Supporting Information

NIR luminescence of a series of benzoyltrifluoroacetone erbium

complexes

Qingyan Sun, Pengfei Yan, Wanying Niu, Wenyi Chu*, Xu Yao, Guanghui An, Guangming Li*

Key Laboratory of Functional Inorganic Material Chemistry (MOE), School of Chemistry and Materials Science, Heilongjiang University, No. 74, Xuefu Road, Nangang District, Harbin, Heilongjiang 150080, P. R. China, E-mail: <u>gmli_2000@163.com</u>. Fax: (+86)451– 86608458; Tel: (+86) 451–86608458

Index				
Fig. S1-3	TG–DSC curves of complex 1–5			
Fig. S4–6	The powder X-ray diffraction spectra of the simulated			
	patterns and experimental for complexes 1-5			
Fig. S7	IR spectra of bta and complexes 1–5			
Fig. S8	Excitation spectra of complexes 1–5			
Table S1	Selected bond lengths (Å) and angles (°) for complexes			
	1–3			
Table S2	Selected bond lengths (Å) and angles (°) for complexes			
	4 and 5 .			



Fig. S1 TG–DSC curves of complexes 1 (left) and 2 (right).



Fig. S2 TG–DSC curves of complexes 3 (left) and 4 (right)



Fig. S3 TG–DSC curves of complex 5



Fig. S4 The powder X-ray diffraction spectra of the simulated patterns and experimental for



complexes 1 (left) and 2 (right).

Fig. S5 The powder X-ray diffraction spectra of the simulated patterns and experimental for



complexes 3 (left) and 4 (right).

Fig. S6 The powder X-ray diffraction spectra of the simulated patterns and experimental for

complex 5.



Fig. S7 IR spectra of bta and complexes 1–5.



Fig. S8 Excitation spectra of complexes 1–5

Table S1 Selected bond lengths (Å) and angles (°) for complexes 1–3.

1		2		3	
O(1)-Er(1)	2.371(3)	O(1)-Er(1)	2.304(4)	O(7)–Er(2)	2.335(4)
O(2) - Er(1)	2.391(3)	O(2) - Er(1)	2.328(3)	O(8)– $Er(2)$	2.294(4)
O(3) - Er(1)	2.338(3)	O(3) - Er(1)	2.301(3)	O(9) - Er(2)	2.263(4)
O(4)-Er(1)	2.293(4)	O(4)-Er(1)	2.278(4)	O(10) - Er(2)	2.313(4)
O(5)– $Er(1)$	2.349(3)	O(5) - Er(1)	2.282(4)	O(11) - Er(2)	2.293(4)
O(6)-Er(1)	2.284(3)	O(6)-Er(1)	2.312(4)	O(12) - Er(2)	2.315(3)
O(7)– $Er(1)$	2.247(3)	N(1)-Er(1)	2.515(4)	N(3)– $Er(2)$	2.525(4)
O(8)– $Er(1)$	2.401(3)	N(2)-Er(1)	2.513(4)	N(4)– $Er(2)$	2.556(4)
O(6) - Er(1) - O(4)	136.56(12)	O(3)-Er(1)- $O(2)$	78.90(13)	O(10)-Er(2)- $O(7)$	116.45(14)
O(7)-Er(1)- $O(3)$	75.56(13)	O(5)-Er(1)- $O(2)$	148.33(13)	O(8) - Er(2) - O(7)	72.00(13)
O(6) - Er(1) - O(5)	72.73(12)	O(4)-Er(1)-N(1)	142.56(13)	O(9)-Er(2)-N(3)	138.28(13)
O(6)-Er(1)-O(1)	73.09(12)	O(6) - Er(1) - N(2)	71.05(14)	O(11)-Er(2)-N(4)	70.13(13)

4		5	
O(1)-Er(1)	2.295(5)	O(1)-Er(1)	2.300(5)
O(2)-Er(1)	2.320(6)	O(2)-Er(1)	2.304(5)
O(3)-Er(1)	2.282(5)	O(3)-Er(1)	2.295(5)
O(4)-Er(1)	2.272(6)	O(4)-Er(1)	2.266(5)
O(5)-Er(1)	2.296(5)	O(5)– $Er(1)$	2.303(5)
O(6)-Er(1)	2.301(6)	O(6)-Er(1)	2.304(5)
N(1)-Er(1)	2.526(7)	N(1)-Er(1)	2.507(5)
N(2)-Er(1)	2.531(8)	N(2)-Er(1)	2.508(5)
O(5)-Er(1)-O(1)	116.0(2)	O(4) - Er(1) - O(3)	73.43(18)
O(5) - Er(1) - O(4)	86.3(2)	O(3) - Er(1) - O(1)	121.88(17)
O(5) - Er(1) - N(1)	149.1(2)	O(4) - Er(1) - N(1)	145.90(18)
O(3)-Er(1)-N(2)	71.7(2)	O(4) - Er(1) - N(2)	149.52(18)

 Table S2 Selected bond lengths (Å) and angles (°) for complexes 4 and 5.