

Supporting Information

Lithium-coordinating ionic conductor for solid-state dye-sensitized solar cells

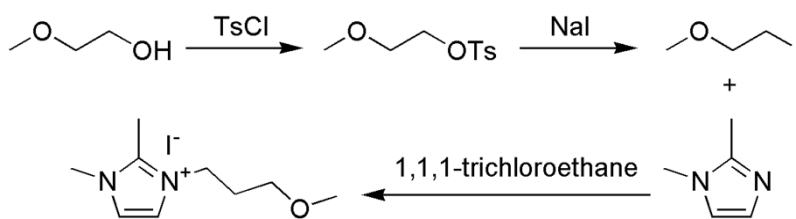
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Scheme S1. Synthetic Route of DOII

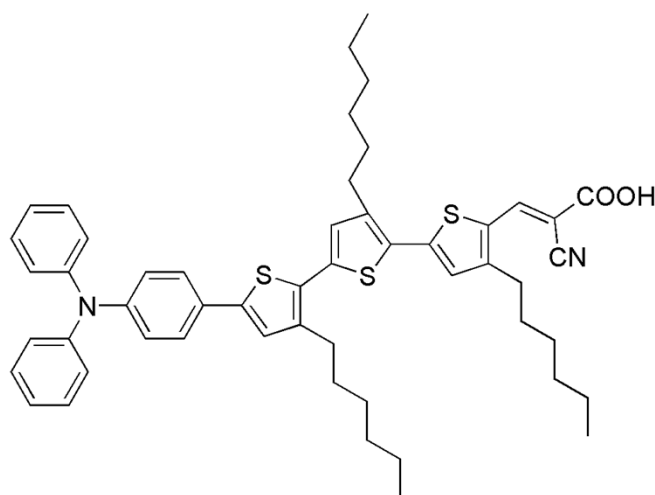


Figure S1. Chemical structure of the metal-free organic dye (FNE29) used in this work.

Table S1. Crystallographic data of DOII

Formula	C ₈ H ₁₅ IN ₂ O
<i>M</i>	282.12
<i>T</i>	293(2) K
Cryst syst	Monoclinic
Space group	P2(1)/n
a (Å)	8.701(3)
b (Å)	10.496(4)
c (Å)	12.077(5)
Alpha (deg)	90
Beta (deg)	91.900(5)
Gamma (deg)	90
V (Å ³)	1102.4(7)
Z	4
Density (mg m ⁻³)	1.700
Absorption coefficient (mm ⁻¹)	2.868
Theta range (deg)	2.57 to 26.01
Reflections collected	4769
Independent reflns	2160 [R(int) = 0.0368]
GOF on <i>F</i> ²	1.018
Final R1, wR2 [<i>I</i> >2σ(<i>I</i>)]	0.0344, 0.0831

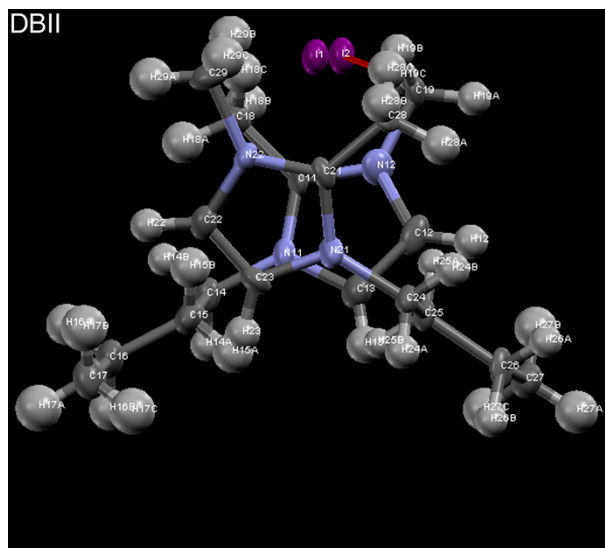


Figure S2. Single crystal structure of DBII

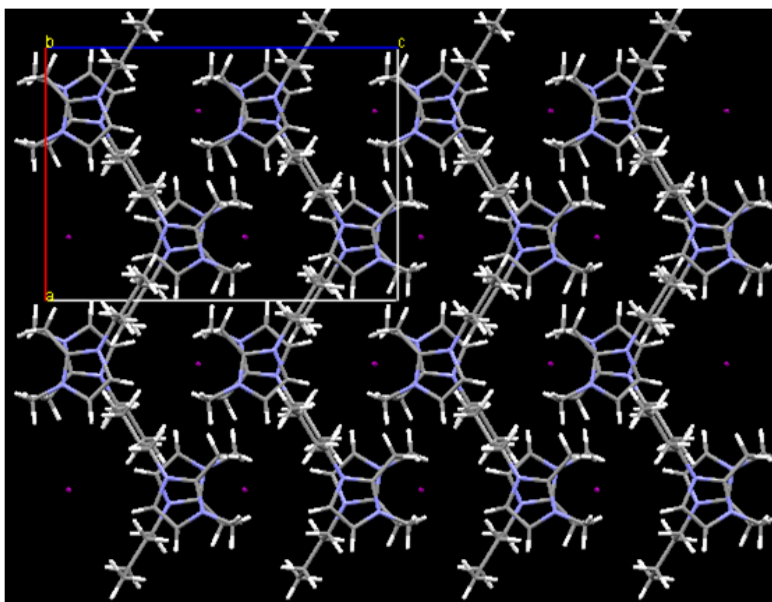


Figure S3. Packing structure of DBII viewed along the *b* axis.

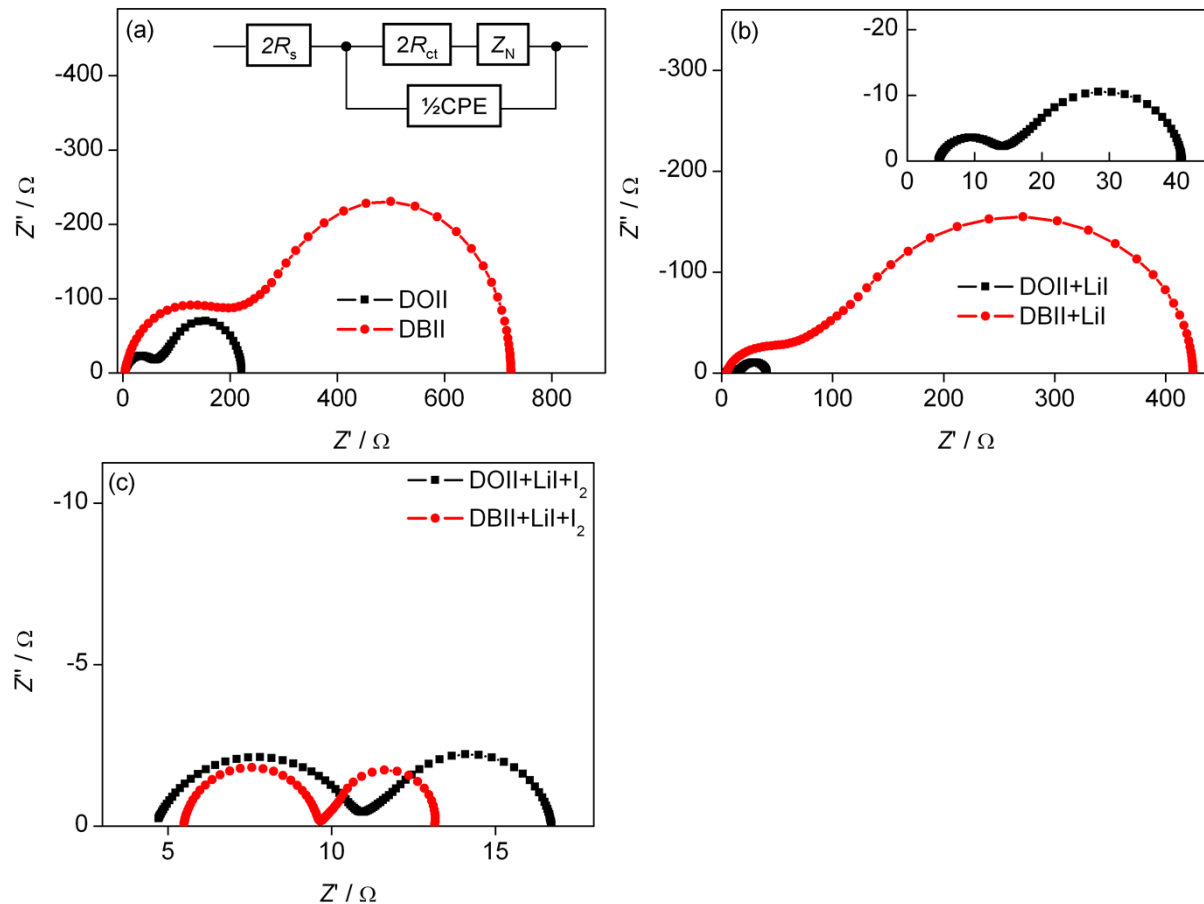


Figure S4. EIS plots for pure ionic conductors (a), mixtures doped with LiI (b), and mixtures doped with LiI and I_2 (c). Inset in (a) is the equivalent circuit diagram for fitting the impedance spectra. The inset in (b) is the magnified impedance spectrum of DOII + LiI. The molar ratio of DOII or DBII and LiI is 2/1; the molar ratio of DOII or DBII, LiI and I_2 is 2/1/0.2.