

**Highly responsive glutathione functionalized green AuNP probe for precise colorimetric  
detection of Cd<sup>2+</sup> contamination in the environment**

Rajarathinam Manjumeena,<sup>\*a</sup> Dhanapal Duraibabu<sup>b</sup>, Thangavelu Rajamuthuramalingam<sup>c</sup>,  
Ramasamy Venkatesan<sup>d</sup> and Puthupalayam Thangavelu Kalaichelvan<sup>a</sup>

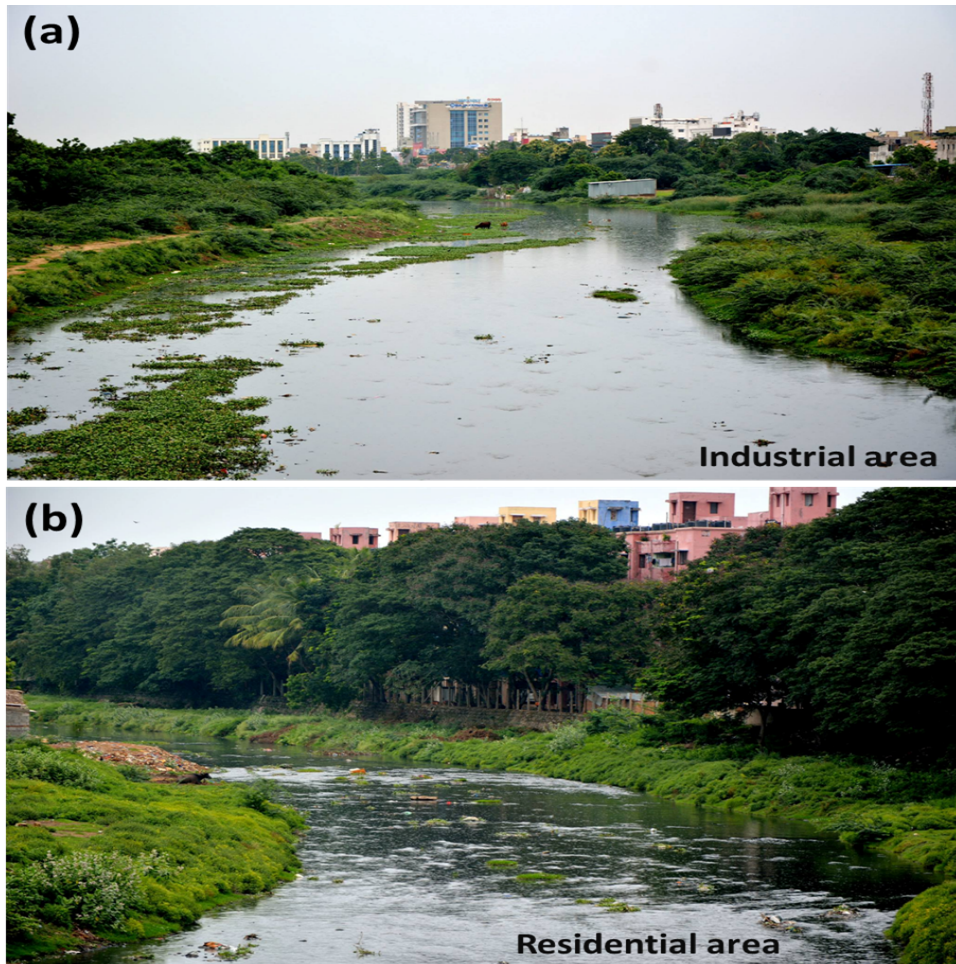
<sup>a</sup>CAS in Botany, University of Madras, Guindy Campus, Chennai- 600 025, India

<sup>b</sup>Department of Chemistry, Anna University, Chennai- 600 025, India

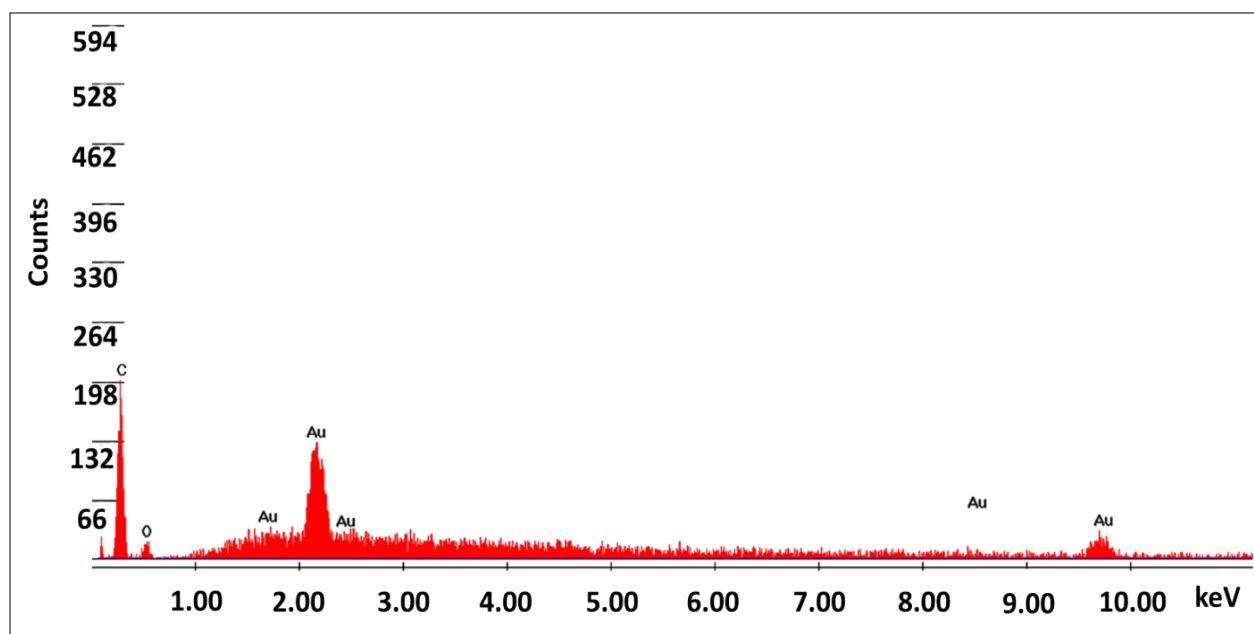
<sup>c</sup>Department of Biotechnology, University of Madras, Guindy Campus, 600 025, India

<sup>d</sup>National Institute of Ocean Technology, Chennai- 600100, India

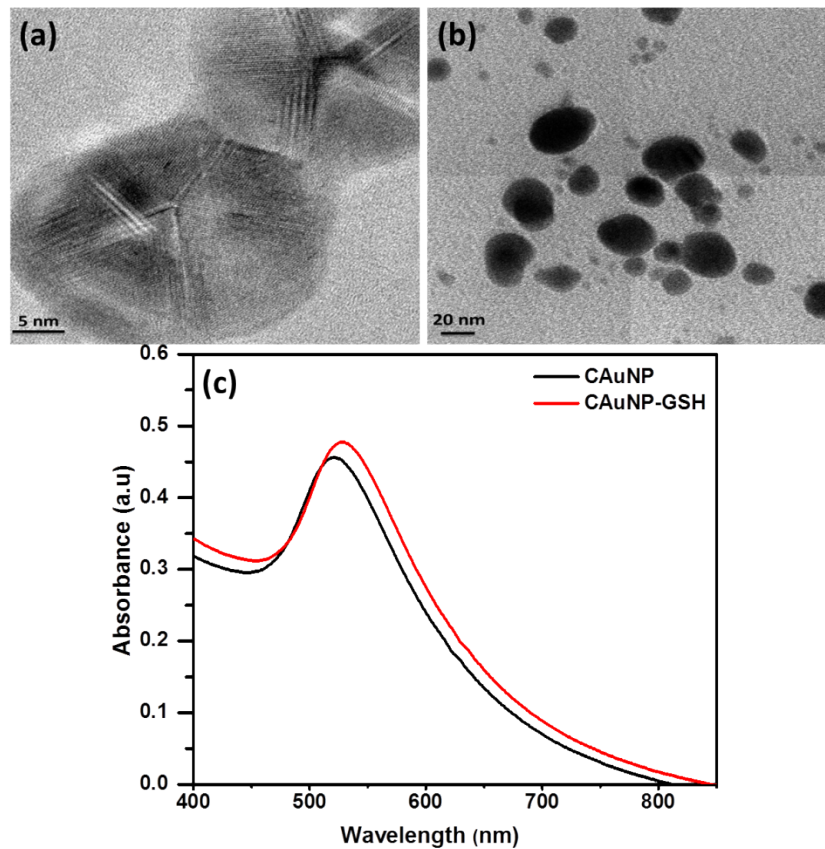
\*manjumeena1989@gmail.com



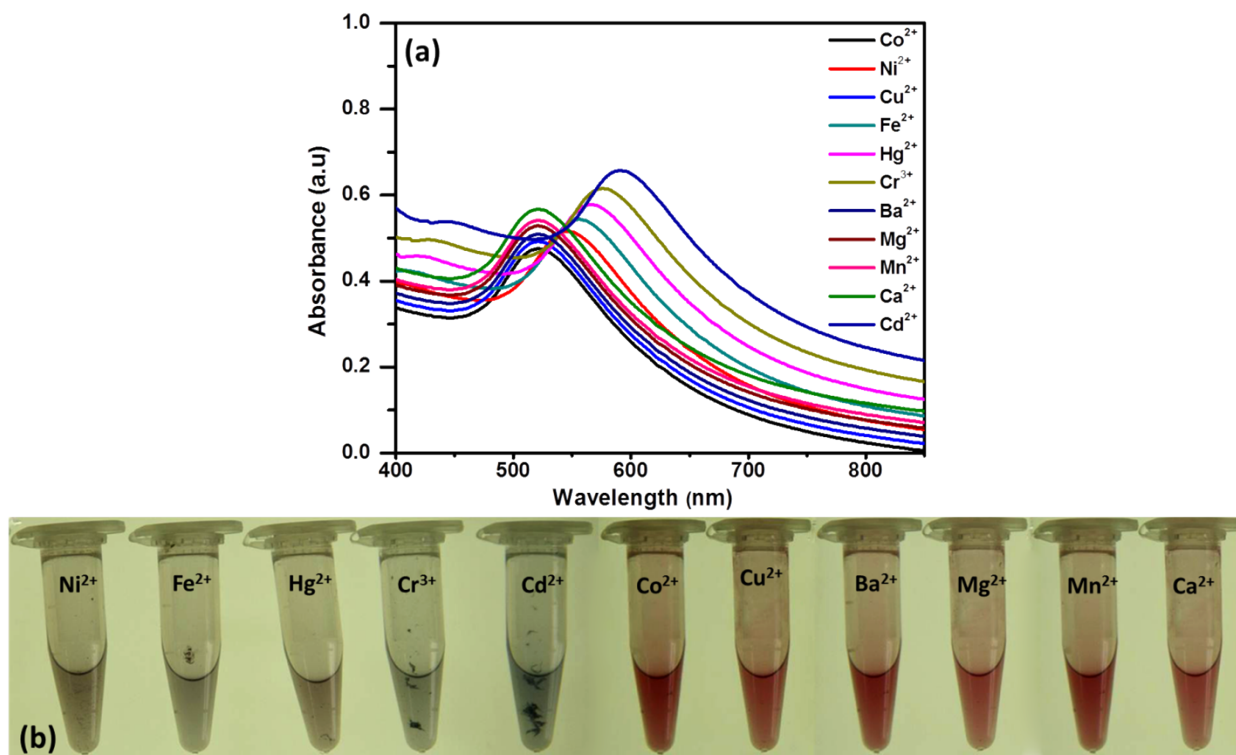
**Fig. S1.** (a) Photograph of Ranipet industrial area (b) Photograph of residential area in the vicinity.



**Fig. S2.** EDAX spectrum of AuNPs.



**Fig. S3.** (a) (b) HRTEM micrographs of the chemically synthesized AuNPs (c) UV-Vis absorption spectra of chemically synthesized AuNps (control) before and after functionalization with GSH



**Fig. S4.** (a) UV- Vis absorption spectra of chemically synthesized AuNPs- GSH (control) and  
 (b) the corresponding color change photo images of chemically synthesized  
 AuNPs- GSH added with different metal ions showing selectivity/interference

**Table S1** Synoptic table of Au-based colorimetric sensor for heavy metal and other analyte detection

<b>Colorimetric Nanosensor Employed</b>	<b>Route of Synthesis</b>	<b>Surface Functionalization</b>	<b>Analytical Detection</b>	<b>References</b>
AuNPs	Citrate reduction of HAuCl <sub>4</sub>	GSH	Cd <sup>2+</sup>	43
AuNPs	citrate reduction of HAuCl <sub>4</sub>	6-mercaptopuronic acid (MNA) and L-Cysteine (LCys)	Cd <sup>2+</sup>	44
Fluorescent gold nanodots	Chemical synthesis using tetrakis(hydroxymethyl) phosphonium chloride (THPC) and HAuCl <sub>4</sub>	GSH	Pb <sup>2+</sup>	45
Fluorescent AuNPs	Encapsulation in bio-compatible poly-(amidoamine)	GSH	Cu <sup>2+</sup>	46
AuNPs	Citrate reduction of HAuCl <sub>4</sub>	-	melamine	47
AuNPs	Citrate reduction of HAuCl <sub>4</sub>	-	organophosphorus pesticides - mathamidophos	48
AuNPs	Citrate reduction of HAuCl <sub>4</sub>	azide-terminal alkyne functionalization	Organophosphate pesticide- Paraoxon	49
AuNPs	Citrate reduction of HAuCl <sub>4</sub>	Papain	Hg <sup>2+</sup> , Pb <sup>2+</sup> and Cu <sup>2+</sup>	50
AuNPs	Citrate reduction of HAuCl <sub>4</sub>	2-mercapto-5-(3-nitrophenyl)-1,3,4-thiadiazole]	Cd <sup>2+</sup>	51

AuNPs	Citrate reduction of H <sub>AuCl</sub> <sub>4</sub>	Alkane Thiol-Tween 20, 16-mercaptohexadecanoic acid	Glucose oxidase	52
AuNPs	Citrate reduction of H <sub>AuCl</sub> <sub>4</sub>	11-mercaptoundecanoic acid	Pb <sup>2+</sup> , Cd <sup>2+</sup> , Hg <sup>2+</sup>	53
AuNPs	Citrate reduction of H <sub>AuCl</sub> <sub>4</sub>	mercaptopropionic acid (MPA) and homocystine	Hg <sup>2+</sup>	54
AuNPs	Citrate reduction of H <sub>AuCl</sub> <sub>4</sub>	Chitosan	Zn <sup>2+</sup> and Cu <sup>2+</sup>	55
AuNPs	Citrate reduction of H <sub>AuCl</sub> <sub>4</sub>	DNA-oligonucleotides	Hg <sup>2+</sup>	56
AuNPs	Citrate reduction of H <sub>AuCl</sub> <sub>4</sub>	Dithiocarbamate derivative of calixarene	Hg <sup>2+</sup>	57
AuNPs	Gallic acid mediated reduction of H <sub>AuCl</sub> <sub>4</sub>	-	Pb <sup>2+</sup>	58
Gold nanorods	N-cetyl trimethylammonium bromide mediated reduction of H <sub>AuCl</sub> <sub>4</sub>	GSH	Pb <sup>2+</sup>	59
Fluorescent gold nanoclusters incorporated into electrospun polyvinyl alcohol nanofibers	Bovine serum albumin mediated reduction of H <sub>AuCl</sub> <sub>4</sub>	-	Hg <sup>2+</sup>	60
AuNPs	Citrate reduction of H <sub>AuCl</sub> <sub>4</sub>	GSH	Pb <sup>2+</sup>	61

**Table S2** Quantification of metal ions in sample 1 and 2 by atomic absorption spectroscopy

<b>S.No</b>	<b>Transitions metals</b>	<b>Sample 1 (<math>\mu\text{g/L}</math>)</b>	<b>Sample 2 (<math>\mu\text{g/L}</math>)</b>
1	$\text{Cr}^{3+}$	32.5 $\pm$ 0.25	24.0 $\pm$ 0.22
2	$\text{Ni}^{2+}$	23.8 $\pm$ 0.95	11.2 $\pm$ 0.30
3	$\text{Mn}^{2+}$	23.0 $\pm$ 0.20	11.9 $\pm$ 0.41
4	$\text{Cu}^{2+}$	32.0 $\pm$ 0.42	31.1 $\pm$ 0.60
5	$\text{Hg}^{2+}$	43.0 $\pm$ 0.95	30.9 $\pm$ 0.87
6	$\text{Co}^{2+}$	37.5 $\pm$ 0.40	31.2 $\pm$ 0.25
7	$\text{Cd}^{2+}$	33.6 $\pm$ 0.48	25.1 $\pm$ 0.50
8	$\text{Fe}^{2+}$	25.0 $\pm$ 0.98	21.5 $\pm$ 0.45
<b>Alkaline earth metals</b>			
9	$\text{Mg}^{2+}$	21.0 $\pm$ 0.25	18.5 $\pm$ 0.57
10	$\text{Ca}^{2+}$	39.8 $\pm$ 0.47	31.0 $\pm$ 0.63
11	$\text{Ba}^{2+}$	20.1 $\pm$ 0.85	19.5 $\pm$ 0.49