

## The influence of pH on the stability of antazoline: Kinetic analysis

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### Supplementary information

**Table S1.** Experimental and theoretical molar absorption coefficient ( $\epsilon$ ) values of the pure components (ANT and ABA) at two characteristic wavelengths (293 and 325 nm)

pH	ANT ( $\lambda = 293$ nm)		ABA ( $\lambda = 293$ nm)		ABA ( $\lambda = 325$ nm)	
	Experimental values, mol·L <sup>-1</sup> ·cm <sup>-1</sup>	Theoretical (optimised) values, mol·L <sup>-1</sup> ·cm <sup>-1</sup>	Experimental values, mol·L <sup>-1</sup> ·cm <sup>-1</sup>	Theoretical (optimised) values, mol·L <sup>-1</sup> ·cm <sup>-1</sup>	Experimental values, mol·L <sup>-1</sup> ·cm <sup>-1</sup>	Theoretical (optimised) values, mol·L <sup>-1</sup> ·cm <sup>-1</sup>
7.4	1990	2001	1791	1795	756	760
6.0	1978	1969	1785	1788	745	741
5.0	1944	1950	1784	1788	742	741
4.0	1923	1920	1782	1774	738	732
3.0	1892	1898	1777	1775	734	735