

Hydrogen Bond Effect on the Photophysical Properties of 2-Ureido-4[1*H*]-pyrimidinone Quadruple Hydrogen Bonded Systems

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Table S1. Summary of steady-state absorption and fluorescence emission measurements.

compound	absorption bands λ_{\max} / nm [$\epsilon \times 10^4 / \text{cm}^{-1} \cdot \text{M}^{-1}$]	λ_{ex} /nm	λ_{emis} /nm
AnUP-DCM	391 [4.59], 370 [4.74], 352 [2.91], 335 [1.34], 259 [62.06]	350	392, 415, 441, 470
AnUP-DMSO	393 [9.11], 372 [9.63], 354 [6.02], 337 [3.05], 287 [12.29], 261 [116.45]	350	395, 418, 443,469
NaUP-DCM	268 [13.88]	280	319, 326, 350, 334
NaUP-DMSO	288 [24.93]	280	319, 335, 350
UPNa-DCM	288 [1.35], 256 [4.08]	280	340, 353
UPNa-DMSO	285 [0.68]	280	358

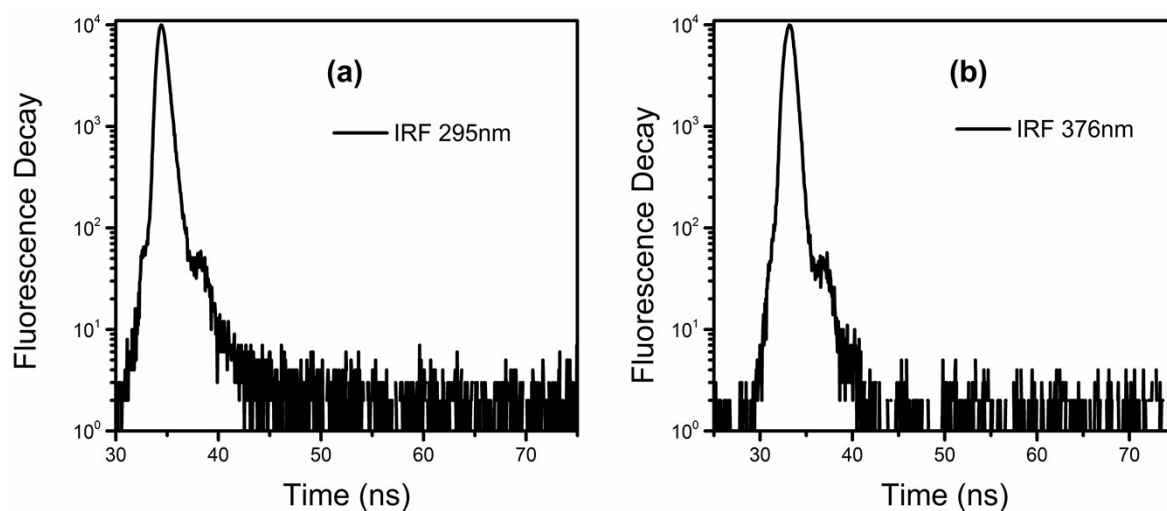


Figure S1. The instrument response functions (IRFs)

Table S2: Calculated important bond lengths (in Å) and bond angles (in degree) for the fully optimized structure of AnUP-dimer (in DCM) and AnUP-DMSO complex (in DMSO) in ground state (S_0) and first excited state (S_1). The corresponding X-ray data¹ are listed in parentheses.

	AnUP(DCM) ^[a]	AnUP(DMSO) ^[a]	AnUP(DCM) ^[b]	AnUP(DMSO) ^[b]
N ₁ -H ₁	1.03 (0.86)	1.02	1.04	1.03
N ₂ -H ₂	1.04 (0.86)	1.04	1.04	1.04
N ₁ '-H ₁ '	1.03 (0.86)	-	1.04	-
N ₂ '-H ₂ '	1.04 (0.86)	-	1.04	-
N ₄ -H ₃	1.03 (0.86)	1.03	1.03	1.03
N ₄ '-H ₃ '	1.03 (0.86)	-	1.03	-
O ₁ '...H ₁	1.71 (1.90)	1.99	1.71	1.99
N ₃ '...H ₂	1.92 (2.12)	-	1.92	-
N ₃ '...H ₂ '	1.92 (2.12)	-	1.92	-
O ₁ '...H ₁ '/H ₂	1.71 (1.90)	1.68	1.72	1.68
O ₂ '...H ₃	1.72 (1.91)	1.83	1.72	1.83
O ₂ '...H ₃ '	1.72 (1.90)	-	1.72	-
O ₁ '...H ₁ -N ₁	173.19 (168.75)	141.08	173.25	141.06
N ₃ '...H ₂ -N ₂	179.70 (171.39)	-	179.79	-
N ₃ '...H ₂ '-N ₂ '	179.71 (171.99)	-	179.62	-
O ₁ '...H ₁ '/H ₂ -N ₁ '/N ₂	173.19 (161.56)	170.67	173.22	170.68
O ₂ '...H ₃ -N ₄	136.60 (133.01)	132.68	136.63	132.77
O ₂ '...H ₃ '-N ₄ '	136.60 (133.68)	-	136.58	-
C ₁ -N ₁ -C ₂ -N ₂	-179.77 (-176.79)	-179.71	-179.87	-179.69
O ₂ -C ₂ -N ₂ -C ₃	0.09 (-8.06)	-9.78	0.72	-9.93
H ₃ -N ₄ -C ₄ -C ₆	-179.38 (177.42)	179.71	-179.65	179.82
C ₅ -C ₆ -C ₇ -C ₈	126.57 (130.37)	-57.78	125.96	-58.08
C ₅ -C ₆ -C ₈ -C ₉	-161.30 (-158.78)	-97.56	-160.44	-97.00
C ₁ '-N ₁ '-C ₂ '-N ₂ '	179.77 (-178.33)	-	179.78	-
O ₂ '-C ₂ '-N ₂ '-C ₃ '	-0.09 (-0.85)	-	0.14	-
H ₃ '-N ₄ '-C ₄ '-C ₆ '	179.39 (-179.25)	-	179.57	-
C ₅ '-C ₆ '-C ₇ '-C ₈ '	-126.57 (-128.67)	-	-126.55	-
C ₅ '-C ₆ '-C ₈ '-C ₉ '	161.31 (161.46)	-	161.20	-

[a] The geometric parameters of the ground state (S_0). [b] The geometric parameters of the first excited state (S_1)

Table S3: Calculated important bond lengths (in Å) and bond angles (in degree) for the fully optimized structure of NaUP-dimer (in DCM) and NaUP-DMSO complex (in DMSO) in ground state (S_0) and first excited state (S_1).

	NaUP(DCM) ^[a]	NaUP(DMSO) ^[a]	NaUP(DCM) ^[b]	NaUP(DMSO) ^[b]
N ₁ -H ₁	1.04	1.03	1.04	1.03
N ₂ -H ₂	1.04	1.04	1.04	1.04
N ₁ '-H ₁ '	1.04	-	1.04	-
N ₂ '-H ₂ '	1.04	-	1.04	-
N ₄ -H ₃	1.03	1.03	1.03	1.03
N ₄ '-H ₃ '	1.03	-	1.03	-
O ₁ '...H ₁	1.72	1.97	1.72	1.98
N ₃ '...H ₂	1.92	-	1.91	-
N ₃ '...H ₂ '	1.92	-	1.92	-
O ₁ '...H ₁ '/H ₂	1.72	1.69	1.71	1.69
O ₂ '...H ₃	1.72	1.83	1.73	1.83
O ₂ '...H ₃ '	1.72	-	1.72	-
O ₁ '...H ₁ -N ₁	173.12	141.38	173.05	141.47
N ₃ '...H ₂ -N ₂	179.26	-	179.76	-
N ₃ '...H ₂ '-N ₂ '	179.80	-	179.61	-
O ₁ '...H ₁ '/H ₂ -N ₁ '/N ₂	173.20	171.27	173.15	171.29
O ₂ '...H ₃ -N ₄	136.56	132.71	136.45	132.71
O ₂ '...H ₃ '-N ₄ '	136.61	-	136.64	-
C ₁ -N ₁ -C ₂ -N ₂	179.70	-179.09	-179.19	-178.96
O ₂ -C ₂ -N ₂ -C ₃	0.34	-8.83	0.75	-8.55
H ₃ -N ₄ -C ₄ -C ₆	-179.52	-179.94	-179.15	179.95
C ₅ -C ₆ -C ₇ -C ₈	-90.53	-83.61	-102.74	-86.10
C ₅ -C ₆ -C ₈ -C ₉	100.54	-115.40	121.76	-118.59
C ₁ '-N ₁ '-C ₂ '-N ₂ '	-179.62	-	-179.70	-
O ₂ '-C ₂ '-N ₂ '-C ₃ '	0.50	-	1.16	-
H ₃ '-N ₄ '-C ₄ '-C ₆ '	179.59	-	-179.84	-
C ₅ '-C ₆ '-C ₇ '-C ₈ '	89.86	-	78.69	-
C ₅ '-C ₆ '-C ₈ '-C ₉ '	-99.24	-	-17.69	-

[a] The geometric parameters of the ground state (S_0). [b] The geometric parameters of the first excited state (S_1)

Table S4: Calculated important bond lengths (in Å) and bond angles (in degree) for the fully optimized structure of UPNa-dimer (in DCM) and UPNa-DMSO complex (in DMSO) in ground state (S_0) and first excited state (S_1).

	UPNa (DCM) ^[a]	UPNa (DMSO) ^[a]	UPNa (DCM) ^[b]	UPNa (DMSO) ^[b]
N ₁ -H ₁	1.04	1.04	1.05	1.05
N ₂ -H ₂	1.04	1.05	1.04	1.05
N ₁ '-H ₁ '	1.04	-	1.04	-
N ₂ '-H ₂ '	1.04	-	1.04	-
N ₄ -H ₃	1.03	1.03	1.03	1.02
N ₄ '-H ₃ '	1.03	-	1.03	-
O ₁ '...H ₁	1.70	1.75	1.66	1.68
N ₃ '...H ₂	1.92	-	1.89	-
N ₃ ...H ₂ '	1.92	-	1.90	-
O ₁ ...H ₁ '/H ₂	1.70	1.71	1.70	1.68
O ₂ ...H ₃	1.74	1.84	1.76	1.86
O ₂ '...H ₃ '	1.74	-	1.75	-
O ₁ '...H ₁ -N ₁	171.54	164.38	171.21	167.71
N ₃ '...H ₂ -N ₂	177.59	-	178.26	-
N ₃ ...H ₂ '-N ₂ '	177.59	-	177.60	-
O ₁ ...H ₁ '/H ₂ -N ₁ '/N ₂	171.54	172.10	172.12	171.63
O ₂ ...H ₃ -N ₄	135.05	131.90	134.10	131.05
O ₂ '...H ₃ '-N ₄ '	135.05	-	134.72	-
N ₂ -C ₁ -N ₁ -C ₂	-178.32	-172.62	-173.29	-168.38
N ₂ -C ₁ -C ₂ -C ₃	-178.03	40.19	-164.44	26.04
N ₂ '-C ₁ '-N ₁ '-C ₂ '	-178.32	-	-177.75	-
N ₂ '-C ₁ '-C ₂ '-C ₃ '	-178.04	-	-173.58	-

[a] The geometric parameters of the ground state (S_0). [b] The geometric parameters of the first excited state (S_1)

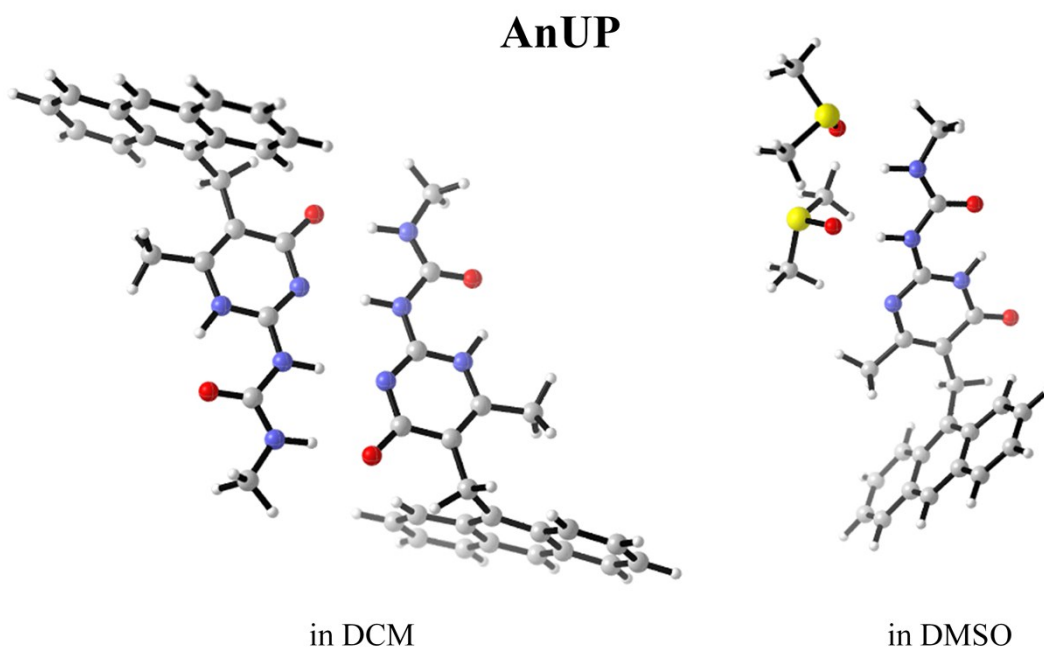


Figure S2. Optimized first excited state (S_1) structures for AnUP dimer (in DCM) and monomer complexes (in DMSO). Gray: C; red: O; blue: N; yellow: S; white: H.

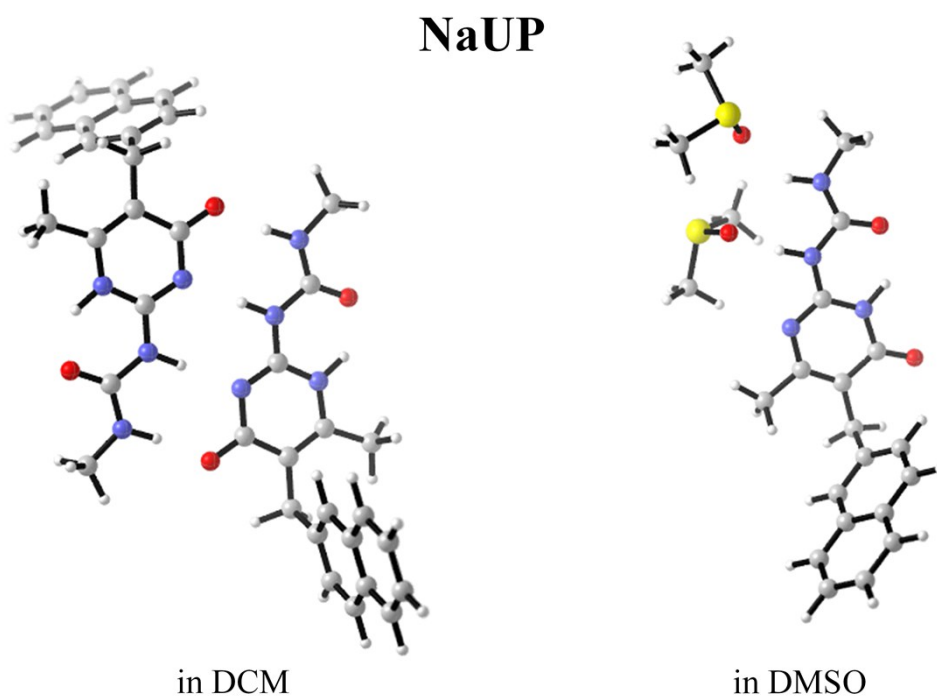


Figure S3. Optimized first excited state (S_1) structures for NaUP dimer (in DCM) and monomer complexes (in DMSO). Gray: C; red: O; blue: N; yellow: S; white: H.

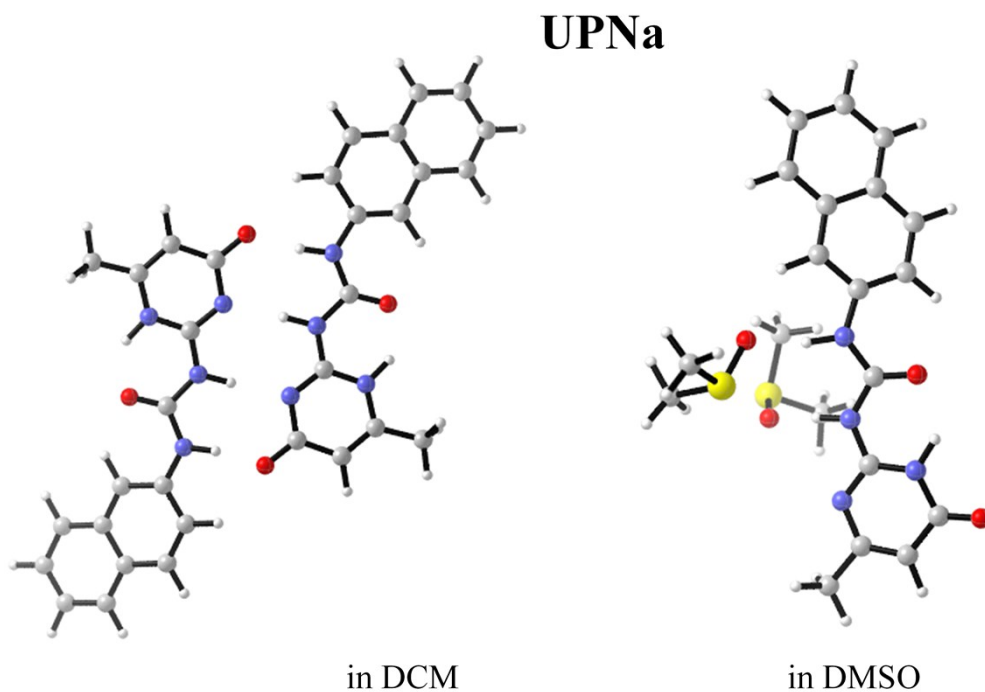


Figure S4. Optimized first excited state (S_1) structures for UPNa dimer (in DCM) and monomer complexes (in DMSO). Gray: C; red: O; blue: N; yellow: S; white: H.

YZL150331--ANUP

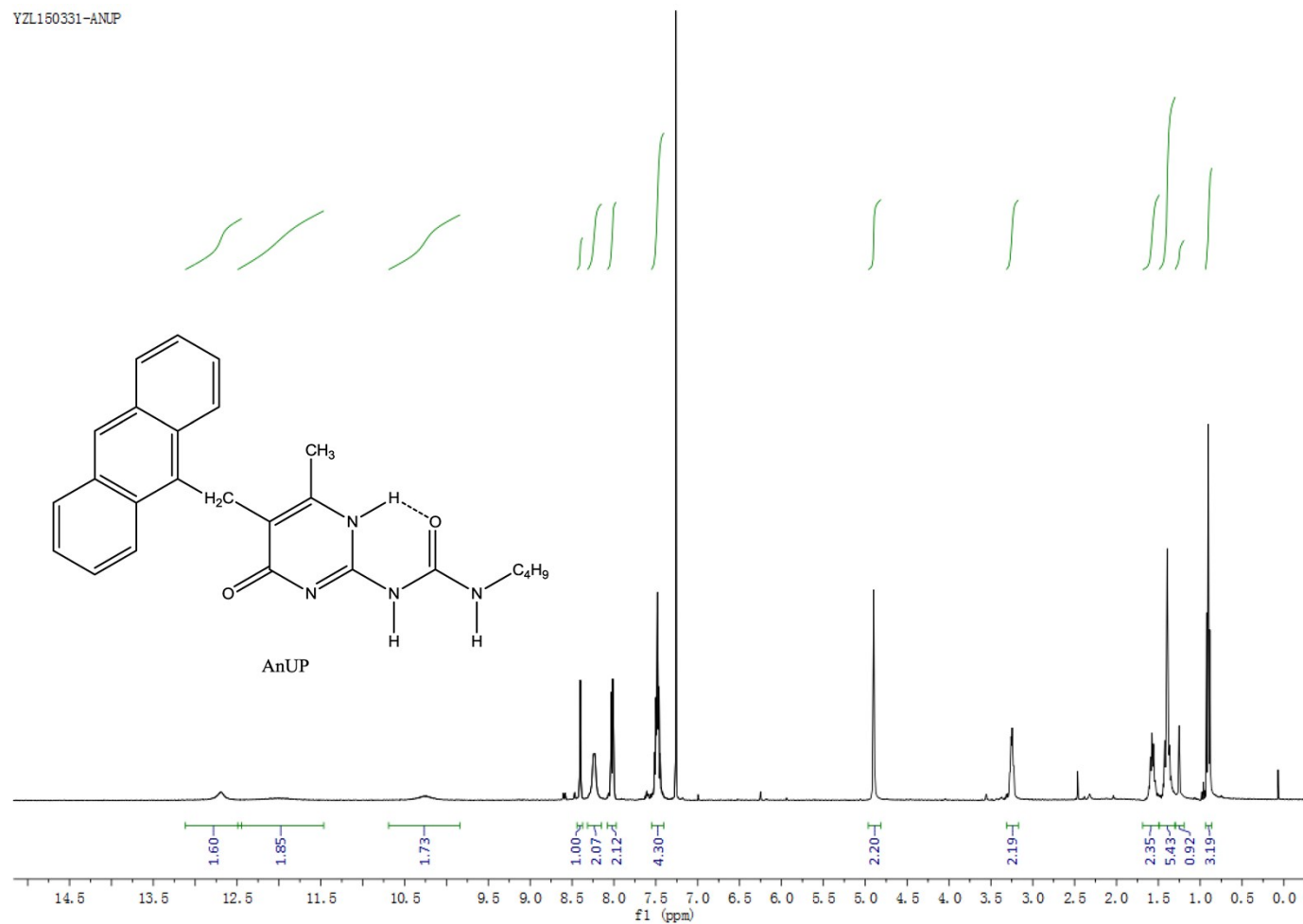


Figure S6. ¹H NMR spectrum of AnUP in CDCl₃

YZL160331-NaUP

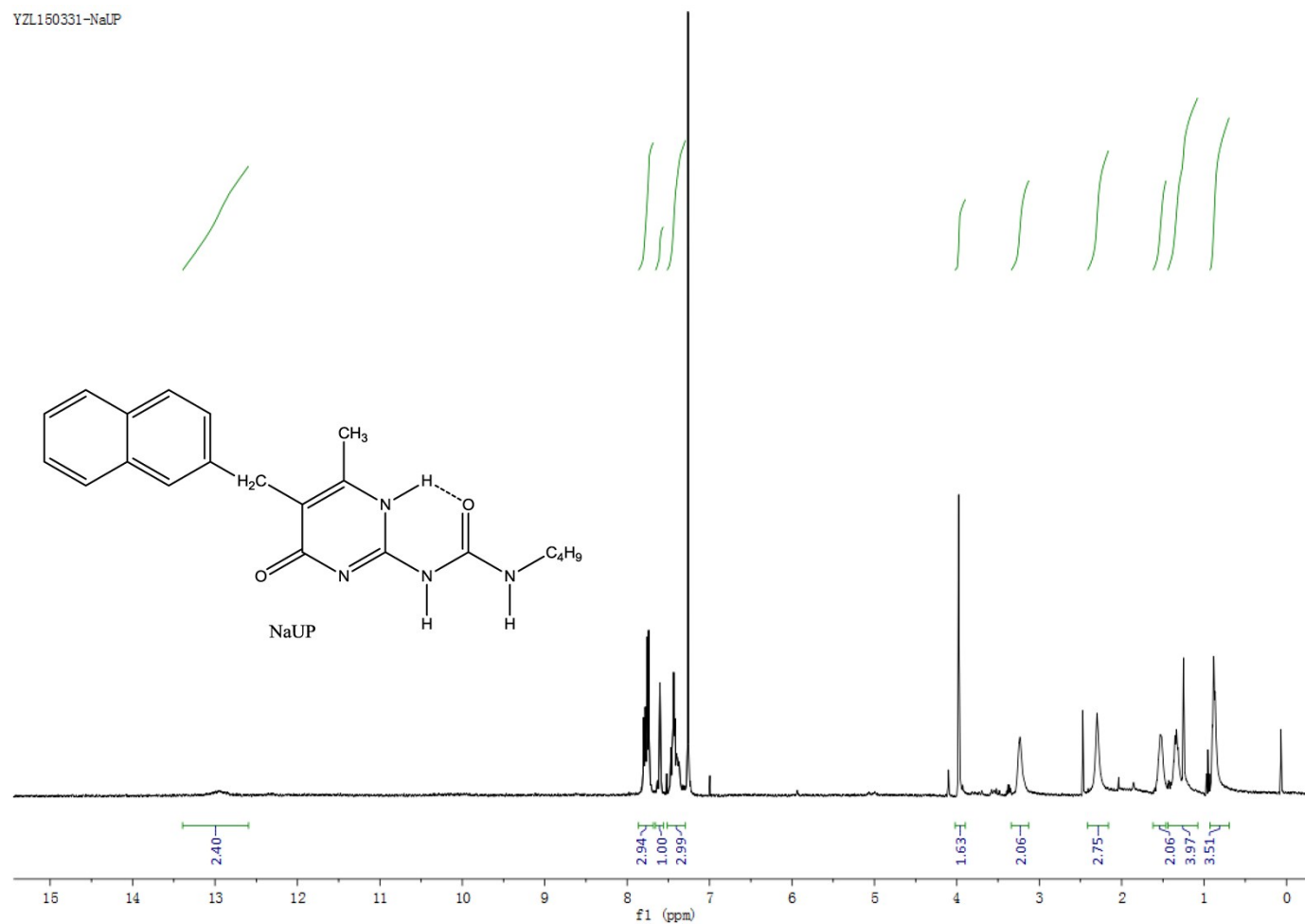


Figure S7. ¹H NMR spectrum of NaUP in CDCl₃

YZL150331-UPNa

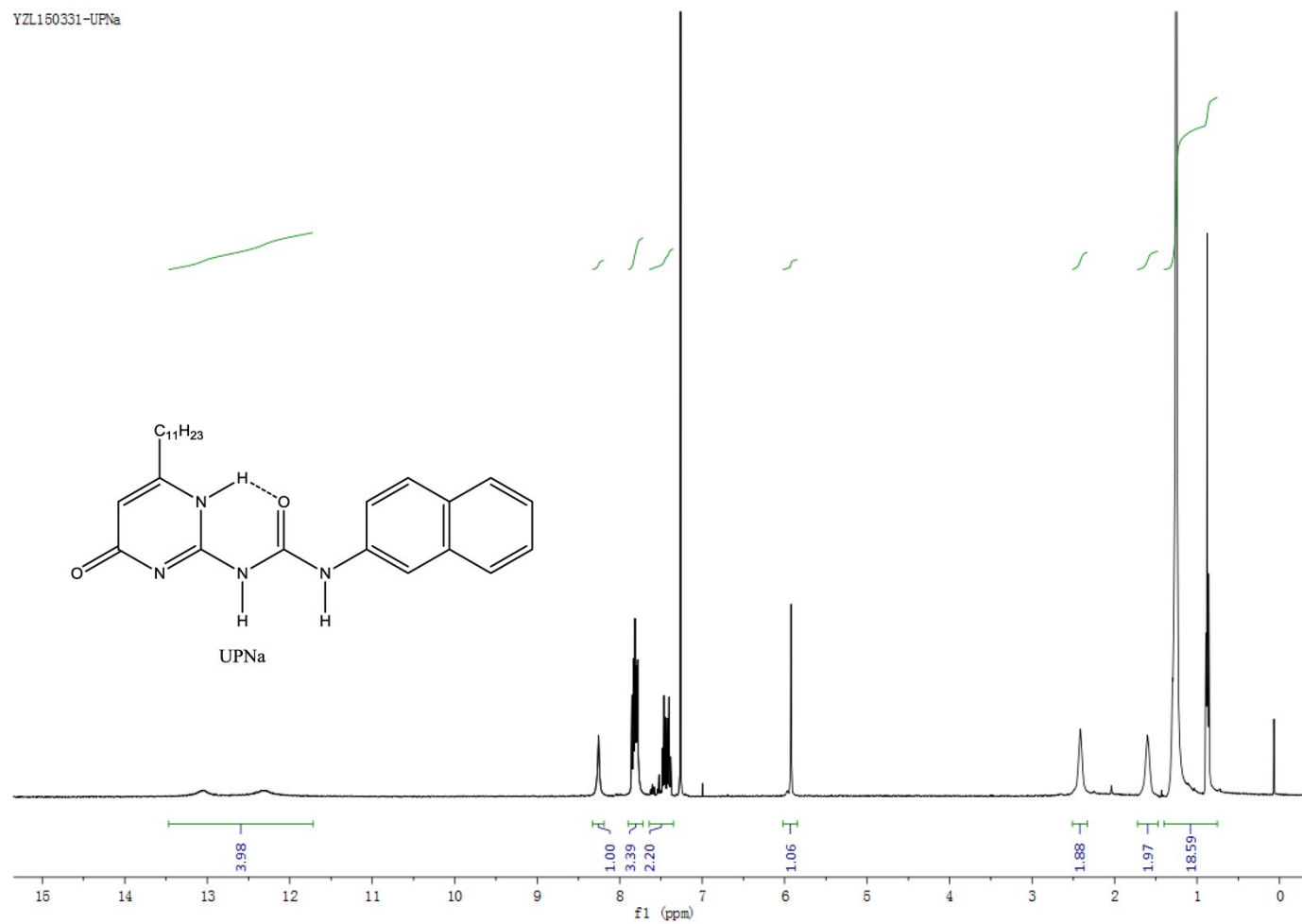


Figure S8. ¹H NMR spectrum of UPNa in CDCl₃

^1H NMR spectra were recorded on a Bruker Avance II 400 MHz NMR spectrometers using CDCl_3 as solvent. Chemical shifts (δ) are reported to the shift-scale calibrated with the residual NMR solvent as internal standard.

Reference:

1. Zhao, Y.-p.; Zhao, C.-c.; Wu, L.-z.; Zhang, L.-p.; Tung, C.-h.; Pan, Y.-j., First Fluorescent Sensor for Fluoride Based on 2-Ureido-4 [1H] -pyrimidinone Quadruple Hydrogen-Bonded AADD Supramolecular Assembly. *J Org Chem* **2006**, *71*, 2143-2146.