

1 Supporting Information for

2 **A Simple Highly Sensitive and Selective TURN-ON Fluorescent Chemosensor for the**
3 **Detection of Cadmium Ions in Physiological Condition**

4 Sundaram Ellairaja^a, Ramar Manikandan^b, Muthunanthevar Vijayan^c, Seenivasan Rajagopal^a, Vairathevar
5 Sivasamy Vasantha^{a*}

6 ^aSchool of Chemistry, Madurai Kamaraj University, Madurai-625021, Tamilnadu, India.

7 ^bUniversity of Madras, Chennai-600035, Tamilnadu, India.

8 ^cCentral Electro Chemical Research Institute, Karaikudi, Tamilnadu, India.

9

10

11

12

13

14

15

16

17

18

19

20

21

22

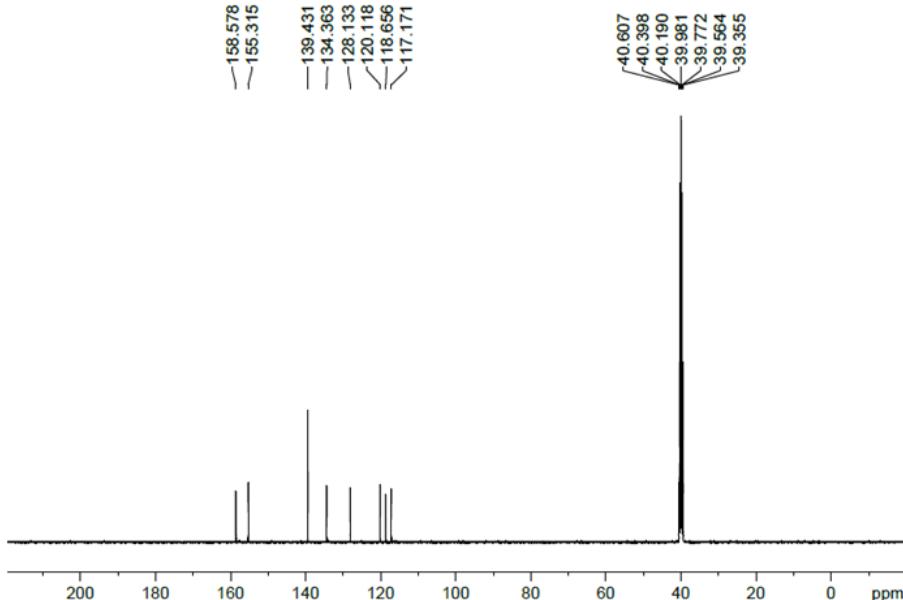
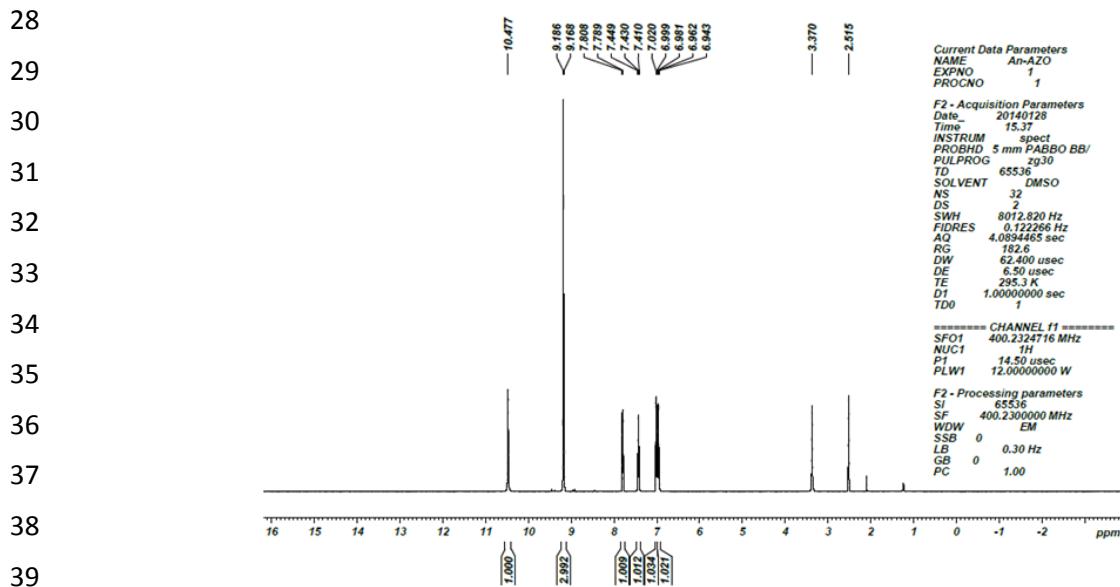
23

24

25 *Corresponding Author

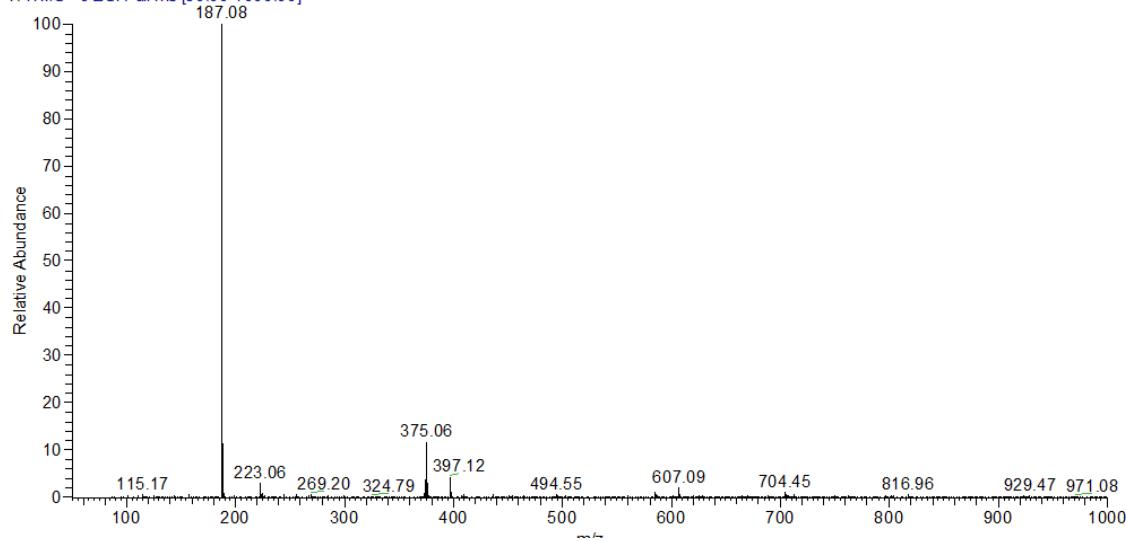
26 Email id: vasantham999@yahoo.co.in (V. S. Vasantha)

27 Tel: 91-452-2458246; Fax: 91-452-2459139



61

TAI-1 #8 RT: 0.10 AV: 1 NL: 7.71E3
 T: ITMS - c ESI Full ms [50.00-1000.00]

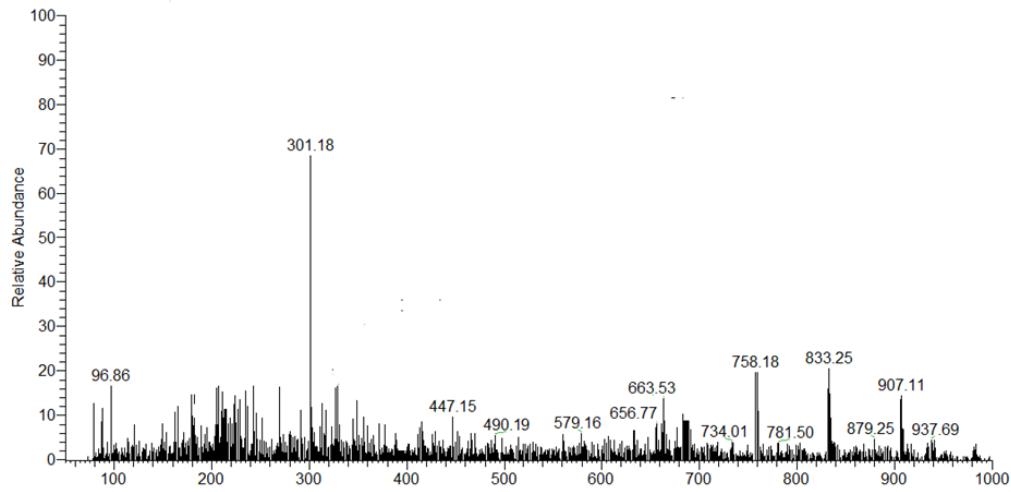


73

74 **Fig. S3** Mass Spectra of the synthesized probe.

75

E (1)#13 RT: 0.17 AV: 1 NL: 1.43E3
 T: ITMS + c ESI Full ms [50.00-1000.00]



86

87 **Fig. S4** Mass Spectra of the synthesized probe- Cd^{2+} ions

88

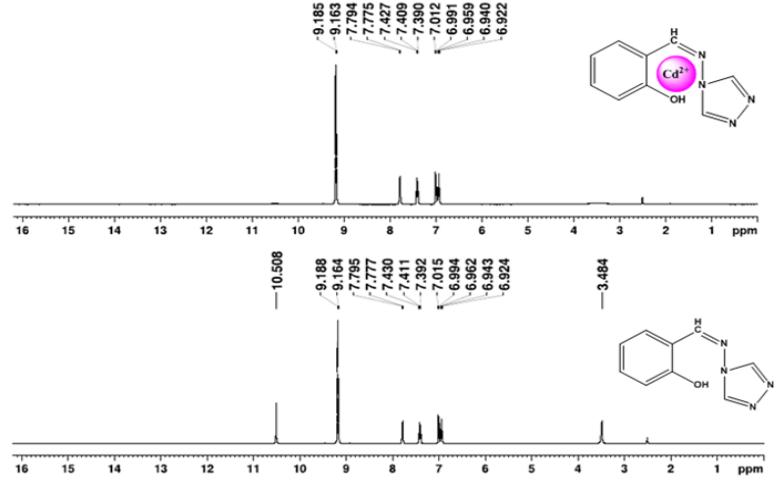
89

90

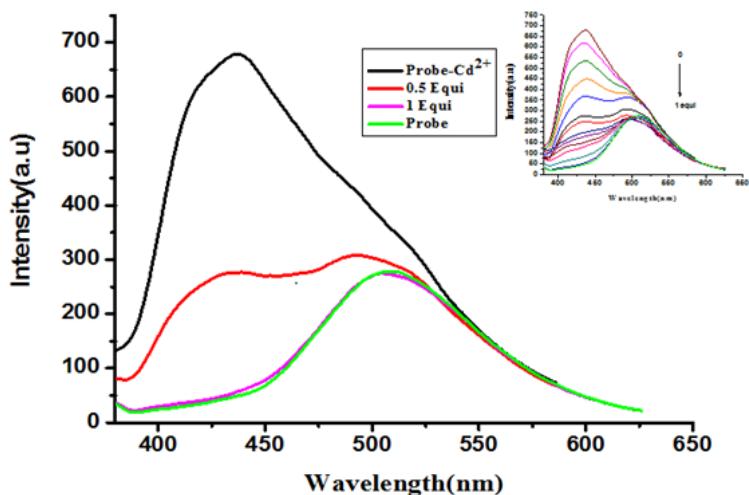
91

92

93

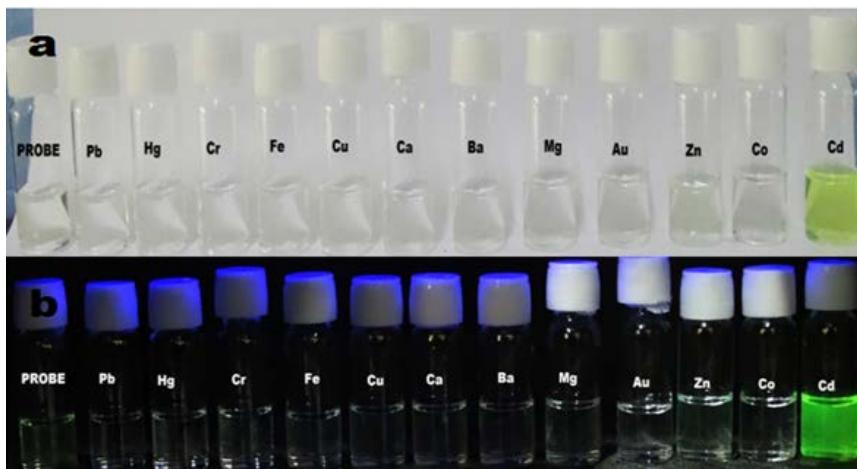


125 **Reversibility test with EDTA**



136 **Fig. S6 Reversibility test of the developed chemosensor with EDTA.** (green) probe alone; (pink)
137 EDTA (1 equivalent) in presence of probe- Cd^{2+} ion ; (Red) EDTA (0.5 equivalent) in presence of
138 probe- Cd^{2+} ion ; (black) the initial fluorescence intensity of probe - Cd^{2+} ion. The concentration of
139 probe and EDTA is 5 μM and the excitation wavelength was 332 nm. Excitation and emission slit
140 widths are 5 nm and 2.5 nm respectively. The emission intensity at 436 nm was used for this study.
141 [Inset: The complete emission spectra for the variation of EDTA with probe - Cd^{2+} ion.

142 **Fluorescence quenching in presence of interfering metal ions**



152 **Fig. S7 Color changes observed during selectivity assay (a) Naked eye (b) Under UV lamp**