

Supporting Information

Fabrication of MIL-120 membranes supported by α - Al_2O_3

Hollow ceramic fibers for H_2 separation

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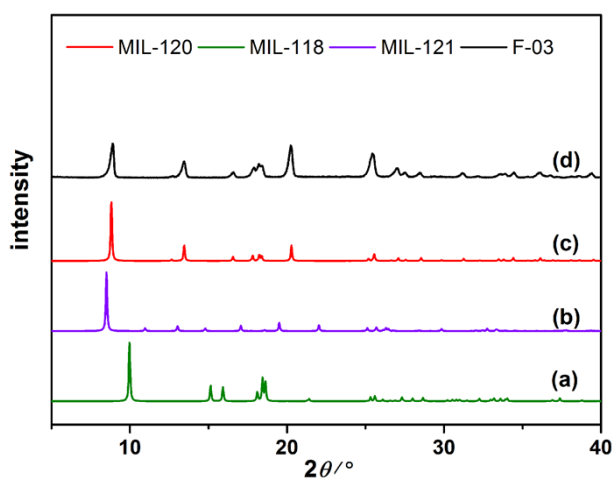


Fig.S1 XRD patterns of simulated (a) MIL-118; (b) MIL-121; (c) MIL-120; and (d) F-03.

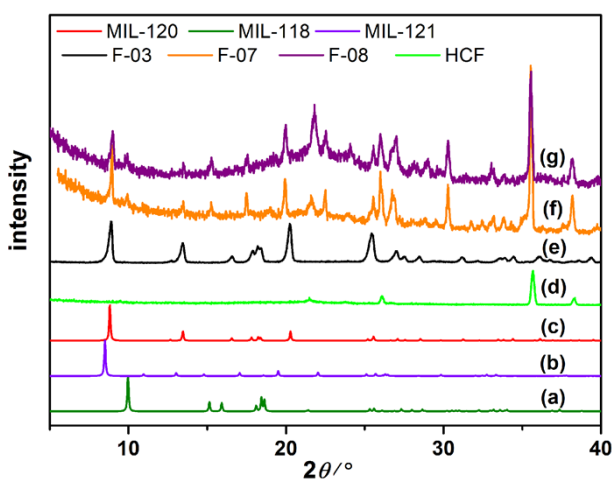


Fig.S2 XRD patterns of simulated (a) MIL-118; (b) MIL-121; (c) MIL-120; (d) HCF; (e) F-03; (f) F-07; (g) F-08.

Table.S1 Single and binary gas permeances ($\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}\cdot\text{Pa}^{-1}$) and separation factors for the MIL-120 membrane at 293K, 313K, 343K and 1 bar. The volume ratio for binary gas systems is 1:1.

Gas(H ₂ /j)	Knudsen constant	Temperature (K)	Single gas			Binary gas		
			Permeances (H ₂)	Permeances (j)	Ideal separation factor	Permeances (H ₂)	Permeances (j)	Separation factor
H ₂ /CH ₄	2.8	293	3.80×10^{-7}	4.9×10^{-8}	7.7	3.50×10^{-7}	4.43×10^{-8}	7.9
		313	5.0×10^{-7}	7.9×10^{-8}	6.3	4.36×10^{-7}	7.30×10^{-8}	6.0
		343	5.86×10^{-7}	1.14×10^{-7}	5.1	5.0×10^{-7}	1.00×10^{-7}	5.0
H ₂ /N ₂	3.7	293	3.80×10^{-7}	5.14×10^{-8}	7.4	3.47×10^{-7}	4.75×10^{-8}	7.3
		313	5.0×10^{-7}	8.33×10^{-8}	6.0	4.21×10^{-7}	6.90×10^{-8}	6.1
		343	5.86×10^{-7}	1.30×10^{-7}	4.5	4.74×10^{-7}	1.05×10^{-7}	4.5
H ₂ /CO ₂	4.7	293	3.80×10^{-7}	4.13×10^{-8}	9.2	3.58×10^{-7}	3.8×10^{-8}	9.4
		313	5.0×10^{-7}	6.80×10^{-8}	7.3	4.27×10^{-7}	5.90×10^{-8}	7.2
		343	5.86×10^{-7}	9.16×10^{-8}	6.4	4.52×10^{-7}	7.00×10^{-8}	6.4

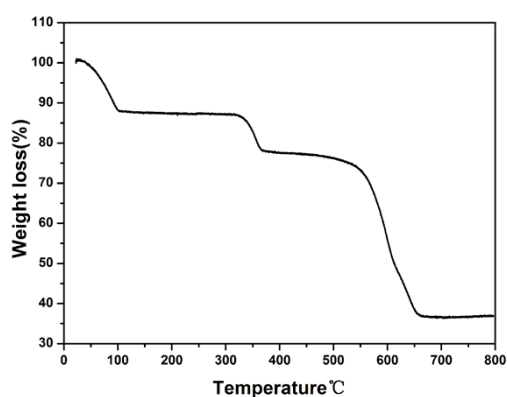


Fig.S3 TG curve of the MIL-120 membrane (F-07) (under air, heating rate: $1^{\circ}\text{C}\cdot\text{min}^{-1}$).

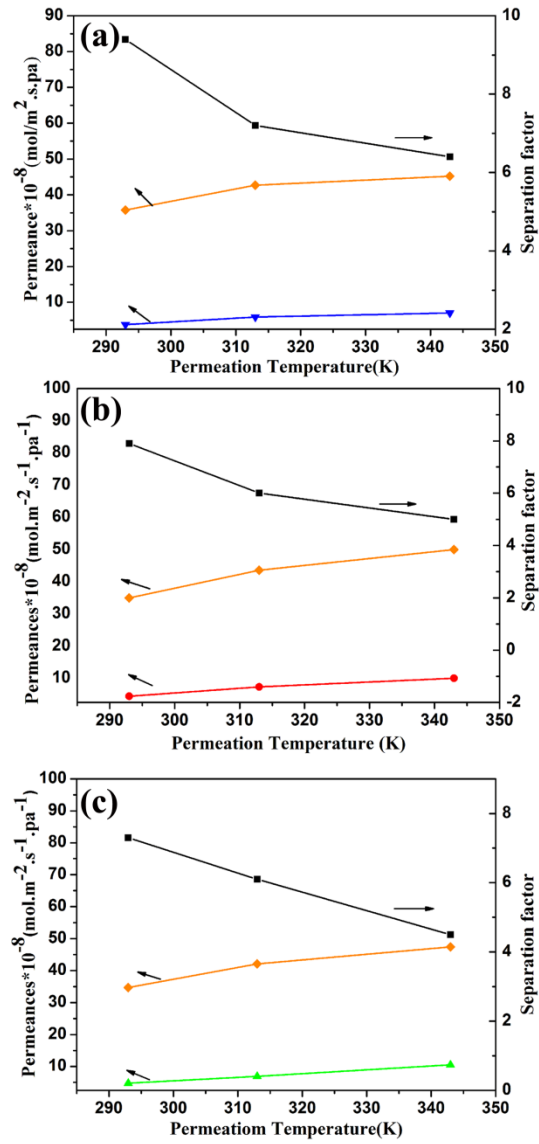


Fig.S4 (a) H₂/CO₂, (b) H₂/CH₄ and (c) H₂/N₂ permeances and separation factors of the MIL-120 membrane as a function of the permeation temperature at a pressure drop of 1bar (inverted triangles: CO₂, triangles: N₂, circles: CH₄, rhombuses: H₂, squares: separation factor).

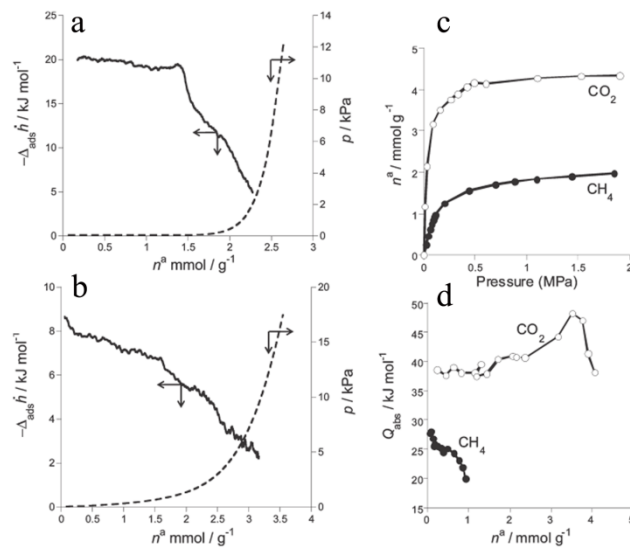


Fig.S5 Differential enthalpies of adsorption and isotherm obtained with (a) nitrogen at 77 K, (b) hydrogen at 77 K on MIL-120, (c) Isotherms and (d) differential enthalpies of adsorption at 303 K obtained with carbon dioxide and methane on MIL-120¹.

Reference

1. C. Volkringer, T. Loiseau, M. Haouas, F. Taulelle, D. Popov, M. Burghammer, C. Riegel, C. Zlotea, F. Cuevas and M. Latroche, *Chemistry of Materials*, 2009, **21**, 5783-5791.