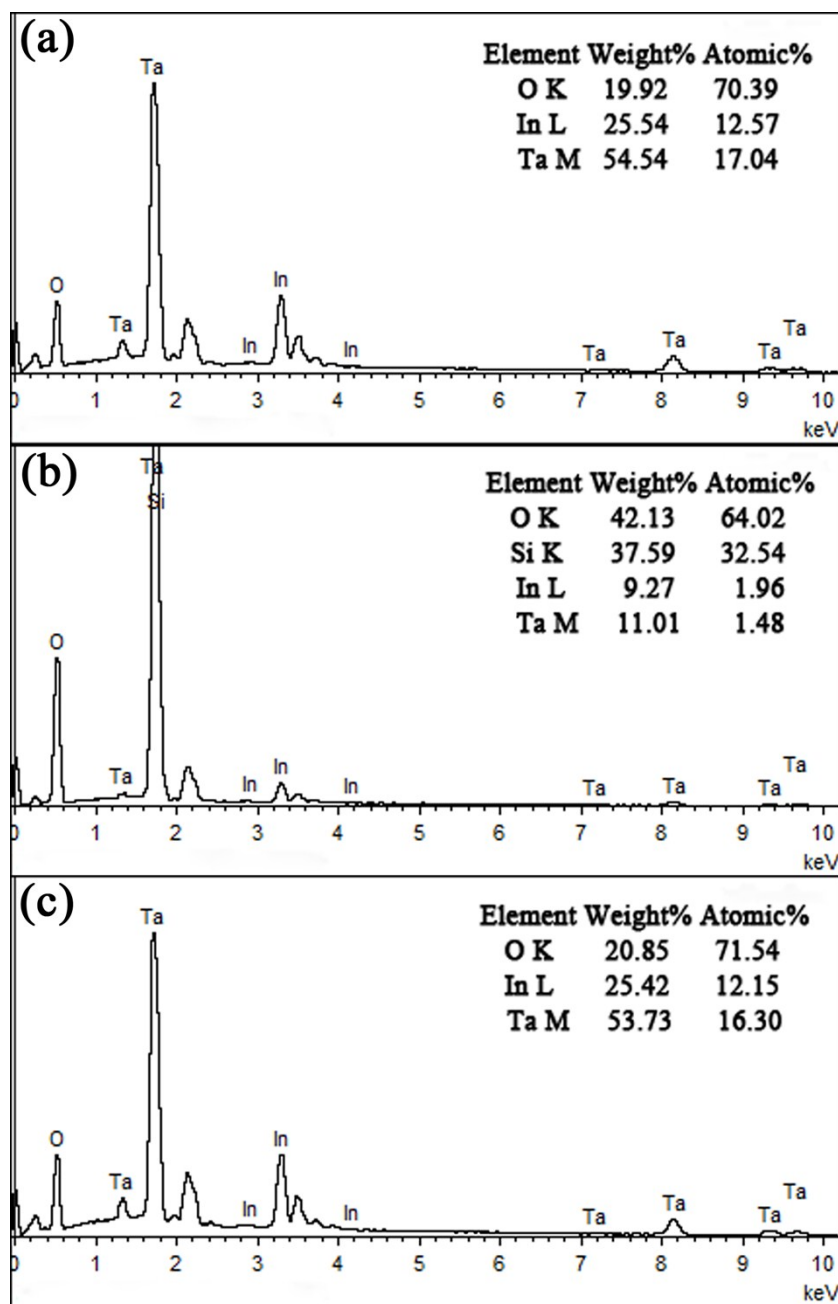


Fig. S1. EDS spectra of (a) InTaO₄ nanoparticles, (b) SiO₂@InTaO₄ core-shell nanospheres, and (c) InTaO₄ nanofibers.

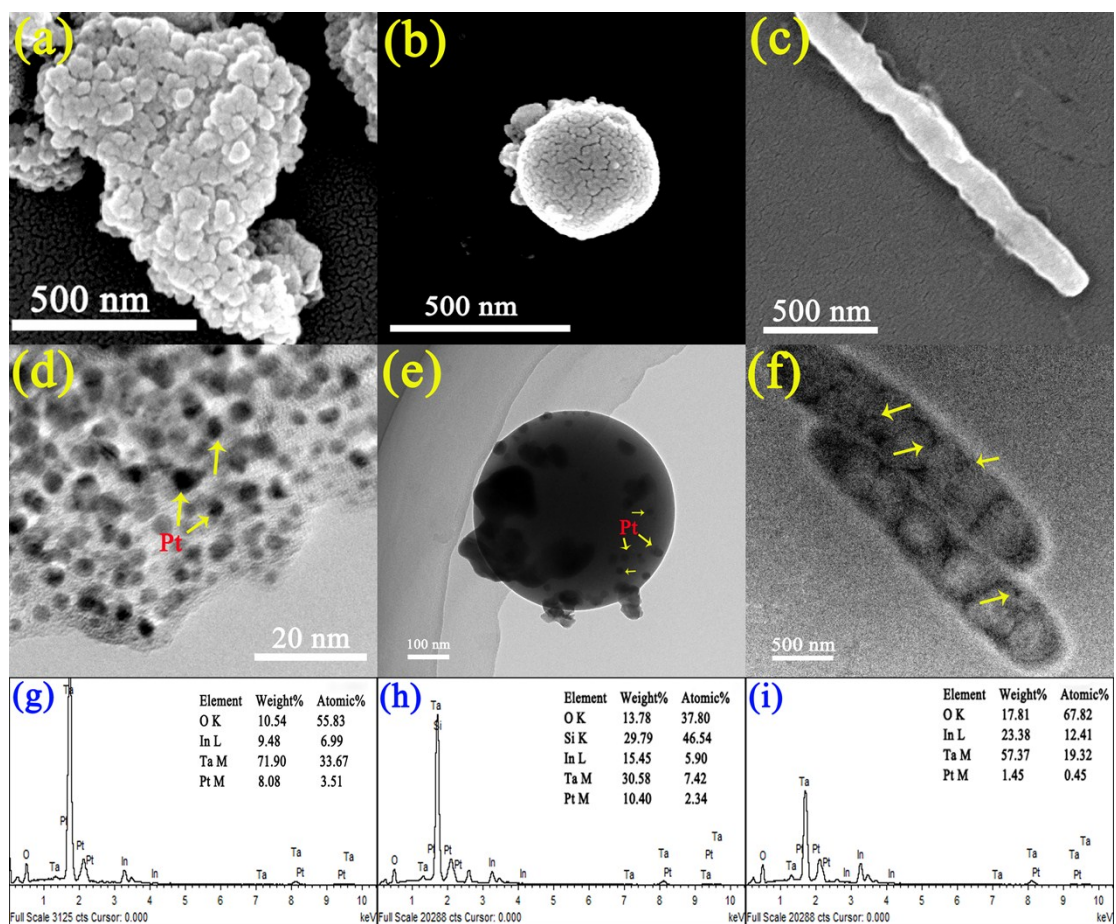


Fig. S2. SEM images of InTaO₄ nanoparticles (a), SiO₂@InTaO₄ nanospheres (b), and InTaO₄ nanofibers (c) after photodeposition of Pt; TEM images of InTaO₄ nanoparticles (d), SiO₂@InTaO₄ nanospheres (e), and InTaO₄ nanofibers (f) after photo photodeposition of Pt; EDS spectra of InTaO₄ nanoparticles (g), SiO₂@InTaO₄ nanospheres (h), and InTaO₄ nanofibers (i) after photo photodeposition of Pt.

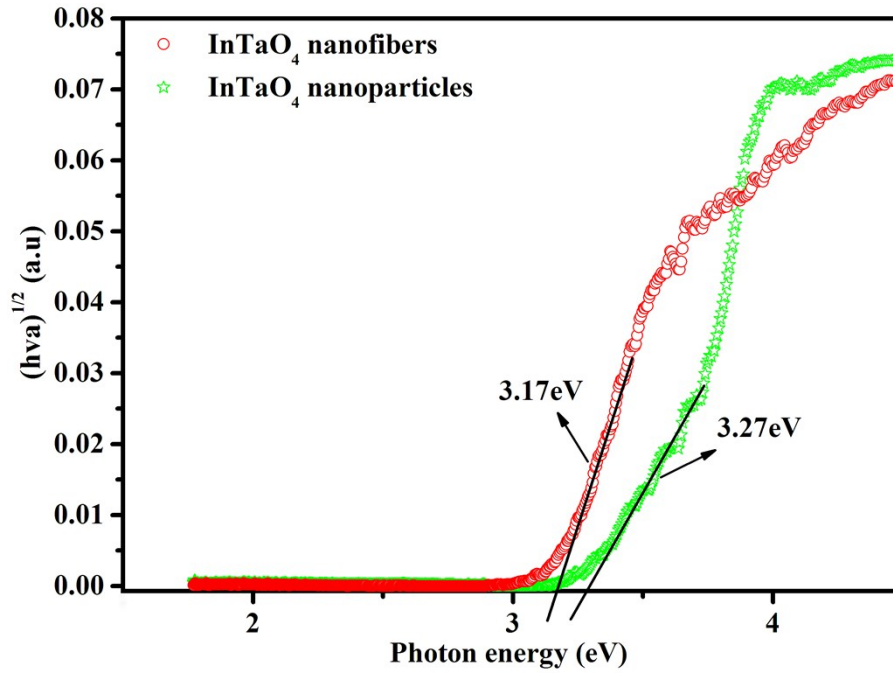


Fig. S3. The band gaps of InTaO₄ nanofibers and InTaO₄ nanoparticles were determined to be 3.17 eV and 3.27 eV, respectively.

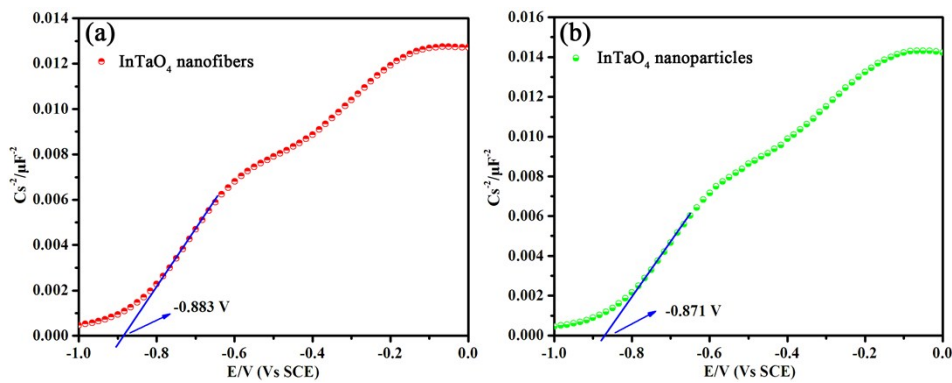


Fig. S4. Mott Schottky plots of (a) InTaO₄ nanofibers and (b) InTaO₄ nanoparticles electrodes using 0.5M Na₂SO₄ as electrolyte at frequency of 1000 Hz.