

Electronic supplementary information (ESI)

Reactive porous composites for chromium(VI) reduction applications based on Fe/carbon obtained from post-consumer PET and iron oxide.

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Fig. S1 (a) XDR pattern; (b) Mössbauer spectrum and (c) Raman spectrum of the hematite prepared by thermal treatment of iron(III) nitrate at 450 °C.

Fig. S2 Profile of temperature along with the evolution of gases of Fe₅₀₀/C_{PET}, Fe₆₀₀/C_{PET}, Fe₈₀₀/C_{PET} and Fe₁₀₀₀/C_{PET} under argon atmosphere.

Fig. S3 EDS mapping of (a) Fe₈₀₀/C_{PET} and (b) Fe₉₀₀/C_{PET}.

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Fig. S5 Pore distribution in the Fe₅₀₀/C_{PET}, Fe₆₀₀/C_{PET}, Fe₈₀₀/C_{PET} and Fe₁₀₀₀/C_{PET} composites.

Fig. S6 Magnetic separation of the Fe₈₀₀/C_{PET} composite from a suspension in water.

Table S1 Mössbauer hyperfine parameters of the composites.

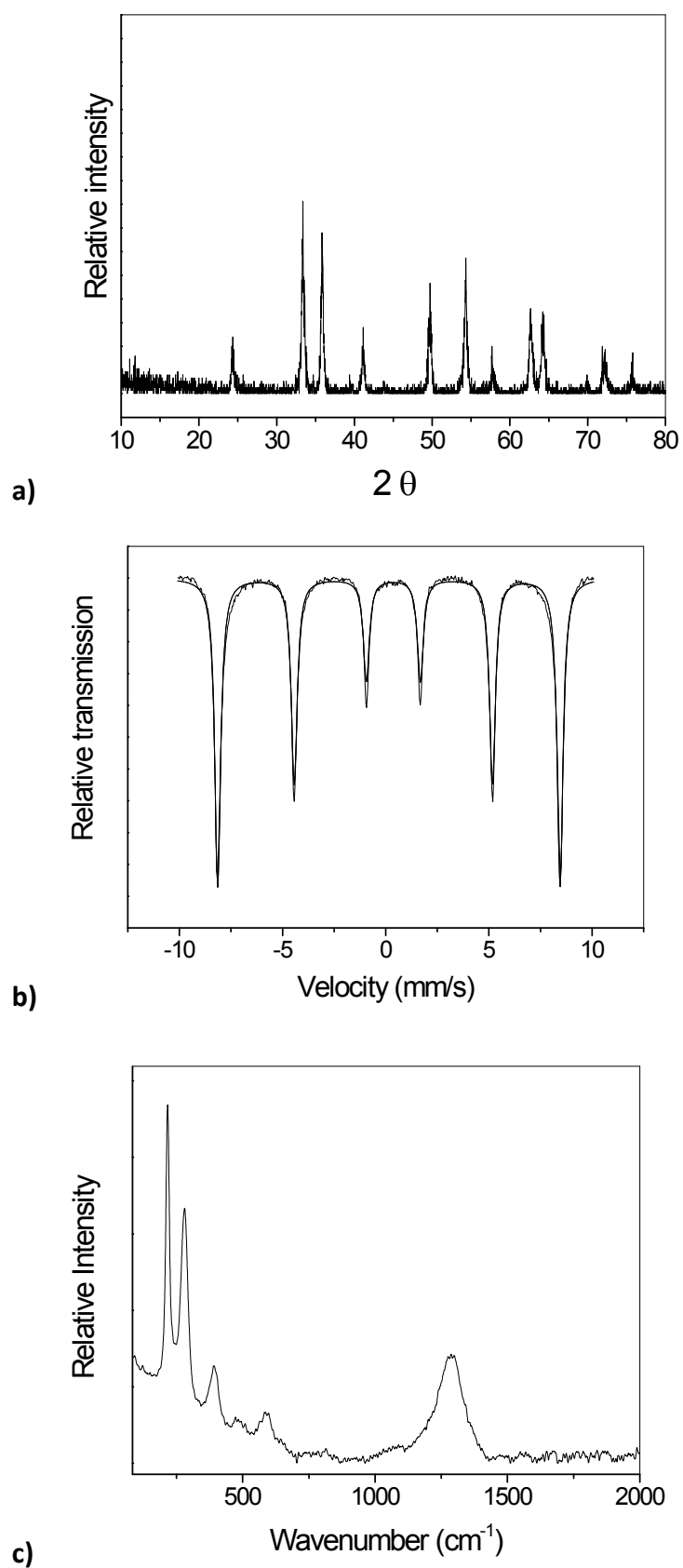


Fig. S1 (a) XDR pattern; (b) Mössbauer spectrum and (c) Raman spectrum of the hematite prepared by thermal treatment of iron(III) nitrate at 450 °C. All the signals and bands are only related to hematite.

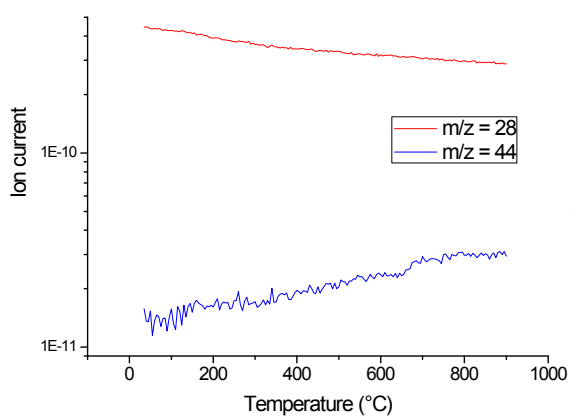
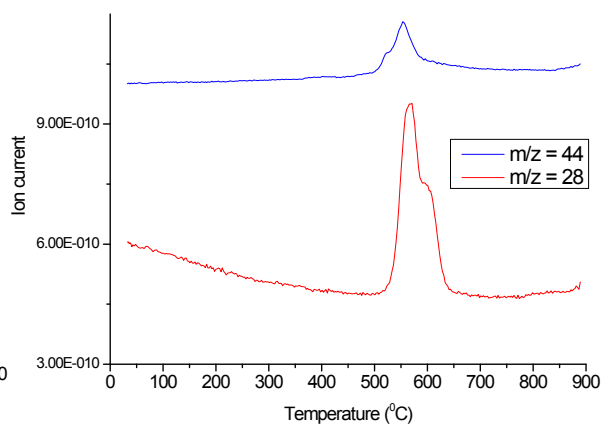
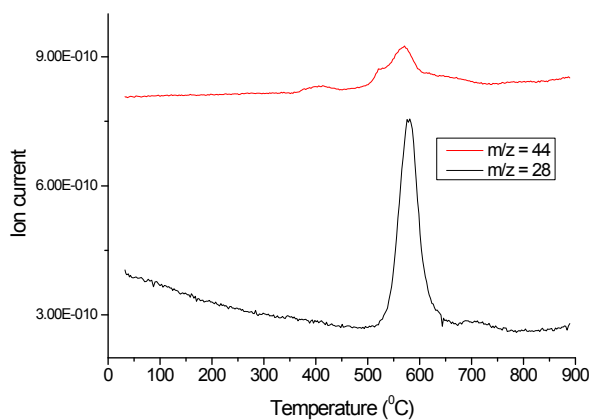
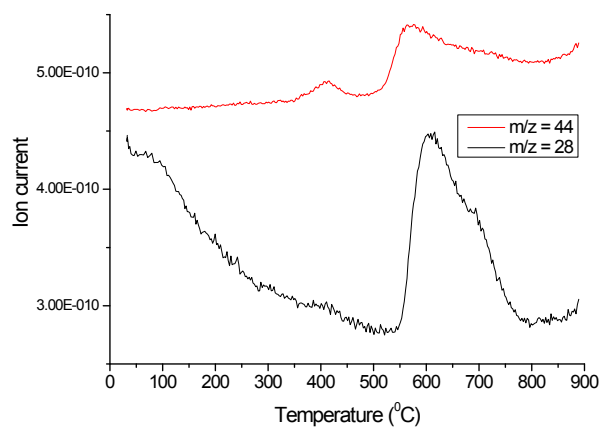
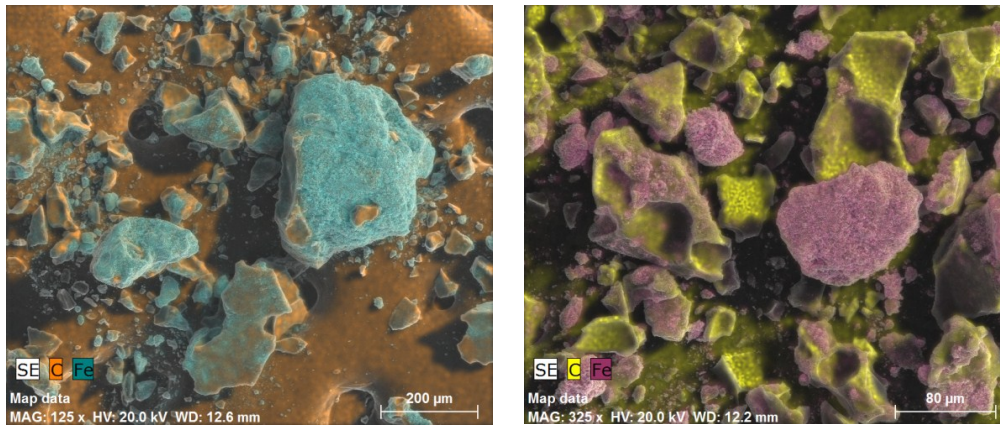
Fe₇₀₀/C_{PET}Fe₈₀₀/C_{PET}Fe₉₀₀/C_{PET}Fe₁₀₀₀/C_{PET}

Fig. S2 Profile of temperature along with the evolution of gases of Fe₅₀₀/C_{PET}, Fe₆₀₀/C_{PET}, Fe₈₀₀/C_{PET} and Fe₁₀₀₀/C_{PET} under argon atmosphere.



a)

b)

Fig. S3 EDS mapping of (a) Fe₈₀₀/C_{PET} and (b) Fe₉₀₀/C_{PET}.

EDS mapping suggests that the Fe is homogeneously distributed throughout the particles, as it may be seen by the blue and pink colors in Fig. S2 (a) and (b), respectively, which shows the mapping of the elemental Fe in the material. The distribution of the carbon in the material is indicated by the orange and yellow colors in Fig. S2 (a) and (b).

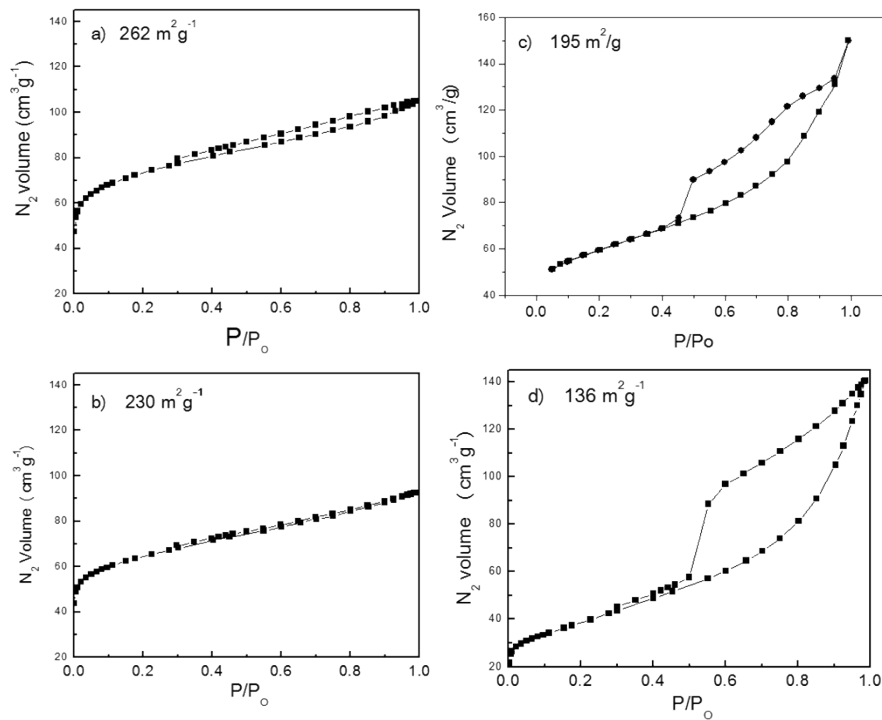


Fig. S4 Nitrogen adsorption and desorption isotherms obtained for (a) Fe₅₀₀/C_{PET}; (b) Fe₆₀₀/C_{PET}; (c) Fe₈₀₀/C_{PET}; (d) Fe₁₀₀₀/C_{PET} and the surface area values.

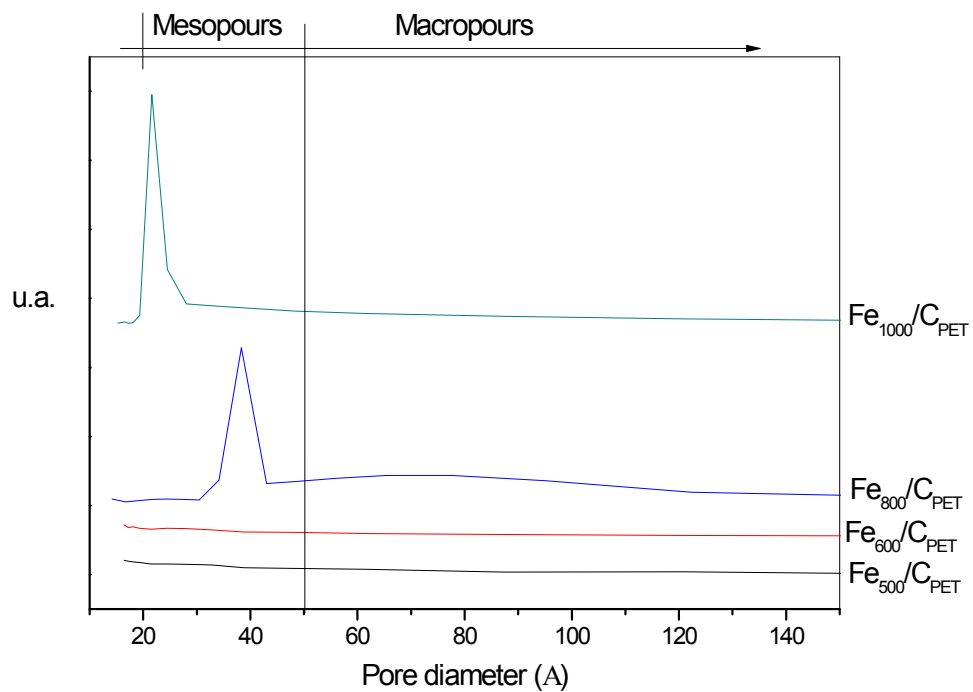


Fig. S5 Pore distribution in the Fe₅₀₀/C_{PET}, Fe₆₀₀/C_{PET}, Fe₈₀₀/C_{PET} and Fe₁₀₀₀/C_{PET} composites.

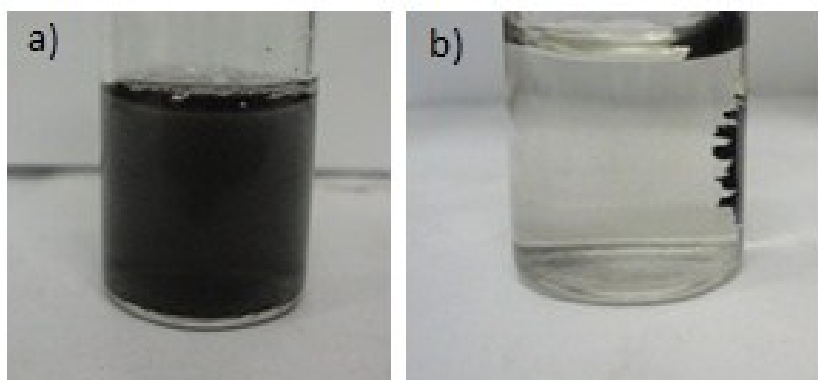


Fig. S6. Magnetic separation of the Fe₈₀₀/C_{PET} composite from a suspension in water.

Table S1 Mössbauer hyperfine parameters of the composites.

Material	Phase	$\delta \pm 0.05$ (mm/s)	$2\varepsilon//\Delta Q \pm 0.05$ (mm/s)	$B_{HF} \pm 0.5$ (Tesla)	Area ± 1 (%)
Fe ₅₀₀ /C _{PET}	mix Fe₃O₄,	0.65	0.04	46.0	28
	γ-Fe₃O₄	0.25	-0.06	49.0	29
	Fe³⁺	0.21	1.34	-	18
	Fe²⁺	0.97	1.08	-	09
	FeOOH	0.63	0.13	38.4	16
Fe ₆₀₀ /C _{PET}	mix Fe₃O₄,	0.53	0.04	45.6	27
	γ-Fe₃O₄	0.27	-0.06	49.2	26
	Fe³⁺	0.34	1.62	-	15
	Fe²⁺	0.97	1.40	-	15
	FeOOH	0.31	0.13	40.9	17
Fe ₇₀₀ /C _{PET}	α-Fe	0.0	0.0	33.0	23
	Fe₃C	0.17	0.02	20.8	69
	γ-Fe(C)	0.41	0.80	-	4
	γ-Fe	-0.20	-	-	4
Fe ₈₀₀ /C _{PET}	α-Fe	0,00	0,00	33,10	65
	Fe₃C	0,20	0,03	20,70	22
	γ-Fe	-0,09	-----	-----	13
Fe ₉₀₀ /C _{PET}	α-Fe	0.0	0.0	33.0	31
	Fe₃C	0.17	0.02	20.8	53
	γ-Fe(C)	0.39	0.58	-	6
	γ-Fe	-0.15	-	-	10
Fe ₁₀₀₀ /C _{PET}	α-Fe	0.0	0.0	33.0	49
	Fe₃C	0.15	0.02	21.0	38
	γ-Fe(C)	0.40	0.69	-	4
	γ-Fe	-0.13	-	-	9