Immobilization of cadmium ions to synthesis hierarchical flowerlike cadmium phosphates microspheres and their application in degradation of organic pollutants under light irradiation

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**Fig. S1** Nitrogen adsorption-desorption isotherms of  $Cd_5(PO_4)_2P_2O_7$  (T500 and T700). The inset is the corresponding pore-size distribution.



Fig. S2 Comparative photocatalytic activity of  $Ag_3PO_4/Cd_5H_2(PO_4)_4\cdot 4H_2O$  with  $Ag_3PO_4/Ca(PO_4)_6(OH)_2$ ) and pure  $Ag_3PO_4$  under the irradiation of visible light ( $\lambda > 400$  nm). The content of  $Ag_3PO_4$  in the composites was about 10 wt%.

The Ag<sub>3</sub>PO<sub>4</sub>/Cd<sub>5</sub>H<sub>2</sub>(PO<sub>4</sub>)<sub>4</sub>·4H<sub>2</sub>O composite was prepared by a simple precipitation method similar to the preparation of Cd<sub>5</sub>H<sub>2</sub>(PO<sub>4</sub>)<sub>4</sub>·4H<sub>2</sub>O. In a typical process, Cd(CH<sub>3</sub>COO)<sub>2</sub>·2H<sub>2</sub>O and a certain amount of Ag(CH<sub>3</sub>COO) were firstly dissolved in deionized water. Then Na<sub>2</sub>HPO<sub>4</sub> aqueous solution was added dropwise to the Cd-Ag aqueous solution under stirring within 30 min. The mixture was stirred for 1 h and the resultant precipitates were collected by filtration, washed with distilled water repeatedly, and dried at 60 °C overnight. The Ag<sub>3</sub>PO<sub>4</sub>/Ca(PO<sub>4</sub>)<sub>6</sub>(OH)<sub>2</sub>) composite was prepared according to the literature.<sup>1</sup> Ag<sub>3</sub>PO<sub>4</sub> was obtained by the reaction of Ag(CH<sub>3</sub>COO) solution and Na<sub>2</sub>HPO<sub>4</sub> solution at room.

Photocatalytic experiments were performed in an aqueous solution at ambient temperature. A 300 W halogen lamp (Philips Electronics) equipped with a composited cut-off filter (400 nm  $< \lambda <$  800 nm) was used as the visible light source. The system was cooled by a fan and circulating water to maintain at room temperature. Briefly, 80 mg of photocatalyst was suspended in 80 mL aqueous solution of RhB (10 ppm). Prior to irradiation, the suspension was magnetically stirred in dark for 0.5 h to establish an adsorption–desorption equilibrium. A 3 mL aliquot was taken at several minutes intervals during the experiment and centrifuged to remove the powders. The degradation percentage is reported as  $C/C_0$ , where  $C_0$  is the concentration of initial RhB, and *C* represents the corresponding concentration at a certain time interval.



Fig. S3 SEM image of  $Cd_5(PO_4)_2P_2O_7$  (T400) nanoparticles after 400 °C calcination.

## References

1 X. T. Hong, X. H. Wu, Q. Y. Zhang, M. F. Xiao, G. L. Yang, M. R. Qiu and G. C. Han, *Appl. Surf. Sci.*, 2012, **258**, 4801-4805.