Supplementary Information for

Shape-controlled synthesis of α-Fe₂O₃ nanocrystals for

efficient adsorptive removal of Congo red

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Synthesis of the ligand 2,4,6-Tris(pyrazol-1-yl)-1,3,5-triazine (Tptz)

Tptz can be easily obtained via the reported method.¹ Typically, cyanuric chloride (1.84 g, 10 mmol) and pyrazole (4.08 g, 60 mmol) were mixed and stirred rapidly without solvent for 5 min at 60 °C. Then, the reaction mixture was dissolved in 100 mL chloroform and washed with 100 mL distilled water. The organic layer was collected and aqueous layer was further extracted with chloroform (3×50 mL). The combined organic layer was dried over anhydrous MgSO₄ and evaporated to dryness. The resulting white powder was collect and the ¹H-NMR data were identical with those reported in literature.



1 D. Azarifar, M. A. Zolfigol and A. Forghaniha, Heterocycles, 2004, 63, 1897-1901.



Fig. S1. The FT-IR spectrum of the obtained products A1, A2 and A3.



Fig. S2. TEM images of the products obtained from pure H_2O . (a, scale bar = 200 nm; b, scale bar = 50 nm).



Fig. S3. PXRD pattern of the synthesized sample in pure H_2O .



Fig S4. PXRD patterns of the synthesized samples in the absence of Tptz in three kinds of mixed solvent DMF/H₂O.

TPTz:FeCl ₃ ·6H ₂ O	DMF/H ₂ O=6:2 mL	DMF/H2O=4:4 mL	DMF/H ₂ O=2:6 mL
0.1:0.2 mmol	20 m		20 m
0.3:0.2 mmol			200 nm
0.4:0.2 mmol	SO AN		
0.5:0.2 mmol			88 - NO

Fig S5. TEM images of the samples obtained from different solvent DMF/H_2O in the presence of 0.1-0.5 mmol Tptz.



Fig. S6. Nitrogen adsorption–desorption isotherms (left) and Barrett-Joyner-Halenda (BJH) pore size distribution profiles (right) of A1, A2 and A3.



Fig. S7. Recycle test of CR removal efficiency of α -Fe₂O₃ A3.