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Photocatalytic degradation of methylene blue with hematite nanoparticles synthesized by thermal decomposition of fluoroquinolones oxalato-iron(III) complexes

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Supporting Information

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c)

Fig. S1: Experimental FT IR spectra of a) ciprofloxacin·HCl, b) lomefloxacin·HCl and c) norfloxacin.







b)



c)

Fig. S2: Experimental FT IR spectra of complexes a) 1, b) 2 and c) 3.



Fig. S3: Experimental electronic absorption spectra of the investigated complexes in DMSO.



Fig. S4: Theoretical TD-DFT spectra of the investigated complexes in DMSO calculated at DFT/B3LYP/6-31G(d) level of theory.



c)

Fig. S5: XRD patterns of nano-hematite obtained from the controlled thermal decomposition of complexes a) 1, b) 2 and c) 3.



a)



c)



Fig. S6: FE-SEM images of α -Fe₂O₃ nanoparticles obtained from thermal decomposition of a) 1, b) 2 and c) 3 by magnification of 40 000x.



Fig. S7: Absorption spectral changes and photo-degradation of MB by H₂O₂ under the effect of UV light at 365 nm (Control experiment).



Fig. S8: Absorption spectral changes and photo-degradation of MB by H₂O₂ in presence of hematite nanoparticles No. 1 under the effect of UV light at 365 nm.

Table S1: Second-order interaction energy (E^2 , kcal/mol) between donor and acceptor orbitals in the studied complexes 1-3 calculated at DFT/B3LYP/6-31G(d) level of theory (selected)

Donor→Acceptor		E^2 (kcal/mol)		
	1	2	3	
$LP(3)O2 \rightarrow RY^*(3)Fe$	1.07	1.08	1.08	
$LP(2)O3 \rightarrow RY^{*}(4)Fe$	0.72	0.70	0.69	
$LP(3)O4 \rightarrow RY^{*}(2)Fe$	1.27	1.28	1.27	
LP(2)O5→RY [*] (3)Fe	1.33	1.33	1.33	
$LP(2)O6 \rightarrow RY^{*}(4)Fe$	0.89	0.91	0.90	
LP(3)O7→RY [*] (2)Fe	1.32	1.33	1.32	