

Supporting Information for RSC Advances:

Enhanced catalytic performance of a Pt-xCeO₂/Graphene catalyst for DMFCs by adjusting the crystal-plane and shape of nanoscale ceria

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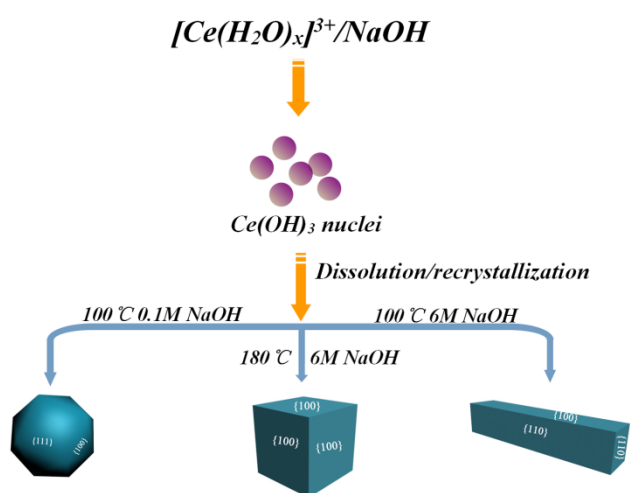


Figure S1. Schematic diagram for the shape-selective synthesis of CeO₂ nanopolyhedra, nanocubes and nanorods.

Figure S1 shows the shape-selective synthesis process of CeO₂ nanopolyhedra, nanocubes and nanorods. We use a simple hydrothermal method to synthesis different shape cerium oxide via adjusting hydrothermal temperature and NaOH concentration of the solution. According to the dissolution/recrystallization mechanism^{2,4}, when Ce³⁺ ions was mixed with NaOH solution, it formed anisotropic Ce(OH)₃ nuclei soon. At a low temperature of 100°C and low OH⁻ concentrations, the dissolution/recrystallization rate was slow, there might exist inadequately high chemical potential for driving the anisotropic Ce(OH)₃ nuclei grew into CeO₂ nanopolyhedras. When the

NaOH concentration was improved, it promoted the $\text{Ce}(\text{OH})_3$ nuclei to grow anisotropically and formed CeO_2 nanorods. When the temperature rise to 180°C , the $\text{Ce}(\text{OH})_3$ became unstable and was readily oxidized into CeO_2 nanocubes.

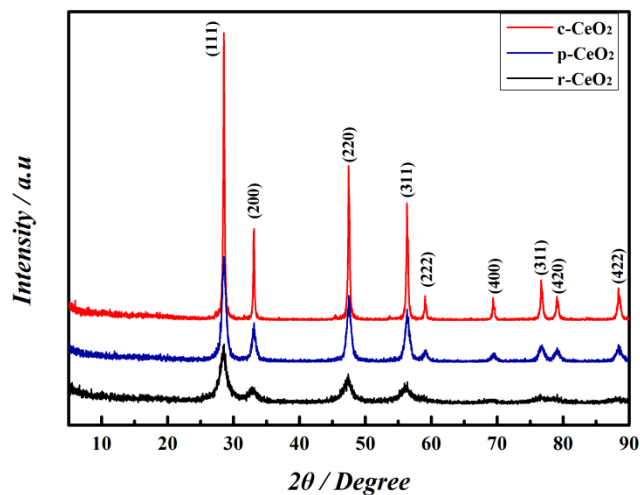


Figure S2. XRD patterns of rod-shaped CeO_2 , cube-shaped CeO_2 , polyhedra-shaped CeO_2 .

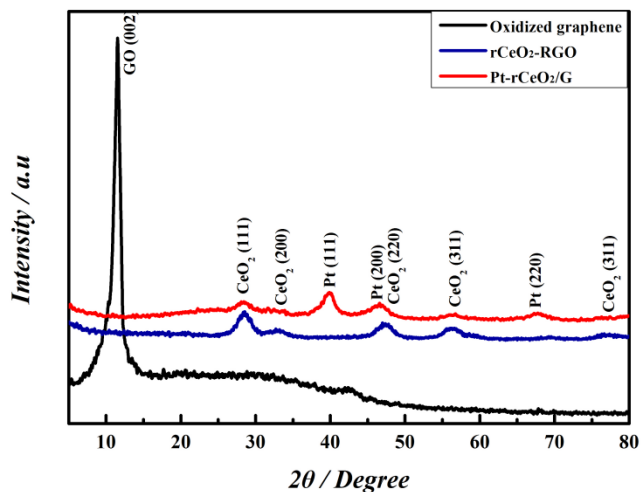


Figure S3. XRD patterns of graphene oxide (GO), rCeO_2/RGO , $\text{Pt-rCeO}_2/\text{Graphene}$ composites.

The XRD patterns of rod-shaped CeO_2 (rCeO_2), cube-shaped CeO_2 (cCeO_2), polyhedra-shaped CeO_2 (pCeO_2) are shown in Figure S2. The sharpness in the diffracted peaks suggests a good crystallinity of the CeO_2 nanocrystals (JCPDS No. 34-0394). Figure S3 shows the XRD patterns

of graphene oxide (GO), rCeO₂/RGO, Pt-rCeO₂/graphene composites. Compared with the typical diffraction peak at ca. 10.06° in the XRD pattern of GO, the XRD pattern of the Pt-rCeO₂/Graphene shows a broad, low intensity peak at ca. 24.9° and it is hardly to find a diffraction peak below 25° in the XRD pattern of rCeO₂/RGO, indicating the reduction of oxidized graphene. The diffraction peaks at ca.39.8°, 46.2° and 67.5° in the Pt-rCeO₂/G catalysts correspond to the face-centered cubic phase of Pt (JCPDS No. 04-0802). The other peaks at ca. 28.6°, 33.1°, 47.5°, 56.4° and 77.7° in Pt-rCeO₂/G catalyst are consistent with the diffraction peaks of rCeO₂/RGO which corresponded to the diffraction peaks of CeO₂ (JCPDS No. 34-0394).^{5,6}

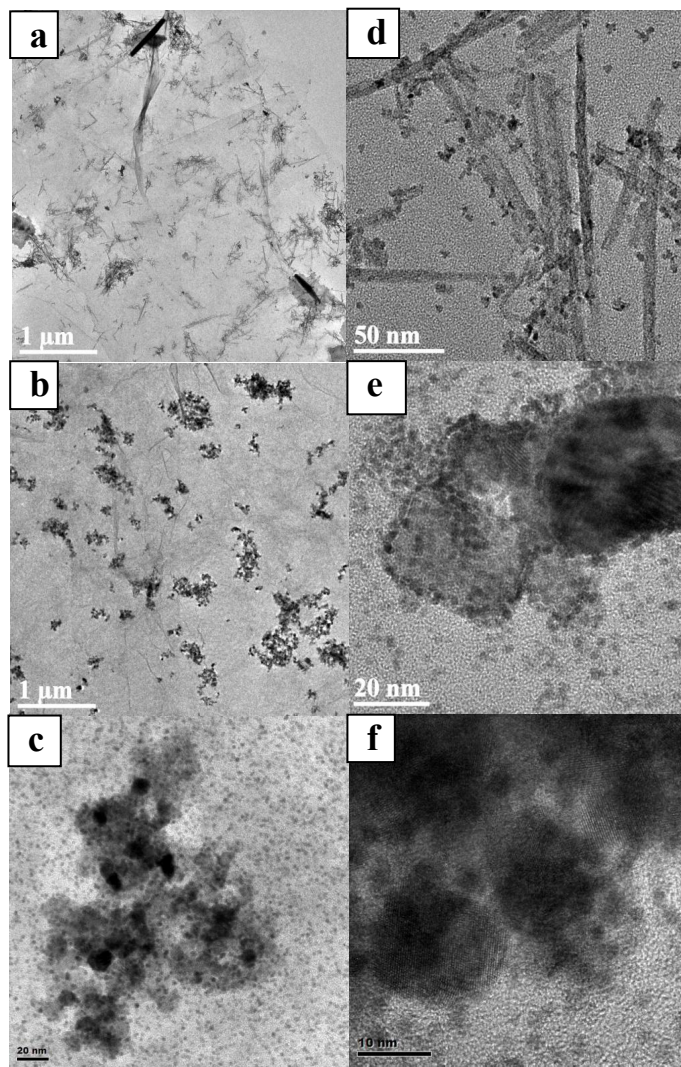


Figure S4. TEM images of Pt-rCeO₂/Graphene (a, d), Pt-cCeO₂/Graphene (b, e), and Pt-pCeO₂/Graphene (c, f)

Reference

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