

Supporting Information

A Stable 3D Porous Coordination Polymer as Multi-Chemosensor to Cr(IV) anion and Fe(III) cation and Its Selective Adsorption of Malachite Green Oxalate Dye

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(1) THE SUPPORTING FIGURES

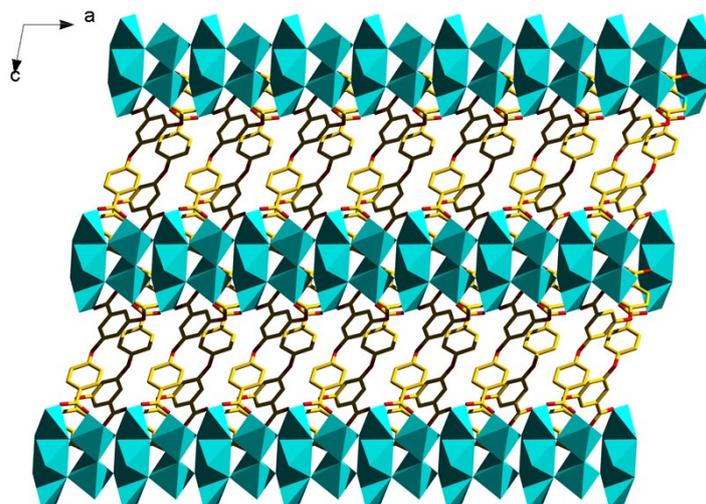


Figure S1. The 3D structure of **1** along the *b* axis.

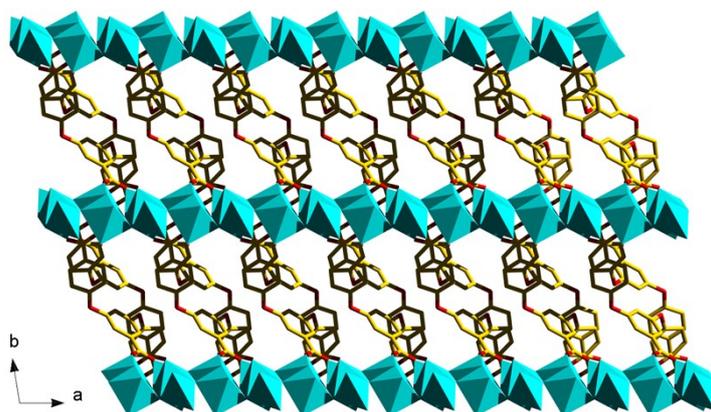


Figure S2. The 3D structure of **1** along the *c* axis.

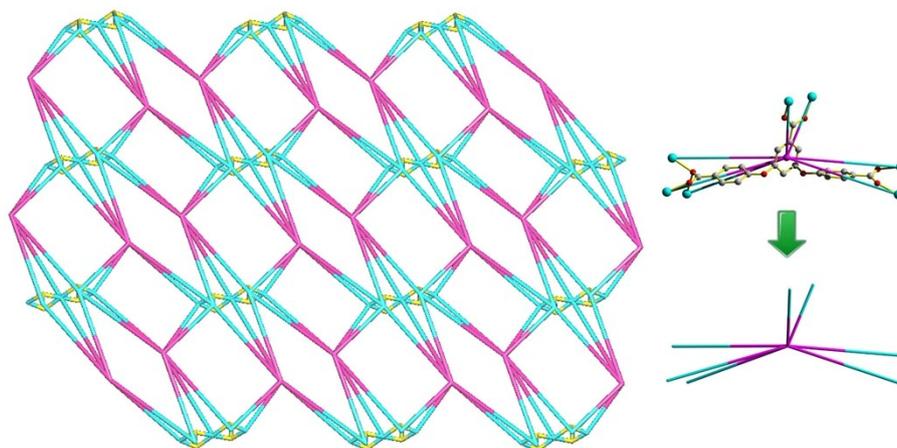


Figure S3. The topology of **1** view along *a* axis.

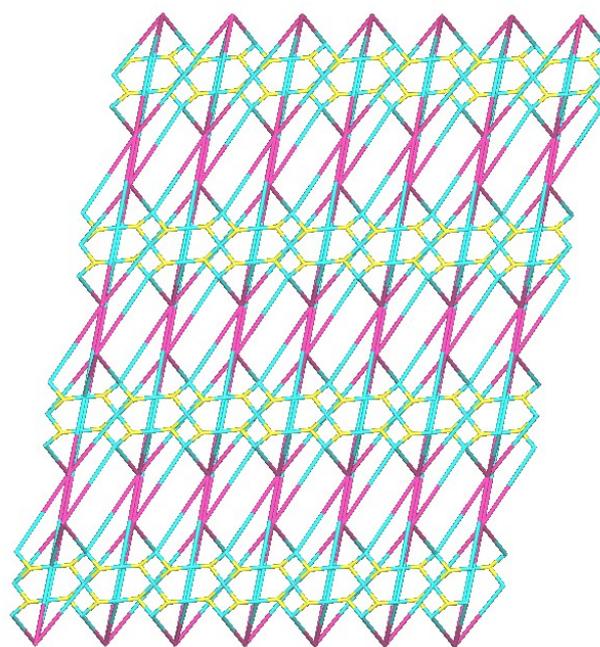


Figure S4. The topology of **1** along the *b* axis.

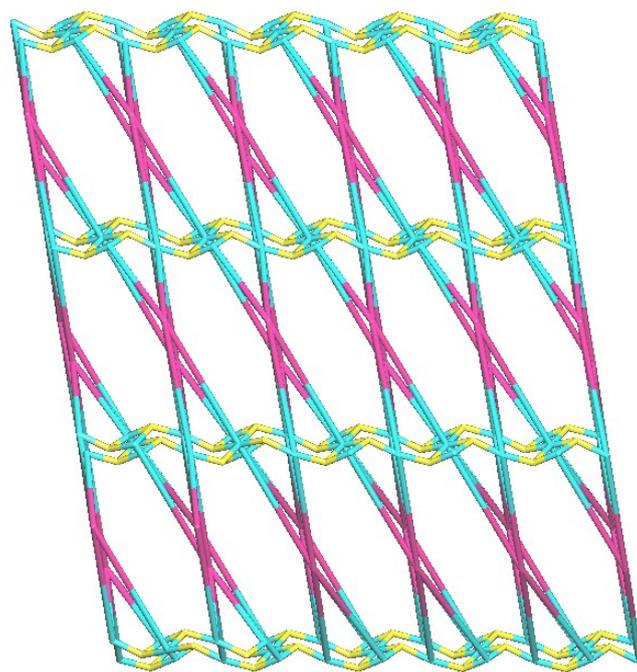


Figure S5. The topology of **1** along the *c* axis.

(2) IR RESULTS

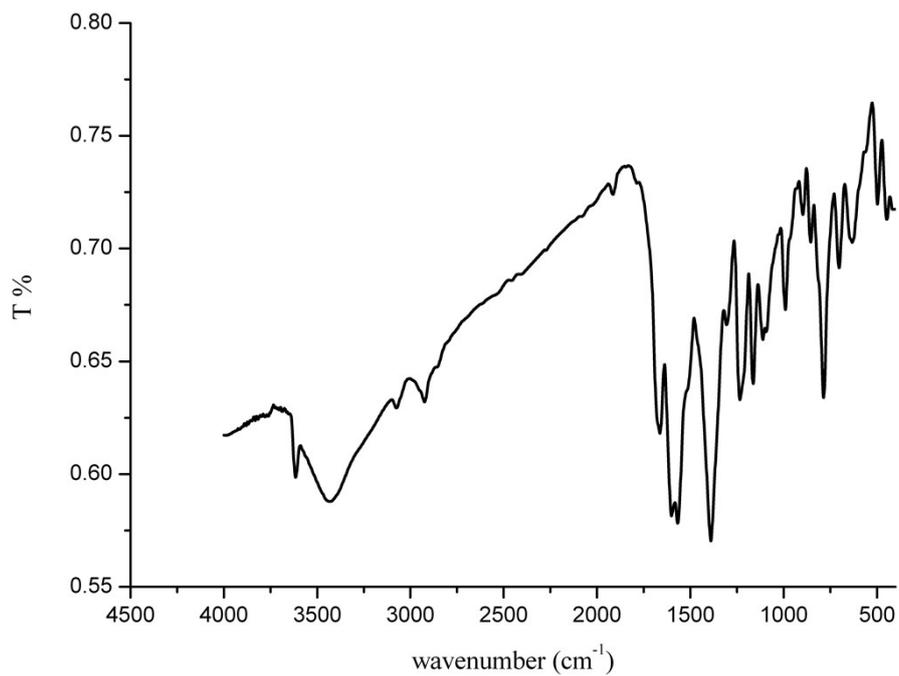


Figure S6. FT-IR spectroscopy of the complex **1-DMF**.

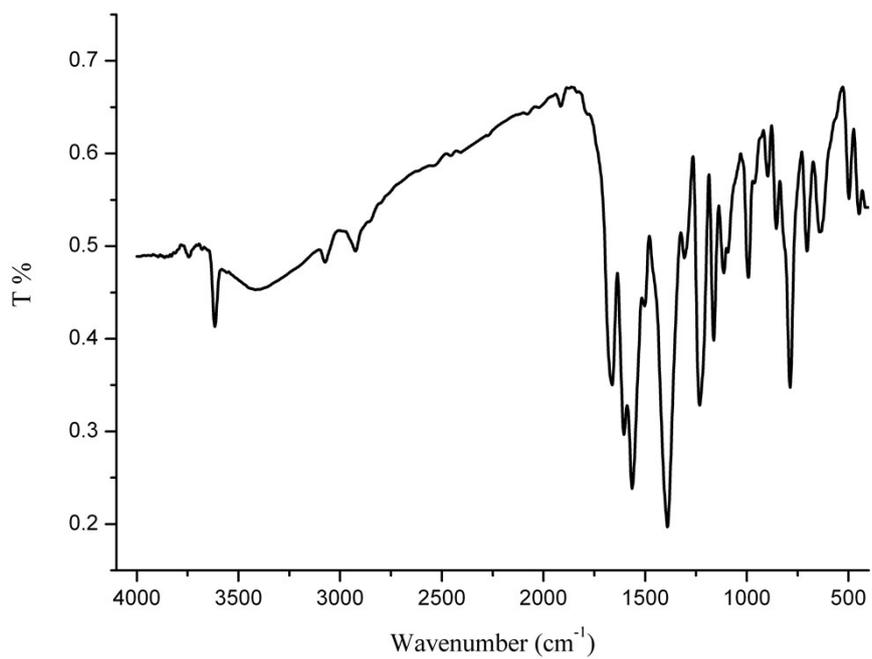


Figure S7. FT-IR spectroscopy of the complex **1-DMA**.

(3) PXRD RESULTS

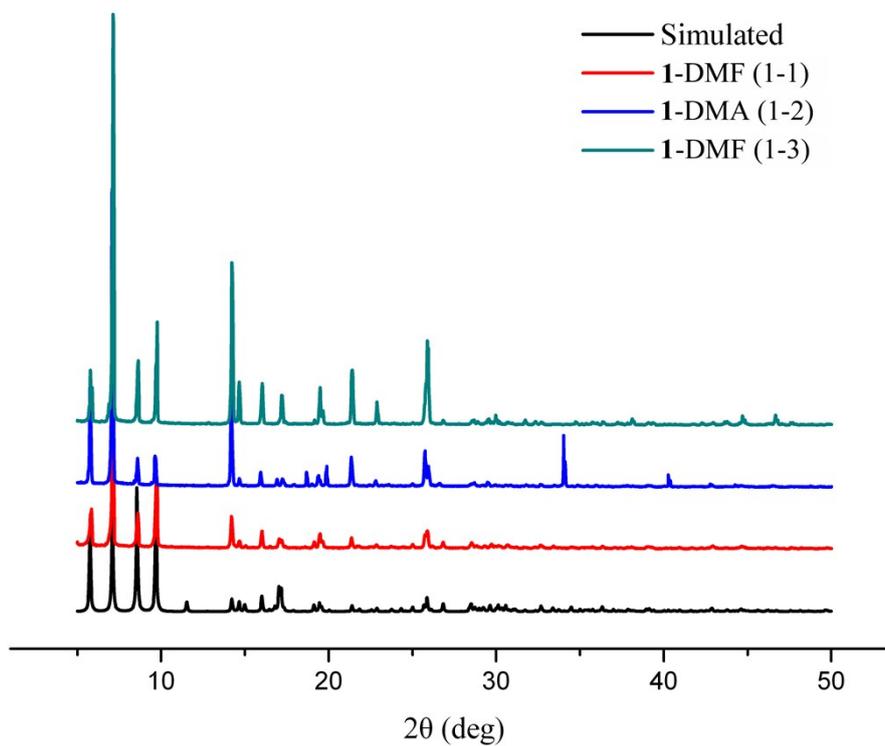


Figure S8. Powder X-ray diffraction patterns of complex **1-DMF** and **1-DMA**.

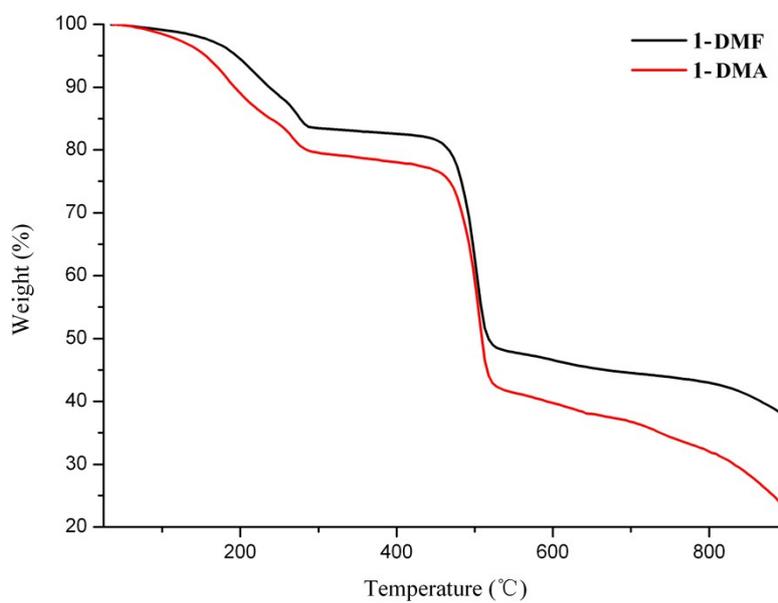


Figure S9. TGA curves of **1-DMF** and **1-DMA**.

(4) LUMINESCENCE SENSING FIGURES

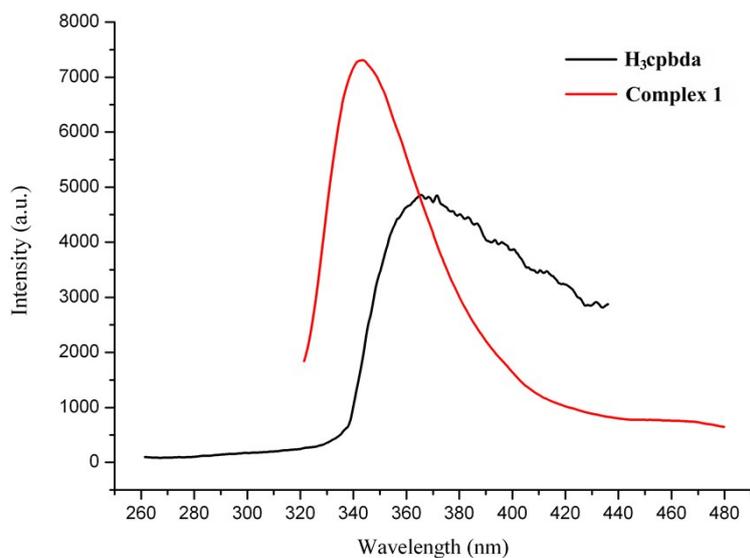


Figure S10. The solid-state fluorescent emission of **1**.

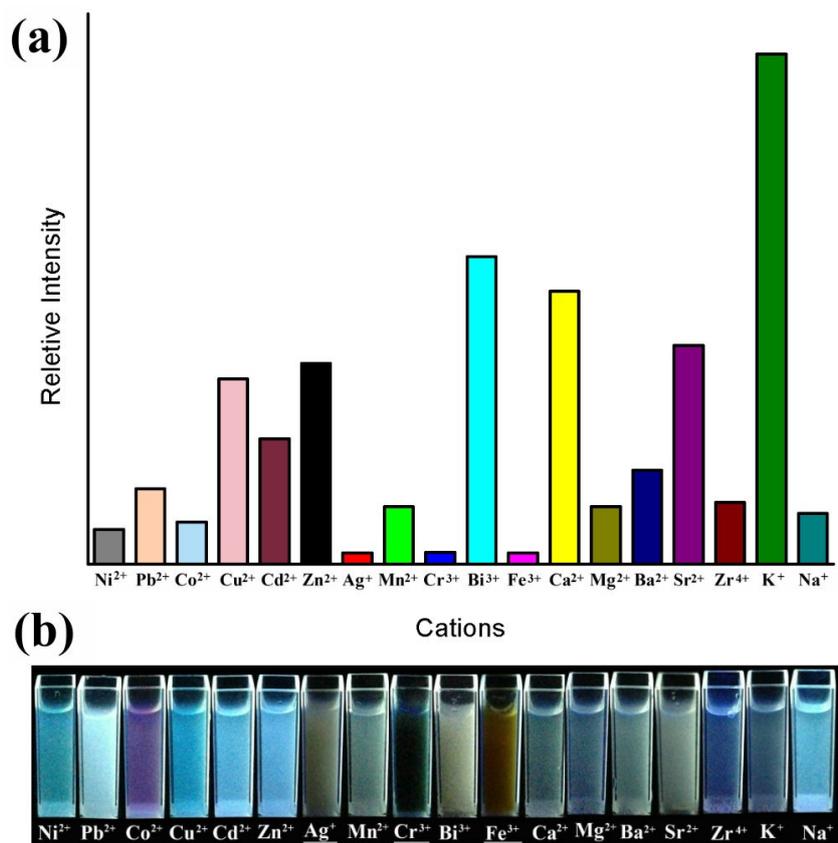


Figure S11. (a) Comparison of the relative luminescence intensities of various metal cations based on **1-DMA**. (b) The visual change on the addition of various metal cations.

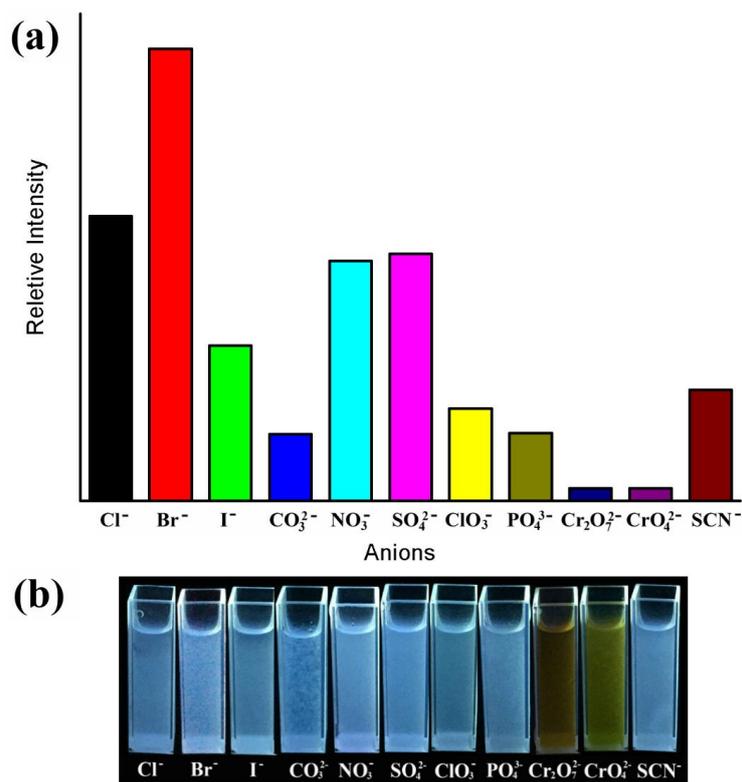


Figure S12. (a) Comparison of the relative luminescence intensities of various anions base on **1**-DMA. (b) The visual change on the addition of various anions.

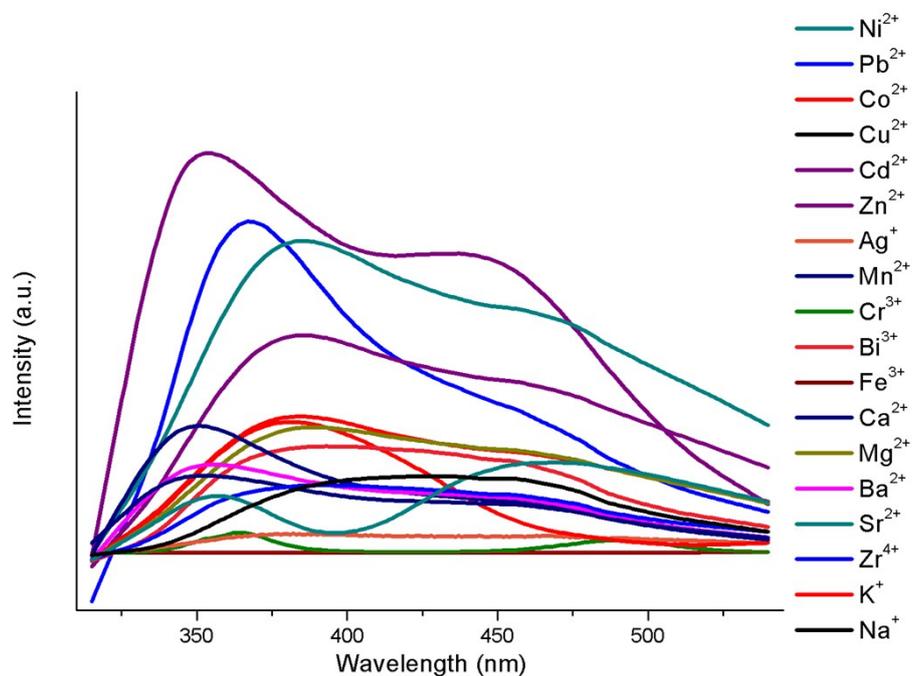


Figure S13. The luminescence emission spectras of complex **1**-DMF suspensions with different metal cations.

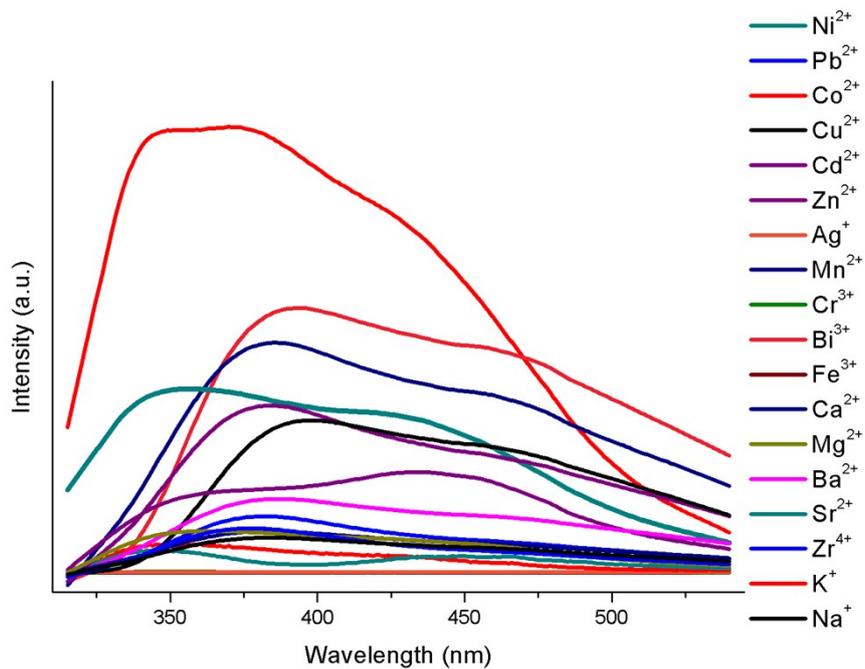


Figure S14. The luminescence emission spectras of complex **1-DMA** suspensions with different metal cations.

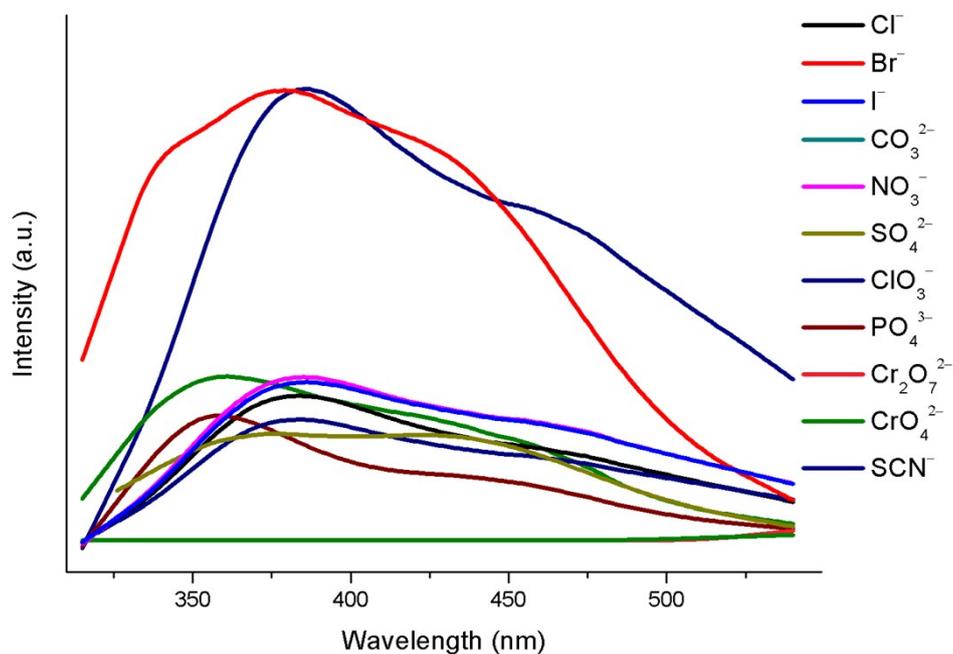


Figure S15. The luminescence emission spectras of complex **1-DMF** suspensions with different anions.

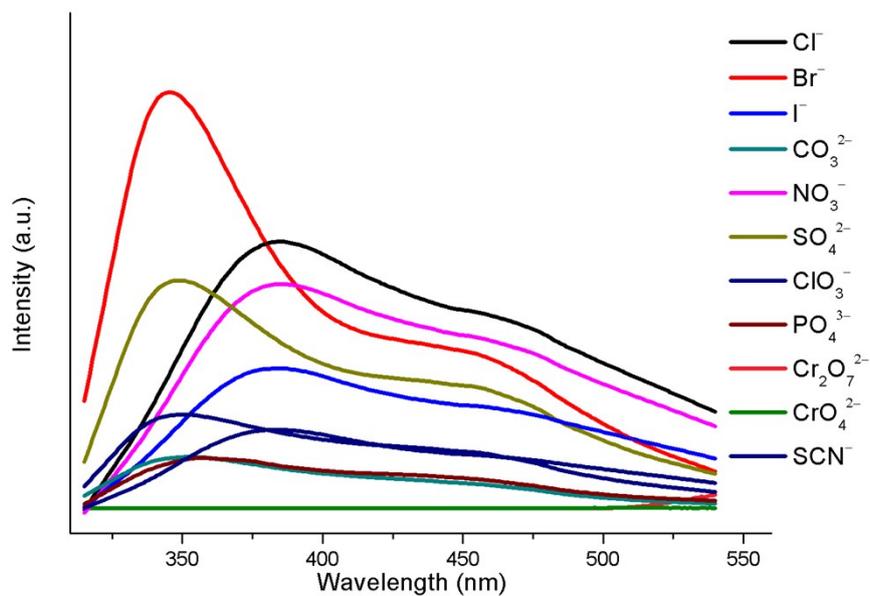


Figure S16. The luminescence emission spectra of complex **1-DMA** suspensions with different anions.

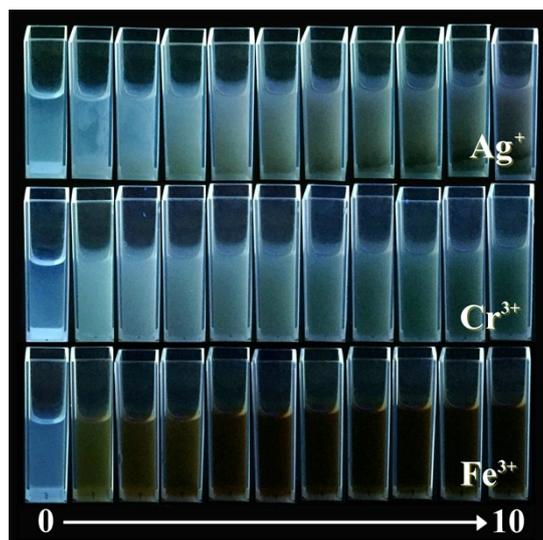


Figure S17. The luminescence intensities of **1-DMA** suspensions with cations concentrations varying from 0 to 1429 ppm.

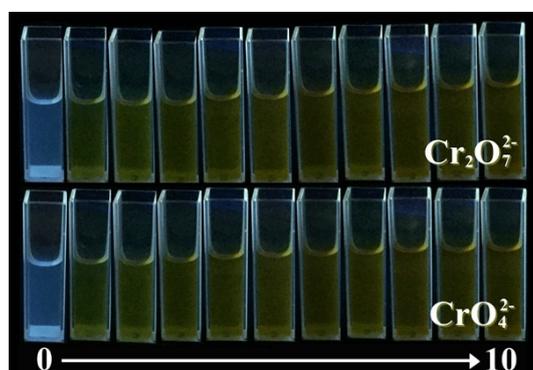
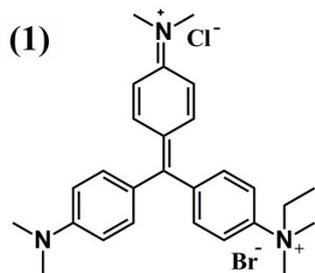
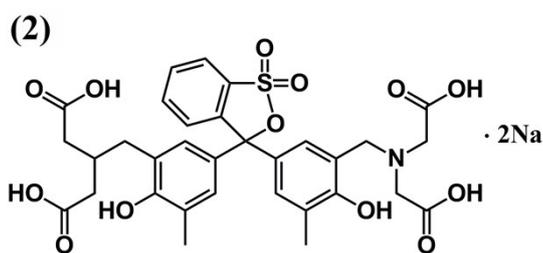


Figure S18. The luminescence intensities of **1-DMA** suspensions with anions concentrations varying from 0 to 1429 ppm.

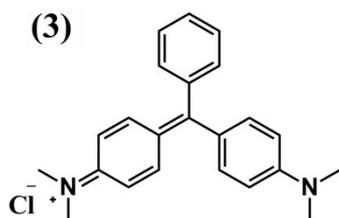
(5) MOLECULAR FORMULA OF DYES



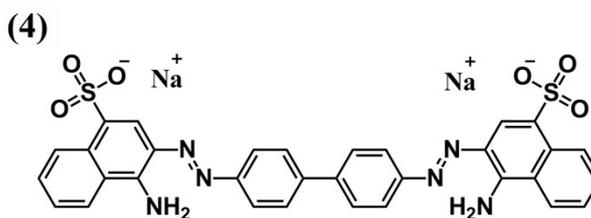
Methyl Green (MG)



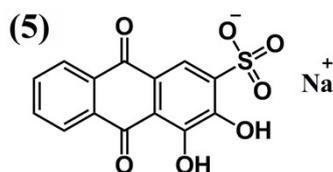
Xylenol Orange (XO)



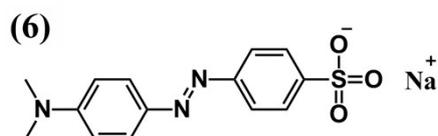
Malachite Green Oxalate (MGO)



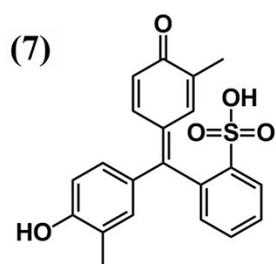
Congo Red (CR)



Alizarin Red (AR)



Methyl Orange (MO)



Cresol Red (CR)

Figure S19. The structures of seven kinds of dyes.

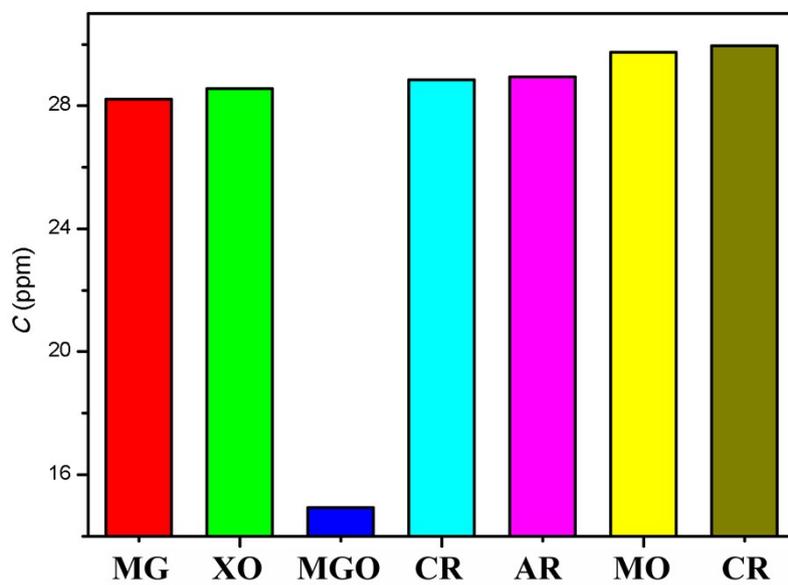


Figure S20. The concentrations of dyes in 2 hours after the additions of complexes 1-DMA.

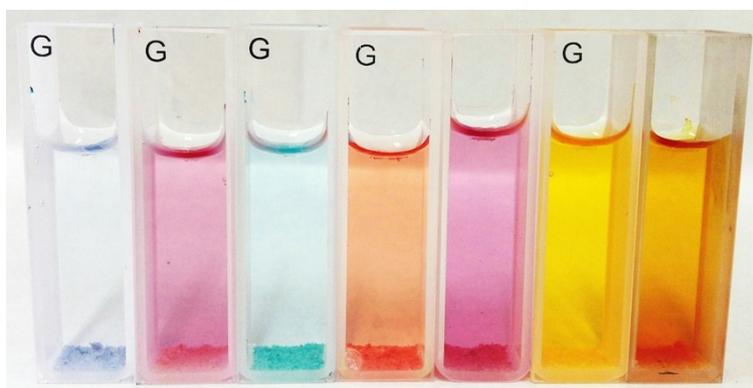


Figure S21. The naked-eye photos of dyes in 2 hours after the additions of complexes 1-DMA.

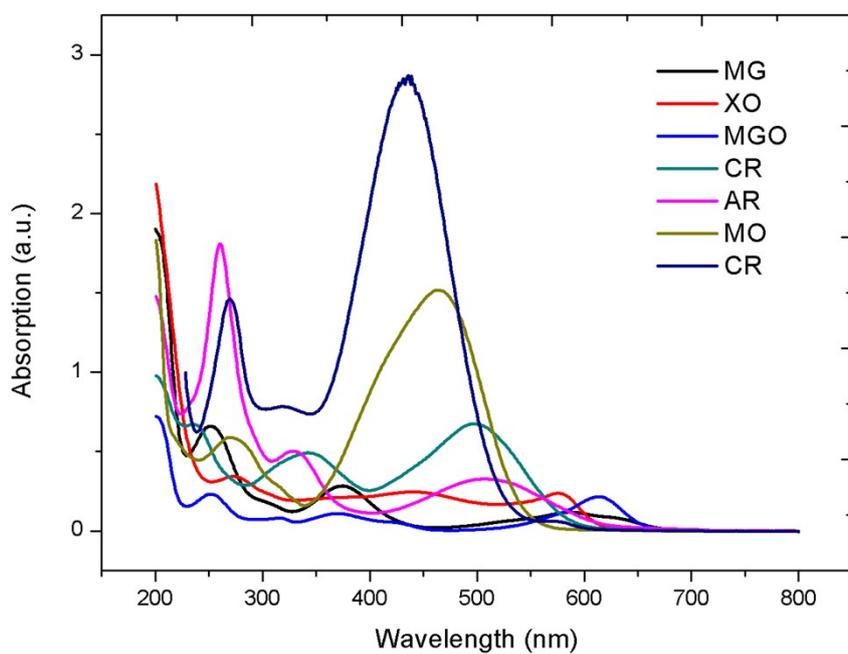


Figure S22. the UV spectra of seven different kinds of dye.

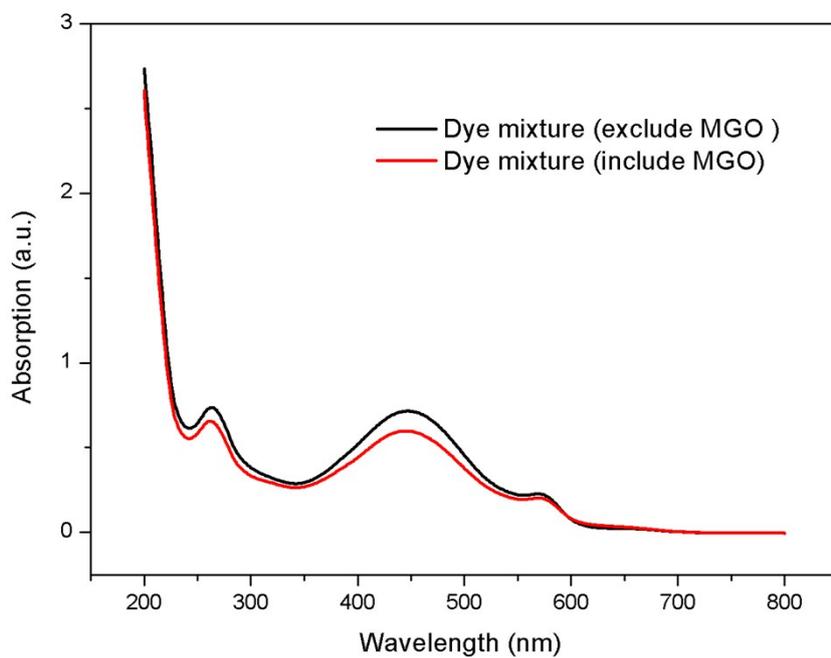


Figure S23. The UV spectrum of six dyes mixture (exclude MGO) and seven dyes mixture (include MGO)

(6) CRYSTALLOGRAPHIC DATAS TABLES

Table S1. Crystallographic data and details of diffraction experiments for complexes **1**.

Complex 1			
Formula	C42 H26 O20 Zn5	V (Å ³)	1217.3
M_r	1177.48	Z	1
Crystal system	triclinic	ρ (g cm ⁻³)	1.606
Space group	$P\bar{1}$	μ (mm ⁻¹)	2.501
a (Å)	6.327	T (K)	296(2)
b (Å)	12.671	Goof	1.001
c (Å)	15.684	R [$I > 2\sigma(I)$]	$R_1 = 0.0432$
α (°)	95.69		$wR_2 = 0.1390$
β (°)	99.77		$R_1 = 0.0530$
γ (°)	98.01	R (all data)	$wR_2 = 0.1454$

Table S2. Selected Bond Lengths (Å) and Angles (deg) for **1**.

Complex 1			
Zn(1)-O(2W)	2.007(3)	Zn(1)-O(8)#4	2.102(4)
Zn(1)-O(1)	2.064(4)	Zn(1)-O(4)#2	2.130(3)
Zn(1)-O(1W)	2.012(3)	Zn(1)-O(4)#7	2.435(3)
Zn(2)-O(2)	1.924(4)	Zn(2)-O(1W)	1.981(3)
Zn(2)-O(7)#5	1.924(4)	Zn(2)-O(2W)#8	1.978(3)
Zn(3)-O(1W)	2.080(3)	Zn(3)-O(2W)#8	2.200(3)
Zn(3)-O(5)#2	2.063(3)		
O(1W)-Zn(1)-O(4)#7	78.40(12)	O(4)#2-Zn(1)-O(4)#7	81.89(13)
O(1W)-Zn(1)-O(8)#4	85.14(14)	O(2W)-Zn(1)-O(4)#2	86.85(13)
O(8)#4-Zn(1)-O(4)#7	88.69(14)	O(2W)-Zn(1)-O(4)#7	90.32(12)
O(2W)-Zn(1)-O(8)#4	92.13(14)	O(1)-Zn(1)-O(8)#4	92.97(16)
O(1W)-Zn(1)-O(4)#2	93.99(13)	O(1W)-Zn(1)-O(1)	94.72(14)
O(2W)-Zn(1)-O(1)	96.64(15)	O(1)-Zn(1)-O(4)#2	96.51(14)
O(2W)-Zn(1)-O(1W)	168.44(14)	O(8)#4-Zn(1)-O(4)#2	170.51(15)
O(1)-Zn(1)-O(4)#7	172.77(12)	O(2W)#8-Zn(2)-O(1W)	91.45(13)
O(2)-Zn(2)-O(7)#5	113.48(17)	O(2)-Zn(2)-O(2W)#8	118.09(17)
O(7)#5-Zn(2)-O(2W)#8	111.92(14)	O(2)-Zn(2)-O(1W)	109.20(15)
O(7)#5-Zn(2)-O(1W)	110.39(15)	O(5)#9-Zn(3)-O(1W)	84.85(13)
O(1W)-Zn(3)-O(2W)#8	82.89(12)	O(5)#9-Zn(3)-O(2W)#8	93.94(13)
O(5)#2-Zn(3)-O(2W)#8	86.06(13)	O(5)#2-Zn(3)-O(1W)	95.15(13)
O(1W)-Zn(3)-O(2W)#11	97.11(12)	O(1W)#10-Zn(3)-O(1W)	180.0
O(5)#9-Zn(3)-O(5)#2	180.00(19)		

Symmetry transformations used to generate equivalent atoms: #1 $x-1, y+1, z$; #2 $-x+2, -y+1, -z+1$; #3 $x, y+1, z$; #4 $-x+2, -y, -z$; #5 $-x+3, -y, -z$; #6 $x-1, y, z$; #7 $x, y-1, z$; #8 $x+1, y, z$; #9 $x+1, y-1, z$; #10 $-x+3, -y, -z+1$; #11 $-x+2, -y, -z+1$.

REFERENCES

[1] Sheldrick G M. *SHELXL-97, Program for Crystal Structure Determination*, University of Göttingen, Germany, 1997.

[2] Sheldrick G M. *SHELXL-97, Program for Crystal Structure Refinement*, University of Göttingen, Germany, 1997.