

Sugar thiacycrown-ether appended calix[4]arene as a selective chemosensor for Fe^{2+} and Fe^{3+} ions

Irene Ling^{*a}, Rauzah Hashim^a and Karem J. Sabah^{*a,b}

Supplementary Information

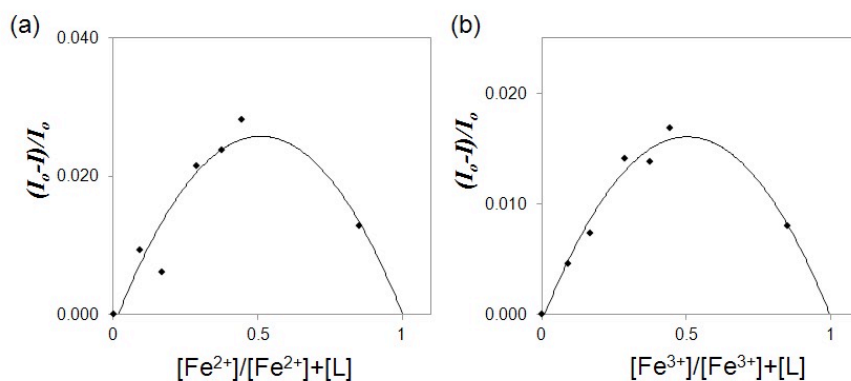


Fig. S1. Job's plot for (a) Fe^{2+} and (b) Fe^{3+} complex in MeCN: CHCl_3 . Ligand, **9**, concentration is 1.0×10^{-5} M.

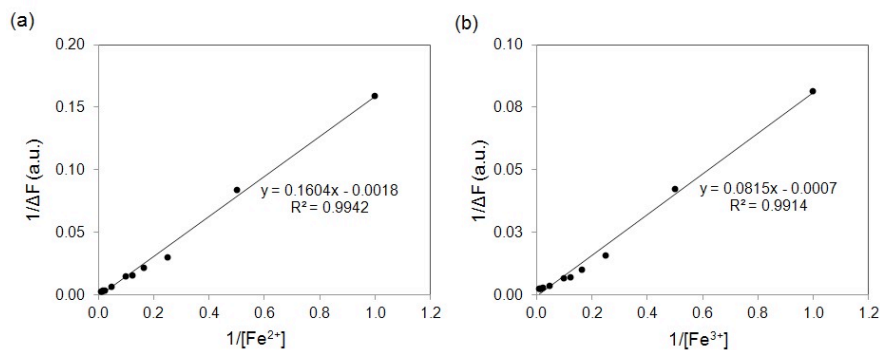


Fig. S2. The binding constants K_a for **9**-metal complexes were estimated to be 900 M^{-1} for Fe^{2+} and 875 M^{-1} for Fe^{3+} , obtained from the slope and intercept in the plot of $1/\Delta F$ against (a) $1/[\text{Fe}^{2+}]$ or (b) $1/[\text{Fe}^{3+}]$.

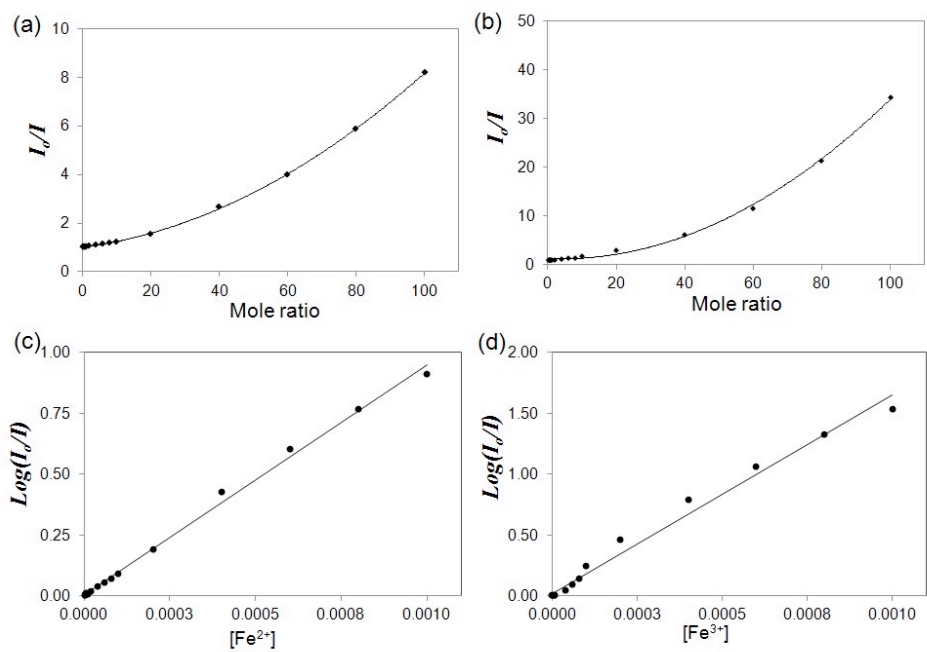


Fig S3. I_0/I versus (a) $[\text{Fe}^{2+}]$ and (b) $[\text{Fe}^{3+}]$ in MeCN/ CHCl_3 (1:1, v/v), excitation at 343 nm. Modified Stern-Volmer relationship between **9** and Fe^{2+} (c) or Fe^{3+} (d).

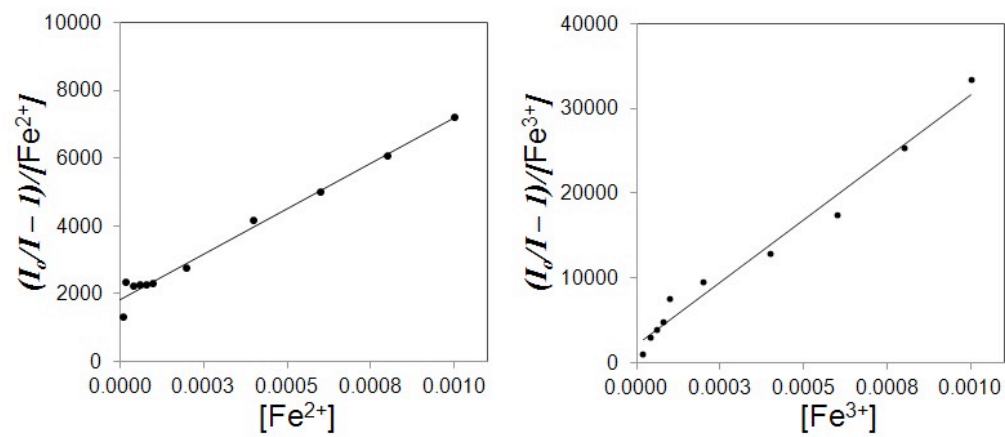


Fig S4. Plots of $(I_0/I - 1)/[\text{Fe}^{2+}]$ versus $[\text{Fe}^{2+}]$ and $(I_0/I - 1)/[\text{Fe}^{3+}]$ versus $[\text{Fe}^{3+}]$.